



Laporan Hasil Penelitian

“Pengaruh Derajat Keasaman (pH) Presipitasi dan Suhu Kalsinasi terhadap Karakterisasi Kalsium Fosfat”

DAFTAR PUSTAKA

- Abd. Rahim, T. *et al.* (2020) ‘Eggshell derived calcium phosphate and its conversion to dense bodies’, *Journal of Advanced Research in Fluid Mechanics and Thermal Sciences*, 65(2), pp. 334–341.
- Alshaaer, M. *et al.* (2022) ‘Effects of magnetite incorporation in a chemically bonded phosphate ceramic’, *Journal of Physics and Chemistry of Solids*, 162(April 2021), p. 110531. Available at: <https://doi.org/10.1016/j.jpcs.2021.110531>.
- Amin, M. and Kurniasih, A. (2016) ‘Pengaruh ukuran dan waktu kalsinasi batu kapur terhadap tingkat perolehan kadar CaO’, *Seminar Nasional Sains Matematika Informatika dan Aplikasinya IV*, 4, pp. 74–82.
- Amjad, Z. (1998) *Calcium Phosphates in Biological and Industrial Systems*. London: Kluwer Academic Publisher.
- Augustijns, P. and Brewster, E.M. (2014) *Solvent Systems and Their Selection in Pharmaceutics and Biopharmaceutics*. 1st edn. Beerse: AAPS Press. Available at: <https://doi.org/10.1007/978-0-387-69154-1>.
- BPS (2022) *Kebutuhan Impor Kalsium Fosfat di Indonesia*. Available at: <https://www.bps.go.id/exim/> (Accessed: 3 July 2023).
- Carella, F. *et al.* (2021) ‘The Use of Calcium Phosphates in Cosmetics, State of the Art and Future Perspectives’, *Materials*, 14(6398), pp. 1–37. Available at: <https://doi.org/10.1201/9780429486579-5>.
- Dermawan, S.K. *et al.* (2022) ‘Effect of the Calcination Temperature on the Properties of Hydroxyapatite from Black Tilapia Fish Bone’, *Journal of Physics: Conference Series*, 2169(1). Available at: <https://doi.org/10.1088/1742-6596/2169/1/012034>.
- Dorozhkin, S. V (2019) ‘Dental Applications of Calcium Orthophosphates (CaPO₄)’, *Journal of Dentistry Research*, 1(1), pp. 024–054.
- Dorozhkina, E.I. and Dorozhkin, S. V. (2002) ‘Mechanism of the solid-state transformation of a calcium-deficient hydroxyapatite (CDHA) into biphasic calcium phosphate (BCP) at elevated temperatures’, *Chemistry of Materials*, 14(10), pp. 4267–4272. Available at: <https://doi.org/10.1021/cm0203060>.
- Drevet, R., Fauré, J. and Benhayoune, H. (2023) ‘Bioactive Calcium Phosphate



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-
- Coatings for Bone Implant Applications: A Review’, *Coatings*, 13(6), pp. 1–27. Available at: <https://doi.org/10.3390/coatings13061091>.
- Edahwati, L. *et al.* (2022) ‘Synthesis of Calcium Phosphate from Boiler Egg Shells as Raw Material for Hydroxyapatite’, 2022, pp. 108–113.
- Estrada, F. *et al.* (2007) ‘Pengambilan Minyak Kemiri dengan Cara Pengepresan dan Dilanjutkan Ekstrasi Cake Oil’, *Jurnal Widya Teknik*, 6(2), pp. 121–130.
- Le Gars Santoni, B. *et al.* (2021) ‘Chemically pure β -tricalcium phosphate powders: Evidence of two crystal structures’, *Journal of the European Ceramic Society*, 41(2), pp. 1683–1694. Available at: <https://doi.org/10.1016/j.jeurceramsoc.2020.09.055>.
- Gras, P. *et al.* (2013) ‘Crystallisation of a highly metastable hydrated calcium pyrophosphate phase’, *CrystEngComm*, 15(12), pp. 2294–2300. Available at: <https://doi.org/10.1039/c2ce26499d>.
- Griesiute, D. *et al.* (2021) ‘Synthesis of β -ca₂p₂o₇ as an adsorbent for the removal of heavy metals from water’, *Sustainability (Switzerland)*, 13(14), pp. 1–9. Available at: <https://doi.org/10.3390/su13147859>.
- Guo, X. *et al.* (2013) ‘Effect of calcining temperature on particle size of hydroxyapatite synthesized by solid-state reaction at room temperature’, *ADVANCED POWDER TECHNOLOGY*, 1(1), pp. 1–5. Available at: <https://doi.org/10.1016/j.apt.2013.03.002>.
- Insiyah and Cahyaningrum, S.E. (2019) ‘Sintesis Dan Karakterisasi Hidroksiapatit Dari Batu Kapur Dengan Metode Pengendapan Basah’, *UNESA Journal of Chemistry*, 8(3), pp. 114–110.
- Ishikawa, K., Garskaite, E. and Kareiva, A. (2020) ‘Sol–gel synthesis of calcium phosphate-based biomaterials—A review of environmentally benign, simple, and effective synthesis routes’, *Journal of Sol-Gel Science and Technology*, 94(3), pp. 551–572. Available at: <https://doi.org/10.1007/s10971-020-05245-8>.
- Islamillennio, A. (2023) ‘PENGARUH SUHU DAN WAKTU KALSINASI TERHADAP KEMURNIAN HIDROKSIAPATIT BERBASIS TULANG AYAM DENGAN METODE PRESIPITASI Mochamad Arif Irfa’i’, *Jurnal*
-



Laporan Hasil Penelitian

“Pengaruh Derajat Keasaman (pH) Presipitasi dan Suhu Kalsinasi terhadap Karakterisasi Kalsium Fosfat”

Teknik Mesin, 11(1), pp. 19–24.

Jadhav, V.R. et al. (2019) ‘Mathematical Treatment to Understanding the Concentration Terms’, *International Journal of Research & Review*, 6(January), pp. 172–175. Available at: www.ijrrjournal.com.

Jauhari and Maming (2014) ‘DETERMINATION OF THE CORAL AGE IN SPERMONDE ARCHIPELAGO MEASUREMENT ACTIVITY USING LSC (Liquid Scintillation Counting) METHOD’, *Marina Chimica Acta*, 15(1), pp. 15–20.

Kalbarczyk, M. et al. (2022) ‘Synthesis and Characterization of Calcium Phosphate Materials Derived from Eggshells from Different Poultry with and without the Eggshell Membrane’, *Materials*, 15(3), pp. 1–12. Available at: <https://doi.org/10.3390/ma15030934>.

Karalkeviciene, R. et al. (2023) ‘Solvothermal Synthesis of Calcium Hydroxyapatite via Hydrolysis of Alpha-Tricalcium Phosphate in the Presence of Different Organic Additives’, *Crystals*, 13(2), pp. 1–11. Available at: <https://doi.org/10.3390/crust13020265>.

Khaira, K. (2011) ‘Pengaruh Temperatur dan Waktu Kalsinasi Batu Kapur Terhadap Karakteristik Precipitated Calcium Carbonate (PCC)’, *Jurnal Sainstek*, III(1), pp. 33–43.

Lee, Y.H. et al. (2021) ‘RSC Advances PAPER Effects of pH and metal composition on selective extraction of calcium from steel slag for Ca (OH)₂’, *RSC Advances*, 11(1), pp. 8306–8313. Available at: <https://doi.org/10.1039/D0RA08497B>.

Lewis, A.E. et al. (2015) *Industrial crystallization: Fundamentals and applications*, *Industrial Crystallization: Fundamentals and Applications*. Available at: <https://doi.org/10.1017/CBO9781107280427>.

Meyer, F. et al. (2018) ‘Overview of Calcium Phosphates used in Biomimetic Oral Care’, *The Open Dentistry Journal*, 12(1), pp. 406–423. Available at: <https://doi.org/10.2174/1874210601812010406>.

Muljani, S., Sumada, K. and Erliyanti, N.K. (2019) ‘Synthesis and Characteristics of Calcium Phosphate from White Mussel Shell’, 2019, pp. 551–557. Available



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“Pengaruh Derajat Keasaman (pH) Presipitasi dan Suhu Kalsinasi terhadap Karakterisasi Kalsium Fosfat”

at: <https://doi.org/10.11594/nstp.2019.0473.551>.

Neves, J.G. *et al.* (2021) ‘Effect of pH level and calcination on the production of calcium phosphates by acidic route of wet precipitation’, *Ceramica*, 67(382), pp. 236–243. Available at: <https://doi.org/10.1590/0366-69132021673822965>.

Nguyen-Trong, D., Pham-Huu, K. and Nguyen-Tri, P. (2019) ‘Simulation on the Factors Affecting the Crystallization Process of FeNi Alloy by Molecular Dynamics’, *ACS Omega*, 4(11), pp. 14605–14612. Available at: <https://doi.org/10.1021/acsomega.9b02050>.

Nugraha, A.W., Nurmalis and Martoprawiro, M.A. (2013) ‘Pengembangan Metode Penentuan Jenis Pelarut Senyawa-Senyawa Organik Berdasarkan Kajian Termodinamika Kimia Melalui Pendekatan Pemodelan Molekul Dan Eksperimen Di Laboratorium’, *Jurnal Saintika*, pp. 48–57.

Nurma Wahyusi, K., Karunia, N. and Satrya, M. (2021) ‘Precipitation Method in Calcium Phosphat Synthesis from Blood Clamshells (*Anadara Granosa*)’, *Journal of Physics: Conference Series*, 1899(1), pp. 1–6. Available at: <https://doi.org/10.1088/1742-6596/1899/1/012057>.

Pangestu, T.O. *et al.* (2021) ‘Sintesis dan Karakterisasi Kalsium Fosfat dari Cangkang Bekicot dengan Metode Presipitasi’, *CHEESA: Chemical Engineering Research Articles*, 4(2), p. 82. Available at: <https://doi.org/10.25273/cheesa.v4i2.8931.82-90>.

Permana, A.J. *et al.* (2019) ‘Facile Sol-Gel Synthesis of Calcium Phosphates: Influence of Ca/P Ratio and Calcination Temperature’, *IOP Conference Series: Earth and Environmental Science*, 217(1), pp. 1–4. Available at: <https://doi.org/10.1088/1755-1315/217/1/012001>.

Prayudo, A. *et al.* (2015) ‘Koefisien transfer massa kurkumin dari temulawak’, *Andalas University Press*, 14(1), pp. 26–31.

Prillaman, J.T., Miyake, N. and Davis, R.J. (2021) ‘Calcium Phosphate Catalysts for Ethanol Coupling to Butanol and Butadiene’, *Catalysis Letters*, 151(3), pp. 648–657. Available at: <https://doi.org/10.1007/s10562-020-03342-5>.

Rahmadina, E. *et al.* (2020) ‘Penggandaan Skala Proses Pengadukan Terhadap Rendemen Patchouli Alcohol Pada Kristalisasi Scale Up The Agitating Process



Laporan Hasil Penelitian

“Pengaruh Derajat Keasaman (pH) Presipitasi dan Suhu Kalsinasi terhadap Karakterisasi Kalsium Fosfat”

Of Patchouli Alcohol ’S Yield On’, *Prosiding Seminar Nasional Agribisnis*, (November), pp. 107–111.

Razak, A., Isa, N.M. and Adzila, S. (2021) ‘Synthesis of calcium phosphate extracted from eggshell waste through precipitation method’, *Biointerface Research in Applied Chemistry*, 11(6), pp. 15058–15067. Available at: <https://doi.org/10.33263/BRIAC116.1505815067>.

Sinambela, F., Windarti, T. and Parsaroan (2012) ‘Jurnal Kimia Sains dan Aplikasi Pengaruh Waktu pada Pembentukan Kalsium Fosfat dengan’, *Journal of Scientific and Applied Chemistry*, 15(3), pp. 105–110.

Sokolova, V. and Epple, M. (2021) ‘Biological and Medical Applications of Calcium Phosphate Nanoparticles’, *Chemistry - A European Journal*, 27(27), pp. 7471–7488. Available at: <https://doi.org/10.1002/chem.202005257>.

Sunardi, Irawati, U. and Wianto, T. (2011) ‘Karakterisasi Kaolin Lokal Kalimantan Selatan Hasil Kalsinasi’, *Jurnal Fisika FLUX*, 8(1), pp. 59–65.

Sunardi, S., Krismawati, E.D. and Mahayana, A. (2020) ‘Sintesis dan Karakterisasi Nanokalsium Oksida dari Cangkang Telur’, *ALCHEMY Jurnal Penelitian Kimia*, 16(2), p. 250. Available at: <https://doi.org/10.20961/alchemy.16.2.40527.250-259>.

Sunarya, Y.J.P.P. et al. (2022) ‘Pengaruh Metode Hidrolisis Terhadap Karakteristik Kimia Senyawa Kalsium Hasil Ekstraksi dari Cangkang Telur Ayam’, *KOVALEN: Jurnal Riset Kimia*, 8(1), pp. 25–31. Available at: <https://doi.org/10.22487/kovalen.2022.v8.i1.15803>.

Tavoni, M. et al. (2021) ‘Bioactive calcium phosphate-based composites for bone regeneration’, *Journal of Composites Science*, 5(9), pp. 1–27. Available at: <https://doi.org/10.3390/jcs5090227>.

Tsikourkitoudi, V. et al. (2020) ‘Flame-made calcium phosphate nanoparticles with high drug loading for delivery of biologics’, *Molecules*, 25(7). Available at: <https://doi.org/10.3390/molecules25071747>.

Windarti, T. et al. (2016) ‘Preface: International Conference on Recent Trends in Physics (ICRTP 2016)’, *Journal of Physics: Conference Series*, 755(1), pp. 1–7. Available at: <https://doi.org/10.1088/1742-6596/755/1/011001>.



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Woyengo, T.A. *et al.* (2022) ‘Calcium and phosphorus digestibility in rock- and bone-derived calcium phosphates for pigs and poultry: A review’, *Animal Feed Science and Technology*, 294. Available at: <https://doi.org/10.1016/j.anifeedsci.2022.115509>.

Yuliani, S.H. *et al.* (2019) ‘Effects of Particle Size, Extraction Time, and Solvent Selection on Daidzein Extracted Amount from Tempeh-A Fermented Product of Soybean’, *Journal of Pharmaceutical Sciences and Community*, 16(1), pp. 44–49. Available at: <https://doi.org/10.24071/jpsc.001794>.