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Glucose Syrup of Annealing Modified of Cocoyam (Xanthosoma sagittifolium) Starch

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ABSTRACT

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> Glucose syrup is made from the hydrolysis of starch which can be hydrolyzed by acid, enzyme, or a combination of both. The liquefaction process in the production of glucose syrup is controlled by the enzymatic hydrolysis by α amylase. The gelatinized starch is hydrolyzed into simpler molecules to be further hydrolyzed by the glucoamylase into glucose. The more effective liquefaction process can facilitate the saccharification process by the glucoamylase. The starch modified by hydrothermal processes such as annealing is hydrolyzed faster by enzymes. Annealing (ANN) method is a regulation of starch suspension that results in a moisture content ranging from 40%-60%. This process is carried out at temperatures above the glass transition temperature, but still below the gelatinization temperature. This study aims to determine the characteristics of glucose syrup made from modified starch of (Xanthosoma sagittifolium) through enzymatic hydrolysis using α-amylase and glucoamylase. Modification is done by annealing method. Annealing is a process in which starch granules are heated in an excess of water at a temperature slightly below the gelatinization temperature for a relatively long time. The annealing process allowed greater accessibility of both enzymes to the amorphous region. This study used a factorial completely randomized design with two factors. The first factor was the concentration of α -amylase (0.01%; 0.02%; and 0.03%). The second factor was liquefaction time (90, 120, and 150 minutes). In conclusion, optimal treatment resulting in glucose syrup was obtained from the combination treatment of 0.03% a-amylase enzyme and liquefaction time of 120 minutes. This characteristics resulted yield of 90.73\%, water content 17.43%, ash 0.0815%, reducing sugar 77 0.48%, dextrose equivalent 83.04%, total dissolved solids 79.3, clarity value 0.0112, viscosity 2.94 Pa.s, sweetness level 4.40 and glucose content 60%.

Keywords: glucose syrup, starch, cocoyam, annealing, α -amylase

1. INTRODUCTION

The high starch content in tubers such as cocoyam tubers (*Xanthosoma sagittifolium*) can be used as raw material for making glucose syrup. That cocoyam has a high starch content of 77.9% [1]. Cocoyam production in Indonesia is 17316.88 tons [2]. The amylose content of cocoyam was 37.98%, higher than cassava (33.84%)[3] and taro tuber (*Colocasia esculenta*) of 30.62% [4].

Sugar syrup made from natural starch resulted in a low yield, reducing sugar and dextrose equivalent (DE) compared to the modified starch. Several factors influence the production of glucose syrup, including temperature, pH, substrate concentration, enzyme concentration and hydrolysis time. [5] The liquefaction process is the core process of enzymatic hydrolysis controlled by α -amylase and process gelatinized. The gelatinized starch is hydrolyzed into simpler molecules to be further hydrolyzed by the glucoamylase enzyme into glucose. [6]The enzyme glucoamylase, which is an exoamylase, can work effectively when the substrate has been hydrolyzed by the α -amylase enzyme. Therefore, the more effective the liquefaction process, the easier the saccharification process by glucoamylase enzymes will be. The enzymatic hydrolysis of glucose syrup is closely related to the amorphous region of starch. [7]The aamylase enzyme hydrolyzes the amorphous region faster than the crystalline region. The crystalline area consists of the amylopectin fraction and the main component of the amorphous area is amylose. [8] Native starch consists of semi-crystalline regions with low amorphous regions. In general, hydrolysis of natural starch by α -amylase enzyme is not very effective because it requires a longer hydrolysis time and produces a low yield. [9]Starch modified by hydrothermal processes such as annealing is hydrolyzed faster by enzymes. Annealing (ANN) is 4 modification by heating the starch suspension for a

certain time with a water content ranging from 40% to 60% w/w. This process is carried out at temperatures above the glass transition temperature, but still below the gelatinization temperature. Excess water content and heating in the annealing process can cause swelling of the granules, high starch chain mobility in the crystalline part and melting of the crystalline part, either partially or completely, followed by a process of separating the double helical structure of amorphous and crystalline regions so that the amorphous regions increase [9].

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The decrease in crystallinity indicated that the crystalline region of the starch was disturbed by the hydrothermal treatment. [10] That double helix motion during hydrothermal treatment can disturb the crystalline area of starch and change the crystal orientation. This explains the changes that occur in starch crystallinity after hydrothermal treatment. [11] A decrease in the degree of crystallinity is closely related to an increase in amorphous regions and starch digestibility. [12] That the increase in amorphous regions of starch modified by the annealing method had a higher susceptibility to the α -amylase enzyme than natural starch.

The annealing modification causes the formation of pores on the surface of the starch granules, so that it can increase the susceptibility to the α-amylase enzyme. The formation of pores on the surface of starch granules after annealing modification is caused by swelling which causes cracks on the surface of the starch granules. The presence of a porous structure makes starch more susceptible to enzymatic hydrolysis. Annealing modifications carried out prior to enzymatic starch hydrolysis treatment can expand the porous structure. As a result, it can facilitate the penetration of enzymes into the granules during hydrolysis so that the degree of starch hydrolysis will increase. Enzyme hydrolysis in annealed starch occurs mainly on the surface of the starch granules and penetrates into the granules through pores or gaps that have been expanded by the annealing modification. Although amorphous and crystalline lamellae become more regular in annealed starch, accessibility to amorphous regions by enzymes is facilitated by the presence of pores and smectic structures formed due to the annealing process [13]. Some changes in the character of starch occur in the annealing process, including the interaction of starch chains in the amorphous and crystalline parts, an increase in gelatinization temperature 32 nd an increase in amorphous regions [14]. Therefore, this study aims to determine the effect of α amylase concentration and liquefaction time on the production of glucose syrup from modified cocoyam starch using the annealing method.

2. MATERIALS AND METHOD

2.1. Material

The material used in this research is Cocoyam (*Xanthosoma sagittifolium*) tubers, the enzymes used are α -amylase (Sigma Aldrich) and glucoamylase (Novozymes). Calcium Chloride, Sodium Hydroxide, Hydrochloric Acid, glucose standard, ethanol 95%, Iodine, Acetate Acid, Potassium Iodide, Fehling A, Fehling B, DNS (2,3-dinitrosalicylic acid), Buffer Potassium Sodium Tartrate 0,2 M pH 6, Sodium sulfate and Sodium Phosphate (Sigma Aldrich).

2.2. Method

This study was arranged using a factorial completely randomized design with two factors, namely the concentration of α -amylase enzyme (0.01%; 0.02%; and 0.03%) and liquefaction time (90, 120 and 150 minutes). Data were analyzed by analysis of variance. The difference between the two treatments was tested by Duncan Multiple Range Test.

2.2.1. Cocoyam starch making

The making of starch begins with stripping and cutting the tubers. Then the tubers were blended for 2 minutes with the addition of 1:3 w/v water. Furthermore, the cocoyam tuber paste is extracted by squeezing and filtering using a filter cloth. This milling and filtering process was repeated four times. Then the starch is left for 24 hours. After the starch precipitate is formed, the filtrate is separated. The starch precipitate was washed with water. The starch precipitate was dried in a cabinet dryer at 45 oC for 6 hours. Dry starch was refined and sifted using an 80 mesh sieve. Then the starch was tested for yield [15], moisture content [16], ash content [17], starch content [15], amylose content [15] and degree of crystallinity (XRD method).

2.2.2. Modification of cocoyam starch using Annealing method

Preparation of starch suspension by mixing 100 grams of starch added with 300 ml of distilled water (1:3). Furthermore, the heating process is carried out at a temperature of 55 °C for 24 hours using a water bath shaker. The starch suspension is drained. The starch precipitate was dried in a cabinet dryer at 60 °C for 6 hours. The modified dry starch was refined and sifted using an 80 mesh sieve. Modified starch was tested for yield [15], moisture content [16], ash content [17], starch content [15], amylose content [15] and degree of crystallinity.



2.2.3. Glucose Syrup Making

The manufacture of glucose syrup by enzymatic hydrolysis consists of two stages, namely the liquefaction stage using the am-amylase enzyme and the saccharification stage using the glucoamylase enzyme. The modified starch suspension with a concentration of 15% (w/v) was heated at 78 °C until the starch solution thickened. The gelatinized starch was conditioned at pH 5.4 with the addition of 0.1 N HCl and 0.1% CaCl2 (w/w). The amylase enzyme was tested for its activity using the DNS method [18]. Hydrolysis was carried out by adding the enzyme -amylase at a concentration of 0.01%; 0.02%; 0.03% and heated at 90 C for 90; 120; and 150 minutes using a water bath shaker. The liquefied starch was conditioned at pH 4.5 by adding 0.1 N HCl and decreasing the temperature to 60°C. The process was followed by hydrolysis by 0.05% (w/w) glucoamylase enzymes at 60°C for 48 hours using a water bath shaker. The resulting glucose syrup filtrate was separated from the unhydrolyzed starch precipitate using filter paper. The glucose syrup was evaporated using a water bath at 100 °C for 11 hours. The glucose obtained was tested for yield [15], moisture content [16], ash content [17], dextrose equivalent (DE)[19], reducing sugar using the Nelson-Somogyi Method [15], total dissolved solids [20], viscosity [21] and clarity [22].

The sugar purity was tested using HPLC. The work procedure refers to the ligoture [23] using HPLC waters 2695 which is equipped with a Refractive index detector (LC-RID) and data handling system was Software OpenLab (LCRID) as well as a shodex column asahipak NH2P-50 4D (4.6 x 150 mm), eluent CH₃CHN/H₂O= 75/25, flow rate 1.0 mL/min, column temperature 25 °C, injection volume 10 μ L and run time 20 minutes. The yield of unmodified starch was 20%, while the modified starch yield of the annealing method was 19.50%. [24]That the addition of water and heating in the annealing modification will cause the starch to dissolve in water, so that the modified starch has a lower yield than natural starch. The amylose content of modified cocoyam starch was higher than that of unmodified starch. The results of starch testing from cocoyam tubers can be seen in Table 1.

The testing results of the crystallinity of annealing modified starch and unmodified starch using XRD (XRay Diffraction) can be seen in Figure 1. Crystallinity can indicate a change in amorphous regions which is indicated by a decrease in crystallinity.

Based on the chromatogram, it can be seen that in the annealing modified starch chromatogram, there are 3 peaks that are gentler than the unmodified cocoyam starch. The degree of crystallinity of modified cocoyam starch was 24.76%, lower than that of unmodified cocoyam starch, which was 32.42% (Figure 1). Modified sorghum starch annealing method obtained a crystallinity degree of 23.03% lower than that of unmodified sorghum starch, which was 36.5%[32]. That hydrogen bonding disturbances between amorphous and crystalline regions by the treatment The heat during the annealing modification process causes the expansion of the amorphous region. The changes produced by heating starch below 100°C with high moisture content cause an increase in the degree of association of starch molecules in the granules. Moisture content is an important factor influencing changes in the amorphous region due to the contact between water and starch because reorientation or rearrangement of molecules in starch granules will cause X-ray diffraction patterns to change[35].

3. RESULTS AND DISCUSSION

3.1. Characteristics of Cocoyam Starch

Table 1. The results of the cocoyam (Xanthosoma sagittifolium) starch test

| Component | Annealing modified cocoyam starch P | | |
|-----------------------------|-------------------------------------|------------------|--|
| | analysis results | analysis results | |
| Yield (%) | 20.00 | 19.50 | |
| Water (%) | 10.15 ± 0.02 | 11.89 ± 0.08 | |
| Ash (%) | 0.19 ± 0.002 | 0.18 ± 0.003 | |
| Starch (%) | 87.48 ± 0.06 | 87.17 ± 0.21 | |
| Amylose (%) | 29.56 ± 0.06 | 34.14 ± 0.18 | |
| Amylopectin (%) | 70.44 ± 15.33 | 65.86 ± 0.18 | |
| Degree of crystallinity (%) | 32.42 | 24.76 | |

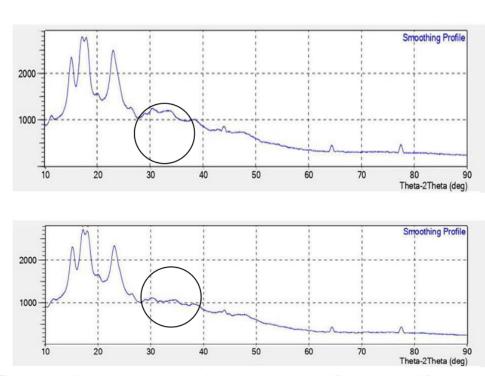


Figure 1. Crystallinity chromatogram results using XRD method: (a) unmodified starch, (b) modified starch annealing

The association between amylose and amylopectin chains in crystal lamellae has a very weak interaction strength, because it involves the interaction of long amylose chains with short amylopectin outer chains. This weak interaction will be disrupted during the reorganization of the starch chain during the annealing process caused by the presence of excessive heat and water conditions, resulting in the cutting of amylopectin long chains located on the outer edge of the amylopectin starch granules into amylose helices and penetration of amylose into the amorphous region. The annealing process causes a decrease in peak intensity and relative crystallinity of starch, so that the relative crystallinity of modified starch using the annealing method is low [36].

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Starch modification by the annealing method can increase the amorphous area as indicated by the results of x-ray diffraction (XRD) with wider peaks than the original starch. The wider the peaks produced resulted the larger the amorphous area[33].

The decrease in crystalline area causes starch granules to swell more easily during the gelatinization process. This is in accordance [37] that the double helix and crystalline structure of starch were damaged during the gelatinization process. However, the large crystalline area causes swelling of the starch granules to be restrained by the crystalline structure which is difficult to break. [38]The amorphous region is the part that can expand in the starch gelatinization process. Water absorbed in the amorphous region causes starch granules to lose their crystalline structure stability. The increase in the amorphous region will cause the α -amylase enzyme to more easily hydrolyze starch [39].

3.2. Physical Characteristics of Glucose Syrup

The yield of glucose syrup is in the range of 84.74% -90.73%. This yield research was quite large because the starch was modified using the annealing method. Modification of starch using annealing can increase amylose content and amorphous area which causes starch to be more easily hydrolyzed by the α -amylase enzyme so that the yield of the resulting product was higher. [12]That an increase in the amorphous region of starch analyzed by X-ray caused starch modified by the annealing method to have a higher susceptibility to the α -amylase enzyme than natural starch.

The addition of 0.03% enzyme concentration with 120 minutes of liquefaction time resulted in higher yields compared to 150 minutes of liquefaction time with the same α -amylase enzyme concentration (Table 2). This is presumably because at the same substrate concentration and enzyme concentration, if the liquefaction time is

increased, the glucose produced does not increase because the active site of the enzyme is full of substrate. The hydrolysis of annealed modified starch at 120 minutes of liquefaction has intensified so that the -1,4 glycosidic bonds in starch have been hydrolyzed by α -amylase maximally. The addition of the enzyme concentration will increase the reaction rate. However, the increase in reaction speed will decrease for each increase in enzyme concentration when the amount of substrate is met at a certain concentration.

The water content of glucose syrup ranged from 17.06% - 19.72% (Table 2). With 0.03% α-amylase enzyme concentration and 150 minutes of liquefaction time, the lowest water content was 17.06%. The decrease in the water content of glucose syrup along with the increase in enzyme concentration and duration of liquefaction is suspected to be hydrolysis using enzymes requiring water. The use of more enzymes and longer hydrolysis time resulted in less glucose syrup water. The longer the hydrolysis temperature and time, the more water in the sample will evaporate. The increasing concentration of a-amylase enzymes will cause the water content of glucose syrup to decrease because more hydroxyl groups are needed during the hydrolysis process. Water will donate its group to complete the monosaccharide group formed from the starch hydrolysis process so that the more starch hydrolyzed by the α amylase enzyme, the lower the water content of glucose syrup. [40]Reported that there was an increase in the hydrolysis of cassava starch along with the addition of enzyme concentration which caused the water content of glucose syrup to decrease.

The ash content of glucose syrup is in the range of 0.157% - 0.178%. The ash content of glucose syrup can come from the starch used or the pH adjustment during the hydrolysis or purification process. The total dissolved solids of glucose syrup ranged from 70.5 0Brix - 79.25 OBrix. The results [41] showed that the total dissolved solids of glucose syrup from cassava starch were almost the same, ranging from 70 to 78 °brix. The addition of increasing enzyme concentration and longer liquefaction caused the brix grade of modified starch glucose syrup to be produced higher. This can happen because the increase in the degree of brix is closely related to the level of reducing sugar produced. The higher the reducing sugar content, the greater the total dissolved solids in the product. Components measured as total dissolved solids can be organic acids, sucrose, reducing sugars, salts and proteins [42]

The gradual increase in the degree of total dissolved solids indicates that α -amylase can convert starch rapidly due to high enzyme activity. The increase in total sugar concentration indicated by total dissolved solids is the same as the increase in reducing sugar concentration [43].

| Trea | tment | | | | | |
|------------------|----------------------------------|-----------------------------|-------------------------------|------------------------------------|--------------------------|---------------------------|
| α-amilase (%) | Time liquefaction (minuts) | yield (%) | Water(%) | Total solid(^o brix) | Viscosity (Pa.s) | clarity |
| | 90 | $84.74^{a} \pm 0.10$ | $19.72^{g} \pm 0.01$ | $70.50^{a} \pm 0.71$ | $2.34^{a} \pm 0.008$ | $7.04^{\rm f} \pm 0.0007$ |
| 0.01 | 120 | $86.43^{\circ} \pm 0.03$ | $19.67^{g} \pm 0.22$ | $73.75^{\circ} \pm 0.35$ | 2.51°+0.003 | $7.35^{\circ} \pm 0.0014$ |
| | 150 | $85.65^{\mathrm{b}}\pm0.18$ | $19.49^{\rm f} \pm 0.04$ | $72.75^{b} \pm 0.35$ | $2.50^{\rm b} \pm 0.019$ | $7.69^{d} \pm 0.0021$ |
| | 90 | $87.62^{\rm d}\pm0.06$ | $19.04^{e} \pm 0.32$ | $74.25^{\circ} \pm 0.35$ | $2.59^{d} \pm 0.016$ | $8.00^{\circ} \pm 0.0014$ |
| 0.02 | 120 | $88.48^{\rm f}\pm0.24$ | $18.62^{\rm d}\pm0.02$ | $77.00^{\circ} \pm 0.00$ | $2.73^{e} \pm 0.013$ | $8.13^{\circ} \pm 0.0007$ |
| | 150 | $87.81^{\circ} \pm 0.24$ | $18.13^{\circ} \pm 0.13$ | $76.25^d \pm 0.35$ | $2.71^{\circ} \pm 0.008$ | $8.47^{\circ} \pm 0.0014$ |
| 0.03 | 90 | $89.43^{g} \pm 0.01$ | $18.03^{\mathrm{b}} \pm 0.13$ | $78.25^{\rm f} \pm 0.35$ | $2.83^{e} \pm 0.003$ | $8.70^{\circ} \pm 0.0014$ |
| | 120 | $90.73^{\rm h}\pm0.08$ | $17.43^{\mathrm{b}} \pm 0.15$ | $79.25^{g} \pm 0.35$ | $2.94^{\circ} \pm 0.011$ | $8.93^{b} \pm 0.0007$ |
| | 150 | $90.04^{\rm h}\pm0.17$ | $17.06^a\pm0.06$ | $78.75^{g} \pm 0.35$ | $2.90^{\circ} \pm 0.011$ | $9.26^{a} \pm 0.0014$ |
| | | | | | | |

| Table 2. The characteristic | c of glucose s | yrup from cocoyam | modification starch |
|-----------------------------|----------------|-------------------|---------------------|
|-----------------------------|----------------|-------------------|---------------------|

The increasing concentration of enzymes and the liquefaction time is causing the value of clarity and reducing sugar of glucose syrup produce higher and less

non-sugar components. The clarity of glucose syrup is influenced by the content of non-sugar components, such as mineral metals, oligosaccharides and other organic materials. The more non-sugar components in the syrup, the lower the transmission value or the higher the absorbance value. Clarity and color quality in starch hydrolyzate are influenced by the content of ISSP (Insoluble Starch Particles) in starch. ISSP is starch particles composed mostly of amylose joined together to form a straight chain (linear), the more starch is hydrolyzed to glucose causing the clarity of glucose syrup to increase.

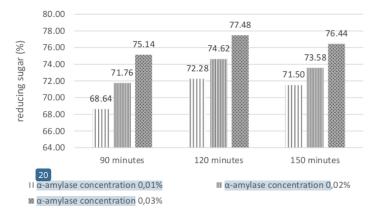
The viscosity of glucose syrup ranges from 2.34 Pa.s - 2.83 Pa.s. The addition of alpha amylase enzymes can increase the viscosity of glucose syrup due to the increase in total solids causing an increase in the viscosity of glucose syrup. The longer the hydrolysis time, the more dissolved solids in the liquid sugar are produced so that the more interactions between the molecules of the solids are formed.

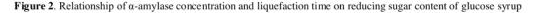
3.3. Reducing Sugar Level

The reducing sugar content of glucose syrup ranged from 68.64% - 77.48% (Figure 2). The treatment of 0.03% α-amylase concentration and 120 minutes liquefaction time obtained the highest reducing sugar content of 77.48%. The reducing sugar content of glucose syrup from modified cocoyam starch was higher. [40] that the starch modification process using the annealing method can increase the amylose content and the area of the amorphous region which causes starch to be more easily hydrolyzed by the α -amylase enzyme so that the yield and the reduced sugar content of the resulting product are higher. [44]Starch modified by the hydrothermal process was hydrolyzed more quickly by enzymes and caused the starch granules to swell with water absorption and the crystal structure was disturbed by increasing the amorphous region.

Alpha-amylase is an endoenzyme that cuts the bonds from the middle or inside the 1,4-glycosidic and 1,6glycosidic bonds, the more alpha amylase enzymes are added, the more non-reducing ends are formed so that the glucoamylase enzyme can hydrolyze more[45]. The relationship between the treatment of -amylase enzyme concentration and duration of liquefaction on the reducing sugar content of glucose syrup can be seen in Figure 2.

In the treatment of 0.03% α-amylase enzyme concentration and 120 minutes of liquefaction time, the highest reducing sugar content was found at 77.48%. Reducing sugar levels will increase along with the addition of the concentration of the α -amylase enzyme. However, at the same substrate concentration and enzyme concentration, if the liquefaction time was increased, the reduced sugar content obtained did not increase because the active site of the enzyme was full of substrate. The addition of the enzyme concentration will increase the reaction rate. The increase in reaction speed is indicated by the increasing number of products formed and the amount of substrate that continues to decrease. [41]The increase in yield, the value of DE and reducing sugar will reach a certain limit point, after that point is exceeded there will be no change in yield, the value of DE and reducing sugar will be higher even though the liquefaction period is extended. When the active site of the enzyme is saturated with the substrate, no more substrate can be attached to the active site of the enzyme. This happens because many substrates have been hydrolyzed into reducing sugars, so the addition of hydrolysis time will not increase the reducing sugar content.





3.4. Dextrose Equivalent (DE)

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The Dextrose Equivalent (DE) value of glucose syrup ranged from 72.84% - 83.04% (Figure 3). The more addition of a-amylase enzyme, the more starch is hydrolyzed into simpler sugars. The a-amylase enzyme cleaves the starch randomly at the glycosidic linkage which is then further hydrolyzed into maltose and glucose. [46]That the α-amylase enzyme can hydrolyze -1,4 glycosidic bonds in starch polymers internally. The mechanism of action of α -amylase is divided into two stages, namely the degradation of amylose and the formation of glucose and maltose. The first step is the degradation of amylose into maltose and maltotriose which occur randomly. The second stage is the formation of glucose and maltose as products and is not random. Both are the action of the α -amylase enzyme on the amylose molecule. In the amylopectin molecule, the action of a-amylase will produce glucose, maltose and a series of -limit dextrins, as well as oligosaccharides consisting of four or more glucose containing -1,6glycosidic bonds.

These results are better than [47] the making of glucose syrup from cassava starch which produces a Dextrose Equivalent (DE) value of 44.3%. The DE value increased with increasing sugar content. The relationship between the treatment of α -amylase enzyme concentration and liquefaction time on the amount of Dextrose Equivalent (DE) glucose syrup can be seen in Figure 3.

Figure 3. shows the treatment with 0.03% α -amylase enzyme concentration and 120 minutes of liquefaction time, the highest Dextrose Equivalent (DE) level is 83.04% (Figure 3). The glucoamylase enzyme which is an exo amylase can work effectively when the substrate has been hydrolyzed by the α -amylase enzyme.

Therefore, the more effective the liquefaction process, the easier the saccharification process by glucoamylase enzymes will be. The Michaelis-Menten model explains that at a certain substrate concentration, increasing the enzyme concentration will increase the reaction rate until finally saturation occurs where the reaction rate does not increase at a certain time.

The DE value of glucose syrup from starch modified using annealing was greater than that using unmodified starch. This can happen because the starch modification process using the annealing method can increase the amylose content and amorphous area, which causes starch to be more easily hydrolyzed by the α -amylase enzyme so that the resulting DE value is higher. [48] Amylose and amylopectin were hydrolyzed simultaneously in the amorphous fraction (amylose and intercrystalline regions of amylopectin) of starch granules. The amorphous region can be easily hydrolyzed by the α -amylase enzyme.

The value of Dextrose Equivalent (DE) in glucose syrup from waxy corn starch modified using annealing was 67.6% higher than native starch waxy corn, which was 66.7%. An increase in the value of Dextrose Equivalent (DE) also occurred in glucose syrup from common corn starch modified using annealing by 52.6%, higher than native starch common corn, which was 48.7% [49].

Modification of annealing can result in the formation of pores in the starch which makes it easier for enzymes to hydrolyze starch. [13]That annealing modification causes the formation of pores on the surface of starch granules, thereby increasing susceptibility to α -amylase enzymes. The formation of pores on the surface of starch granules after annealing m16 fication is caused by swelling which causes cracks on the surface of the starch granules. The presence of a porous structure in starch makes starch more susceptible to enzymatic hydrolysis.

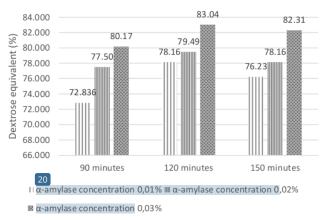


Figure 3. The relationship of α -amylase concentration and liquefaction time on the dextrose equivalent (DE) of glucose syrup

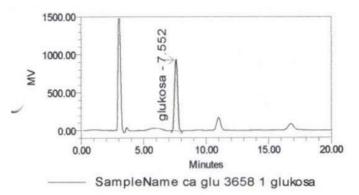


Figure 4. Chromatogram of glucose syrup from annealing modified cocoyam starch

To determine the purity of glucose syrup in treatment of α -amylase enzyme concentration of 0.03% and liquefaction time of 120 minutes were carried out using high performance liquid chromatography (HPLC) in Figure 4. The glucose standard used was 5 different concentrations:

| Table 3. | The g | lucose | analysis | by | using | HPLC |
|----------|-------|--------|----------|----|-------|------|
|----------|-------|--------|----------|----|-------|------|

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| Volume of glucose standard | area of the curve |
|-------------------------------|-------------------|
| 0.1 ml | 757295 |
| 0.2 ml | 1647336; |
| 0.5 ml | 3912445; |
| 1.0 ml | 8059445 |
| 2 ml | 18342587 |

Then from the area of the chromatogram curve, the regression equation y = 11733786 x - 27341.4 was obtained. The glucose chromatogram curve was formed with an average area of 14445415 so that it was known that the glucose purity of the annealed modified starch glucose syrup was 60.70% (Figure 4).

4. CONCLUSION

The degree of crystallinity of modified cocoyam starch was 24.76%, lower than that of unmodified cocoyam starch, which was 32.42%. Disruption of hydrogen bonds between the amorphous and crystalline regions by heat treatment during the annealing modification process causes expansion of the amorphous regions. The increase in the amorphous region will cause the α -amylase enzyme to more easily hydrolyze starch so

that it can produce glucose syrup from the aggregated starch more optimally.

This study obtained the optimal result of glucose syrup from annealing modified cocoyam starch with the treatment of α -amylase enzyme concentration of 0.03% and liquefaction time of 120 minutes. This glucose syrup had a yield of 90.73%, water content 17.43%, ash 0.0815%, reducing sugar 77.48%, dextrose equivalent 83.04, total dissolved solids 79.3, clarity value 0.0112, viscosity 2.94 Pa.s, and glucose purity 60.70%.

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