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Tracking of Dissolved Effluent Organic Matter (dEfOM) in wastewater treatment plant by using fluorescence method

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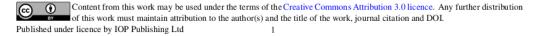
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Tostract. Discharge of wastewater into water body can change the quantity and quality of natural organic matter (NOM) in source w12, because wastewater may contain dissolved effluent organic matter (dEfOM). dEfOM consists of a heterogenous mixture of organic compounds with various structures, such 20 dissolved organic matter, microbial by-products, endocrine disrupting compound, etc. The objectives of this study is to track the fate of dEfOM by using fluorescence excitation emission matrices (FEEMs) in order to apply of reclaimed water from industrial wastewater. Tracking of NOM was observed in industrial park of wastewater treatment, which is consist of equalization tank 19 oagulation, flocculation, sedimentation, activated sludge tank and clarifier tank. FEEMs with parallel factor analysis (PARAFAC) analysis decomposed the organic matter in wastewater into two components, namely component C1 at (Ex/Em): 330(235)/410 is humic acid like and fulvic, component C2 at (Ex/Em): 280(230)/350 is soluble microbial products and aromatic protein. Characteristic of dEfOM depends on the wastewater characteristic and its treatment processes.

1. Introduction

Discharging of effluting from treated wastewater is an one of the contributed source of organic matter to aquatic systems. The contribution of organic matter from wastewater effluent discharge severely affects water quality, par 16 larly in water treatment processes. The sustained increase in quantity and quality of organic matter in raw water will decrease the performance of water treatment processes and add the advanced technology of water purification, such as dosage increasing for coagulant, oxidant, desinfectant, and also membrane fouling, decreasing adsorption capacity of activated carbon [1]. Effluent organic matter (EfOM) is the organics in the sewage that cannot be removed during wastewater treatment. Constituens of EfOM in wastewater can be classified as synthetic organic compounds and microbial by-products derived from biological wastewater treatment processes. EfOM are typically cell fragments and macromoleculs, such as polysaccharides, proteins, lipids, carbohydrates, amino acids, vitamins and chlorophyll [2].

EfOM from different types of wastewater could have various compounds, which differ from terrestrial organic matter. Component hydrophobic and transphilic of EfOM are derived from microbial sources, EfOM contains between 60-80% hydrophilic material [3]. Coagulation, flocculation, sedimentation and rapid sand filtra [15] is mostly used and feasible method to treat organic matter. The remaining organic matter can be further removed by advanced treatment processes, such as adsorption with activated carbon [4], biofiltration, ion exchange [5], pellet softening



[6], ozonation [7], membrane technology [8], advanced oxidation process (AOP), such as hydroxyl radial.

Combined treatment process and hybrids with different combination of those mentioned methods have been employed for removal of EfOM. The objective of these combined systems is to maximise the removal of different fractions of the NOM. Some of these methods include: coagulation followed by ultrafiltration, ozonation followed by granular activated carbon, activated carbon filter followed by reverse osmosis, biofiltration followed by nanofiltration, ion exchange followed by activated carbon filtration and ozonation followed by biofiltration and membrace processes [6,9]. However, some researchs showed that the biological treatment processes will produce soluble microbial products (SMPs) and extracellular polymeric substances (EPS) material [10,11]. Several research projects had investigated using biological processes or combined chemical processes with biological processes to remove organic matter [12,13].

Several methods for organic matter composition and structre have been developed in the last decades, such as NMR, FTIR, HPSEC, EEM spectroscopy, which detect organic matter based on their properties [14]. Nevertheless, characterization of EfOM in wastewater plant in Indonesia and concern to water reuse or wastewater reclamation is still a few. Therefore, an interest in characterizing EfOM has become more important.

2. Research Metho

Wastewater effluent was collected from wastewater treatment plant of industrial park. The wastewater treatment train include inlet chamber, equalization tank, coagulation, flocculation, sedimentation, activated sludge tank, clarifier and indicator pond. Sample was taken in each effluent unit process twice per-week for one 10 pnth. As this study focused on characterization of EfOM, the collected effluent wastewater was filtered through $1.2/0.5 \,\mu$ m and $0.45 \,\mu$ m disposable capsule filters (both from Millipore Corporation, USA) to remove any suspended solids before further usage.

Wastewater effluent samples were analysed for bulk parameter characteristics, including dissolved organic carbon (DOC) by using a total org14 c carbon analyzer (Model TOC-5000A, Shimadzu, Japan), ultraviolet absorbance at wavelength 254 nm (UV₂₅₄) by using a UV-VIS spectrophotometer (Model UV-2600, Shimadzu, Japan), and pH 18 asurement [15].

Fluorescence measurements have been undertaken using a Perkin Elmer LS-55 luminescence spectrometer was used to measure fluorescence signal with excitation wavelengths 200-400 nm at 10nm and emission wavelength 2.5-nm increment from 300 to 547.5 nm. EEMs were measured using 1cm path ength quartz cells. EEM of Milli-Q water samples was collected throughout the experiment period. Stater Raman scatter peak were eliginated by substracting the EEMs of Milli-Q water blank and norma 23 d to the area under the water Raman peak of the blank at excitation wavelength 350 nm, producing data in Raman Units (R.U.).

PARAFAC analysis applied all EEM datasets fluorescence, which uses an alternating least-square algorithm to decompose the data signal into a set of trilinear terms and a residual array, as described with:

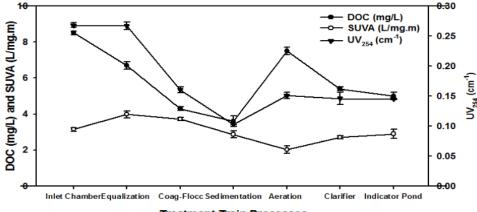
 $x_{ijk} = \sum_{f=1}^{F} \alpha_{if} \ b_{jf} \ c_{kf} + e_{ijk}$ ³ i = 1,2,...,I; j = 1, 2,..., J; k = 1,2,...,K; f = 1,2,...,F

where x_{ijk} is fluorescence intensity of the *i*th samples at the *j*th variable (emission mode) and at *k*th variable (exc 7 tion mode); a_{if} is the concentration of the *f*th analyte in the *i*th sample; b_{if} is the fluorescence quantum efficiency of the *n*th analyte at emission wavelength *j*; c_{kf} is the specific absorption 2 oefficient at excitation wavelength *k*; F is the number of fluorophores (components) and e_{ijk} is the residual noise, representing the variability not accounted for by the model [16]. drEEM toolbox in Matlab® (http://www.models.life.ku.dk/drEEM) [17], was used to generate parallel factor analysis (PARAFAC) models.

3. Result and Discussion

3.1. Bulk wastewater quality for industrial wastewater treatment train

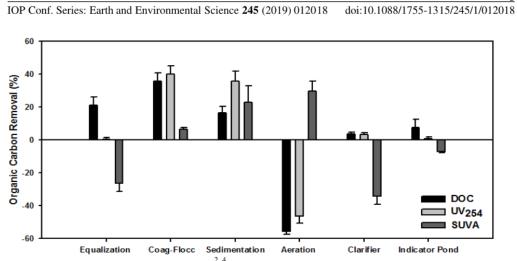
Figure 1 shows the variation of organic matter surrogates in term of DOC concentration, UV_{254} and SUVA value of treated wastewater in each unit process. The variation was more pronounced for the treatment by coagulation, flocculation, sedimentation, and activated sludge processes. Unit process of inlet chamber and equalization tank has the highest value, while the effluent of sedimentation has the lowest value for organic carbon and UV_{254} . It indicated that aromatic process, which composed organic carbon in wastewater, are more amenable to be removed in the physical treatment, such as coagulation, flocculation and sedimentation. It is consistent with that reviewed in the previous study [18,19].



Treatment Train Processes

Figure 1. The variation of DOC, UV₂₅₄, and SUVA value of effluent wastewater treatment of industrial park.

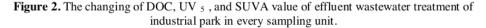
Figure 2 shows the percentage reduction of organic matter TOC, UV₂₅₄ and SUVA value across the wastewater treatment processes. The general trend is a reduction in DOC through equalization, coagulation, flocculation, sedimentation, and insignificantly removal in clarifier after activated process and in the indicator pond. Activated sludge process could increase the organic compound, which represented through increasing DOC and UV₂₅₄ concentration. In addition, biological process could degrade hydrophobic and hydrophilic compound as detected by SUVA value. Biological processes, which involved microbiology for biodegradation, could release organic matter, such as soluble microbial products (SMPs) as by-product during their metabolism processes. SMPs will be derived during substrate utilization and may released from cell lysis during biomass decay [20].



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Treatment Train Processes



3.2. Using Fluorescence Excitation-Emission Matrices (F-EEMs) to Characterize dEfOM Contained 1 the Wastewater 21

Parallel Factor Analysis (PARAFAC) was needed to model the dataset of fluorescence excitation emission matrices (F-EEMs). The specific G1a set with non-negativity constraints were used on all modes during preprocessing dataset. The F-EEMs were also Raman calibrated by normalizing to the area under the Raman scatter peak of Milli-Q water samples, in order to remove scatter. Measurement at excitation wavelengths below 220 nm were excluded, because 5 samples were identified by leverange plots (about 9% of 56 samples) as outlier. Each dataset was normalized to its total signal before PARAFAC modelling in order to reach data. A series of PARAFAC models consisting of 2 components were generated and those number of fluorescence components was validated using split half analysis and split half validation. Finally, PARAFAC analysis decomposed the organic matter in treated wastewater into two components, as shown in Figure 3.

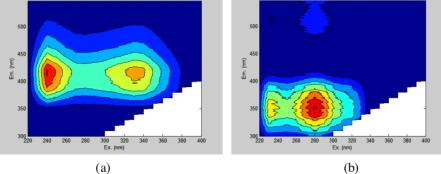


Figure 3. Fluorescence (a) Component 1 and (b) Component 2 have been found from effluent of wastewater treatment in industrial park.

The fluorescence components in this study are comparable with previously classified components [21,22]. The component C1 comprises two peaks centered at excitation/emission (Ex/Em): 230/410 nm, which is identified as fulvic acid-like and also as the main peak, the secore peak of component C2 at (Ex/Em): 330/410 nm, as humic acid-like. The component C2 had first and secondary excitation peak, about 280 nm and 230 nm, respectively, and with single emission peak about 350 nm, and it is

dentified as soluble microbial products (SMPs) and aromatic protein, respectively. Fa ther, the component distribution in the source and treated water was des 12 ed in terms of Fmax in order to attempt the relative contribution of the PARAFAC flue rophore to the tracking of dissolved organic matter. However, the higher Fmax does not indicate a major existence of the component in the all sample [23].

Effluent of wastewater treatment was composed of fluorescence soluble microbial products as dominant component and followed by humic acid-like and aromatic proten. The presence of those component probably contributed by the biological activities which released their metabolite products. Previous studies have mentioned that biological process might released organic compound during metabolism and decay [20], whether as extracellular polymeric substances (EPS), SMPs, and inert biomass [10].

4. Conclusion

This study concludes that FEEMs analysis revealed that two component, (Ex/Em): 230/410 nm; (Ex/Em): 330/410 nm, derived from PARAFAC modelling were the major organic component in effluent wastewater from industrial park. F-EEMs and dEfOM bulk parameter demonstrated that wastewater treatment train had applied, significantly reduced the aromaticity of organic carbon matter in water. Coagulation-flocculation and sedimentation removed organic carbon significantly. Biological process gives higher contribution to the existence of dEfOM than other processes.

5. Acknowledgment

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