



PERANCANGAN PABRIK

Pabrik Diocetyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

APPENDIX A NERACA MASSA

Kapasitas produk = 30000 ton/tahun
= 90909 kg/hari
= 3787.9 kg/jam
Waktu operasi = 330 hari/tahun
= 24 jam/hari
Basis perhitungan = 1445.3 kg/jam

Data Berat Molekul masing-masing Komponen

Komponen	BM (kg/kmol)
$C_8H_4O_3$	148.0000
$C_8H_{18}O$	130.0000
H_2O	18.0000
$C_{16}H_{36}O_4Ti$	340.0000
$C_{24}H_{38}O_4$	390.0000
$C_8H_6O_4$	166.0000
NaOH	40.0000
Karbon Aktif	12.0000
$C_{16}H_{35}O_4TiN_2$	362
$C_8H_4O_4Na_2$	210
C_6H_6	78
Na_2CO_3	106

1. Melter

Fungsi : Untuk melelehkan $C_8H_4O_3$ sebelum masuk ke reaktor



Komposisi Phthalic anhydride

Komposisi	Kadar
$C_8H_4O_3$	99.7%
$C_8H_6O_4$	0.2%
H_2O	0.1%



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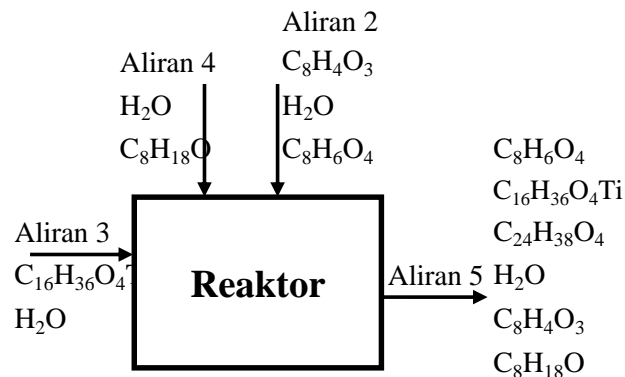
Kebutuhan phthalic anhydride

$$\begin{aligned} \text{C}_8\text{H}_4\text{O}_3 &= 1449.7 \times 99.7\% \\ &= 1445.3 \text{ Kg/Jam} \\ \text{C}_8\text{H}_6\text{O}_4 &= 1449.70 \times 0.2\% \\ &= 2.8994 \text{ Kg/Jam} \\ \text{H}_2\text{O} &= 1449.70 \times 0.1\% \\ &= 1.44970 \text{ Kg/Jam} \end{aligned}$$

NERACA MASSA MELTER			
Komponen Masuk		Komponen Keluar	
Phthalic Anhydride Dari Me		Phthalic anhydride masuk reak	
Komponen	Massa (Kg/Jam)	Komponen	Massa (Kg/Jam)
$\text{C}_8\text{H}_4\text{O}_3$	1445.34966	$\text{C}_8\text{H}_4\text{O}_3$	1445.34966
$\text{C}_8\text{H}_6\text{O}_4$	2.89939751	$\text{C}_8\text{H}_6\text{O}_4$	2.89939751
H_2O	1.44970	H_2O	1.44969876
TOTAL	1449.69876	TOTAL	1449.69876

2. Reaktor

Fungsi : Tempat untuk mereaksikan $\text{C}_8\text{H}_4\text{O}_3$ dengan $\text{C}_8\text{H}_{18}\text{O}$ menggunakan katalis $\text{C}_{16}\text{H}_{36}\text{O}_4\text{Ti}$



Stokiometri

Konversi Re ϵ = 99.8%

	$\text{C}_8\text{H}_4\text{O}_3$	$2\text{C}_8\text{H}_{18}\text{O}$	$\text{C}_{24}\text{H}_{38}\text{O}_4$	H_2O
m	9.7659	19.532	-	-
r	9.7463	19.4927	9.7463	9.7463
s	0.0195	0.0391	9.7463	9.7463



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Kebutuhan Bahan Baku

1. Kebutuhan Phthalic Anhydride

Komponen	Massa (Kg/Jam)
$C_8H_4O_3$	1445.34966
$C_8H_6O_4$	2.89939751
H_2O	1.44969876
TOTAL	1449.69876

2. Kebutuhan 2-Ethyl Hexanol

Komposisi	Kadar
$C_8H_{18}O$	99.5%
H_2O	0.5%

$$\begin{aligned} \text{Massa } C_8H_{18}O &= 19.5318 \text{ kmol/Jar} \times 130.000 \\ &= 2539.1 \text{ Kg/Jam} \end{aligned}$$

$$\begin{aligned} \text{Massa } C_8H_{18}O + &= \frac{2539.1 \text{ Kg/Jam}}{99.5\%} \\ &= 2551.9 \text{ Kg/Jam} \end{aligned}$$

$$\text{Massa } H_2O = 12.759 \text{ Kg/Jam}$$

Komposisi	Massa (Kg/Jam)
$C_8H_{18}O$	2539.12778
H_2O	12.7594361
Total	2551.88722

3. Kebutuhan Katalis

Komposisi	Kadar
$C_{16}H_{36}O_4Ti$	99.6%
H_2O	0.4%

$$\begin{aligned} \text{Massa Katalis} &= \text{Total Bahan} \times 0.002\% \\ C_{16}H_{36}O_4Ti &= 3984.47744 \times 0.002\% \\ &= 0.05976716 \text{ Kg/Jam} \end{aligned}$$

$$\begin{aligned} \text{massa } C_{16}H_{36}O_4Ti + &= \frac{0.05976716 \text{ Kg/Jam}}{99.6\%} \\ &= 0.06000719 \text{ Kg/Jam} \end{aligned}$$

$$\text{massa } H_2O = 0.00024003 \text{ Kg/Jam}$$

Komposisi	Massa (Kg/Jam)
$C_{16}H_{36}O_4Ti$	0.05976716
H_2O	0.00024003
Total	0.06000719



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Pabrik Diocetyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Sisa Bahan Baku

1. Sisa Phthalic Anhydride

$$\text{Massa } C_8H_4O_3 = 2.89070 \text{ Kg/Jam}$$

2. Sisa 2-Ethyl Hexanol

$$\text{Massa } C_8H_{18}O = 5.07826 \text{ Kg/Jam}$$

Produk Reaksi

$$\text{Massa } C_{24}H_{38}O_4 = 3801.07 \text{ Kg/Jam}$$

$$\text{Massa } H_2O = 175.43 \text{ Kg/Jam}$$

NERACA MASSA REAKTOR			
Komponen Masuk		Komponen Keluar	
Phthalic Anhydride Dari Me		1. Produk Reaktor	
Komponen	Massa (Kg/Jam)	Komponen	Massa (Kg/Jam)
$C_8H_4O_3$	1445.34966	$C_8H_4O_3$	2.89069932
$C_8H_6O_4$	2.89939751	$C_8H_6O_4$	2.89939751
H_2O	1.44969876	$C_8H_{18}O$	5.07825556
Total	1449.69876	$C_{16}H_{36}O_4Ti$	0.05976716
		$C_{24}H_{38}O_4$	3801.07429
2. 2-Ethyl Hexanol		H_2O	189.643573
Komponen	Massa (Kg/Jam)	Total	4001.64598
$C_8H_{18}O$	2539.12778		
H_2O	12.7594361		
Total	2551.88722		
3. Katalis			
Komponen	Massa (Kg/Jam)		
$C_{16}H_{36}O_4Ti$	0.05976716		
H_2O	0.00024003		
Total	0.06000719		
TOTAL	4001.6460	TOTAL	4001.6460

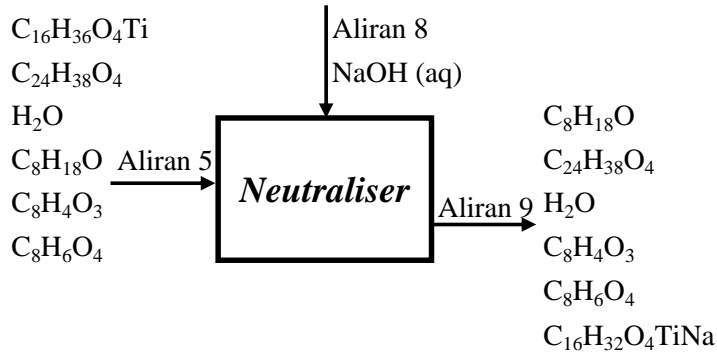


PERANCANGAN PABRIK

Pabrik Diocetyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

3. Neutralizer

Fungsi : Untuk menetralkan katalis $C_{16}H_{36}O_4Ti$ menggunakan NaOH



Komponen Masuk

Komponen	Massa (Kg/Jam)
$C_8H_4O_3$	#####
$C_8H_6O_4$	2.89939751
$C_8H_{18}O$	5.07825556
$C_{16}H_{36}O_4Ti$	0.05976716
$C_{24}H_{38}O_4$	3801.07429
Komponen	Massa (Kg/Jam)
H_2O	189.643573
Total	4001.64598

Stokimetri Netrallizer

Reaksi 1

	$C_{16}H_{36}O_4Ti$	+	NaOH		$C_{16}H_{35}O_4TiNa$	+	H_2O
m	0.00017579		0.0002		-		-
r	0.00017579		0.0002		0.00017579		0
s	0		0		0.00017579		0

Kebutuhan Larutan NaOH 0.4%

Mol NaOH Total = 0.00017579 Kmol/Jam

Massa NaOH Total = 0.00703143 Kg/Jam

Produk Terbentuk

Massa $C_{16}H_{35}O_4T$ = 0.06363445 Kg/Jam

Massa H_2O RX 1 = 0.00316414 Kg/Jam



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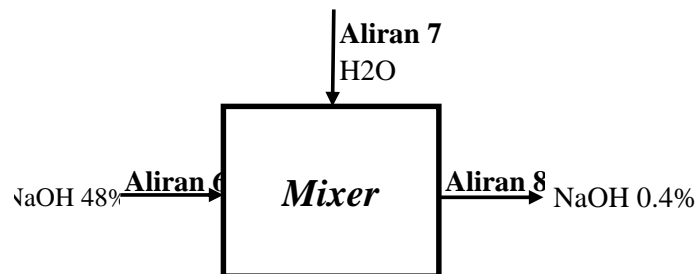
Pabrik Diocetyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\text{Massa H}_2\text{O Total} = 0.00316414 \text{ Kg/Jam}$$

NERACA MASSA NETRALLIZER			
Komponen Masuk		Komponen Keluar	
1. Produk Reaktor		1. Produk Netrallizer	
Komponen	Massa (Kg/Jam)	Komponen	Massa (Kg/Jam)
C ₈ H ₄ O ₃	2.8906993	C ₈ H ₁₈ O	5.07825556
C ₈ H ₆ O ₄	2.89939751	C ₂₄ H ₃₈ O ₄	3801.07429
C ₈ H ₁₈ O	5.07825556	H ₂ O	189.646737
C ₁₆ H ₃₆ O ₄ Ti	0.05976716	C ₁₆ H ₃₅ O ₄ TiN ₄	0.063634
C ₂₄ H ₃₈ O ₄	3801.07429	C ₈ H ₄ O ₃	2.890699
H ₂ O	189.643573	C ₈ H ₆ O ₄	2.89939751
Total	#####	Total	4001.65301
2. Larutan NaOH 0.4%			
Komponen	Massa (Kg/Jam)		
NaOH _(aq)	0.00703143		
Total	0.00703143		
TOTAL	4001.65301	TOTAL	4001.65301

4. Tangki Pengenceran

Fungsi : Untuk mengencerkan NaOH 48% menjadi 0.4%



Kebutuhan NaOH Untuk Proses Netrallizer

$$\text{Massa NaOH} = 0.00703143 \text{ Kg/Jam}$$

$$\rho \text{ Campuran} = 2.13 \text{ Kg/L}$$

Kebutuhan NaOH 48%

$$\frac{10 \times \% \times \rho \times V1}{Mr} = \frac{10 \times \% \times \rho \times V2}{Mr}$$

$$48\% \times V1 = 0.40\% \times V2$$



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Pabrik Diocetyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$V1 = \frac{0.40\% \times V2}{48\%}$$

$$V1 = 2.751E-05 \text{ L/Jam}$$

$$= 5.8595E-05 \text{ Kg/Jam}$$

$$\text{Kebutuhan H}_2\text{O} = 0.00697284 \text{ Kg/Jam}$$

Kebutuhan H₂O Untuk Pengenceran

$$\text{Massa NaOH } 0.4\% = 0.00703143 \text{ Kg/Jam}$$

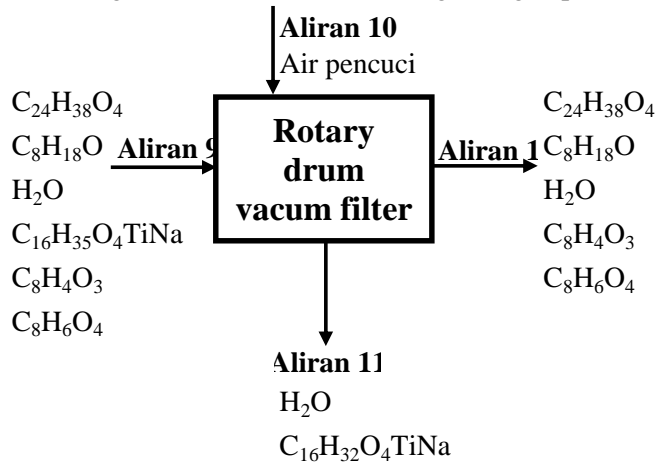
$$\text{Massa NaOH } 48\% = 5.8595E-05 \text{ Kg/Jam}$$

$$\text{Massa H}_2\text{O} = 0.00697284 \text{ Kg/Jam}$$

NERACA MASSA TANGKI PENGECERAN			
Komponen Masuk		Komponen Keluar	
1. NaOH 48%		1. NaOH 0.4%	
Komponen	Massa (Kg/Jam)	Komponen	Massa (Kg/Jam)
NaOH	2.8126E-05	NaOH _(aq)	0.00703143
H ₂ O	3.047E-05	Total	0.00703143
Total	5.8595E-05		
2. Air Proses			
Komponen	Massa (Kg/Jam)		
H ₂ O	0.00697284		
Total	0.00697284		
Total	0.00703143	Total	0.00703143

5. Rotary Drum Vacum Filter

Fungsi : Untuk memisahkan slage dengan produk C₂₄H₃₈O₄





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Pabrik Diocetyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Kondisi operasi

Tekanan = 1 atm
Suhu Masuk = 95 °C
Suhu Keluar = 95 °C
Asumsi H₂O untuk pencuci = ##
Asumsi Liquid yang terikut = 2%

Aliran masuk dari Netralizer

Komponen	Massa (Kg/Jam)
C ₈ H ₁₈ O	5.07825556
C ₂₄ H ₃₈ O ₄	3801.07429
H ₂ O	189.646737
C ₁₆ H ₃₅ O ₄ TiN ₄	0.06363445
C ₈ H ₄ O ₃	#####
C ₈ H ₆ O ₄	2.89939751
Total	4001.65301

Kebutuhan H₂O Pencucian

Air pencuci = ## x 4001.7 kg/jam
= 1400.6 kg/jam
Massa H₂O pencuci pada fi = ## x 1400.6 kg/jam
= 1372.6 kg/jam
Massa H₂O pencuci pada c = 2% x 1400.6 kg/jam
= 28.012 kg/jam

Asumsi komponen 98% padatan dan 2% mengand

Komponen	Massa (Kg/Jam)
C ₁₆ H ₃₅ O ₄ TiN ₄	0.001298662
H ₂ O	0.001298662

C₁₆H₃₅O₄TiN₄ = 0.06363445 kg/jam

H₂O = $\frac{2\%}{98\%} \times 0$ kg/jam
= 0.00129866 kg/jam



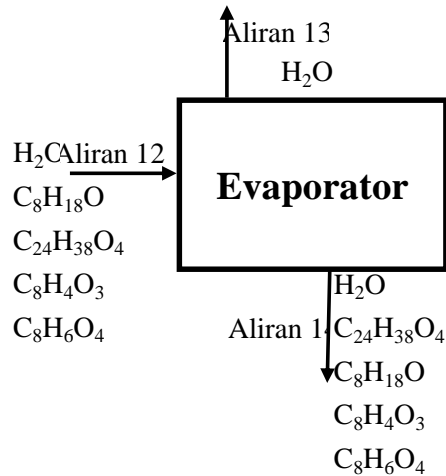
PERANCANGAN PABRIK

Pabrik Diocetyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

PERACA MASSA ROTARY DRUM VACUM FILTER			
Komponen Masuk		Komponen Keluar	
1. Produk Netralizer		Produk Rotary drum vacum filter	
Komponen	Massa (Kg/Jam)	Komponen	Massa (Kg/Jam)
C ₈ H ₁₈ O	5.07825556	C ₈ H ₁₈ O	5.07825556
C ₂₄ H ₃₈ O ₄	3801.07429	C ₂₄ H ₃₈ O ₄	3801.07429
H ₂ O	189.646737	H ₂ O	1562.2124
C ₁₆ H ₃₅ O ₄ TiN ₃	0.06363445	C ₈ H ₄ O ₃	2.89069932
C ₈ H ₄ O ₃	2.8906993	C ₈ H ₆ O ₄	2.89939751
C ₈ H ₆ O ₄	2.89939751	Total	5374.15506
Total	4001.65301		
2. Air Pencuci		2. Aliran Slurry	
Komponen	Massa (Kg/Jam)	Komponen	Massa (Kg/Jam)
H ₂ O	1400.57855	C ₁₆ H ₃₅ O ₄ TiN ₃	0.06363445
Total	1400.57855	H ₂ O	28.0128697
		Total	28.0765042
Total	5402.2316	Total	5402.2316

6. Evaporator

Fungsi : Untuk menguapkan H₂O





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Pabrik Diethyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

Aliran masuk dari rotary drum vacuum filter

Komponen	Massa (Kg/Jam)
$C_8H_{18}O$	5.07825556
$C_{24}H_{38}O_4$	3801.07429
H_2O	1562.21242
$C_8H_4O_3$	2.89069932
$C_8H_6O_4$	2.89939751

Aliran keluaran dari kondensor

Asumsi H_2O 99,99% terpisahkan

$$\begin{aligned} H_2O &= \frac{100}{100} \times 1562.21 \text{ kg/jam} \\ &= 1562.06 \text{ kg/jam} \end{aligned}$$

Aliran menuju tangki dekolorisasi

Asumsi $C_{24}H_{38}O_4$ 99,8%, sedangkan 0,2% mengandung H_2O

$$\begin{aligned} C_{24}H_{38}O &= \frac{100}{100} \times 3801.1 \text{ kg/jam} \\ &= 3793.47 \text{ kg/jam} \end{aligned}$$

$$\begin{aligned} H_2O &= \frac{0.2}{100} \times 3793.47 \text{ kg/jam} \\ &= 7.6021 \text{ kg/jam} \\ &= 7.7584 \text{ kg/jam} \end{aligned}$$

$$\begin{aligned} C_8H_{18}O &= \frac{100}{100} \times 5.07826 \text{ kg/jam} \\ &= 5.0783 \text{ kg/jam} \end{aligned}$$

$$\begin{aligned} C_8H_4O_3 &= \frac{100}{100} \times 2.89070 \text{ kg/jam} \\ &= 2.8907 \text{ kg/jam} \end{aligned}$$

$$\begin{aligned} C_8H_6O_4 &= \frac{100}{100} \times 2.89940 \text{ kg/jam} \\ &= 2.8994 \text{ kg/jam} \end{aligned}$$



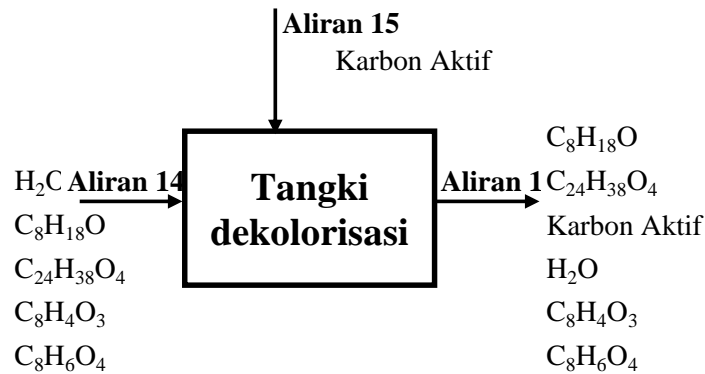
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Pabrik Diocetyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

NERACA MASSA EVAPORATOR			
Komponen Masuk		Komponen Keluar	
Produk rotary drum vacum filter		1. Produk Evaporator	
Komponen	Massa (Kg/Jam)	Komponen	Massa (Kg/Jam)
$C_{24}H_{38}O_4$	#####	$C_8H_{18}O$	5.0782556
$C_8H_{18}O$	#####	$C_{24}H_{38}O_4$	3793.4721
H_2O	#####	H_2O	7.7583698
$C_8H_4O_3$	2.890699	$C_8H_4O_3$	2.8906993
$C_8H_6O_4$	2.899398	$C_8H_6O_4$	2.8993975
Total	5374.1551	Total	3812.09886
		2. Aliran menuju kondensator	
		Komponen	Massa (Kg/Jam)
		H_2O	1562.0562
		Total	1562.0562
Total	5374.1551	Total	5374.1551

7. Tangki Dekolorisasi

Fungsi : Untuk menghilangkan warna pada produk dengan menambahkan karbon aktif



Aliran masuk dari evaporator

$C_8H_{18}O$	5.0782556
$C_{24}H_{38}O_4$	3793.4721
H_2O	7.7583698
$C_8H_4O_3$	2.8906993
$C_8H_6O_4$	2.8993975
Total	#####



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Pabrik Diocetyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Penambahan karbon aktif sebanyak 0.05% dari total massa produk

$$\begin{aligned}\text{Karbon aktif} &= \frac{0.05}{100} \times 3812.10 \text{ kg/jam} \\ &= 1.906 \text{ kg/jam}\end{aligned}$$

Produk yang hilang karena proses adsorpsi sebesar 0,2-0,5% dari total massa produk

$$\begin{aligned}\text{Karbon Aktif} &= 0.2\% \times 3812.10 \text{ kg/jam} \\ &= 7.6242 \text{ kg/jam} \\ \text{Massa total} &= 7.6242 \text{ kg/jam} + 1.9060 \text{ kg/jam} \\ &= 9.5302 \text{ kg/jam}\end{aligned}$$

$$\begin{aligned}\text{C}_{24}\text{H}_{38}\text{O}_4 \text{ teradsor} &= 0.2\% \times 3793.47 \text{ kg/jam} \\ &= 7.5869 \text{ kg/jam}\end{aligned}$$

$$\begin{aligned}\text{C}_{24}\text{H}_{38}\text{O}_4 \text{ total} &= 3793.47 \text{ kg/jam} - 7.5869 \text{ kg/jam} \\ &= 3785.89 \text{ kg/jam}\end{aligned}$$

$$\begin{aligned}\text{C}_8\text{H}_{18}\text{O} \text{ teradsorp} &= 0.2\% \times 5.07826 \text{ kg/jam} \\ &= 0.0102 \text{ kg/jam}\end{aligned}$$

$$\begin{aligned}\text{C}_8\text{H}_{18}\text{O} \text{ total} &= 5.078 \text{ kg/jam} - 0.0102 \text{ kg/jam} \\ &= 5.068 \text{ kg/jam}\end{aligned}$$

$$\begin{aligned}\text{C}_8\text{H}_4\text{O}_3 \text{ teradsorp} &= 0.2\% \times 2.8907 \text{ kg/jam} \\ &= 0.0058 \text{ kg/jam}\end{aligned}$$

$$\begin{aligned}\text{C}_8\text{H}_4\text{O}_3 \text{ total} &= 2.891 \text{ kg/jam} - 0.0058 \text{ kg/jam} \\ &= 2.885 \text{ kg/jam}\end{aligned}$$

$$\begin{aligned}\text{C}_8\text{H}_6\text{O}_4 \text{ teradsorp} &= 0.2\% \times 2.89940 \text{ kg/jam} \\ &= 0.0058 \text{ kg/jam}\end{aligned}$$

$$\begin{aligned}\text{C}_8\text{H}_6\text{O}_4 \text{ total} &= 2.899 \text{ kg/jam} - 0.0058 \text{ kg/jam} \\ &= 2.894 \text{ kg/jam}\end{aligned}$$

$$\begin{aligned}\text{H}_2\text{O} \text{ teradsorpsi} &= 0.2\% \times 7.76 \text{ kg/jam} \\ &= 0.0155 \text{ kg/jam}\end{aligned}$$

$$\begin{aligned}\text{H}_2\text{O} \text{ total} &= 7.758 \text{ kg/jam} - 0.0155 \text{ kg/jam} \\ &= 7.743 \text{ kg/jam}\end{aligned}$$



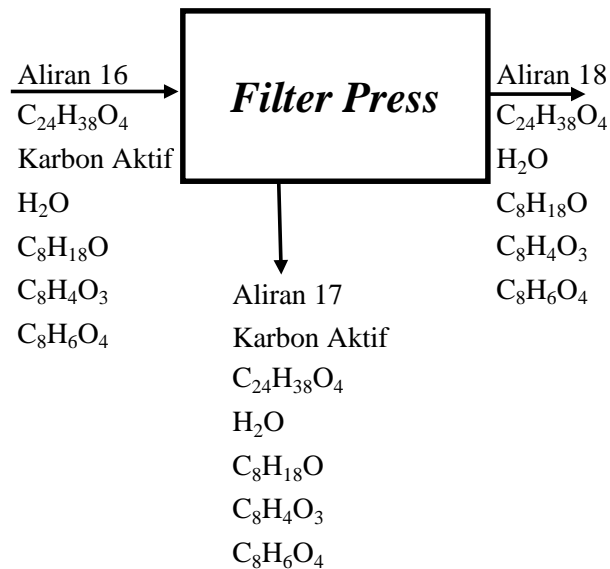
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NERACA MASSA TANGKI DEKOLORISASI			
Komponen Masuk		Komponen Keluar	
1. Produk evaporator		. Produk tangki dekolorisasi	
Komponen	Massa (Kg/Jam)	Komponen	Massa (Kg/Jam)
$C_{24}H_{38}O_4$	3793.4721	$C_8H_{18}O$	5.0681
$C_8H_{18}O$	5.0783	$C_{24}H_{38}O_4$	3785.8852
$C_8H_4O_3$	2.8906993	$C_8H_4O_3$	2.8849179
$C_8H_6O_4$	2.8993975	$C_8H_6O_4$	2.8935987
H_2O	7.7584	H_2O	7.7428531
Total	3812.0989	Karbon aktif	9.5302472
		Total	3814.0049
2. Aliran karbon aktif			
Komponen	Massa (Kg/Jam)		
Karbon aktif	1.90604943		
Total	1.90604943		
Total	3814.0049	Total	3814.0049

8. Filter Press

Fungsi : Untuk memisahkan produk dari karbon aktif





PERANCANGAN PABRIK

Pabrik Diocetyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Aliran masuk dari tangki dekolorisasi

Komponen	Massa (kg/jam)
C ₈ H ₁₈ O	5.0681
C ₂₄ H ₃₈ O ₄	3785.8852
C ₈ H ₄ O ₃	2.8849179
C ₈ H ₆ O ₄	2.8935987
H ₂ O	7.7428531
Karbon aktif	9.5302472

Aliran keluaran berupa cake dari *filter press*

Asumsi efisiensi pemisahan pada *filter press* 99,5%.

Jumlah filtrat yang tertahan dalam *cake* sebesar 0,5% dari jumlah padatan yang tertahan

$$\begin{aligned} \text{C}_{24}\text{H}_{38}\text{O}_4 &= \frac{0.5}{\#\#} \times 3785.9 \text{ kg/jam} \\ &= 18.9294 \text{ kg/jam} \end{aligned}$$

$$\begin{aligned} \text{C}_8\text{H}_{18}\text{O} &= \frac{0.5}{\#\#} \times 5.06810 \text{ kg/jam} \\ &= 0.0253 \text{ kg/jam} \end{aligned}$$

$$\begin{aligned} \text{C}_8\text{H}_4\text{O}_3 &= \frac{0.5}{\#\#} \times 2.88492 \text{ kg/jam} \\ &= 0.0144 \text{ kg/jam} \end{aligned}$$

$$\begin{aligned} \text{C}_8\text{H}_6\text{O}_3 &= \frac{0.5}{\#\#} \times 2.89360 \text{ kg/jam} \\ &= 0.0145 \text{ kg/jam} \end{aligned}$$

$$\begin{aligned} \text{H}_2\text{O} &= \frac{0.5}{\#\#} \times 7.74285 \text{ kg/jam} \\ &= 0.0387 \text{ kg/jam} \end{aligned}$$

$$\begin{aligned} \text{Karbon Aktif} &= \frac{\#\#}{\#\#} \times 9.53025 \text{ kg/jam} \\ &= 9.5302 \text{ kg/jam} \end{aligned}$$

Aliran keluaran filtrat masuk tangki penyimpanan

$$\begin{aligned} \text{C}_{24}\text{H}_{38}\text{O} &= 3785.89 - 18.9294 \\ &= 3766.96 \text{ kg/jam} \end{aligned}$$

$$\text{C}_8\text{H}_{18}\text{O} = 5.0681 - 0.0253$$



PERANCANGAN PABRIK

Pabrik Diocetyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$= 5.0428 \text{ kg/jam}$$

$$\begin{aligned} \text{C}_8\text{H}_4\text{O}_3 &= 2.8849 - 0.0144 \\ &= 2.8705 \text{ kg/jam} \end{aligned}$$

$$\begin{aligned} \text{C}_8\text{H}_6\text{O}_4 &= 2.8936 - 0.0145 \\ &= 2.8791 \text{ kg/jam} \end{aligned}$$

$$\begin{aligned} \text{H}_2\text{O} &= 7.7429 - 0.0387 \\ &= 7.7041 \text{ kg/jam} \end{aligned}$$

NERACA MASSA <i>FILTER PRESS</i>			
Komponen Masuk		Komponen Keluar	
. Produk tangki dekolourisasi		1. Aliran cake	
Komponen	Massa (Kg/Jam)	Komponen	Massa (Kg/Jam)
$\text{C}_{24}\text{H}_{38}\text{O}_4$	3785.8852	$\text{C}_{24}\text{H}_{38}\text{O}_4$	18.9294
$\text{C}_8\text{H}_{18}\text{O}$	5.0681	$\text{C}_8\text{H}_{18}\text{O}$	0.02534
$\text{C}_8\text{H}_4\text{O}_3$	2.8849	$\text{C}_8\text{H}_4\text{O}_3$	0.01442
$\text{C}_8\text{H}_6\text{O}_4$	2.8936	$\text{C}_8\text{H}_6\text{O}_4$	0.01447
H_2O	7.7429	H_2O	0.03871
Karbon Aktif	9.5302	Karbon Aktif	9.53025
Total	3814.0049	Total	28.5526
		2. Produk <i>filter press</i>	
		Komponen	Massa (Kg/Jam)
		$\text{C}_{24}\text{H}_{38}\text{O}_4$	3766.9558
		$\text{C}_8\text{H}_{18}\text{O}$	5.0428
		$\text{C}_8\text{H}_4\text{O}_3$	2.8705
		$\text{C}_8\text{H}_6\text{O}_4$	2.8791
		H_2O	7.7041
		Total	3785.4523
Total	3814.0049	Total	3814.0049



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

APPENDIX B NERACA PANAS

$$\begin{aligned} \text{Kapasitas produksi} &= 30000 \text{ ton/jam} \\ &= 30000 \times 1000 \text{ kg/tahun} \\ &= 30000000 \text{ kg/tahun} \\ &= \frac{30000000}{330} \text{ kg/hari} \\ &= 90909.09091 \text{ kg/hari} \\ &= \frac{90909.09091}{24} \text{ kg/jam} \\ &= 3787.878788 \text{ kg/jam} \\ \text{Waktu operasi} &= 330 \text{ hari/tahun} \\ &= 24 \text{ jam/hari} \end{aligned}$$

Kapasitas panas gas (Himmelblau, 2004 : 692) :

$$C_p = A + BT + CT^2 + DT^3 + ET^4$$

$$\text{Sehingga, } \int C_p dT = AT + \frac{B}{2}T^2 + \frac{C}{3}T^3 + \frac{D}{4}T^4 + \frac{E}{5}T^5$$

Kapasitas panas *liquid* (Himmelblau, 2004 : 692) :

$$C_p = A + BT + CT^2 + DT^3$$

$$\text{Sehingga, } \int C_p dT = AT + \frac{B}{2}T^2 + \frac{C}{3}T^3 + \frac{D}{4}T^4$$

Kapasitas panas *solid* (Himmelblau, 2004 : 692) :

$$C_p = A + BT + CT^2$$

$$\text{Sehingga, } \int C_p dT = AT + \frac{B}{2}T^2 + \frac{C}{3}T^3$$

keterangan :

$$\begin{aligned} C_p &= \text{kapasitas panas (kJ/mol.K)} \\ A, B, C, D, E &= \text{konstanta} \\ T &= \text{suhu (K)} \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Tabel B.1 Data Berat Molekul masing-masing Komponen

Komponen	BM (kg/kmol)
C ₈ H ₄ O ₃	148.1140
C ₈ H ₁₈ O	130.2257
H ₂ O	18.0147
C ₁₆ H ₃₆ O ₄ Ti	340.3164
C ₂₄ H ₃₈ O ₄	390.5507
C ₈ H ₆ O ₄	166.1286
NaOH	39.9966

Tabel B.2 Data Kapasitas Panas *Liquid, solid* dan gas (J/mol.K) :

Komponen	A	B	C	D	E	Literatur
C ₈ H ₄ O _{3(l)}	-105.6270	1.98400	-0.00388	2.851E-06	-	(Yaws, 1999)
C ₈ H ₄ O _{3(s)}	26.3200	0.39060	0.00021	-	-	(Yaws, 1999)
C ₈ H ₁₈ O _(l)	-105.6270	0.97903	-0.00262	3.104E-06	-	(Yaws, 1999)
C ₁₆ H ₃₆ O ₄ Ti _(l)	Cp=	711.00000				(NIST, 2019)
C ₂₄ H ₃₈ O _{4(l)}	370.5240	1.98040	-0.00416	4.28E-06	-	(Yaws, 1999)
NaOH _(l)	87.6390	-0.00048	0.00000	1.19E-09	-	(Yaws, 1999)
H ₂ O _(g)	33.93300	-0.00842	0.000030	-1.78E-08	3.69E-12	(Yaws, 1999)
H ₂ O _(l)	92.0530	-0.03995	-0.00021	5.35E-07	-	(Yaws, 1999)
H ₂ O _(s)	9.6590	0.0750	-0.000016	-	-	(Yaws, 1999)
C ₈ H ₆ O _{4(l)}	-561.0000	3.46000	-0.00461	2.28E-06	-	(Yaws, 1999)
C ₈ H ₆ O _{4(s)}	22.0000	0.55600	0.00000	-	-	(Yaws, 1999)

Konversi Data Kapasitas Panas *Liquid, solid* dan gas (kkal/mol.K) :

Komponen	A	B	C	D	E
C ₈ H ₄ O _{3(l)}	-0.0252	0.0004742	-9.28E-07	6.815E-10	-
C ₈ H ₄ O _{3(s)}	0.0063	9.335E-05	5.081E-08	-	-
C ₈ H ₁₈ O _(l)	-0.0252	0.000234	-6.26E-07	7.419E-10	-
C ₁₆ H ₃₆ O ₄ Ti _(l)	Cp=	0.00004			
C ₂₄ H ₃₈ O _{4(l)}	0.08856	0.0004733	-9.94E-07	1.02E-09	-
NaOH _(l)	0.02095	1.156E-07	-1.09E-09	2.84E-13	-
H ₂ O _(g)	0.00811	-2.01E-06	7.148E-09	-4.26E-12	8.83E-16
H ₂ O _(l)	0.02200	-9.55E-06	-5.04E-08	1.28E-10	-
H ₂ O _(s)	0.00231	1.791E-05	-3.72E-09	-	-
C ₈ H ₆ O _{4(l)}	-0.13408	0.0008269	-1.1E-06	5.45E-10	-
C ₈ H ₆ O _{4(s)}	0.00526	0.0001329	0	-	-



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

Tabel B.4 Data Entalpi Pembentukan

Komponen	ΔH_f (kJ/mol)	ΔH_f (kkal/kmol)	
$C_8H_4O_3$	-393.1300	-93960.4288	(Yaws, 1999)
$C_8H_{18}O$	-365.3000	-87308.8918	(Yaws, 1999)
$C_{16}H_{36}O_4Ti$	-1668.3000	-398733.7098	(NIST, 2019)
$C_{24}H_{38}O_4$	-966.7200	-231051.8803	(Yaws, 1999)
$C_8H_6O_4$	-663.3300	-158539.8500	(Yaws, 1999)
NaOH	-423.1200	-101128.2187	(Chase, 1998)
H_2O	-240.5600	-57495.2834	(Yaws, 1999)

Komponen	ΔH_v (kJ/gmol)
$C_8H_4O_3$	74.1900
$C_8H_{18}O$	48.6800
$C_{16}H_{36}O_4Ti$	47.6000
C_3H_8O	39.8700
$C_{24}H_{38}O_4$	66.1300
$C_8H_6O_4$	74.1900
NaOH	200.6700
H_2O	39.5000

Tabel B.5 Entalpi Penguapan (ΔH_v) ((kJ/gmol))

Komponen	T_c (K)	T_b (K)	n	A	Literatur
$C_8H_4O_3$	800.0000	598.000	0.3740	124.140	(Yaws, 1999)
$C_8H_{18}O$	640.2500	457.750	0.6610	111.600	(Yaws, 1999)
$C_{16}H_{36}O_4Ti$	ΔH_v		47.6000		(NIST, 2019)
C_3H_8O	508.31	355.410	0.3260	58.9830	(Yaws, 1999)
$C_{24}H_{38}O_4$	806.0000	657.150	0.4680	145.786	(Yaws, 1999)
$C_8H_6O_4$	800.0000	598.000	0.3740	124.140	(Yaws, 1999)
NaOH	2820.0000	1663.15	0.3800	281.536	(Yaws, 1999)
H_2O	647.1300	373.150	0.3210	52.0530	(Yaws, 1999)

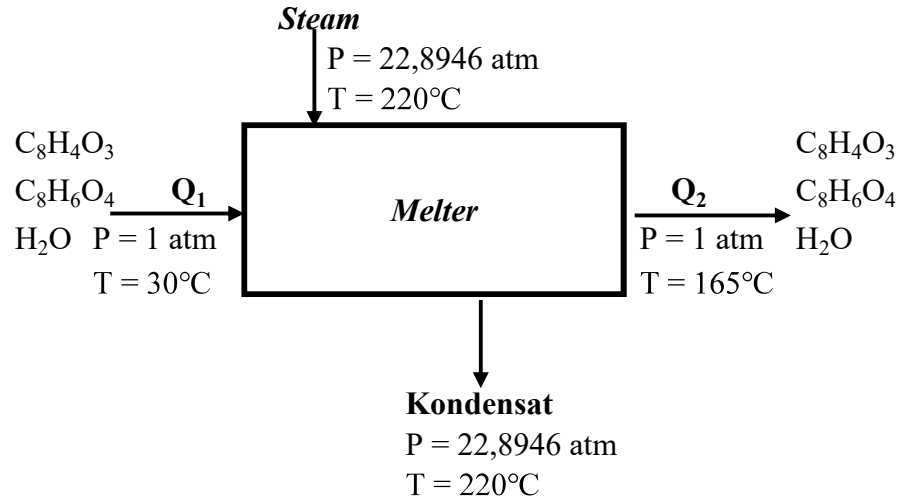


PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraethyl Titanate Melalui Proses Esterifikasi

1. Melter $C_8H_4O_3$

Fungsi : mengubah fase $C_8H_4O_3(s)$ ke fase $C_8H_4O_3(l)$



Panas masuk Melter

Tekanan	=	1	atm
Suhu Operasi	=	30	°C
	=	303.15	K
Treff	=	25	°C
	=	298.15	K

$$\int_{CpdT} C_8H_4O_3 = AT + \frac{B}{2}T^2 + \frac{C}{3}T^3$$

$$= 0.0314524 + 0.1403335 + 0.0229648$$

$$= 0.1947507 \text{ kkal/kmol.K}$$

$$\Delta H = n \times \int_{CpdT}$$

$$= 9.7583621 \times 1.948E-01$$

$$= 1.90 \text{ kkal}$$

$$\int_{CpdT} C_8H_6O_4 = AT + \frac{B}{2}T^2 + \frac{C}{3}T^3$$

$$= 0.02629 + 0.1997579 + 0.0$$

$$= 0.2 \text{ kkal/kmol.K}$$

$$\Delta H = n \times \int_{CpdT}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$= 0.0174527 \times 2.3E-01$$

$$= 0.0039 \text{ kkal}$$

$$\int_{CpdT} H_2O = AT + \frac{B}{2}T^2 + \frac{C}{3}T^3$$

$$= 0.0115425 + 0.0269296 + 0.0$$

$$= 0.03679 \text{ kkal/kmol.K}$$

$$\Delta H = n \times \int_{CpdT}$$

$$= 0.0804732 \times 3.7E-02$$

$$= 0.00296 \text{ kkal}$$

Enthalpi Masuk

Komponen	Massa	n	\int_{CpdT}	ΔH
	(kg)	(kmol)	(kkal/kmol.K)	(kkal)
C ₈ H ₄ O ₃	1445.34966	9.758362143	0.1947507	1.9004483
C ₈ H ₆ O ₄	2.899397513	0.017452725	0.2260479	0.0039452
H ₂ O	1.449698756	0.080473189	0.0367887	0.0029605
Total				1.907354

Panas keluar Melter

Tekanan = 1 atm

Suhu Operasi = 165 °C

= 438.15 K

Treff = 25 °C

= 298.15 K

$$\int_{CpdT} C_8H_4O_3 = AT + \frac{B}{2}T^2 + \frac{C}{3}T^3 + \frac{D}{4}T^4$$

$$= -3.534279 + 24.4 + -17.82935 + 4.9325015$$

$$= 8.0083796 \text{ kkal/kmol.K}$$

$$\Delta H = n \times \int_{CpdT}$$

$$= 9.7583621 \times 8.0E+00$$

$$= 78.15 \text{ kkal}$$

$$\int_{CpdT} C_8H_6O_4 = AT + \frac{B}{2}T^2 + \frac{C}{3}T^3 + \frac{D}{4}T^4$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$= -18.77106 + 42.6 + -21.15821 + 3.9442021$$

$$= 6.636248 \text{ kkal/kmol.K}$$

$$\Delta H = n \times \int C_{pd} T$$

$$= 0.0174527 \times 6.6E+00$$

$$0.12 \text{ kkal}$$

$$\int C_{pd} T \text{ H}_2\text{O} = AT + \frac{B}{2} T^2 + \frac{C}{3} T^3 + \frac{D}{4} T^4$$

$$= 3.0800934 + -0.492153 + -21.15821 + 0.9249673$$

$$= -17.6453 \text{ kkal/kmol.K}$$

$$\Delta H = n \times \int C_{pd} T$$

$$= 0.0804732 \times -1.8E+01$$

$$-1.42 \text{ kkal}$$

Enthalpi Masuk

Komponen	Massa	n	$\int C_{pd} T$	ΔH
	(kg)	(kmol)	(kkal/kmol.K)	(kkal)
C ₈ H ₄ O ₃	1445.34966	9.758362143	8.0083796	78.148668
C ₈ H ₆ O ₄	2.899397513	0.017452725	6.6362480	0.1158206
H ₂ O	1.449698756	0.080473189	-17.6453010	-1.419974
Total				76.844515

Neraca Energi Total :

$$\Delta H \text{ Bahan Masuk} + Q_{\text{supply}} = \Delta H \text{ Bahan Keluar} + Q_{\text{loss}}$$

Asumsi :

$$Q_{\text{loss}} = 5\% Q_{\text{supply}}$$

$$\Delta H \text{ Bahan Masuk} + Q_{\text{supply}} = \Delta H \text{ Bahan Keluar} + Q_{\text{loss}}$$

$$1.907353978 + Q_{\text{supply}} = 76.84451506 + Q_{\text{loss}}$$

$$95\% Q_{\text{supply}} = 74.93716108 \text{ kkal/Jam}$$

$$Q_{\text{supply}} = 78.88122219 \text{ kkal/Jam}$$

$$Q_{\text{loss}} = 3.94406111 \text{ kkal/Jam}$$

(Ulrich : hal.432)

Kebutuhan steam:

Dipakai steam pada tekanan steam 22,9 atm dengan suhu steam 220°C

$$H_L = 943.7 \text{ kJ/kg} \quad (\text{Smith : Steam table F-1})$$



PERANCANGAN PABRIK

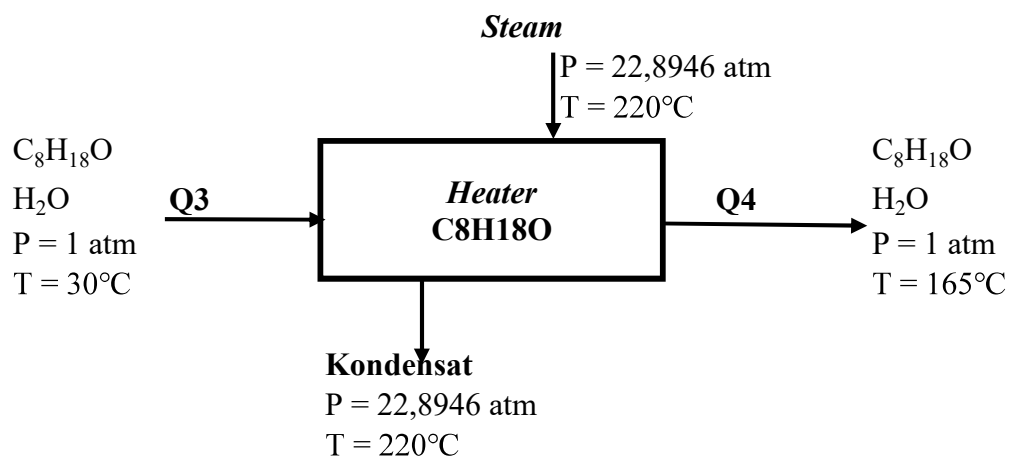
Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
 H_v &= 2799.9 \text{ kJ/kg} && (\text{Smith : Steam table F-1}) \\
 \lambda_{\text{steam}} &= H_v - HL \\
 &= 2799.9 - 943.7 \\
 &= 1856.2 \text{ kJ/kg} \\
 &= 443.64294 \text{ kkal/kg} \\
 Q_{\text{steam}} &= M_{\text{steam}} \times \lambda_{\text{steam}} \\
 M_{\text{steam}} &= \frac{Q_{\text{steam}}}{\lambda_{\text{steam}}} \\
 &= \frac{78.88122219 \text{ kkal/jam}}{443.6429372 \text{ kkal/kg}} \\
 &= 0.177803399 \text{ kg/jam}
 \end{aligned}$$

NERACA PANAS MELTER			
Panas Masuk		Panas Keluar	
Komponen	ΔH (kkal)	Komponen	ΔH (kkal)
$C_8H_4O_3$	1.900448321	$C_8H_4O_3$	78.1486681
$C_8H_6O_4$	0.003945151	$C_8H_6O_4$	0.115820613
H_2O	0.002960506	H_2O	-1.419973647
Q_{supply}	78.88122219	Q_{loss}	3.94406111
Total	80.78857617	Total	80.78857617

2. Heater $C_8H_{18}O$

Fungsi : Untuk memanaskan $C_8H_{18}O$ sebelum masuk kedalam reaktor





PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

Panas masuk *Heater*

$$\begin{aligned} \text{Tekanan} &= 1 \text{ atm} \\ \text{Suhu Operasi} &= 30 \text{ }^\circ\text{C} \\ &= 303.15 \text{ K} \\ \text{Treff} &= 25 \text{ }^\circ\text{C} \\ &= 298.15 \text{ K} \\ \text{Cp C}_8\text{H}_{18}\text{O (l)} &= A + BT + CT^2 + DT^3 \\ &= -105.627 + 297 + -240.5664 + 8.6E+01 \\ &= 37.080858 \text{ J/mol.K} \\ &= 37.080858 \times 0.24 \\ &= 8.8625475 \text{ kal/mol.k} \\ &= 8.8625475 \text{ kkal/kmol.k} \\ &= \frac{\text{Cp}}{\text{BM C}_8\text{H}_{18}\text{O}} \\ &= \frac{8.862547494 \text{ kkal/kmol.k}}{130.2257 \text{ kg/kmol}} \\ &= \mathbf{0.068055277 \text{ kkal/kg.K}} \\ \Delta H &= m \times \text{cp} \times \Delta T \\ &= 2539.127781 \times 0.068055277 \times 5 \\ &= 864.005227 \text{ kkal} \\ \text{Cp H}_2\text{O (l)} &= A + BT + CT^2 + DT^3 \\ &= 92.053 + -12 + -19.39364 + 1.49E+01 \\ &= 7.54E+01 \text{ J/mol.K} \\ &= 75.443783 \times 0.24 \\ &= 18.031517 \text{ kal/mol.k} \\ &= 18.031517 \text{ kkal/kmol.k} \\ &= \frac{18.03151677 \text{ kkal/kmol.k}}{18.0147 \text{ kg/kmol}} \\ &= \mathbf{1.000934614 \text{ kkal/kg.K}} \\ \Delta H &= m \times \text{cp} \times \Delta T \\ &= 12.75943609 \times 1.000934614 \times 5 \\ &= 63.85680616 \text{ kkal} \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Enthalpi Masuk

Komponen	Massa (kg)	Cp (kkal/kg.k)	ΔT (K)	ΔH
				(kkal)
C ₈ H ₁₈ O	2539.127781	0.068055277	5	864.00523
H ₂ O	12.75943609	1.000934614	5	63.856806
Total				927.86203

Panas keluar Heater

Tekanan = 1 atm
 Suhu Operasi = 165 °C
 = 438.15 K
 Treff = 25 °C
 = 298.15 K

$$\begin{aligned}
 C_p \text{ C}_8\text{H}_{18}\text{O} \text{ (l)} &= A + BT + CT^2 + DT^3 \\
 &= -105.627 + 429 + -502.5341 + 2.6E+02 \\
 &= 81.907707 \text{ J/mol.K} \\
 &= 81.907707 \times 0.24 \\
 &= 19.576433 \text{ kal/mol.k} \\
 &= 19.576433 \text{ kkal/kmol.k} \\
 &= \frac{C_p}{\text{BM C}_8\text{H}_{18}\text{O}} \\
 &= \frac{19.57643347 \text{ kkal/kmol.k}}{130.2257 \text{ kg/kmol}} \\
 &= \mathbf{0.150326936 \text{ kkal/kg.K}}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H &= m \times c_p \times \Delta T \\
 &= 2539.127781 \times 0.150326936 \times 140 \\
 &= 53437.9019 \text{ kkal}
 \end{aligned}$$

$$\begin{aligned}
 C_p \text{ H}_2\text{O} \text{ (l)} &= A + BT + CT^2 + DT^3 \\
 &= 92.053 + -17.50541 + -40.51257 + 4.50E+01 \\
 &= 7.90E+01 \text{ J/mol.K} \\
 &= 79.009951 \times 0.24 \\
 &= 18.883852 \text{ kal/mol.k} \\
 &= 18.883852 \text{ kkal/kmol.k} \\
 &= \frac{18.88385237 \text{ kkal/kmol.k}}{18.0147 \text{ kg/kmol}} \\
 &= \mathbf{1.048248005 \text{ kkal/kg.K}}
 \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned} \Delta H &= m \times c_p \times \Delta T \\ &= 12.75943609 \times 1.048248005 \times 140 \\ &= 1872.507479 \text{ kkal} \end{aligned}$$

Enthalpi keluar

Komponen	Massa (kg)	Cp (kkal/kg.k)	ΔT (K)	ΔH (kkal)
C ₈ H ₁₈ O	2539.127781	0.150326936	140	53437.902
H ₂ O	12.75943609	1.048248005	140	1872.5075
Total				55310.409

Neraca Energi Total :

$$\begin{aligned} \Delta H_{\text{bahan masuk}} + Q_{\text{supply}} &= \Delta H_{\text{bahan keluar}} + Q_{\text{loss}} \\ 927.8620332 + Q_{\text{supply}} &= 55310.40938 + 5\% Q_{\text{supply}} \\ Q_{\text{supply}} &= 57244.78668 \text{ Kkal/jam} \\ Q_{\text{loss}} &= 2862.239334 \text{ Kkal/jam} \end{aligned}$$

(Ulrich : hal.432)

Kebutuhan steam:

Dipakai steam pada tekanan steam 22,9 atm dengan suhu steam 220°C

$$\begin{aligned} \lambda_{\text{steam}} &= 1856.2 \text{ kj/kg} \quad (\text{Smith : Steam table F-1}) \\ &= 443.6318 \text{ kkal/kg} \\ Q_{\text{steam}} &= M_{\text{steam}} \times \lambda \\ M_{\text{steam}} &= \frac{Q_{\text{steam}}}{\lambda_{\text{steam}}} \\ &= \frac{57244.78668}{443.6318} \\ &= 129.0367072 \text{ kg/jam} \end{aligned}$$

NERACA PANAS HEATER			
Panas Masuk		Panas Keluar	
Komponen	ΔH (kkal)	Komponen	ΔH (kkal)
C ₈ H ₁₈ O	864.005227	C ₈ H ₁₈ O	53437.9019
H ₂ O	63.85680616	H ₂ O	1872.507479
Qsupply	57244.78668	Qloss	2862.239334
Total	58172.64871	Total	58172.64871

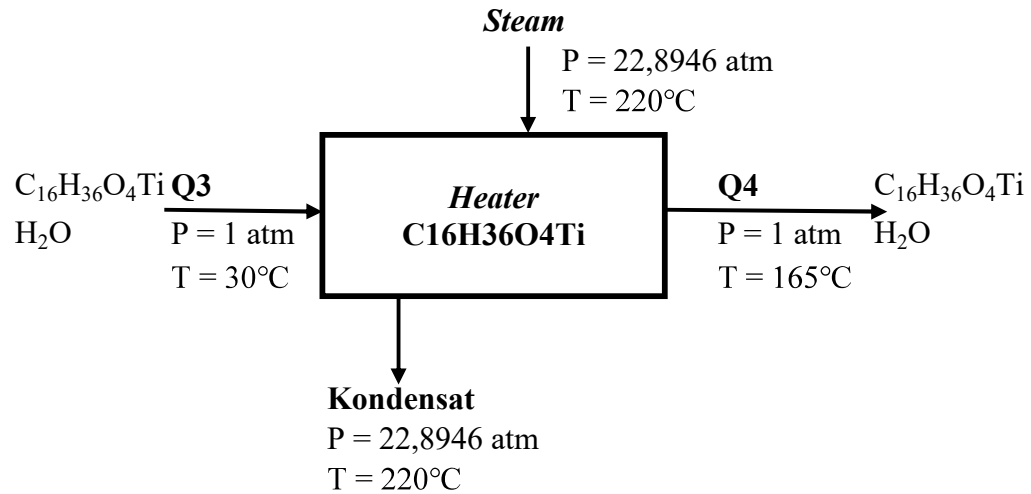


PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

3. Heater $C_{16}H_{36}O_4Ti$

Fungsi : Untuk memanaskan $C_{16}H_{36}O_4Ti$ sebelum masuk kedalam reaktor



Panas masuk Heater

Tekanan	=	1	atm	
Suhu Operasi	=	30	°C	
	=	303.15	K	
Treff	=	25	°C	
	=	298.15	K	
$C_p C_{16}H_{36}O_4Ti$	=	711.00000	J/mol.K	
	=	711.0000	x 0.24	
	=	169.93327	kal/mol.k	
	=	169.93327	kkal/kmol.k	
	=	$\frac{C_p}{BM C_{16}H_{36}O_4Ti}$		
	=	$\frac{169.933266}{340.3164}$	kkal/kmol.k	
	=	0.499338986	kkal/kg.K	
ΔH	=	m	x	cp
	=	0.059767162	x	0.499338986
	=	0.149220369	kkal	x ΔT
				x 5



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
 C_p \text{ H}_2\text{O} &= A + BT + CT^2 + DT^3 \\
 &= 92.053 + -12.11175 + -19.39364 + 1.49E+01 \\
 &= 75.443783 \text{ J/mol.K} \\
 &= 75.443783 \times 0.24 \quad \text{(konversi ke Kal/mol.k)} \\
 &= 18.031517 \text{ kal/mol.k} \\
 &= 18.031517 \text{ kkal/kmol.k} \\
 &= \frac{C_p}{\text{BM H}_2\text{O}} \\
 &= \frac{18.03151677 \text{ kkal/kmol.k}}{18.0147 \text{ kg/kmol}} \\
 &= 1.000934614 \text{ kkal/kg.K}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H &= m \times c_p \times \Delta T \\
 &= 0.000240029 \times 1.000934614 \times 5 \\
 &= 0.001201265 \text{ kkal}
 \end{aligned}$$

Enthalpi Masuk

Komponen	Massa (kg)	Cp (kkal/kg.k)	ΔT (K)	ΔH
				(kkal)
C ₁₆ H ₃₆ O ₄ Ti	0.059767162	0.499338986	5	0.1492204
H ₂ O	0.000240029	1.000934614	5	0.0012013
Total				0.1504216

Panas keluar Heater

$$\begin{aligned}
 \text{Tekanan} &= 1 \text{ atm} \\
 \text{Suhu Operasi} &= 165 \text{ }^\circ\text{C} \\
 &= 438.15 \text{ K} \\
 \text{Treff} &= 25 \text{ }^\circ\text{C} \\
 &= 298.15 \text{ K}
 \end{aligned}$$

$$\begin{aligned}
 C_p \text{ C}_{16}\text{H}_{36}\text{O}_4\text{Ti} &= 711.00000 \text{ J/mol.K} \\
 &= 711.0000 \times 0.24 \quad \text{(konversi ke Kal/mol.k)} \\
 &= 169.93327 \text{ kal/mol.k} \\
 &= 169.93327 \text{ kkal/kmol.k} \\
 &= \frac{C_p}{\text{BM C}_{16}\text{H}_{36}\text{O}_4\text{Ti}} \\
 &= \frac{169.933266 \text{ kkal/kmol.k}}{340.3164 \text{ kg/kmol}}
 \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
 &= \mathbf{0.499338986 \text{ kkal/kg.K}} \\
 \Delta H &= m \times cp \times \Delta T \\
 &= 0.059767162 \times 0.499338986 \times 140 \\
 &= 4.178170341 \text{ kkal} \\
 \\
 C_p \text{ H}_2\text{O} &= A + BT + CT^2 + DT^3 \\
 &= 92.0530 + (-17.50541) + (-40.51257) + 4.50E+01 \\
 &= 79.0100 \text{ J/mol.K} \\
 &= 79.009951 \times \mathbf{0.24} \quad \text{(konversi ke Kal/mol.k)} \\
 &= 18.883852 \text{ kal/mol.k} \\
 &= 18.883852 \text{ kkal/kmol.k} \\
 &= \frac{C_p}{\text{BM H}_2\text{O}} \\
 &= \frac{18.88385237 \text{ kkal/kmol.k}}{18.0147 \text{ kg/kmol}} \\
 &= \mathbf{1.048248005 \text{ kkal/kg.K}} \\
 \Delta H &= m \times cp \times \Delta T \\
 &= 0.000240029 \times 1.048248005 \times 140 \\
 &= 0.035225354 \text{ kkal}
 \end{aligned}$$

Enthalpi keluar

Komponen	Massa (kg)	Cp (kkal/kg.k)	ΔT (K)	ΔH (kkal)
C ₁₆ H ₃₆ O ₄ Ti	0.059767162	0.499338986	140	4.1781703
H ₂ O	0.000240029	1.048248005	140	0.0352254
Total				4.2133957

Neraca Energi Total :

$$\begin{aligned}
 \Delta H_{\text{bahan masuk}} + Q_{\text{supply}} &= \Delta H_{\text{bahan keluar}} + Q_{\text{loss}} \\
 0.150421635 + Q_{\text{supply}} &= 4.213395695 + 0.05 Q_{\text{supply}} \\
 Q_{\text{supply}} &= 4.2768148 \text{ Kkal/jam} \\
 Q_{\text{loss}} &= 0.21384074 \text{ Kkal/jam} \\
 &\text{(Ulrich : hal.432)}
 \end{aligned}$$

Kebutuhan steam:

Dipakai steam pada tekanan steam 22,9 atm dengan suhu steam 220°C

$$\begin{aligned}
 \lambda_{\text{steam}} &= 1856.2 \text{ kj/kg} \quad \text{(Smith : Steam table F-1)} \\
 &= 443.6318 \text{ kkal/kg} \\
 Q_{\text{steam}} &= M_{\text{steam}} \times \lambda_{\text{steam}} \\
 M_{\text{steam}} &= \frac{Q_{\text{steam}}}{\lambda_{\text{steam}}}
 \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

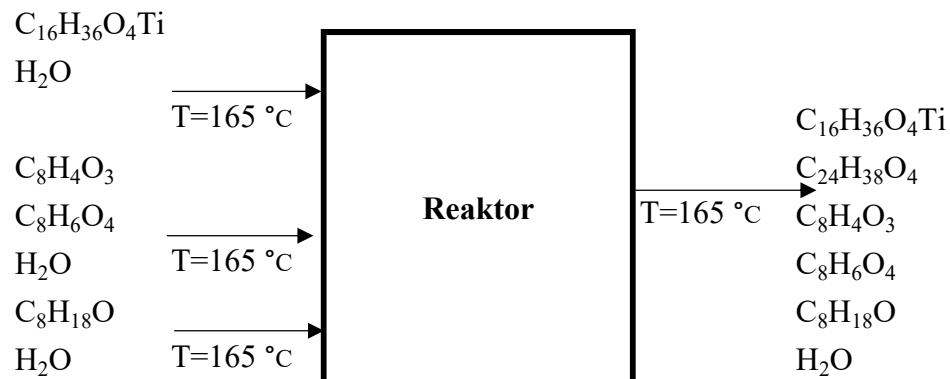
$$= \frac{\lambda_{\text{steam}}}{443.6318}$$

$$= 0.00964046 \quad \text{kg/jam}$$

NERACA PANAS <i>HEATER</i>			
Panas Masuk		Panas Keluar	
Komponen	ΔH (kkal)	Komponen	ΔH (kkal)
C ₈ H ₁₈ O	0.149220369	C ₈ H ₁₈ O	4.178170341
H ₂ O	0.001201265	H ₂ O	0.035225354
Q _{supply}	4.2768148	Q _{loss}	0.21384074
Total	4.427236435	Total	4.427236435

4. Reaktor

Fungsi : Mereaksikan phthalic anhydride dan 2-Ethylhexanol menggunakan katalis tetrabutyl titanate



Enthalpi bahan masuk

Tekanan = 1 atm

$$T_{in} = 165 \text{ } ^\circ\text{C} = 438.15 \text{ K}$$

$$T_{ref} = 25 \text{ } ^\circ\text{C} = 298.15 \text{ K}$$

Panas masuk (phthalic anhydride)



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Komponen	Feed in (kg/jam)	n (kmol/jam)	$\int C_p dT$ (kkal/kmol)	ΔH (kkal/Jam)
C ₈ H ₄ O ₃	1445.35	9.76	8008.58	78150.63
C ₈ H ₆ O ₄	2.89940	0.02	6636.41	0.1158235
H ₂ O	1.44970	0.08	2544.42	204.75770
Total				78355.5

Panas masuk (Katalis tetrabutyl titanate)

Komponen	Feed in (kg/jam)	n (kmol/jam)	$\int C_p dT$ (kkal/kmol)	ΔH (kkal/Jam)
C ₁₆ H ₃₆ O ₄ Ti	0.059767	0.000176	169.933	0.029844
H ₂ O	0.000240	0.000013	2544.421	0.033902
Total				0.0637

Panas masuk (2-Ethyl hexanol)

Komponen	Feed in (kg/jam)	n (kmol/jam)	$\int C_p dT$ (kkal/kmol)	ΔH (kkal/Jam)
C ₈ H ₁₈ O	2539.1278	19.4979	1881.47	36684.63
H ₂ O	12.7594	0.7083	2544.42	1802.1625
Total				38486.8

Panas masuk total = 116842.36

Menghitung Entalpi Panas Reaksi

$$T_{out} = 165 \text{ }^\circ\text{C} = 438 \text{ K}$$

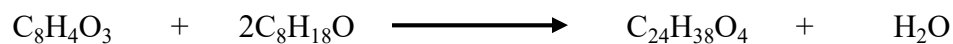
$$T_{ref} = 25 \text{ }^\circ\text{C} = 298 \text{ K}$$

$$\Delta H = H_{out} - H_{in}$$

Untuk menghitung entalpi panas reaksi digunakan persamaan :

$$H_{f,reaksi\ 298K} = \sum(n \times H_f)_{produk} - \sum(n \times H_f)_{reaktan}$$

Reaksi yang terjadi :



Tabel B.18 Entalpi Pembentukan Reaktan pada 298 K

Komponen	Massa (kg/jam)	n (kmol/jam)	H _f (kkal/kmol)	H _{f,Reaktan} (kkal/Jam)
C ₈ H ₄ O ₃	1445.3497	9.7584	-93960.4	-916899.9
C ₈ H ₄ O ₃	2.8994	0.0175	-158539.8	-2767.0



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

C ₈ H ₁₈ O	2539.1278	19.4979	-87308.9	#####
H ₂ O	14.2094	0.7888	-57495.3	-45350.3
Total				#####

Tabel B.19 Entalpi Pembentukan Produk pada 298 K

Komponen	Feed (kg/jam)	n (kmol/jam)	Hf (kkal/kmol)	Hf _{Reaktan} (kkal/Jam)
C ₂₄ H ₃₈ O ₄	3801.0743	9.7326	#####	#####
H ₂ O	189.6436	10.5272	-57495.28	-605263
Total				#####

$$\begin{aligned}
 H_{f_{\text{Reaksi}(298\text{K})}} &= \sum(n \times H_f)_{\text{Produk}} - \sum(n \times H_f)_{\text{Reaktan}} \\
 &= -2853998.2770 - -2667357.0436 \\
 &= -186641.2334 \text{ kkal/jam}
 \end{aligned}$$

Menghitung Entalpi Panas Produk Reaktor

$$T_{\text{out}} = 165 \text{ }^{\circ}\text{C} = 438 \text{ K}$$

$$T_{\text{ref}} = 25 \text{ }^{\circ}\text{C} = 298 \text{ K}$$

Panas Keluar Reaktor

Komponen	Feed in (kg/jam)	n (kmol/jam)	$\int C_p dT$ (kkal/kmol)	ΔH (kkal/Jam)
C ₂₄ H ₃₈ O ₄	3801.0743	9.7326	25107.22	244358.55
C ₈ H ₄ O ₃	2.8907	0.0195	17802.408	347.4447
C ₈ H ₆ O ₄	2.8994	0.0175	15208.09	265.4226
C ₈ H ₁₈ O	5.0783	0.0390	-9226.43	-359.7919
C ₁₆ H ₃₆ O ₄ Ti	0.0598	0.0002	169.933	0.0298
H ₂ O	189.6436	10.5272	7700.68	81066.37
Total				325678.0

$$\begin{aligned}
 H_{f_{\text{reaksi}(438,15\text{K})}} &= (XA \times n + \Delta H_{298}) + (H_{\text{out}} - H_{\text{in}}) \\
 &= -185658.7588 + 208835.6584 \\
 &= 23176.8996 \text{ kkal}
 \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Menghitung Kebutuhan Pendingin

Q yang diterima pendingin

$$\begin{aligned}
 Q_{lepas} &= H_{Reaksi} + H_{Produk} - H_{Reaktan} \\
 &= 23176.8996 + 325678.0212 - 116842.3628 \\
 &= 232012.5580 \text{ kkal}
 \end{aligned}$$

Pendingin menggunakan *cooling water* pada suhu 30°C dan keluar pada suhu 45°C.

$$T_{in} = 30 \text{ }^{\circ}\text{C} = 303.15 \text{ K} = 86 \text{ }^{\circ}\text{F}$$

$$T_{out} = 45 \text{ }^{\circ}\text{C} = 318.15 \text{ K} = 113 \text{ }^{\circ}\text{F}$$

$$C_p = 4.1868 \text{ kJ/kg. }^{\circ}\text{K} \text{ (Smith et al., 2005)}$$

$$= 1.0000004 \text{ kkal/kg.K}$$

$$\begin{aligned}
 m_{pendingin} &= \frac{Q_{lepas}}{C_p \times \Delta T} \\
 &= \frac{232012.5580 \text{ kkal}}{15.000006492 \text{ kkal/kg}} \\
 &= 15467.497 \text{ kg/jam}
 \end{aligned}$$

NERACA PANAS REAKTOR			
Masuk		Keluar	
Komponen	ΔH (kkal)	Komponen	ΔH (kkal)
Dari melter			
C ₈ H ₄ O ₃	78150.63	C ₂₄ H ₃₈ O ₄	244358.5457
C ₈ H ₆ O ₄	0.1158235	C ₈ H ₄ O ₃	347.4446886
H ₂ O	204.75770	C ₈ H ₆ O ₄	265.422643
Dari tangki 2-Ethyl Hexanol		C ₈ H ₁₈ O	-359.7919087
C ₈ H ₁₈ O	36684.63	C ₁₆ H ₃₆ O ₄ Ti	0.029844074
H ₂ O	1802.1625	H ₂ O	81066.37032
Dari tangki Tetrabutyl titanate		ΔH reaksi	23176.89961
C ₁₆ H ₃₆ O ₄ Ti	0.029844	Qserap	-232012.558
H ₂ O	0.033902		
Total	116842.3628	Total	116842.3628

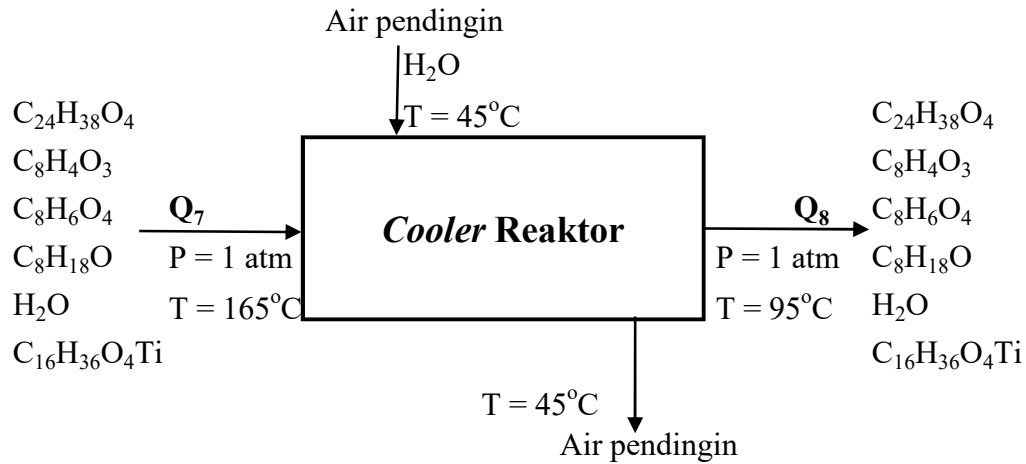


PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetra-butyl Titanate Melalui Proses Esterifikasi

5. Cooler Reaktor

Fungsi : untuk menurunkan suhu produk keluar reaktor



Panas masuk Melter

Tekanan	=	1 atm
Suhu Operasi	=	165 °C
	=	438.15 K
Treff	=	25 °C
	=	298.15 K

$$\begin{aligned}
 C_p \text{ C}_{24}\text{H}_{38}\text{O}_4 &= A + BT + CT^2 + CT^3 \\
 &= 370.5240 + 867.71226 + -798.66 + 360.18 \\
 &= 799.756 \text{ J/mol.K} \\
 &= 799.756 \times 0.24 \\
 &= 191.14658 \text{ kal/mol.k} \\
 &= 191.14658 \text{ kkal/kmol.k} \\
 &= \underline{191.14658 \text{ kkal/kmol.k}}
 \end{aligned}$$

$$= 390.5507 \text{ kg/kmol}$$

$$= \mathbf{0.4894283 \text{ kkal/kg.K}}$$

$$\begin{aligned}
 \Delta H &= m \times c_p \times \Delta T \\
 &= 3801.074288 \times 0.489428302 \times 140 \\
 &= 260449.4668 \text{ kkal}
 \end{aligned}$$

$$\begin{aligned}
 C_p \text{ C}_8\text{H}_4\text{O}_3 &= A + BT + CT^2 + CT^3 \\
 &= -105.6270 + 869.28960 + -745.7669 + -326757.8 \\
 &= -326740 \text{ J/mol.K}
 \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned} &= \text{#####} \times 0.24 \\ &= -78092.79 \text{ kal/mol.k} \\ &= -78092.79 \text{ kkal/kmol.k} \\ &= \frac{-78092.79 \text{ kkal/kmol.k}}{148.1140 \text{ kg/kmol}} \\ &= \mathbf{-527.248 \text{ kkal/kg.K}} \end{aligned}$$
$$\begin{aligned} \Delta H &= m \times cp \times \Delta T \\ &= 2.89069932 \times -527.2480206 \times 140 \\ &= -213376.1693 \text{ kkal} \end{aligned}$$
$$\begin{aligned} \text{Cp C}_8\text{H}_6\text{O}_4 &= A + BT + CT^2 + CT^3 \\ &= -561.000 + 1515.9990 + -885.0067 + 191.7800 \\ &= \text{##### J/mol.K} \\ &= 261.77229 \times 0.24 \\ &= 62.565149 \text{ kal/mol.k} \\ &= 62.565149 \text{ kkal/kmol.k} \\ &= \frac{62.565149 \text{ kkal/kmol.k}}{166.1286 \text{ kg/kmol}} \\ &= \mathbf{0.3766066 \text{ kkal/kg.K}} \end{aligned}$$
$$\begin{aligned} \Delta H &= m \times cp \times \Delta T \\ &= 2.899397513 \times 0.37660664 \times 140 \\ &= 152.8705296 \text{ kkal} \end{aligned}$$
$$\begin{aligned} \text{Cp C}_8\text{H}_{18}\text{O} &= A + BT + CT^2 + CT^3 \\ &= -105.6270 + 428.9620 + -502.5341 + 261.1068 \\ &= 81.9077 \text{ J/mol.K} \\ &= 81.9077 \times 0.24 \\ &= 19.576433 \text{ kal/mol.k} \\ &= 19.576433 \text{ kkal/kmol.k} \\ &= \frac{19.576433 \text{ kkal/kmol.k}}{130.2257 \text{ kg/kmol}} \\ &= \mathbf{0.1503269 \text{ kkal/kg.K}} \end{aligned}$$
$$\begin{aligned} \Delta H &= m \times cp \times \Delta T \\ &= 5.078255562 \times 0.150326936 \times 140 \\ &= 106.8758038 \text{ kkal} \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
 C_p \text{ H}_2\text{O} &= A + BT + CT^2 + CT^3 \\
 &= 92.0530 + -17.5054 + -40.5126 + 44.9749 \\
 &= 79.00995 \text{ J/mol.K} \\
 &= 79.00995 \times 0.24 \\
 &= 18.883852 \text{ kal/mol.k} \\
 &= 18.883852 \text{ kkal/kmol.k} \\
 &= \frac{18.883852}{18.0147} \text{ kkal/kmol.k} \\
 &= \mathbf{1.048248 \text{ kkal/kg.K}} \\
 \Delta H &= m \times c_p \times \Delta T \\
 &= 189.6435728 \times 1.048248005 \times 140 \\
 &= 27831.08955 \text{ kkal}
 \end{aligned}$$

$$\begin{aligned}
 C_p \text{ C}_{16}\text{H}_{36}\text{C} &= 711.00000 \text{ J/mol.K} \\
 &= 711.00000 \times 0.24 \\
 &= 169.93327 \text{ kal/mol.k} \\
 &= 169.93327 \text{ kkal/kmol.k} \\
 &= \frac{169.93327}{340.3164} \text{ kkal/kmol.k} \\
 &= \mathbf{0.499339 \text{ kkal/kg.K}} \\
 \Delta H &= m \times c_p \times \Delta T \\
 &= 0.059767162 \times 0.499338986 \times 140 \\
 &= 4.178170341 \text{ kkal}
 \end{aligned}$$

Enthalpi Masuk

Komponen	Massa (kg)	Cp (kkal/kg.k)	ΔT (K)	ΔH
				(kkal)
C ₂₄ H ₃₈ O ₄	3801.074288	0.489428302	140	260449.47
C ₈ H ₄ O ₃	2.890699320	-527.2480206	140	-213376.2
C ₈ H ₆ O ₄	2.899397513	0.376606664	140	152.87053
C ₈ H ₁₈ O	5.078255562	0.150326936	140	106.8758
H ₂ O	189.6435728	1.048248005	140	27831.09
C ₁₆ H ₃₆ O ₄ Ti	0.059767162	0.499338986	140	4.1781703
Total				75168.312



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetra-butyl Titanate Melalui Proses Esterifikasi

Panas keluar Melter

$$\begin{aligned} \text{Tekanan} &= 1 \text{ atm} \\ \text{Suhu Operasi} &= 95 \text{ }^\circ\text{C} \\ &= 368.15 \text{ K} \\ \text{Treff} &= 25 \text{ }^\circ\text{C} \\ &= 298.15 \text{ K} \end{aligned}$$

$$\begin{aligned} \text{Cp C}_{24}\text{H}_{38}\text{C} &= A + BT + CT^2 + CT^3 \\ &= 370.5240 + 729.08426 + -563.85 + 213.66 \\ &= 749.4169 \text{ J/mol.K} \\ &= 749.4169 \times 0.24 \\ &= 179.11514 \text{ kal/mol.k} \\ &= 179.11514 \text{ kkal/kmol.k} \\ &= \frac{179.11514 \text{ kkal/kmol.k}}{390.5507 \text{ kg/kmol}} \\ &= \mathbf{0.458622 \text{ kkal/kg.K}} \end{aligned}$$

$$\begin{aligned} \Delta H &= m \times \text{cp} \times \Delta T \\ &= 3801.074288 \times 0.458621957 \times 70 \\ &= 122027.929 \text{ kkal} \end{aligned}$$

$$\begin{aligned} \text{Cp C}_8\text{H}_4\text{O}_3 &= A + BT + CT^2 + CT^3 \\ &= -105.6270 + 730.40960 + -526.5106 + -193834.9 \\ &= \text{#####} \text{ J/mol.K} \\ &= \text{#####} \times 0.24 \\ &= -46304.21 \text{ kal/mol.k} \\ &= -46304.21 \text{ kkal/kmol.k} \\ &= \frac{-46304.21 \text{ kkal/kmol.k}}{148.1140 \text{ kg/kmol}} \\ &= \mathbf{-312.6256 \text{ kkal/kg.K}} \end{aligned}$$

$$\begin{aligned} \Delta H &= m \times \text{cp} \times \Delta T \\ &= 2.89069932 \times -312.6255524 \times 70 \\ &= -63259.45302 \text{ kkal} \end{aligned}$$

$$\begin{aligned} \text{Cp C}_8\text{H}_6\text{O}_4 &= A + BT + CT^2 + CT^3 \\ &= -561.000 + 204.6914 + -624.8137 + 113.7652 \\ &= -867.357 \text{ J/mol.K} \\ &= -867.3571 \times 0.24 \\ &= -207.3036 \text{ kal/mol.k} \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned} &= -207.3036 \text{ kkal/kmol.k} \\ &= \frac{-207.3036 \text{ kkal/kmol.k}}{166.1286 \text{ kg/kmol}} \\ &= \mathbf{-1.24785 \text{ kkal/kg.K}} \end{aligned}$$
$$\begin{aligned} \Delta H &= m \quad \times \quad cp \quad \times \quad \Delta T \\ &= 2.899397513 \quad \times \quad -1.247849612 \quad \times \quad 70 \\ &= -253.2608443 \quad \text{kkal} \end{aligned}$$
$$\begin{aligned} C_p C_8H_{18}O &= A \quad + \quad BT \quad + \quad CT^2 \quad + \quad CT^3 \\ &= -105.6270 \quad + \quad 360.4299 \quad + \quad -354.7885 \quad + \quad 154.8903 \\ &= 54.90470 \text{ J/mol.K} \\ &= 54.90470 \quad \times \quad \mathbf{0.24} \\ &= 13.122552 \text{ kal/mol.k} \\ &= 13.122552 \text{ kkal/kmol.k} \\ &= \frac{13.122552 \text{ kkal/kmol.k}}{130.2257 \text{ kg/kmol}} \\ &= \mathbf{0.1007677 \text{ kkal/kg.K}} \end{aligned}$$
$$\begin{aligned} \Delta H &= m \quad \times \quad cp \quad \times \quad \Delta T \\ &= 5.078255562 \quad \times \quad 0.100767744 \quad \times \quad 70 \\ &= 35.82070482 \quad \text{kkal} \end{aligned}$$
$$\begin{aligned} C_p H_2O &= A \quad + \quad BT \quad + \quad CT^2 \quad + \quad CT^3 \\ &= 92.0530 \quad + \quad -14.7087 \quad + \quad -28.6018 \quad + \quad 26.6794 \\ &= 75.421900 \text{ J/mol.K} \\ &= 75.421900 \quad \times \quad \mathbf{0.24} \\ &= 18.026287 \text{ kal/mol.k} \\ &= 18.026287 \text{ kkal/kmol.k} \\ &= \frac{18.026287 \text{ kkal/kmol.k}}{18.0147 \text{ kg/kmol}} \\ &= \mathbf{1.0006443 \text{ kkal/kg.K}} \end{aligned}$$
$$\begin{aligned} \Delta H &= m \quad \times \quad cp \quad \times \quad \Delta T \\ &= 189.6435728 \quad \times \quad 1.000644281 \quad \times \quad 70 \\ &= 13283.60296 \quad \text{kkal} \end{aligned}$$
$$\begin{aligned} C_p C_{16}H_{36}C &= 711.00000 \text{ J/mol.K} \\ &= 711.00000 \quad \times \quad \mathbf{0.24} \\ &= 169.93327 \text{ kal/mol.k} \\ &= 169.93327 \text{ kkal/kmol.k} \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetra-butyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
 &= \frac{169.93327 \text{ kkal/kmol.k}}{340.3164 \text{ kg/kmol}} \\
 &= \mathbf{0.499339 \text{ kkal/kg.K}} \\
 \Delta H &= m \quad \times \quad cp \quad \times \quad \Delta T \\
 &= 0.059767162 \quad \times \quad 0.499338986 \quad \times \quad 70 \\
 &= 2.089085171 \quad \text{kkal}
 \end{aligned}$$

Enthalpi Keluar

Komponen	Massa (kg)	Cp (kkal/kg.k)	ΔT (K)	ΔH (kkal)
C ₂₄ H ₃₈ O ₄	3801.074288	0.458621957	70	122027.93
C ₈ H ₄ O ₃	2.89069932	-312.6255524	70	-63259.45
C ₈ H ₆ O ₄	2.899397513	-1.247849612	70	-253.2608
C ₈ H ₁₈ O	5.078255562	0.100767744	70	35.820705
H ₂ O	189.6435728	1.000644281	70	13283.603
C ₁₆ H ₃₆ O ₄ Ti	0.059767162	0.499338986	70	2.0890852
Total				71836.728

Q serap

$$\begin{aligned}
 \text{Suhu air pendingin masuk} &= 30 \text{ }^\circ\text{C} = 303 \text{ K} \quad (\text{Ulrich : 427}) \\
 \text{Suhu reference} &= 25 \text{ }^\circ\text{C} = 298 \text{ K} \\
 \text{Suhu air pendingin keluar} &= 45 \text{ }^\circ\text{C} = 318 \text{ K} \quad (\text{Ulrich : 427})
 \end{aligned}$$

Perhitungan kapasitas panas H₂O_(g) 70

$$\begin{aligned}
 \int Cp \, dT &= AT + \frac{B}{2}T^2 + \frac{C}{3}T^3 + \frac{D}{4}T^4 \\
 &\quad + \frac{D}{5}T^5 \\
 &= 170 + -12.65526 + 13.516391 + -2.422218 \\
 &\quad + 0.1509042 \\
 &= 168.25482 \text{ J/mol.K} \\
 &= 168.25482 \times \mathbf{0.24} \\
 &= 40.213911 \text{ kal/mol.k} \\
 &= 40.213911 \text{ kkal/kmol.k} \\
 &= \frac{40.213911 \text{ kkal/kmol.k}}{18.0147 \text{ kg/kmol}} \\
 &= \mathbf{2.2322856 \text{ kkal/kg.K}}
 \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
 Q_{\text{serap}} &= m_{\text{air pendingin}} \times C_p_{\text{air}} \times \Delta T \\
 &= m_{\text{air pendingin}} \times 2.2323 \times 15 \\
 &= m_{\text{air pendingin}} \times 33.4843 \\
 Q_{\text{loss}} &= 5\% \times Q_{\text{serap}} \\
 &= 5\% \times m_{\text{air pendingin}} \times 33.4843 \\
 &= m_{\text{air pendingin}} \times 1.6742
 \end{aligned}$$

Neraca Panas Total :

$$\begin{aligned}
 \Delta H_{\text{bahan masuk}} &= \Delta H_{\text{bahan keluar}} + Q_{\text{serap}} + Q_{\text{loss}} \\
 75168.31157 &= 71836.72788 + (m_{\text{air pendingin}} \times 33.484284) \\
 &\quad + (m_{\text{air pendingin}} \times 1.6742142) \\
 m_{\text{air pendingin}} &= 94.758987 \text{ kg} \\
 \text{sehingga, } Q_{\text{serap}} &= 3172.9368 \text{ kkal} \\
 Q_{\text{loss}} &= 158.6468 \text{ kkal}
 \end{aligned}$$

NERACA PANAS COOLER			
Panas Masuk		Panas Keluar	
Komponen	ΔH (kkal)	Komponen	ΔH (kkal)
$C_{24}H_{38}O_4$	260449.4668	$C_{24}H_{38}O_4$	122027.929
$C_8H_4O_3$	-213376.1693	$C_8H_4O_3$	-63259.45302
$C_8H_6O_4$	152.8705296	$C_8H_6O_4$	-253.2608443
$C_8H_{18}O$	106.8758038	$C_8H_{18}O$	35.82070482
H_2O	27831.08955	H_2O	13283.60296
$C_{16}H_{36}O_4Ti$	4.178170341	$C_{16}H_{36}O_4Ti$	2.089085171
		Q_{serap}	3172.9368
		Q_{loss}	158.6468
Total	75168.31157	Total	75168.31157

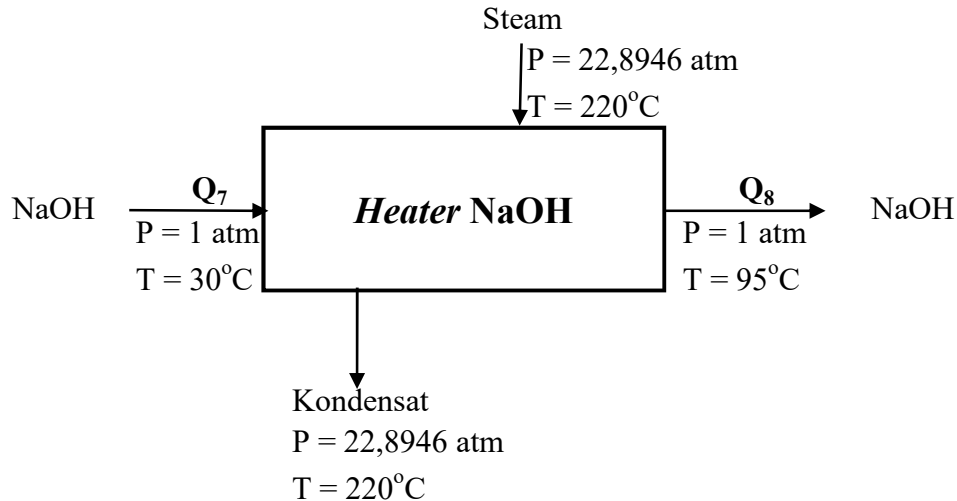


PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

6. Heater NaOH

Fungsi: Memanaskan NaOH hingga 95°C sebelum masuk *neutraliser*



Panas masuk Heater

Tekanan	=	1 atm
Suhu Operasi	=	30 °C
	=	303.15 K
Treff	=	25 °C
	=	298.15 K

$$\begin{aligned}
 C_p \text{ NaOH} &= A + BT + CT^2 + DT^3 \\
 &= 87.639 + (-0.14663)T + (-0.41744)T^2 + 3.30E-02 T^3 \\
 &= 87.10799 \text{ J/mol.K} \\
 &= 87.1080 \times 0.24 \\
 &= 20.819331 \text{ kal/mol.k} \\
 &= 20.819331 \text{ kkal/kmol.k} \\
 &= \frac{20.819331 \text{ kkal/kmol.k}}{39.9966 \text{ kg/kmol}} \\
 &= \mathbf{0.520527 \text{ kkal/kg.K}} \\
 \Delta H &= m \times c_p \times \Delta T \\
 &= 0.007031431 \times 0.520527001 \times 5 \\
 &= 0.018300248 \text{ kkal}
 \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

Enthalpi Masuk

Komponen	Massa (kg)	Cp (kkal/kg.k)	ΔT (K)	ΔH
				(kkal)
NaOH	0.00703143	0.520527001	5	0.0183002
Total				0.0183002

Panas keluar Heater

Tekanan = 1 atm
 Suhu Operasi = 95 °C
 = 368.15 K
 = 25 °C
 = 298.15 K

$$\begin{aligned}
 C_p \text{ NaOH} &= A + BT + CT^2 + DT^3 \\
 &= 87.639 + (-0.17807) + (-0.61564) + 5.92E-02 \\
 &= 86.90449 \text{ J/mol.K} \\
 &= 86.9045 \times 0.24 \\
 &= 20.770694 \text{ kkal/mol.k} \\
 &= 20.770694 \text{ kkal/kmol.k} \\
 &= \frac{20.770694 \text{ kkal/kmol.k}}{39.9966 \text{ kg/kmol}} \\
 &= \mathbf{0.519311 \text{ kkal/kg.K}} \\
 \Delta H &= m \times c_p \times \Delta T \\
 &= 0.007031431 \times 0.519310974 \times 70 \\
 &= 0.255604941 \text{ kkal}
 \end{aligned}$$

Enthalpi Keluar

Komponen	Massa (kg)	Cp (kkal/kg.k)	ΔT (K)	ΔH
				(kkal)
NaOH	0.007031431	0.519310974	70	0.2556049
Total				0.2556049

Neraca Energi Total :

$$\begin{aligned}
 \Delta H_{\text{bahan masuk}} + Q_{\text{supply}} &= \Delta H_{\text{bahan keluar}} + Q_{\text{loss}} \\
 0.018300248 + Q_{\text{supply}} &= 0.255604941 + 5\% Q_{\text{supply}} \\
 Q_{\text{supply}} &= 0.249794414 \text{ Kkal/jam} \\
 Q_{\text{loss}} &= 0.012489721 \text{ Kkal/jam}
 \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetra-butyl Titanate Melalui Proses Esterifikasi

Kebutuhan steam:

Dipakai steam pada tekanan steam 22,9 atm dengan suhu steam 220°C

(Ulrich : hal.432)

$$\lambda \text{ steam} = 1856.2 \text{ kJ/kg} \quad (\text{Smith : Steam table F-1})$$

$$= 443.6318 \text{ kkal/kg}$$

$$Q \text{ steam} = M \text{ steam} \times \lambda_{\text{steam}}$$

$$M \text{ steam} = \frac{Q \text{ steam}}{\lambda_{\text{steam}}}$$

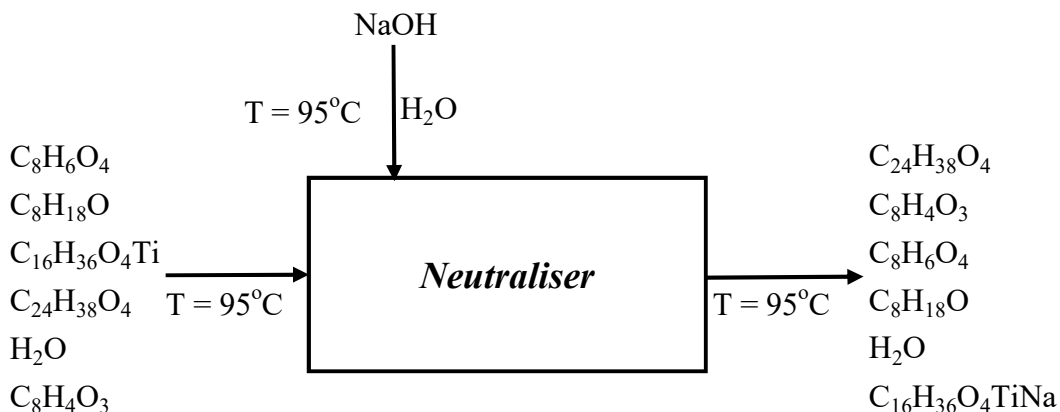
$$= \frac{0.2497944143 \text{ kkal/jam}}{443.6318000000 \text{ kkal/kg}}$$

$$= 0.000563067 \text{ kg/jam}$$

NERACA PANAS <i>HEATER</i>			
Panas Masuk		Panas Keluar	
Komponen	ΔH (kkal)	Komponen	ΔH (kkal)
NaOH	0.018300248	NaOH	0.255604941
Qsupply	0.249794414	Qloss	0.012489721
Total	0.268094662	Total	0.268094662

B.7 Neutraliser

Fungsi : untuk menetralkan katalis $C_{16}H_{36}O_4Ti$ menggunakan NaOH





PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Enthalpi bahan masuk

$$T_{in} = 95 \text{ }^{\circ}\text{C} = 368.15 \text{ K}$$

$$T_{ref} = 25 \text{ }^{\circ}\text{C} = 298.15 \text{ K}$$

$$\begin{aligned} \int C_{pd} T \text{ C}_{24}\text{H}_{38}\text{O}_4 &= AT + \frac{B}{2} T^2 + \frac{C}{3} T^3 + \frac{D}{4} T^4 \\ &= 6.2 + 11.037956 + -7.753265 + 2.68E+00 \\ &= 1.22E+01 \text{ kkal/kmol.K} \end{aligned}$$

$$\begin{aligned} \Delta H &= n \times \int C_{pd} T \\ &= 9.733 \times 1.2E+01 \\ &= 118.36 \text{ kkal} \end{aligned}$$

$$\begin{aligned} \int C_{pd} T \text{ C}_8\text{H}_4\text{O}_3 &= AT + \frac{B}{2} T^2 + \frac{C}{3} T^3 + \frac{D}{4} T^4 \\ &= -1.8 + 11.058021 + -7.239823 + 1.8E+00 \\ &= 3.8343633 \text{ kkal/kmol.K} \end{aligned}$$

$$\begin{aligned} \Delta H &= n \times \int C_{pd} T \\ &= 0.02 \times 3.8E+00 \\ &= 0.074834 \text{ kkal} \end{aligned}$$

$$\begin{aligned} \int C_{pd} T \text{ C}_8\text{H}_6\text{O}_4 &= AT + \frac{B}{2} T^2 + \frac{C}{3} T^3 + \frac{D}{4} T^4 \\ &= -9.4 + 19.284654 + -8.591547 + 1.43E+00 \\ &= 2.73E+00 \text{ kkal/kmol.K} \end{aligned}$$

$$\begin{aligned} \Delta H &= n \times \int C_{pd} T \\ &= 0.02 \times 2.7E+00 \\ &= 0.05 \text{ kkal} \end{aligned}$$

$$\begin{aligned} \int C_{pd} T \text{ C}_8\text{H}_{18}\text{O} &= AT + \frac{B}{2} T^2 + \frac{C}{3} T^3 + \frac{D}{4} T^4 \\ &= -1.8 + 5.4567211 + -4.878545 + 1.9E+00 \\ &= 7.525E-01 \text{ kkal/kmol.K} \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetra-butyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}\Delta H &= n \times \int C_{pd} T \\ &= 0.0390 \times 7.5E-01 \\ &= 0.029345 \text{ kkal}\end{aligned}$$

$$\int C_{pd} T \text{ C}_{16}\text{H}_{36}\text{O}_4\text{T} = 0.00004 \text{ kkal/kmol.K}$$

$$\begin{aligned}\Delta H &= n \times \int C_{pd} T \\ &= 0.000176 \times 4.1E-05 \\ &= 7.133E-09 \text{ kkal}\end{aligned}$$

$$\begin{aligned}\int C_{pd} T \text{ H}_2\text{O} &= AT + \frac{B}{2} T^2 + \frac{C}{3} T^3 + \frac{D}{4} T^4 \\ &= 1.54 + -0.222682 + -0.393292 + 3.34E-01 \\ &= 1.2584872 \text{ kkal/kmol.K}\end{aligned}$$

$$\begin{aligned}\Delta H &= n \times \int C_{pd} T \\ &= 10.53 \times 1.3E+00 \\ &= 13.25 \text{ kkal}\end{aligned}$$

Panas masuk (dari reaktor)

Komponen	massa (kg/jam)	n (kmol/jam)	$\int C_{pd} T$ (kkal/kmol)	ΔH (kkal)
C ₂₄ H ₃₈ O ₄	3801.0743	9.733	12.162	118.36
C ₈ H ₄ O ₃	2.8907	0.02	3.8344	0.075
C ₈ H ₆ O ₄	2.8994	0.02	2.7336	0.048
C ₈ H ₁₈ O	5.0783	0.0390	0.752513	0.029
C ₁₆ H ₃₆ O ₄ Ti	0.059767	0.000176	0.752513	0.000
H ₂ O	189.64357	10.53	1.258487	13.25
Total				131.76

$$\begin{aligned}\int C_{pd} T \text{ H}_2\text{O} &= AT + \frac{B}{2} T^2 + \frac{C}{3} T^3 + \frac{D}{4} T^4 \\ &= 1.4662005 + 0.0026958 + -0.008465 + 7.42E-04 \\ &= 1.46E+00 \text{ kkal/kmol.K}\end{aligned}$$

$$\begin{aligned}\Delta H &= n \times \int C_{pd} T \\ &= 0.00018 \times 1.5E+00 \\ &= 0.0002569 \text{ kkal}\end{aligned}$$



PERANCANGAN PABRIK

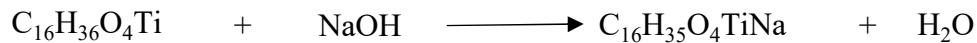
Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

Panas masuk (NaOH)

Komponen	massa (kg/jam)	n (kmol/jam)	$\int C_p dT$ (kkal/kmol)	ΔH (kkal/jam)
NaOH	0.0070	0.0002	0.5193	0.2556
Total				0.2556

$$\begin{aligned}
 \text{Total panas masuk} &= \text{panas masuk (reaktor)} + \text{panas masuk (NaOH)} \\
 &= 131.76 + 0.2556 \\
 &= 132.02
 \end{aligned}$$

$$Q_{\text{penetralan}} = Q_{\text{reaksi}}$$



Tabel B.30 Entalpi Pembentukan Reaktan Reaksi pada 298 K

Komponen	massa (kg/jam)	n (kmol/jam)	Hf (kkal/kmol)	Hf _{Reaktan} (kkal/Jam)
C ₁₆ H ₃₆ O ₄ Ti	0.0598	0.0002	-398733.7	-70.0265
H ₂ O	189.6436	10.5272	-57495.3	-605262.5
NaOH	0.0070	0.0002	-101128.2	-17.7784
Total				-605350.3

Tabel B.31 Entalpi Pembentukan Produk Reaksi pada 298 K

Komponen	massa (kg/jam)	n (kmol/jam)	Hf (kkal/kmol)	Hf _{Reaktan} (kkal/Jam)
C ₁₆ H ₃₅ O ₄ TiNa	0.0636	0.0002	-398733.7	-74.5577
H ₂ O	0.0032	0.0002	-57495.28	-10.0986
Total				-84.6563

$$\begin{aligned}
 Hf_{\text{reaksi}(298K)} &= \sum(n \times Hf)_{\text{produk}} - \sum(n \times Hf)_{\text{reaktan}} \\
 &= -84.656 - (-605350.3439) \\
 &= 605265.688 \text{ kkal/jam}
 \end{aligned}$$

Menghitung Panas Keluar *Neutraliser*

$$T_{\text{Out}} = 95 \text{ }^\circ\text{C} = 368 \text{ K}$$

$$T_{\text{Ref}} = 25 \text{ }^\circ\text{C} = 298 \text{ K}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraethyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned} \int C_{pd} T C_{24}H_{38}O_4 &= AT + \frac{B}{2} T^2 + \frac{C}{3} T^3 + \frac{D}{4} T^4 \\ &= 6.2 + 11.037956 + -7.753265 + 2.68E+00 \\ &= 1.22E+01 \text{ kkal/kmol.K} \end{aligned}$$

$$\begin{aligned} \Delta H &= n \times \int C_{pd} T \\ &= 9.733 \times 1.2E+01 \\ &= 118.36 \text{ kkal} \end{aligned}$$

$$\begin{aligned} \int C_{pd} T C_8H_4O_3 &= AT + \frac{B}{2} T^2 + \frac{C}{3} T^3 + \frac{D}{4} T^4 \\ &= -1.8 + 11.058021 + -7.239823 + 1.8E+00 \\ &= 3.8E+00 \text{ kkal/kmol.K} \end{aligned}$$

$$\begin{aligned} \Delta H &= n \times \int C_{pd} T \\ &= 0.02 \times 3.8E+00 \\ &= 0.074834 \text{ kkal} \end{aligned}$$

$$\begin{aligned} \int C_{pd} T C_8H_6O_4 &= AT + \frac{B}{2} T^2 + \frac{C}{3} T^3 + \frac{D}{4} T^4 \\ &= -9.4 + 19.284654 + -8.591547 + 1.43E+00 \\ &= 2.73E+00 \text{ kkal/kmol.K} \end{aligned}$$

$$\begin{aligned} \Delta H &= n \times \int C_{pd} T \\ &= 0.02 \times 2.7E+00 \\ &= 0.05 \text{ kkal} \end{aligned}$$

$$\begin{aligned} \int C_{pd} T C_8H_{18}O &= AT + \frac{B}{2} T^2 + \frac{C}{3} T^3 + \frac{D}{4} T^4 \\ &= -1.8 + 5.4567211 + -4.878545 + 1.9E+00 \\ &= 7.525E-01 \text{ kkal/kmol.K} \end{aligned}$$

$$\begin{aligned} \Delta H &= n \times \int C_{pd} T \\ &= 0.0390 \times 7.5E-01 \\ &= 0.029345 \text{ kkal} \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\int C_{pd}T \text{ C}_{16}\text{H}_{35}\text{O}_4\text{T} = 0.00004 \text{ kkal/kmol.K}$$

$$\begin{aligned} \Delta H &= n \times \int C_{pd}T \\ &= 0.000187 \times 4.1\text{E-}05 \\ &= 7.594\text{E-}09 \text{ kkal} \end{aligned}$$

$$\begin{aligned} \int C_{pd}T \text{ H}_2\text{O} &= AT + \frac{B}{2}T^2 + \frac{C}{3}T^3 + \frac{D}{4}T^4 \\ &= 1.54 + -0.222682 + -0.393292 + 3.34\text{E-}01 \\ &= 1.26\text{E+}00 \text{ kkal/kmol.K} \end{aligned}$$

$$\begin{aligned} \Delta H &= n \times \int C_{pd}T \\ &= 10.53 \times 1.3\text{E+}00 \\ &= 13.25 \text{ kkal} \end{aligned}$$

Tabel B.32 Panas Keluar *Neutraliser*

Komponen	massa (kg/jam)	n (kmol/jam)	$\int C_{pd}T$ (kkal/kmol)	ΔH (kkal/Jam)
C ₂₄ H ₃₈ O ₄	3801.0743	9.7326	12.162	118.36
C ₈ H ₄ O ₃	2.890699	0.0195	3.834	0.0748
C ₈ H ₆ O ₄	2.899398	0.0175	2.734	0.0477
C ₈ H ₁₈ O	5.078256	0.0390	0.753	0.0293
C ₁₆ H ₃₅ O ₄ TiNa	0.063634	0.0002	4.061E-05	0.0000
H ₂ O	189.6467	10.5273	1.2585	13.249
Total				131.8

Panas reaksi *neutraliser* :

$$\begin{aligned} \Delta H_{368,15} &= (X_A \times n + \Delta H_{298}) + (H_{out} - H_{in}) \\ &= -484212.5507 + -0.2553839 \\ &= -484212.8060 \text{ kkal/Jam} \end{aligned}$$

Nilai panas reaksi penetralan bernilai negatif (-) menunjukkan reaksi berlangsung endotermis

Menghitung Kebutuhan Pendingin

Q yang diterima pendingin

$$\begin{aligned} Q_{Lepas} &= (Q_{out} - Q_{in}) + Q_{Reaksi} \\ &= -0.255384 + -484212.8 \\ &= -484213 \text{ kkal/Jam} \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

Sebagai pendingin digunakan *cooling water* pada suhu 30°C dan keluar pada suhu 45°C.

$$T_{in} = 30\text{ }^{\circ}\text{C} = 303.15\text{ K} = 86\text{ }^{\circ}\text{F}$$

$$T_{out} = 45\text{ }^{\circ}\text{C} = 318.15\text{ K} = 113\text{ }^{\circ}\text{F}$$

$$C_p = 4.1868\text{ kJ/kg}\cdot^{\circ}\text{K} \quad (\text{Smith } et\ al.,\ 2005)$$

$$= 1.0000004\text{ kkal/kg}\cdot\text{K}$$

$$\begin{aligned} m_{\text{pendingin}} &= \frac{Q_{\text{lepas}}}{C_p \times \Delta T} \\ &= \frac{484213.0614\text{ kkal}}{62.8020\text{ kkal/kg}} \\ &= 7710.1535\text{ kg/jam} \end{aligned}$$

NERACA PANAS NEUTRALISER			
Masuk		Keluar	
Komponen	ΔH (kkal)	Komponen	ΔH (kkal)
Dari Neutraliser			
$C_{24}H_{38}O_4$	118.3647032381910	$C_{24}H_{38}O_4$	118.36470324
$C_8H_4O_3$	0.0748342118156	$C_8H_4O_3$	0.0748342118
$C_8H_6O_4$	0.0477082522919	$C_8H_6O_4$	0.0477082523
$C_8H_{18}O$	0.0293448536690	$C_8H_{18}O$	0.0293448537
$C_{16}H_{36}O_4Ti$	0.0000000071326	$C_{16}H_{32}O_4TiNa$	0.0000000076
H_2O	13.248306784780	H_2O	13.248527829
Dari tangki NaOH		ΔH reaksi	-484212.8060
NaOH	0.255605	Qserap	484213.0614
Total	132.0205023	Total	132.0205023

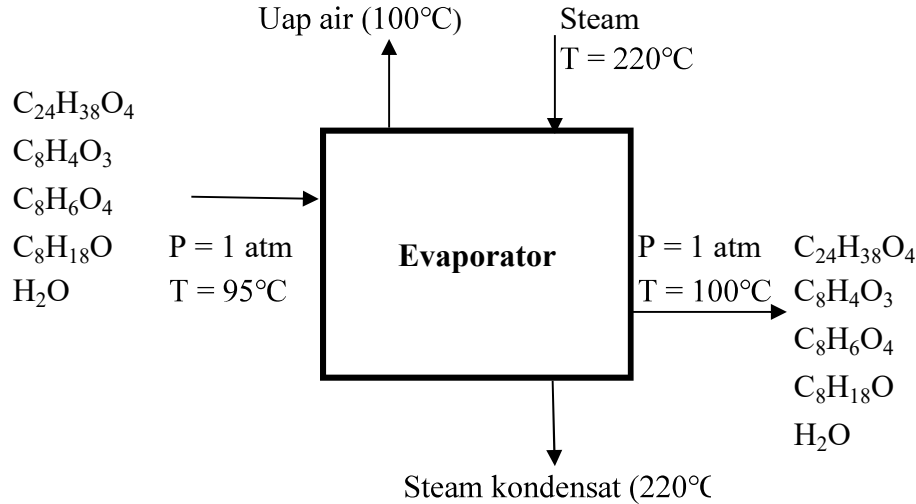


PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetra-butyl Titanate Melalui Proses Esterifikasi

B.8 Evaporator

Fungsi : Untuk memisahkan kandungan Dioctyl phthalate dari air



Enthalpi bahan masuk

Kondisi operasi

Tekanan	=	1 atm
Suhu bahan masuk	=	95 °C
	=	368.15 K
Treff	=	25 °C
	=	298.15 K

$$\begin{aligned}
 C_p \text{ C}_{24}\text{H}_{38}\text{C} &= A + BT + CT^2 + CT^3 \\
 &= 370.5240 + 729.08426 + (-563.85) + 213.66 \\
 &= 749.4169 \text{ J/mol.K} \\
 &= 749.4169 \times 0.24 \\
 &= 179.11514 \text{ kal/mol.k} \\
 &= 179.11514 \text{ kkal/kmol.k} \\
 &= \underline{179.11514} \text{ kkal/kmol.k} \\
 &\quad 390.5507 \text{ kg/kmol} \\
 &= \underline{0.458622} \text{ kkal/kg.K}
 \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned} \Delta H &= m \times cp \times \Delta T \\ &= 3801.074288 \times 0.458621957 \times 70 \\ &= 122027.929 \text{ kkal} \end{aligned}$$
$$\begin{aligned} C_p C_8H_4O_3 &= A + BT + CT^2 + CT^3 \\ &= -105.6270 + 730.40960 + -526.5106 + 142.271 \\ &= 240.54 \text{ J/mol.K} \\ &= 240.54 \times 0.24 \\ &= 57.491301 \text{ kal/mol.k} \\ &= 57.491301 \text{ kkal/kmol.k} \\ &= \underline{57.491301} \text{ kkal/kmol.k} \\ &\quad 148.1140 \text{ kg/kmol} \\ &= \mathbf{0.3881559 \text{ kkal/kg.K}} \end{aligned}$$
$$\begin{aligned} \Delta H &= m \times cp \times \Delta T \\ &= 2.89069932 \times 0.388155857 \times 70 \\ &= 78.54293108 \text{ kkal} \end{aligned}$$
$$\begin{aligned} C_p C_8H_6O_4 &= A + BT + CT^2 + CT^3 \\ &= -561.000 + 1273.7990 + -624.8137 + 113.7652 \\ &= ##### \text{ J/mol.K} \\ &= 201.75047 \times 0.24 \\ &= 48.219572 \text{ kal/mol.k} \\ &= 48.219572 \text{ kkal/kmol.k} \\ &= \underline{48.219572} \text{ kkal/kmol.k} \\ &\quad 166.1286 \text{ kg/kmol} \\ &= \mathbf{0.2902544 \text{ kkal/kg.K}} \end{aligned}$$
$$\begin{aligned} \Delta H &= m \times cp \times \Delta T \\ &= 2.899397513 \times 0.29025442 \times 70 \\ &= 58.90940609 \text{ kkal} \end{aligned}$$
$$\begin{aligned} C_p C_8H_{18}O &= A + BT + CT^2 + CT^3 \\ &= -105.6270 + 360.4299 + -354.7885 + 154.8903 \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
 &= 54.9047 \text{ J/mol.K} \\
 &= 54.9047 \times 0.24 \\
 &= 13.122552 \text{ kal/mol.k} \\
 &= 13.122552 \text{ kkal/kmol.k} \\
 &= \underline{13.122552} \text{ kkal/kmol.k} \\
 &\quad 130.2257 \text{ kg/kmol} \\
 &= \mathbf{0.1007677 \text{ kkal/kg.K}}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H &= m \times cp \times \Delta T \\
 &= 5.078255562 \times 0.100767744 \times 70 \\
 &= 35.82070482 \text{ kkal}
 \end{aligned}$$

$$\begin{aligned}
 Cp \text{ H}_2\text{O} &= A + BT + CT^2 + CT^3 \\
 &= 92.0530 + -14.7087 + -28.6018 + 26.6794 \\
 &= 75.42190 \text{ J/mol.K} \\
 &= 75.42190 \times 0.24 \\
 &= 18.026287 \text{ kal/mol.k} \\
 &= 18.026287 \text{ kkal/kmol.k} \\
 &= \underline{18.026287} \text{ kkal/kmol.k} \\
 &\quad 18.0147 \text{ kg/kmol} \\
 &= \mathbf{1.0006443 \text{ kkal/kg.K}}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H &= m \times cp \times \Delta T \\
 &= 1562.212421 \times 1.000644281 \times 70 \\
 &= 109425.3248 \text{ kkal}
 \end{aligned}$$

Enthalpi Masuk

Komponen	Massa (kg)	Cp (kkal/kg.k)	ΔT (K)	ΔH
				(kkal)
C ₂₄ H ₃₈ O ₄	3801.074288	0.458621957	70	122027.93
C ₈ H ₄ O ₃	2.890699	0.388155857	70	78.542931
C ₈ H ₆ O ₄	2.899398	0.29025442	70	58.909406
C ₈ H ₁₈ O	5.078255562	0.100767744	70	35.820705
H ₂ O	1562.212421	1.000644281	70	109425.32
Total				231626.53



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Enthalpi bahan keluar

Kondisi operasi

$$\text{Tekanan} = 1 \text{ atm}$$

$$\text{Suhu bahan keluar} = 100 \text{ }^\circ\text{C}$$

$$= 373.15 \text{ K}$$

$$\text{Treff} = 25 \text{ }^\circ\text{C}$$

$$= 298.15 \text{ K}$$

$$\begin{aligned} \text{Cp } C_{24}H_{38}C &= A + BT + CT^2 + CT^3 \\ &= 370.5240 + 738.98626 + -579.27 + 222.48 \\ &= 752.7233 \text{ J/mol.K} \\ &= 752.7233 \times 0.24 \\ &= 179.90538 \text{ kal/mol.k} \\ &= 179.90538 \text{ kkal/kmol.k} \\ &= \underline{179.90538} \text{ kkal/kmol.k} \\ &\quad 390.5507 \text{ kg/kmol} \\ &= \mathbf{0.4606454 \text{ kkal/kg.K}} \end{aligned}$$

$$\begin{aligned} \Delta H &= m \times cp \times \Delta T \\ &= 3793.47214 \times 0.460645357 \times 75 \\ &= 131058.3995 \text{ kkal} \end{aligned}$$

$$\begin{aligned} \text{Cp } C_8H_4O_3 &= A + BT + CT^2 + CT^3 \\ &= -105.6270 + 740.32960 + -540.9092 + 148.1 \\ &= 241.94 \text{ J/mol.K} \\ &= 241.94 \times 0.24 \\ &= 57.825236 \text{ kal/mol.k} \\ &= 57.825236 \text{ kkal/kmol.k} \\ &= \underline{57.825236} \text{ kkal/kmol.k} \\ &\quad 148.1140 \text{ kg/kmol} \\ &= \mathbf{0.3904104 \text{ kkal/kg.K}} \end{aligned}$$

$$\begin{aligned} \Delta H &= m \times cp \times \Delta T \\ &= 2.89069932 \times 0.39041044 \times 75 \\ &= 84.64193953 \text{ kkal} \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}C_p C_8H_6O_4 &= A + BT + CT^2 + CT^3 \\&= -561.000 + 1291.0990 + -641.9007 + 118.4637 \\&= 206.662 \text{ J/mol.K} \\&= 206.66202 \times 0.24 \\&= 49.393462 \text{ kal/mol.k} \\&= 49.393462 \text{ kkal/kmol.k} \\&= \underline{49.393462} \text{ kkal/kmol.k} \\&\quad 166.1286 \text{ kg/kmol} \\&= \mathbf{0.2973206 \text{ kkal/kg.K}}\end{aligned}$$
$$\begin{aligned}\Delta H &= m \times cp \times \Delta T \\&= 2.899397513 \times 0.297320572 \times 75 \\&= 64.65378946 \text{ kkal}\end{aligned}$$
$$\begin{aligned}C_p C_8H_{18}O &= A + BT + CT^2 + CT^3 \\&= -105.6270 + 365.3250 + -364.4910 + 161.2872 \\&= 56.49433 \text{ J/mol.K} \\&= 56.49433 \times 0.24 \\&= 13.502484 \text{ kal/mol.k} \\&= 13.502484 \text{ kkal/kmol.k} \\&= \underline{13.502484} \text{ kkal/kmol.k} \\&\quad 130.2257 \text{ kg/kmol} \\&= \mathbf{0.1036852 \text{ kkal/kg.K}}\end{aligned}$$
$$\begin{aligned}\Delta H &= m \times cp \times \Delta T \\&= 5.078255562 \times 0.103685231 \times 75 \\&= 39.49050747 \text{ kkal}\end{aligned}$$
$$\begin{aligned}C_p H_2O &= A + BT + CT^2 + CT^3 \\&= 92.0530 + -14.9085 + -29.3840 + 27.7813 \\&= 75.541816 \text{ J/mol.K} \\&= 75.541816 \times 0.24 \\&= 18.054947 \text{ kal/mol.k} \\&= 18.054947 \text{ kkal/kmol.k}\end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$= \frac{18.054947 \text{ kkal/kmol.k}}{18.0147 \text{ kg/kmol}}$$

$$= 1.0022352 \text{ kkal/kg.K}$$

$$\begin{aligned} \Delta H &= m \quad \times \quad cp \quad \times \quad \Delta T \\ &= 7.758369819 \quad \times \quad 1.002235243 \quad \times \quad 75 \\ &= 583.1783747 \quad \text{kkal} \end{aligned}$$

Enthalpi Keluar

Komponen	Massa (kg)	Cp (kkal/kg.k)	ΔT (K)	ΔH
				(kkal)
C ₂₄ H ₃₈ O ₄	3793.4721	0.460645357	75	131058.4
C ₈ H ₄ O ₃	2.8906993	0.39041044	75	84.64194
C ₈ H ₆ O ₄	2.8993975	0.297320572	75	64.653789
C ₈ H ₁₈ O	5.0782556	0.103685231	75	39.490507
H ₂ O	7.7583698	1.002235243	75	583.17837
Total				131830.36

Entalpi uap air ke kondensor pada suhu 100°C (373,15 K),

$$\text{Massa uap air} = 1,562.0562 \text{ kg/jam} = 86.7102 \text{ kmol/jam}$$

$$\lambda \text{ H}_2\text{O} = 9,717.0000 \text{ kkal/kmol} \text{ (Sherwood, Appendix A)}$$

$$\begin{aligned} Cp \text{ H}_2\text{O} &= A + BT + CT^2 + CT^3 \\ &\quad + CT^4 \\ &= 33.9 + -3.1414 + 4.1641 + -9.26E-01 \\ &\quad + 0.00000 \\ &= 34.029592 \text{ J/mol.K} \\ &= 34.029592 \times 0.24 \\ &= 8.133277 \text{ kal/mol.k} \\ &= 8.133277 \text{ kkal/kmol.k} \end{aligned}$$

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PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\text{Enthalpi bahan} = \int_{T_{ref}}^T C_p \Delta T + n \times \lambda$$

$$\begin{aligned} H \text{ H}_2\text{O uap} &= (\text{Massa uap air} \times C_p \text{ H}_2\text{O suhu } 100) + (\text{Massa uap air} \times \lambda) \\ &= \mathbf{843268.0865} \text{ kkal/jam} \end{aligned}$$

$$\begin{aligned} \text{Total entalpi bahan keluar} &= 131,830.36 + 843268.09 \\ &= 975098.45 \text{ kkal} \end{aligned}$$

Neraca Energi Total :

$$\text{Entalpi bahan masuk} + Q \text{ supply} = \text{Entalpi bahan keluar} + Q \text{ loss}$$

$$\begin{aligned} \text{Asumsi } Q \text{ loss} &= 5\% \text{ dari } Q \text{ supply} \\ &(\text{kehilangan maksimum} = 10\%; \quad \text{Ulrich : 432}) \end{aligned}$$

$$\begin{aligned} 231,626.5 + Q \text{ supply} &= 975,098 + 5\% \text{ dari } Q \text{ supply} \\ Q \text{ supply} &= 782,602 \text{ kkal/jam} \\ Q \text{ loss} &= 39130.101 \text{ kkal/jam} \end{aligned}$$

Kebutuhan steam :

Digunakan steam dengan suhu 220°C dan tekanan 22,9 atm (Ulrich, app B hal. 426)

Kebutuhan steam pada tekanan 22,9 atm dan suhu steam 220°C. (Ulrich, App. B)

$$\begin{aligned} \lambda_{\text{steam}} &= 1856.2 \text{ kJ/kg} \\ &= 443.6318 \text{ kkal/kg} \\ Q_{\text{steam}} &= m_{\text{steam}} \times \lambda_{\text{steam}} \\ M_{\text{steam}} &= \frac{Q_{\text{steam}}}{\lambda_{\text{steam}}} \\ &= \frac{782,602.0250}{443.631800000} \\ &= 1764.080089 \text{ kg/jam} \end{aligned}$$

$$\begin{aligned} \text{Steam ekonomi} &= \frac{V}{S} = \frac{\text{H}_2\text{O menguap}}{\text{Kebutuhan steam}} \\ &= \frac{1,562.0562}{1764.0801} \\ &= \mathbf{0.885479} \end{aligned}$$

(Steam economy < 1,6 maka menggunakan Single Effect Evaporator)
(Wallas)



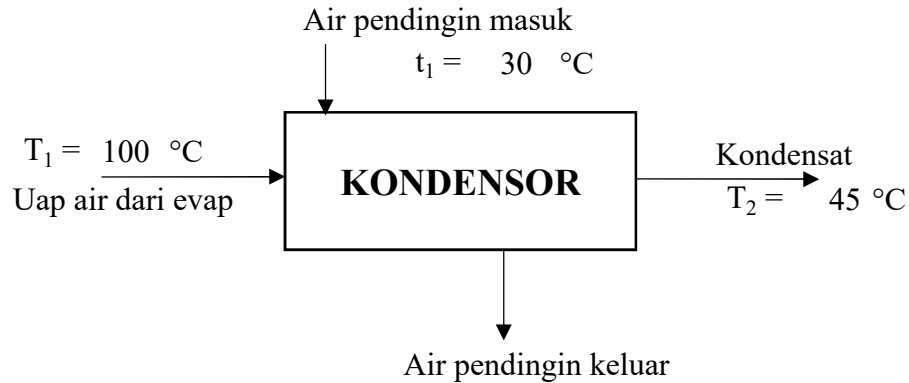
PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

NERACA PANAS EVAPORATOR			
Panas Masuk		Panas Keluar	
Komponen	ΔH (kkal)	Komponen	ΔH (kkal)
$C_{24}H_{38}O_4$	122027.929	$C_{24}H_{38}O_4$	131058.3995
$C_8H_4O_3$	78.54293108	$C_8H_4O_3$	84.64193953
$C_8H_6O_4$	58.90940609	$C_8H_6O_4$	64.65378946
$C_8H_{18}O$	35.82070482	$C_8H_{18}O$	39.49050747
H_2O	109425.3248	H_2O	583.1783747
Total	231626.5268	Total	131830.3641
		$H_2O(g)$	843268.0865
Q_{serap}	782602.0250	Q_{loss}	39130.1013
Total	1014228.5518	Total	1014228.5518

9. Kondensor

Fungsi : Untuk mengubah uap air menjadi fase cair



Kondisi operasi:

Tekanan operasi = 1 atm (Tekanan atmosfer)

Suhu operasi = 30 °C

Entalpi bahan masuk:

Entalpi bahan masuk kondensor $\Sigma H = \Sigma H_{uap}$ keluar evaporator

$H_{2O(g)} = 843,268$ kkal/jam



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned} \text{Hv H}_2\text{O pada } 100^\circ\text{C} &= 2,676.00 \text{ Kj/kg} \\ &= 639.5793 \text{ Kkal/kg} \\ &(\text{Smith } 7^{\text{ed}}; \text{ Steam Table F.1}) \end{aligned}$$

Asumsi :

$$\text{Uap air yang lolos (non-condensat)} = 1\% \text{ uap air yang masuk} \\ [\text{Ludwig : 376}]$$

$$\begin{aligned} \text{Uap air yang lolos (non-condensat)} &= 1\% \times 1,562.056 \\ &= 15.6206 \text{ kg/jam} \\ &= 0.8671 \text{ kmol/jam} \end{aligned}$$

$$\begin{aligned} \text{Entalpi Uap air yang lolos} &= 639.5793 \times 15.6206 \\ &= 9990.5889 \text{ kkal/jam} \end{aligned}$$

$$\begin{aligned} \text{Kondensat} &= \text{Uap air yang masuk} - \text{Uap air non-condensat} \\ \text{Kondensat} &= 1,562.056200233 - 15.6206 \\ \text{Kondensat} &= 1,546.4356 \text{ kg/jam} \\ &= 85.8431 \text{ kmol/jam} \end{aligned}$$

Entalpi bahan keluar:

Uap air ke udara pada suhu 100°C

$$T_{\text{masuk}} = 100 \text{ }^\circ\text{C} = 373.15 \text{ K}$$

$$T_{\text{reff}} = 25 \text{ }^\circ\text{C} = 298.15 \text{ K}$$

$$\begin{aligned} \text{H}_2\text{O}_{(g)} &= A + BT + CT^2 + DT^3 \\ &\quad + ET^4 \\ &= 33.9 + -3.1414 + 4.1641 + -9.26E-01 \\ &\quad + 0.07161 \\ &= 34.101199 \text{ J/mol.K} \\ &= 34.101199 \times \mathbf{0.24} \\ &= 8.150391 \text{ kal/mol.k} \\ &= 8.150391 \text{ kkal/kmol.k} \end{aligned}$$

$$\begin{aligned} \text{H}_2\text{O}_{(g)} &= n \int_{T_{\text{reff}}}^T C_p \Delta T \\ &= 85.8431 \times 8.150391 \times 75 \\ &= 52474.102 \text{ kkal/jam} \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

Kondensat H₂O_(l) ke Steam Condensat (SC) pada suhu 100°C

$$\text{Massa kondensat} = 1,546.4356 \text{ kg/jam} = 85.843081 \text{ kmol/jam}$$

$$\lambda \text{ H}_2\text{O} = 5,456.8480 \text{ kkal/kmol (sherwood, Appendix A)}$$

$$\begin{aligned} \text{H}_2\text{O}_{(l)} &= A + BT + CT^2 + DT^3 \\ &\quad + ET^4 \\ &= 33.9 + -3.1414 + 4.1641 + 0.00E+00 \\ &\quad + 0.00E+00 \\ &= 34.955738 \text{ J/mol.K} \\ &= 34.955738 \times 0.24 \\ &= 8.354631 \text{ kal/mol.k} \\ &= 8.354631 \text{ kkal/kmol.k} \end{aligned}$$

$$\begin{aligned} \text{H}_2\text{O}_{(l)} &= n \int_{T_{ref}}^T C_p \Delta T \\ &= 85.8431 \times 8.354631 \times 75 \\ &= 53789.046 \text{ kkal/jam} \end{aligned}$$

$$\begin{aligned} \text{H}_2\text{O}_{(l)} &= \Delta H = n \int_{T_{ref}}^T [C_p \Delta T] + n \times \lambda \\ &\quad \text{(Terjadi perubahan fase)} \\ &= 8.354631 + 85.8431 \times 5,456.85 \\ &= 468441 \text{ kkal/jam} \end{aligned}$$

$$\begin{aligned} \text{Total entalpi bahan keluar} &= 52,474.10 + 468,441 \\ &= 520,915 \text{ kkal/jam} \end{aligned}$$

Neraca Energi Total :

$$\begin{aligned} Q \text{ masuk} &= Q \text{ keluar} + Q \text{ serap} \\ 843,268.0865 &= 520,915 + Q \text{ serap} \\ Q \text{ serap} &= 322,353.0 \text{ kkal/jam} \end{aligned}$$

Kebutuhan air pendingin

$$\begin{aligned} t \text{ air pendingin masuk } (t_1) &= 45 \text{ }^\circ\text{C} \\ t \text{ air pendingin keluar } (t_2) &= 30 \text{ }^\circ\text{C} \\ \Delta T \text{ air pendingin} &= 45 - 30 \\ &= 15 \text{ }^\circ\text{C} \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

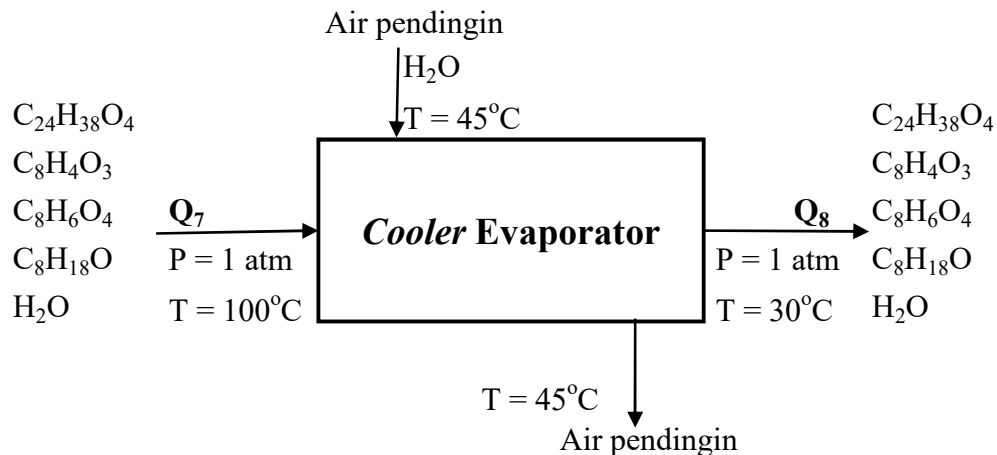
Dari Perry 5th ed. T. 3-176 P. 3-126, diketahui :

$$\begin{aligned}
 C_p \text{ air pendingin pada } 30^\circ\text{C} &= 0.99866 \text{ kkal/kg.}^\circ\text{C} \\
 \text{Kebutuhan air pendingin} &= \frac{Q}{C_p \times \Delta T} \\
 &= \frac{322,352.9836}{0.99866 \times 15} \\
 &= 21,519.0344 \text{ kg/jam}
 \end{aligned}$$

NERACA PANAS KONDENSOR			
Masuk (kkal/jam)		Keluar (kkal/jam)	
Uap dari evaporator		Uap air yang lolos ke udara bebas	
H ₂ O _(g)	843,268.0865	H ₂ O _(g)	52,474.1022939323
		Kondensat	
		H ₂ O _(l)	468,441.0006514740
		Q serap	322,352.9835671930
Total	843,268.0865	Total	843,268.0865

10. Cooler Evaporator

Fungsi : untuk menurunkan suhu produk keluaran evaporator



Panas masuk Cooler

$$\begin{aligned}
 \text{Tekanan} &= 1 \text{ atm} \\
 \text{Suhu Operasi} &= 100 \text{ }^\circ\text{C} \\
 &= 373.15 \text{ K} \\
 \text{Treff} &= 25 \text{ }^\circ\text{C} \\
 &= 298.15 \text{ K}
 \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}C_p C_{24}H_{38}C &= A + BT + CT^2 + CT^3 \\&= 370.5240 + 738.98626 + -579.27 + 222.48 \\&= 752.7233 \text{ J/mol.K} \\&= 752.7233 \times 0.24 \\&= 179.90538 \text{ kal/mol.k} \\&= 179.90538 \text{ kkal/kmol.k} \\&= \frac{179.90538}{390.5507} \text{ kkal/kmol.k} \\&= \mathbf{0.4606454 \text{ kkal/kg.K}}\end{aligned}$$

$$\begin{aligned}\Delta H &= m \times cp \times \Delta T \\&= 3793.47214 \times 0.460645357 \times 75 \\&= 131058.3995 \text{ kkal}\end{aligned}$$

$$\begin{aligned}C_p C_8H_4O_3 &= A + BT + CT^2 + CT^3 \\&= -105.6270 + 740.32960 + -540.9092 + 148.147 \\&= 241.94 \text{ J/mol.K} \\&= 241.94 \times 0.24 \\&= 57.825236 \text{ kal/mol.k} \\&= 57.825236 \text{ kkal/kmol.k} \\&= \frac{57.825236}{148.1140} \text{ kkal/kmol.k} \\&= \mathbf{0.3904104 \text{ kkal/kg.K}}\end{aligned}$$

$$\begin{aligned}\Delta H &= m \times cp \times \Delta T \\&= 2.89069932 \times 0.39041044 \times 75 \\&= 84.64193953 \text{ kkal}\end{aligned}$$

$$\begin{aligned}C_p C_8H_6O_4 &= A + BT + CT^2 + CT^3 \\&= -561.000 + 1291.0990 + -641.9007 + 118.4637 \\&= ##### \text{ J/mol.K} \\&= 206.66202 \times 0.24 \\&= 49.393462 \text{ kal/mol.k} \\&= 49.393462 \text{ kkal/kmol.k} \\&= \frac{49.393462}{166.1286} \text{ kkal/kmol.k} \\&= \mathbf{0.2973206 \text{ kkal/kg.K}}\end{aligned}$$

$$\begin{aligned}\Delta H &= m \times cp \times \Delta T \\&= 2.899397513 \times 0.297320572 \times 75 \\&= 64.65378946 \text{ kkal}\end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
 C_p C_8H_{18}O &= A + BT + CT^2 + CT^3 \\
 &= -105.6270 + 365.3250 + -364.4910 + 161.2872 \\
 &= 56.4943 \text{ J/mol.K} \\
 &= 56.4943 \times 0.24 \\
 &= 13.502484 \text{ kal/mol.k} \\
 &= 13.502484 \text{ kkal/kmol.k} \\
 &= \frac{13.502484}{130.2257} \text{ kkal/kmol.k} \\
 &= \mathbf{0.1036852 \text{ kkal/kg.K}}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H &= m \times c_p \times \Delta T \\
 &= 5.07825562 \times 0.103685231 \times 75 \\
 &= 39.49050747 \text{ kkal}
 \end{aligned}$$

$$\begin{aligned}
 C_p H_2O &= A + BT + CT^2 + CT^3 \\
 &= 92.0530 + -14.9085 + -29.3840 + 27.7813 \\
 &= 75.54182 \text{ J/mol.K} \\
 &= 75.54182 \times 0.24 \\
 &= 18.054947 \text{ kal/mol.k} \\
 &= 18.054947 \text{ kkal/kmol.k} \\
 &= \frac{18.054947}{18.0147} \text{ kkal/kmol.k} \\
 &= \mathbf{1.0022352 \text{ kkal/kg.K}}
 \end{aligned}$$

$$\begin{aligned}
 \Delta H &= m \times c_p \times \Delta T \\
 &= 7.758369819 \times 1.002235243 \times 75 \\
 &= 583.1783747 \text{ kkal}
 \end{aligned}$$

Enthalpi Masuk

Komponen	Massa (kg)	Cp (kkal/kg.k)	ΔT (K)	ΔH
				(kkal)
C ₂₄ H ₃₈ O ₄	3793.4721	0.460645357	75	131058.4
C ₈ H ₄ O ₃	2.8906993	0.39041044	75	84.64194
C ₈ H ₆ O ₄	2.8993975	0.297320572	75	64.653789
C ₈ H ₁₈ O	5.0782556	0.103685231	75	39.490507
H ₂ O	7.7583698	1.002235243	75	583.17837
Total				131830.36



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

Panas keluar Cooler

$$\begin{aligned}\text{Tekanan} &= 1 \text{ atm} \\ \text{Suhu Operasi} &= 30 \text{ }^\circ\text{C} \\ &= 303.15 \text{ K} \\ \text{Treff} &= 25 \text{ }^\circ\text{C} \\ &= 298.15 \text{ K}\end{aligned}$$

$$\begin{aligned}\text{Cp C}_{24}\text{H}_{38}\text{C} &= A + BT + CT^2 + CT^3 \\ &= 370.5240 + 600.35826 + -382.32 + 119.29 \\ &= 707.8544 \text{ J/mol.K} \\ &= 707.8544 \times 0.24 \\ &= 169.18145 \text{ kal/mol.k} \\ &= 169.18145 \text{ kkal/kmol.k} \\ &= \frac{169.18145 \text{ kkal/kmol.k}}{390.5507 \text{ kg/kmol}} \\ &= \mathbf{0.4331869 \text{ kkal/kg.K}}\end{aligned}$$

$$\begin{aligned}\Delta H &= m \times cp \times \Delta T \\ &= 3793.47214 \times 0.433186892 \times 5 \\ &= 8216.412037 \text{ kkal}\end{aligned}$$

$$\begin{aligned}\text{Cp C}_8\text{H}_4\text{O}_3 &= A + BT + CT^2 + CT^3 \\ &= -105.6270 + 601.44960 + -357.0036 + 79.4 \\ &= 218.25 \text{ J/mol.K} \\ &= 218.25 \times 0.24 \\ &= 52.164172 \text{ kal/mol.k} \\ &= 52.164172 \text{ kkal/kmol.k} \\ &= \frac{52.164172 \text{ kkal/kmol.k}}{148.1140 \text{ kg/kmol}} \\ &= \mathbf{0.3521894 \text{ kkal/kg.K}}\end{aligned}$$

$$\begin{aligned}\Delta H &= m \times cp \times \Delta T \\ &= 2.89069932 \times 0.352189434 \times 5 \\ &= 5.090368788 \text{ kkal}\end{aligned}$$

$$\begin{aligned}\text{Cp C}_8\text{H}_6\text{O}_4 &= A + BT + CT^2 + CT^3 \\ &= -561.000 + 1048.8990 + -423.6586 + 63.5196 \\ &= 127.760 \text{ J/mol.K} \\ &= 127.75993 \times 0.24 \\ &= 30.53539 \text{ kal/mol.k}\end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned} &= 30.53539 \text{ kkal/kmol.k} \\ &= \frac{30.53539 \text{ kkal/kmol.k}}{166.1286 \text{ kg/kmol}} \\ &= \mathbf{0.1838057 \text{ kkal/kg.K}} \end{aligned}$$
$$\begin{aligned} \Delta H &= m \quad \times \quad cp \quad \times \quad \Delta T \\ &= 2.899397513 \quad \times \quad 0.183805692 \quad \times \quad 5 \\ &= 2.664628838 \quad \text{kkal} \end{aligned}$$
$$\begin{aligned} C_p C_8H_{18}O &= A \quad + \quad BT \quad + \quad CT^2 \quad + \quad CT^3 \\ &= -105.6270 \quad + \quad 296.7929 \quad + \quad -240.5664 \quad + \quad 86.4813 \\ &= 37.08086 \text{ J/mol.K} \\ &= 37.08086 \quad \times \quad \mathbf{0.24} \\ &= 8.8625475 \text{ kal/mol.k} \\ &= 8.8625475 \text{ kkal/kmol.k} \\ &= \frac{8.8625475 \text{ kkal/kmol.k}}{130.2257 \text{ kg/kmol}} \\ &= \mathbf{0.0680553 \text{ kkal/kg.K}} \end{aligned}$$
$$\begin{aligned} \Delta H &= m \quad \times \quad cp \quad \times \quad \Delta T \\ &= 5.078255562 \quad \times \quad 0.068055277 \quad \times \quad 5 \\ &= 1.728010454 \quad \text{kkal} \end{aligned}$$
$$\begin{aligned} C_p H_2O &= A \quad + \quad BT \quad + \quad CT^2 \quad + \quad CT^3 \\ &= 92.0530 \quad + \quad -12.1118 \quad + \quad -19.3936 \quad + \quad 14.8962 \\ &= 75.443783 \text{ J/mol.K} \\ &= 75.443783 \quad \times \quad \mathbf{0.24} \\ &= 18.031517 \text{ kal/mol.k} \\ &= 18.031517 \text{ kkal/kmol.k} \\ &= \frac{18.031517 \text{ kkal/kmol.k}}{18.0147 \text{ kg/kmol}} \\ &= \mathbf{1.0009346 \text{ kkal/kg.K}} \end{aligned}$$
$$\begin{aligned} \Delta H &= m \quad \times \quad cp \quad \times \quad \Delta T \\ &= 7.758369819 \quad \times \quad 1.000934614 \quad \times \quad 5 \\ &= 38.8281045 \quad \text{kkal} \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

Enthalpi Keluar

Komponen	Massa (kg)	Cp (kkal/kg.k)	ΔT (K)	ΔH
				(kkal)
C ₂₄ H ₃₈ O ₄	3793.4721	0.433186892	5	8216.412
C ₈ H ₄ O ₃	2.8907	0.352189434	5	5.0903688
C ₈ H ₆ O ₄	2.8994	0.183805692	5	2.6646288
C ₈ H ₁₈ O	5.0783	0.068055277	5	1.7280105
H ₂ O	7.7584	1.000934614	5	38.828104
Total				8264.7231

Q serap

Suhu air pendingin masuk = 30 °C = 303 K (Ulrich : 427)
 Suhu reference = 25 °C = 298 K
 Suhu air pendingin keluar = 45 °C = 318 K (Ulrich : 427)

Perhitungan kapasitas panas H₂O_(g)

$$\int Cp dT = AT + \frac{B}{2}T^2 + \frac{C}{3}T^3 + \frac{D}{4}T^4 + \frac{D}{5}T^5$$

$$= 170 + -12.65526 + 13.516391 + -2.422218 + 0.1509042$$

$$= 168.25482 \text{ J/mol.K}$$

$$= 168.25482 \times 0.24$$

$$= 40.213911 \text{ kal/mol.k}$$

$$= 40.213911 \text{ kkal/kmol.k}$$

$$= \frac{40.213911}{18.0147} \text{ kkal/kg.K}$$

$$= 2.2322856 \text{ kkal/kg.K}$$

Qserap = m air pendingin x Cp air x ΔT
 = m air pendingin x 2.2323 x 15
 = m air pendingin x 33.4843

Qloss = 5% x Qserap
 = 5% x m air pendingin x 33.4843
 = m air pendingin x 1.6742



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetra-butyl Titanate Melalui Proses Esterifikasi

Neraca Panas Total :

$$\begin{aligned} \Delta H \text{ bahan masuk} &= \Delta H \text{ bahan keluar} + Q_{\text{serap}} + Q_{\text{loss}} \\ 131830.3641 &= 8264.723149 + (m_{\text{air pendingin}} \times 33.484284) \\ &+ (m_{\text{air pendingin}} \times 1.6742142) \end{aligned}$$

$$m_{\text{air pendingin}} = \frac{\Delta H \text{ bahan masuk} - \Delta H \text{ bahan keluar}}{33.484284 - 1.6742142}$$

$$\begin{aligned} m_{\text{air pendingin}} &= 3514.5312 \text{ kg} \\ \text{sehingga, } Q_{\text{serap}} &= 117681.56 \text{ kkal} \\ Q_{\text{loss}} &= 5884.0781 \text{ kkal} \end{aligned}$$

NERACA PANAS COOLER			
Panas Masuk		Panas Keluar	
Komponen	ΔH (kkal)	Komponen	ΔH (kkal)
$C_{24}H_{38}O_4$	131058.3995	$C_{24}H_{38}O_4$	8216.412037
$C_8H_4O_3$	84.64193953	$C_8H_4O_3$	5.090368788
$C_8H_6O_4$	64.65378946	$C_8H_6O_4$	2.664628838
$C_8H_{18}O$	39.49050747	$C_8H_{18}O$	1.728010454
H_2O	583.1783747	H_2O	38.8281045
		Q_{serap}	117681.5628
		Q_{loss}	5884.0781
Total	131830.3641	Total	131830.3641



PERANCANGAN PABRIK

Pabrik Diocetyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

APPENDIX C SPESIFIKASI ALAT

1 GUDANG PENYIMPANAN KARBON AKTIF (F-110)

Temperatur = 30 C
Tekanan = 1 atm
Waktu penyimpanan = 30 hari
Fungsi = Untuk menyimpan bahan baku berupa karbon aktif
Tipe = berbentuk balok beratap asbes

Komponen karbon aktif masuk

Komponen	%Berat	Massa (Kg)	ρ (gr/ml)
Karbon aktif	1	1.906049431	1.29568
Total	1	1.906049431	

$$\rho \text{ Campura} = \frac{1.000}{\sum \text{Fraksi berat}} \times 62$$

$$= \frac{1}{0.771796} \text{ gr} \times 62 \frac{\text{lbm/cuft}}{\text{gr/ml}}$$

$$= 80.8893 \text{ lbm/cuft}$$

$$\text{Rate massa} = 1.906049 \text{ kg/jam}$$

$$= 1.906049 \times 2.205 \text{ (konversi ke lb/jam)}$$

$$= 4.202839 \text{ lb/jam}$$

$$\text{Rate volumet} = \frac{\text{Rate Massa}}{\rho \text{ Campuran}}$$

$$= \frac{4.202839 \text{ lb/jam}}{80.8893 \text{ lb/cuft}}$$

$$= 0.051958 \text{ cuft/jam}$$

DITENTUKAN

$$\text{Waktu tinggal} = 30 \text{ hari}$$

$$= 720 \text{ jam}$$

$$\text{Tinggi} = 1 \text{ H}$$

$$\text{Panjang} = \text{Lebar} = 2 \text{ H}$$

$$\text{Volume bahan} = \text{Rate volumet} \times \text{Waktu tinggal (jam)}$$

$$= 0.052 \times 720$$

$$= 37.41 \text{ cuft}$$

Volume gudang direncanakan 80% terisi bahan

$$\text{Volume gudang} = \frac{100}{80} \times 37$$

$$= 46.76 \text{ cuft}$$

$$\text{Volume gudang} = P \times L \times H$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned} 46.76212 \text{ cuft} &= 2H \times 2H \times H \\ 46.76212 \text{ cuft} &= 4 H^3 \\ H^3 &= 11.69 \text{ ft} \\ &= 2.27 \\ &= 3.28 \quad \text{(Konversi ft ke m)} \\ &= 0.692 \text{ m} \end{aligned}$$

Sehingga

$$\begin{aligned} \text{Panjang} &= 2H = 1.384 \text{ m} = 4.539 \text{ ft} \\ \text{Lebar} &= 2H = 1.384 \text{ m} = 4.539 \text{ ft} \\ \text{Tinggi} &= H = 0.692 \text{ m} = 2.27 \text{ ft} \end{aligned}$$

SPEKIFIKASI GUDANG KARBON AKTIF

Nama	= Gudang Garam Rakyat
Fungsi	= Untuk menyimpan bahan baku berupa karbon aktif
Kapasitas	= 37.40969 cuft
Panjang	= 4.539152 ft
Lebar	= 4.539152 ft
Tinggi	= 2.269576 ft
Bahan Konstruksi	= Beton
Jumlah	= 1 buah

2 SCREW CONVEYOR (J-111)

Fungsi	= memindahkan karbon aktif dari gudang ke bucket elevator
Type	= Plain spouts or chutes.
Dasar pemilihan	= Umum digunakan untuk padatan dengan sistem tertutup.

Perhitungan :

$$\begin{aligned} \text{Rate massa bahan masuk} &= 1.9060 \text{ kg/jam} \\ &= 4.2021 \text{ lb/jam} \\ \rho \text{ campuran} &= 80.8893 \text{ lb/cuft} \\ \text{Rate volumetri} &= \frac{\text{Rate massa}}{\rho \text{ campuran}} \\ &= \frac{4.2021}{80.8893} \\ &= 0.0519 \text{ cuft/jam} \\ &= 0.0009 \text{ cuft/menit} \end{aligned}$$

Asumsi jarak screw conveyer = 5 ft = 1.524 m.

Untuk densitas 80.8893 lb/cuft, bahan termasuk kelas D dengan

F = 4 (Badger, Tabel 16-6)



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\text{Power motor} = \frac{C \cdot L \cdot W \cdot F}{33,000} \quad (\text{Badger, pers. 16-4})$$

dengan : C = kapasitas ; cuft/menit

L = panjang ; ft

W = densitas bahan ; lb/cuft

F = faktor bahan

Asumsi panjang screw = 20 ft = 6.096 m

$$\begin{aligned} \text{Power motor} &= \frac{C \cdot L \cdot W \cdot F}{33,000} \\ &= \frac{0.0009 \times 20 \times 80.8893 \times 4}{33,000} \\ &= 0.0002 \text{ hp} \end{aligned}$$

Untuk power < 2 hp, maka dikalikan 2 (Badger : 713)

$$\begin{aligned} \text{Power motor} &= 0.0002 \times 2 \\ &= 0.0003 \text{ hp} \end{aligned}$$

Effisiensi motor = 80%

Maka :

$$\begin{aligned} \text{Power motor} &= \frac{0.0003}{80\%} \\ &= 0.0004 \\ &\approx 0 \text{ hp} \end{aligned}$$

Badger, fig. 16-20 untuk kapasitas = 0.051949 cuft/jam digunakan ukuran :

Diameter = 4 in Dasar pemilihan ini(?)

Kecepatan putarar = 27 rpm

SPEKIFIKASI SCREW CONVEYOR

Kapasitas : 0.0519 cuft/jam

Panjang : 20 ft = 6.0960 m

Diameter : 4 in = 0.1016 m

Kecepatan putarar : 27 rpm

Power : 0.000424 hp

Jumlah : 1 buah

3 BUCKET ELEVATOR (J-112)

Tempe = 30 C

Tekana = 1 atm

Fungsi = Mengangkut karbon aktif dari belt conveyor menuju hopper

Tipe = Continuous bucket elevator



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

PERHITUNGAN

$$\begin{aligned}
 \text{Rate massa} &= 1.906 \text{ kg/jam} \\
 &= 0.002 \text{ ton/jam} \\
 \rho \text{ Campura} &= 80.89 \text{ lbm/cuft} \\
 \text{Dari perry 7 Ed. T 21-9 dipilih bucket elevator dengan spesifikasi :} \\
 \text{Tinggi bucket} &= 18.71 \text{ ft} \\
 \text{Putaran head shaft} &= 28 \text{ rpm} \\
 \text{kapasitas maksimum} &= 35 \text{ ton/jam} \\
 \text{bucket linear speed} &= 150 \text{ ft/mnt} \\
 \text{maka, kecepatan bucket elevator} &= \frac{\text{Rate massa}}{\text{pasitas maksimu}} \times \text{bucket} \\
 &= \frac{0.001906049}{35} \times 150 \text{ ft/mnt} \\
 &= 0.008 \text{ ft/mnt} \\
 \text{Power head shaft} &= 1.8 \text{ hp} \\
 \text{power tambahan} &= 0.06 \text{ hp tiap ft} \\
 &= 0.06 \times 19 \\
 &= 1.12239 \text{ hp} \\
 \text{power total} &= \text{Pwr.head shaft} + \text{Pwr tambahan} \\
 &= 1.8 + 1 \\
 &= 2.92239 \text{ hp} \\
 \text{ukuran bucket} &= \text{Lebar} \times \text{Proyeksi} \times \text{Kedalaman} \\
 &= "x5.5"x7.75 \\
 \text{bucket spacing} &= 8 \text{ inch} \\
 \text{efisiensi motor} &= 0.8 \\
 \text{maka, motor penggerak yang} &= \frac{\text{Power total}}{\text{fisiensi moto}} \\
 &= \frac{2.92239}{0.8} \\
 &= 4 \text{ hp} = 4 \text{ hp}
 \end{aligned}$$

SPESIFIKASI BUCKET ELEVATOR

Nama	=	Bucket Elevator 1
Fungsi	=	Memindahkan kristal garam dari gudang menuju tangki pelarutan garam
Tipe	=	Continuous bucket elevator
Kapasitas	=	0.001906 ton/jam
Bucket = tinggi buck	=	18.7065 ft
kecepatan	=	0.0081688 ft/menit
bucket spa	=	8 in
ukuran buc	=	"x5.5"x7.75
putaran he	=	28 rpm



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

power = 3.6529875 hp
jumlah = 1 buah

4 HOPPER KARBON AKTIF (F-113)

Temperatu = 30 C
Tekanan = 1 atm
Waktu ting = 1 jam
Fungsi = Menampung sementara karbon aktif sebelum masuk tangki dekolonisasi
Tipe = Silinder dengan tutup bawah berbentuk conical dalam posisi vertikal

PERHITUNGAN

Rate massa = 1.906049 kg/jam
= 4.202839 lb/cuft
 ρ Campuran = 80.8893 lbm/cuft
Rate volumetrik = $\frac{\text{Rate Massa}}{\rho \text{ Campura}}$
= $\frac{4.202839 \text{ lb/jam}}{80.8893 \text{ lb/cuft}}$
= 0.051958 cuft/jam

DITENTUKAN

Waktu tinggal = 1 jam
Tinggi (Hs) = 1.5 D
 α = 45 °
Volume Bahan = 0.052 cuft
Volume tangki direncanakan 80% terisi bahan, maka
Volume tangki = $\frac{100}{80} \times 0.052$
= 0.065 cuft
Volume silinde = $(\pi/4) \times D^2 \times Hs$
= 3.14 x 2 D³
= 9.42 D³
Volume konis = $(\pi D^3) / 24 \text{ tgc (Hesse, 92)}$
= $\frac{3.14 \times D^3}{12 \times 1}$
= 0.262 D³
Jadi, Volume t = Vsilind + V konis
0.064947 cuf = 9.42 D³ + 0.2617 D³
0.064947 cuf = 9.682 D³
D = 5.215 ft



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$H = 2D = 10.429 \text{ ft}$$

MENENTUKAN UKURAN HOPPER DAN KETEBALANNYA

Tinggi Conical

$$H = \frac{\text{tg}\alpha \times (D - m)}{2} \text{ (Hesse, 92)}$$

Keterangan

$$\begin{aligned} \alpha &= \text{sudut conis} &= 45^\circ \\ D &= \text{diameter tangki} &= \text{ft} \\ m &= \text{flat spot center} &= 12 \text{ in} \\ & &= 1 \text{ ft} \end{aligned}$$

Maka

$$\begin{aligned} H &= \frac{\text{tg}\alpha \times (D - m)}{2} \\ &= \text{tg}45 \times \frac{5.214566238 - 1}{2} \\ &= 1.607283119 \text{ ft} \end{aligned}$$

$$\begin{aligned} \text{Volume Conical} &= 0,262 H (D^2 + D.m) \text{ (Hesse, 92)} \\ &= 0.262 \times 1.607283 \times 33.40627 \\ &= 14.06765233 \text{ cuft} \end{aligned}$$

$$\begin{aligned} \text{Volume Shell} &= \text{Volume Conical} - \text{Volume tangki} \\ &= 0.064947386 - 14.06765233 \\ &= 14.00270494 \text{ cuft} \end{aligned}$$

$$\begin{aligned} \text{Volume Shell} &= (\pi/4) \times D^2 \times H \\ 14.00270494 &= 3.14 \times 27.1917 \times H \\ H &= 0.16400078 \text{ ft} \end{aligned}$$

$$\begin{aligned} \text{Jadi tinggi total} &= 10.42913248 + 0.16400078 \\ &= 10.59313326 \text{ ft} \end{aligned}$$

$$\begin{aligned} \text{Tekanan Bahan} &= \frac{\text{Density} \times H}{144} \\ &= \frac{1.29568 \times 10.59313326}{144} \\ &= 0.095314659 \text{ psig} \end{aligned}$$

$$P \text{ operasi} = 87.4385 \text{ psig}$$

P design diambil 10% lebih besar dari P operasi untuk faktor keamanan.

$$\begin{aligned} P \text{ design} &= 110\% \times P \text{ perencanaan} \\ &= 110\% \times (87.4385 + 0.095315) \\ &= 96.2872 \text{ psig} \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

MENENTUKAN TEBAL MINIMUM SHELL

Tebal shell berdasarkan ASME Code untuk cylindrical tank :

$$t_{min} = \frac{P \times r}{f_e - 0.6P} + C \quad (\text{Brownell, pers 13-1 hal 266})$$

Dipilih

f = 12650 bahan konstruksi carbon stell SA-283 grade C

e = 0.8

c = 0.125 in

r = 0.5 D in

= 0.5 x 5.214566238 ft

= 2.607283119 ft

= 31.28739743 in

t_{min} = $\frac{96.28719612 \times 31.28739743}{10120 - 57.77231767} + 0.125$

= 10120 - 57.77231767

= 0.424 in

Maka dipilih ts = 7/16 in

PENENTUAN TEBAL HEAD

Jenis = Conical

Tipe Las = Single welded butt joint tanpa backing up strip dengan efisiensi

th = $\frac{P \times D}{2 \cos \alpha - (f.E - 0.6 P)} + C$

= $\frac{96.28719612 \times 31.28739743}{2 \cos \alpha - (10120 - 57.77231767)} + 0.125$

= 1.09548465

maka digunakan th = 1 2/16 in

SPESIFIKASI HOPPER

Fungsi = Menampung garam rakyat sementara sebelum menuju tangki pelarutan garam

Tipe = Silinder dengan tutup bawah berbentuk conical dalam posisi vertikal

Kapasitas = 0.051957909 cuft

Diameter dalam silinder = 5.214566238 ft

Tinggi silinder = 10.42913248 ft

Tebal silinder = 7/16 in

Diameter atas conical = 5.214566238 ft

Diameter bawah conical = 1 ft

Tinggi conical = 1.607283119 ft

Cone angel = 45 °

Tebal angel = 1 2/16 in



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Jumlah = 1 buah

5 TANGKI PENYIMPANAN NaOH

Fungsi = Menampung NaOH 48%

Type = silinder tegak, tutup bawah datar dan tutup atas dishead

Dasar pem = umum digunakan pada tekanan atmosfer

Jumlah = 1 buah

Kondisi O_f = $T = 30\text{ }^\circ\text{C} = 303\text{ K}$

$P = 1\text{ atm} = 760\text{ mmHg} = 14\text{ psi}$

Feed masuk = 0.007 kg/jam

= 0.016 lb/jam

Komponen	Massa (kg/jam)	Fraksi Massa	Sg	Fraksi/sg
NaOH	2.81257E-05	0.48	2.13	0.225352
H ₂ O	3.04695E-05	0.52	1	0.52
Total	5.85953E-05			0.745352

Densitas refer = 62.43700000 lb/cuft

rho campuran = $\frac{62.43700}{\sum \text{fr massa/sg}}$

= 83.768 lb/cuft

rate volumetrik = $\frac{0.015504}{83.7685}$

= 0.0002 cuft/jam

direncanakan penyimpanan untuk 7 hari dengan 1 buah tangki

volume tangki = $\frac{0.00019 \times 7.00 \times 24.000}{1.000}$

= 0.031 cuft

direncanakan tangki terisi 80%

sehingga volur = 0.03109

0.80000

= 0.03887 cuft

menentukan ukuran tangki dan ketebalannya :

asumsi $\frac{H}{D} = 1.000$

volume = $\frac{\pi}{4} \times D^2 \times H$

0.039 = 0.785 x D² x 2.000 D

0.039 = 1.570 D³

D³ = 22.997

D = 10.000 ft = 120.00 in

H = 4.000 ft = 48.000 in



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

penentuan tinggi cairan dalam tangki :

$$\text{Vol shell} = \text{Vol tangki} = \text{Vol baha} = 0.038868 \text{ cuft}$$

$$\text{tinggi bahan} = \text{tinggi shell} = 4.000000 \text{ ft}$$

menentukan tebal minimum shell

tebal shell berdasarkan ASME Code untuk cylindrical tank

$$t_{\min} = \frac{P \times r_i + C}{f E - 0,6P}$$

dengan :

t_{mi} = tebal shell minimum, in

P = tekanan tangki, psi

r_i = jari2 tangki, in (1/2 D)

C = faktor korosi, in (digunakan 1/8 in)

E = faktor pengelasan, digunakan double welded butt join, E=0,8

f = stress allowable, bahan komstruksi Carbon steel SA-283 grade C, maka f = 12650 psi (brownell&young)

Poperasi = Phidrostatic + Patmospheric

$$= \frac{\rho \times g \times h}{144.000}$$

$$= 2.327 \text{ psi}$$

Poperasi = 2.327 + 14.700

$$= 17.027 \text{ psi}$$

P_{design} diambil 10% lebih besar dari dari Poperasi untuk faktor keamanan

$$P_{\text{design}} = 1.100 \times 17.027$$

$$= 18.730 \text{ psi}$$

$$r_i = 60$$

$$= 60 \text{ in}$$

$$t_{\min} = \frac{18.730 \times 60}{(12650 \times 1) - (1 \times 19)} + 0$$

$$= 0.236 \text{ in}$$

maka digunakan = 1/4 in

untuk tebal tutup atas karena tekanan atmospheric,

maka tebal tutup atas disamakan dengan tebal tutup

$$s_{\text{he}} = 1/4 \text{ in}$$

untuk tebal tutup bawah datar karena tutup menumpang di atas semen,

maka tebal = 1/4 in (brownell, hal 58)

Tangki Penampung NaOH 48%

fungsi = Menampung NaOH 48%

type = silinder tegak, tutup bawah datar dan tutup atas dishead

$$\text{volume} = 0.031 \text{ cuf} = 0.880 \text{ L}$$

$$\text{diameter} = 10 \text{ ft} = 3.048 \text{ m}$$

$$\text{tinggi} = 4 \text{ ft} = 1.219 \text{ m}$$



PERANCANGAN PABRIK

Pabrik Diocetyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Tebal shell = 1/4 in
tebal tutup = 1/4 in
tebal tutup = 1/4 in
bahan kons = carbon steel SA-283 grade C (brownell)
jumlah = 1 buah

6 POMPA NaOH (L-121)

Fungsi : Mengalirkan NaOH dari tangki penyimpanan NaOH menuju *mixer*
Tipe : *Centrifugal pump*
Jumlah : 1 buah

Menghitung Tenaga Pompa

Rate masu = 1449.699 kg/jam = 0.8878 lb/s
Densitas = 1876.707 kg/m³ = 117.1628 lb/ft³
 μ = 1.164 cP = 0.0008 lb/ft.s

$$\begin{aligned} \text{Rate volumetrik (Q)} &= \frac{\text{massa}}{r} \\ &= \frac{0.887779 \text{ lb/s}}{117.1628 \text{ lb/ft}^3} \\ &= 8.E-03 \text{ ft}^3/\text{s} \\ &= 3.4009 \text{ gpm} \end{aligned}$$

Diperkirakan aliran fluida laminar ($Re < 2100$), sehingga digunakan persamaan

$Di < 1$ in yaitu :

$$Di_{opt} = 4 Q^{0.40} \mu^{0.20} \quad \text{Eq. 48. (Peters \& Timmerhaus, 1991:365)}$$

Dimana :

Di_{opt} = diameter dalam optimum, in

Q = kecepatan volumetrik, ft³/s

μ = viskositas fluida, lb/ft.s

Sehingga:

$$\begin{aligned} Di_{opt} &= 3.6 \times (8.E-03)^{0.40} \times (0.001)^{0.20} \\ &= 0.122 \text{ in} \end{aligned}$$

Dari App. A.5-1 (Geankoplis, 1993:892) dipilih spesifikasi :

NPS = 1/8 in
OD = 0.405 in = 0.034 ft = 0 m
ID = 0.269 in = 0.022 ft = 0 m
A = 0.000 ft² = 0.372 in²
 Sch = 40



PERANCANGAN PABRIK

Pabrik Diocetyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Menghitung kecepatan linier :

Kecepatan linier fluida dapat dicari dengan menggunakan persamaan sebagai berikut :

$$V = \frac{Q}{A}$$

Sehingga :

$$V = \frac{0.008 \text{ ft}^3/\text{s}}{0.000 \text{ ft}^2} = 18.94 \text{ ft/s} = 5.774 \text{ m/s}$$

Menghitung *Reynold Number* (Nre):

$$Nre = \frac{\rho v D}{\mu}$$

Sehingga:

$$\begin{aligned} Nre &= \frac{117.2 \times 18.94 \times 0.02}{0.00} \\ &= 63594.45 \quad (\text{asumsi aliran laminar benar}) \end{aligned}$$

Instalasi pipa

Dari **Fig. 127**. (Brown, 1950:141) dengan NPS = 1/8 in

- 1 buah <i>gate valve fully open</i> ; Le	=	0.2	ft
$\sum Le$	=	1 × 0.2	= 0.2 ft
- 4 buah <i>standard elbow</i> ; Le	=	1	ft
$\sum Le$	=	4 × 1	= 4 ft
- 1 buah <i>sudden enlargement</i> ; Le	=	4.5	ft
$\sum Le$	=	1 × 4.5	= 4.5 ft
- 1 buah <i>sudden contraction</i> ; Le	=	2.5	ft
$\sum Le$	=	1 × 2.5	= 2.5 ft
- 1 buah <i>swing check valve</i> ; Le	=	2.5	ft
$\sum Le$	=	1 × 2.5	= 2.5 ft
- Panjang ekivalen pipa lurus, $\sum Le$	=	13.7	ft
Panjang pipa lurus	=	2	m
	=	6.562	ft
Panjang pipa total	=	20.26	ft
	=	6.176	m

Menghitung *friction loss*

$$\Sigma F = \frac{4f \cdot v^2 \cdot \Delta L}{2 \cdot ID \cdot gc} \quad (\text{Eq. 2.10-6 Geankoplis, 1993: 89})$$

Dimana :

$$\begin{aligned} \Sigma F &= \text{Friction lo. (ft.lbf/lbm)} \\ f &= \text{Faktor friksi} \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}v &= \text{Kecepatan linier fluida (ft/s)} \\ \Delta L &= \text{Panjang pipa (ft)} \\ ID &= \text{Diameter dalam tangki (ft)} \\ gc &= 32.174 \text{ lbf}\cdot\text{ft/lb}\cdot\text{s}^2\end{aligned}$$

Menghitung *Fanning Friction Factor* (f) :

Dari **Fig. 2.10-3.** (Geankoplis, 1993:88) didapat :

$$\begin{aligned}\text{Untuk commercial steel} \quad \rightarrow \quad \varepsilon &= 0.000048 \text{ m} \\ &= 0.000157 \text{ ft}\end{aligned}$$

Sehingga :

$$\frac{\varepsilon}{D} = \frac{0.0002}{0.022} = 0.0070$$

Dari **Fig. 2.10-3.** (Geankoplis, 1993:88) didapat $\Sigma F = 63594.45$

didapatkan = 0.005 sehingga :

$$\begin{aligned}\Sigma F &= \frac{4 \times 0.005 \times (18.94)^2 \times 20.26}{2 \times 0.0224 \times 32.174} \\ &= 100.8111 \frac{\text{lb}_f \cdot \text{ft}}{\text{lb}_m}\end{aligned}$$

Menghitung *Static Head* :

$$Z_1 = 1 \text{ m} = 3.28084 \text{ ft}$$

$$Z_2 = 0.2 \text{ m} = 0.656168 \text{ ft}$$

$$\Delta Z = Z_1 - Z_2 = 2.6247 \text{ ft}$$

$$g/gc = 1 \text{ lbf/lbm}$$

$$\begin{aligned}\Delta Z (g/gc) &= 2.625 \text{ ft} \times 1 \text{ lbf/lbm} \\ &= 2.625 \text{ ft lbf/lbm}\end{aligned}$$

Menghitung *Velocity Head* :

$$V_1 = \text{kecepatan linier fluida dari tangki hidrogen peroksida ke pipa}$$

$$V_2 = \text{kecepatan linier fluida ke heater}$$

$$\text{Karena pada 2 titik reference dianggap sa } 18.9433 \text{ ft/s}$$

$$\text{Sehingga velocity head } (\Delta V^2 / 2gc) = 5.5767$$

Menghitung *Pressure Head* :

$$P_1 = P_2 = 14.7 \text{ Psi}$$

$$\Delta P = 0$$

$$\text{Sehingga, } \Delta P/\rho = 0$$



PERANCANGAN PABRIK

Pabrik Diocetyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Menghitung Energi Mekanik Pompa :

$$- Wf \frac{\Delta V^2}{2 \times \alpha \times} + \Delta z \frac{g}{gc} + \frac{\Delta P}{\rho} + \Sigma F$$

Dimana :

Wf = tenaga yang ditambahkan ke dalam sistem per satuan massa

Sehingga :

$$\begin{aligned} - Wf &= 5.58 + 3 \text{ ft. lbf/l} + 0 + 100.8 \text{ ft. lbf/lbm} \\ &= 109.0124 \text{ ft. lbf/lbm} \end{aligned}$$

Menghitung *Broke Horse Power* (BHP):

$$\text{BHP} = \frac{m. (-Wf)}{550 \cdot \eta}$$

dari **Fig. 14-37.** (Peters & Timmerhaus, 1991:520), untuk Q_f 3.4009 gpm diperoleh η pompa = 0.40

Sehingga:

$$\begin{aligned} \text{BHP} &= \frac{0.888 \times 109.0}{550 \times 0.400} \\ &= 0.439905 \text{ Hp; maka digunakan } power \\ &= 1 \text{ Hp} \end{aligned}$$

Menghitung Tenaga Motor

Fig. 14.38., (Peters & Timmerhaus, 1991:521) untuk BH = 1.0 Hp diperoleh η_{mc} = 0.80

Sehingga *power* motor yang diperlukan :

$$\begin{aligned} P \text{ motor} &= \frac{\text{BHP}}{\eta} \\ &= \frac{1}{0.80} \text{ Hp} \\ &= 1.3 \text{ Hp} \end{aligned}$$

Dipilih motor standar dengan = 2 Hp (*Standard NEMA*)

Spesifikasi Pompa NaOH

Nama Alat : Pompa NaOH
Kode : L-121
Fungsi : Mengalirkan NaOH dari tangki penyimpanan NaOH menuju *mixer*
Tipe : *Centrifugal pump*
Bahan Konstruksi : *Commercial steel*
Jumlah : 1 buah
Rate Volumetrik : 8.E-03 ft³/s
Kecepatan Aliran : 18.9433 ft/s
Ukuran Pipa : NPS = 1/8 in



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

: *Sch. Numb* = 40
 : OD = 0.0338 ft = 0.01 m
 : ID = 0.0224 ft = 0.007 m
 : *Flow Area* = 0.0004 ft² = 4E-05 m²

Power Pompa : 1 Hp
Power Motor : 2 Hp

7 MIXER NaOH (M-122)

Fungsi : Menambahkan H₂O untuk mengencerkan NaOH
Tipe : Tangki berbentuk silinder tegak, tutup atas dan bawah berbentuk *torispherical*
Jumlah : 1 buah
Kondisi Op : T = 30 °C = 303 K
 P = 1 atm = 760 mmHg = 14.7 psi

Menentukan Kapasitas Tangki

Massa Masuk		Massa Keluar	
1. NaOH 48%		1. NaOH 0.4%	
Komponen	massa (Kg/Ja)	Komponen	massa (Kg/Ja)
NaOH	2.81E-05	NaOH _(aq)	0.0070314
H ₂ O	3.05E-05	Total	0.0070314
Total	5.86E-05		
2. Air Proses			
Komponen	massa (Kg/Ja)		
H ₂ O	0.006973		
Total	0.006973		
Total	0.007031	Total	0.007031

T = 30 °C = 303.15 K

Density : $A \cdot B^{-(1-T/T_c)^n}$ (Yaws, 1989)

Komponen	A	B	n	T _c (K)	ρ (kg/m ³)
NaOH	0.2	0.098	0.254	2820	1909.231
H ₂ O	0.347	0.274	0.286	647.1	1022.875

Viscosity : $\log_{10} \mu_{liq} = A + B/T + C/T + D/T^2$ (Yaws, 1989)

Komponen	A	B	C	D	μ (cP)
NaOH	-4.19	#####	0.003	-6E-07	2307.170
H ₂ O	-10.2	1793	0.018	-1E-05	0.815



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Komponen	%Berat	ρ (kg/m ³)	ρ_{Campuran}	μ (cP)	μ_{campuran} (cP)
			(kg/m ³)		
NaOH	0.0040	1909.231	0.0000	2307.170	2.E-06
H ₂ O	0.9960	1022.875	0.0010	0.8150	1.2220
Total	1.0000	2932.107	0.0010	2307.985	1.2220

$$\rho_{\text{campuran}} = \frac{\sum x_i}{\sum x_i / \rho} = \frac{1.0000}{0.0010} = 1024.778 \text{ kg/m}^3$$

$$= 63.9769 \text{ lb/ft}^3$$

$$\mu_{\text{campuran}} = \frac{\sum x_i}{\sum x_i / \mu} = \frac{1.0000}{1.2220} = 0.8183 \text{ cP}$$

$$= 0.0005 \text{ lb/ft.s}$$

Aliran keluar *mixer* dijaga selalu berada pada 0.007 kg/jam. Sehingga tidak terjadi defisit aliran pada *mixer*, karena beroperasi kontinyu (tidak ada waktu tinggal) dianggap pengadukan sempurna 100%. Maka :

$$V_{\text{cairan}} = \frac{m}{\rho} = \frac{0.007 \text{ kg/jam}}{1025 \text{ kg/m}^3} = 0.000007 \text{ m}^3$$

$$= 0.0002 \text{ ft}^3$$

Untuk perancangan diasumsikan tangki berisi 80% sehingga volume desain tangki sebagai berikut :

$$\text{Volume tai} = 0.000007 \text{ m}^3 \div 80\% = 0.000009 \text{ m}^3$$

$$= 0.000303 \text{ ft}^3$$

$$= 0.002266 \text{ galon}$$

Menentukan Diameter dan Tinggi *Shell*

Tangki dirancang dengan tutup atas berupa *torispherical*.

Untuk ukuran optimum tangki silinder, harga $H_s/ID = 2$

$$\text{Asumsi: } H_s/ID = 2$$

$$H_s = 2 \text{ ID}$$

$$V_{\text{tangki}} = V_{\text{silinder}} + 2 \times V_{\text{tutup}}$$

$$V_{\text{tangki}} = \left(\frac{1}{4} \times (\pi \times ID^2)\right) \times H_s + 2 \times (0,000049 \text{ ID}^3)$$

$$0.000009 = \left(\frac{1}{4} \times (3,14 \times ID^2)\right) \times 2 \text{ ID} + 2 \times (0,000049 \text{ ID}^3)$$

$$0.000009 = 1,57 \text{ ID} + 0,000098 \text{ ID}^3$$

$$ID^3 = 0.000005$$

$$ID = 0.0176 \text{ m} = 0.058 \text{ ft} = 0.692 \text{ in}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}H_S &= 2 \times ID \\ &= 2 \times 0.0176 \text{ m} \\ &= 0.035 \text{ m} = 0.1156 \text{ ft}\end{aligned}$$

Menentukan Tinggi Cairan pada Tangki

$$\begin{aligned}H_L &= \frac{V_L}{\frac{1}{4} \times \pi \times ID^2} = \frac{0.0002}{\frac{1}{4} \times \pi \times (0.0578)^2} = 0.092 \text{ ft} \\ &= 0.028 \text{ m}\end{aligned}$$

Menentukan Tekanan Desain

$$\begin{aligned}P_{\text{operasi}} &= 1 \text{ atm} = 14.7 \text{ psi} \\ P_{\text{hidrostatik}} &= \rho_{\text{campuran cair}} \times \frac{g}{gc} \times H_S \\ &= 63.98 \text{ lb/ft}^3 \times 1 \times 0.1156 \text{ ft} \\ &= 7.39 \text{ lb/ft}^2 = 0.0513 \text{ psi}\end{aligned}$$

$$\begin{aligned}P_{\text{desain}} &= P_{\text{operasi}} + P_{\text{hidrostatik}} \\ &= 14.7 + 0.0513 \\ &= 14.7473 \text{ psi}\end{aligned}$$

Menentukan Tebal Shell

Dengan faktor keamanan 20%, maka :

$$\begin{aligned}P_d &= \text{tekanan desain} = 1.2 \times 14.75 \text{ psi} = 17.70 \text{ Psi} \\ r &= \text{jari-jari shell} = 0.692 : 2 = 0.3459 \text{ in} \\ f &= \text{allowable stress} = 12650 \text{ psi} \\ &\quad \text{Carbon steel SA-283 GrA (Brownell \& Young, 1959:251)} \\ E &= \text{efisiensi pengelasan double-we} = 80\% \\ c &= \text{faktor koreksi} = 0.125\end{aligned}$$

Sehingga :

$$\begin{aligned}t_s &= \frac{P_d \cdot r}{f \cdot E - 0.6 P_d} + c \\ &\quad \text{(Pers 13.1 Brownell \& Young, 1959 : 254)} \\ &= \frac{17.6968 \times 0.3459}{12650 \times 80\% - 0.6 \times 17.70} + 0.125 \\ &= 0.1256 \text{ in} \\ &\approx 3/16 \text{ in} \text{ Tabel 5.7 (Brownell \& Young, 1959:89)}\end{aligned}$$

Menentukan Diameter dan Tinggi yang Distandarkan

Diameter standar (Ds)

$$OD = ID + 2 t_s$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
 &= 0.692 \text{ in} + 2 \times 3/16 \text{ in} \\
 &= 1.067 \text{ in}
 \end{aligned}$$

Sehingga :

$$OD_{\text{standar}} = 28 \text{ in} \text{ Tabel 5.7 (Brownell \& Young, 1959:89)}$$

$$\begin{aligned}
 ID &= OD_{\text{standar}} - 2 t_s \\
 &= 28 \text{ in} - 2 \times 3/16 \text{ in} \\
 &= 28 \text{ in} \\
 &= 0.7017 \text{ m}
 \end{aligned}$$

Diperoleh OD 28 in dengan :

$$i_{cr} = 1 \frac{3}{4} \text{ in}$$

$$r = 26 \text{ in}$$

Tinggi standar (H_s)

$$\begin{aligned}
 H_s &= 2 \times 28 \text{ in} \\
 &= 55 \text{ in} \\
 &= 1.4034 \text{ m}
 \end{aligned}$$

Menentukan Dimensi pada Tutup

$$\begin{aligned}
 th &= \frac{0.885 P_d r}{f \cdot E - 0.1 P_d} + c \\
 &= \frac{0.885 \times 17.70 \times 0.35}{12650 \times 80\% - 0.1 \times 17.70} + 0.125 \\
 &= 0.1255 \text{ in}
 \end{aligned}$$

Digunakan tebal standar = 3/16 in (Brownell & Young, 1959:89)

Dari Tabel 5.6 (Brownell & Young, 1959:88) untuk tebal tutup 3/16 in didapat nilai sf 1 1/2 -2 in Diambil sf: 1 1/2 in

Dari Fig. 5.8 (Brownell & Young, 1959:87), dihitung ukuran-ukuran sebagai berikut :

$$\begin{aligned}
 a &= \frac{1}{2} ID = \frac{1}{2} \times 28 = 13.8125 \text{ in} \\
 AB &= a - i_{cr} = 13.813 - 1 \frac{3}{4} = 12.06 \text{ in} \\
 BC &= r - i_{cr} = 26 - 1 \frac{3}{4} = 24.25 \text{ in} \\
 AC &= \sqrt{(BC)^2 - (AB)^2} = 21.0371 \text{ in} \\
 b &= r - AC = 26 - 21.04 = 4.9629 \text{ in} \\
 OA &= th + b + sf \\
 &= 0 + 4.9629 + 1 \frac{1}{2} = 6.4629 \text{ in} \\
 &= 0.1642 \text{ m}
 \end{aligned}$$

$$\begin{aligned}
 \text{Tinggi tangki } t &= H_s + 2 OA \\
 &= 0.0352 + 2 \times 0.1642 \\
 &= 0.3635 \text{ m}
 \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

Menentukan Jumlah Pengaduk

Dipilih jenis pengaduk *propeler three-blade*

$$SpGr = \frac{\rho_{\text{campuran}}}{\rho_{\text{air}}} = \frac{63.977 \text{ lb/ft}^3}{63.858 \text{ lb/ft}^3} = 1.0019$$

$$\begin{aligned} \text{jumlah pengaduk} &= \frac{H_L \times S_j}{ID} = \frac{0.0282 \text{ m} \times 1.002}{0.7017} \text{ m} \\ &= 0.0402 \\ &\approx 1 \text{ buah} \end{aligned}$$

Menentukan Diameter dan Tinggi Pengaduk

Diameter pengaduk

$ID/d = 3$ (Brown, 1950:507)

$$\begin{aligned} Da &= \frac{ID}{3} = \frac{0.702}{3} = 0.2339 \text{ m} \\ &= 9.2083 \text{ in} \\ &= 0.7674 \text{ ft} \end{aligned}$$

Jarak pengaduk dari dasar tangki

$Z_i/d = 0,75 - 1,3$; (dipilih 0,75) (Brown, 1950:507)

$$\begin{aligned} Z_i &= 0,75 \times Da = 0.75 \times 0.2339 = 0.1754 \text{ m} \\ &= 6.9062 \text{ in} \\ &= 0.5755 \text{ ft} \end{aligned}$$

Menentukan Dimensi Pengaduk

Tabel 3.4-1. (Geankoplis, 1993:144)

$$W/Da = 1/5$$

$$Dd/Da = 2/3$$

$$H/Dt = 1$$

$$L/Da = 1/4$$

$$J/Dt = 1/12$$

$$H = Dt = 28 \text{ in} = 0.711 \text{ m} = 2.333 \text{ ft}$$

$$\begin{aligned} Dd &= 2/3 \times Da = 2/3 \times 9.2083 \text{ in} \\ &= 6.139 \text{ in} = 0.156 \text{ m} = 0.512 \text{ ft} \end{aligned}$$

$$\begin{aligned} L &= 1/4 \times Da = 1/4 \times 9.2083 \text{ in} \\ &= 2.302 \text{ in} = 0.058 \text{ m} = 0.192 \text{ ft} \end{aligned}$$

$$W = 1/5 \times Da = 1/5 \times 9.2083 \text{ in}$$



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$$= 1.842 \text{ in} = 0.047 \text{ m} = 0.153 \text{ ft}$$

$$J = 1/12 \times OD = 1/12 \times 28 \text{ in} \\ = 2.333 \text{ in} = 0.059 \text{ m} = 0.194 \text{ ft}$$

Menentukan Power Motor Pengaduk

Kecepatan putar pengaduk

$$WELH = \frac{\rho_{\text{campuran}}}{\rho_{\text{air}}} \times H_L = \frac{63.98 \text{ lb/ft}^3}{63.86 \text{ lb/ft}^3} \times 0.092 \text{ ft} \\ = 0.09 \text{ ft}$$

$$N = \frac{600}{\pi d} \sqrt{\frac{WELH}{2d}} \quad \text{Eq. 8.8 (Rase \& Holmes, 1977:338)}$$

Keterangan:

Da = diameter pengaduk (ft)

N = kecepatan putar pengaduk (rpm)

WELH = *water equivalent liquid height* (ft)

$$N = \frac{600}{\pi d} \sqrt{\frac{WELH}{2d}} = \frac{600}{3.14 \times 0.7674} \sqrt{\frac{0.0926}{2 \times 0.7674}} \\ = 7.5145 \text{ rpm}$$

Diambil kecepatan pengaduk standar 320 rpm, maka (Walas, 1990:288) :

$$N = 320 \text{ rpm} = 5.3333 \text{ rps}$$

Bilangan Reynold

$$N_{Re} = \frac{\rho \times Da^2 \times N}{\mu} \quad \text{Pers. 3.4-1 (Geankoplis, 1993:158)} \\ = \frac{63.98 \text{ lb/f} \times 0.7674^2 \text{ ft} \times 5.333 \text{ rps}}{0.0005 \text{ lb/ft.s}} \\ = 365915.88$$

Karena $N_{Re} > 10.000$, maka sesuai untuk digunakan *baffle* (Perry, 1997:9-8)

Power pengaduk

Karena $N_{Re} > 10.000$ maka power tidak tergantung pada bilangan Reynold dan bukan merupakan fungsi viskositas. Maka untuk pengaduk jenis *propeller 3 blades* didapat: 0.87 **Tabel 9.3** (Mc Cabe et al., 1993:254). Sehingga :



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$$P = \frac{N_p \cdot \rho \cdot N^3 \cdot D_a^5}{g_c} \quad \text{Pers. 9.20 (Mc Cabe et al., 1993:253)}$$

Keterangan :

P = daya pengaduk (lb.ft/s)

N_p = power number

ρ = densitas campuran (lb/ft³)

N = kecepatan putar pengaduk (rps)

D_a = diameter pengaduk (ft)

g_c = percepatan gra = 32 lb.ft/lbf.s²

$$\begin{aligned} P &= \frac{0.87 \times 63.98 \times 5.3333^3 \times 0.77^5}{32.1740} \\ &= 69.83 \text{ lb.ft/s} \\ &= 0.13 \text{ hp} \end{aligned}$$

Selama proses pengadukan, pengaduk mengalami :

- *Gain Losses* diperhitungkan 10% dari daya masuk (dengan mempertimbangkan adanya kebocoran daya pada proses dan *bearing*)
= 10% × P_o *Gain* = 0.013 hp
- *Transmission Losses* diperhitungkan 20% dari daya masuk (dengan meninjau kemungkinan terjadinya kebocoran *belt* atau *gear*) = 20% × P_o
Transmission i = 0.0254 hp
Power input = 0.1270 + 0.0127 + 0.0254
= 0.1650 hp

Fig . 14.38. (Peters & Timmerhaus, 1991:521) didapat efisiensi motor pengaduk 80%, maka *power* motor pengaduk sebesar :

$$\text{Power motor} = \frac{P}{\eta} = \frac{0.1650}{80\%} = 0.206 \text{ hp}$$

Maka digunakan *power* motor pengad = 1 hp

Menentukan Poros Pengaduk

Jarak antara pengaduk

$$H_i = \frac{H_L - Z_i}{4} = \frac{0.028 - 0.175}{4} = -0.0368 \text{ m}$$

Panjang $p_o = Z_T + Z_p - Z_i$

Keterangan Z_T = tinggi total tangki

Z_p = jarak motor ke *bearing*

Z_i = jarak pengaduk dari dasar tangki

Panjang $p_o = 0.364 + 0.3048 - 0.175 = 0.493 \text{ m}$

Diameter poros



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$$\tau = \frac{HP \times 75 \times 60}{2 \pi v}$$

Keterangan HP = daya pengaduk

v = kecepatan putaran pengaduk

$$\tau = \frac{1 \times 75 \times 60}{2 \times 3.14 \times 320} = 2.239 \text{ kg.m}$$

Dengan faktor keamanan 50% maka:

$$\text{Momen pu} = 2 \times 2.239 = 3.359 \text{ kg.m}$$

$$\frac{\tau}{f_s} = \frac{\pi Dp^3}{16}$$

$$Dp^3 = \frac{16 \tau}{\pi f_s}$$

Keterangan:

$$f_s = \text{shear stres} = 6E+06 \text{ kg/m}^2$$

Dp = diameter poros

$$Dp^3 = \frac{16 \times 3.359}{3.14 \times \text{#####}}$$

$$Dp^3 = 0.0000 \text{ m}^3$$

$$Dp = 0.0146 \text{ m}$$

$$= 1.4600 \text{ cm}$$

$$= 0.5748 \text{ in}$$

Spesifikasi *Mixer* NaOH

Fungsi : Menambahkan H₂O untuk mengencerkan NaOH

Tipe : Tangki berbentuk silinder tegak, tutup atas dan bawah berbentuk *torispherical*

Tipe Pengaduk : *Propeler three-blade*

Bahan Konstru : *Carbon Steel SA-283 Grade C*

Jumlah : 1 buah

Kondisi Opera : Temperatur 30°C dan Tekanan 1 atm

Kapasitas : 0.00001 m³

Dimensi

Shell

Diameter : 0.01761 m

Tinggi : 0.03522 m

Tebal : 0.12561 in



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Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

Tutup

Tinggi : 0.16416 m
Tebal : 0.12554 in
Tinggi Tangki : 0.36354 m
Jumlah Pengac : 2 buah
Power Motor : 1 hp

8 POMPA MIXER (L-123)

Fungsi : Mengalirkan NaOH dari *mixer* menuju *heater*
Tipe : *Centrifugal pump*
Jumlah : 1 buah

Menghitung Tenaga Pompa

Rate masu = 0.007 kg/jam = 0.0000 lb/s lb/s
Densitas = 1025 kg/m³ = 63.977 lb/f lb/ft³
 μ = 1.164 cP = 0.0008 lb/f lb/ft.s

$$\begin{aligned} \text{Rate volumetr} &= \frac{\text{massa}}{\rho} \\ &= \frac{0.000004 \text{ lb/s}}{63.98 \text{ lb/ft}^3} = 0.00000007 \text{ ft}^3/\text{s} \\ &= 0.00003021 \text{ gpm} \end{aligned}$$

Diperkirakan aliran fluida turbulen ($N_{re} < 2100$), sehingga digunakan persamaan $Di \geq 1$ in yaitu :

$$Di_{opt} = 3.9 Q^{0.45} \mu^{0.13} \quad \text{Eq. 45. (Peters \& Timmerhaus, 1991:365)}$$

Dimana :

Di_{opt} = diameter dalam optimum, in
 Q = kecepatan volumetrik, ft³/s
 μ = viskositas fluida, lb/ft.s

Sehingga :

$$\begin{aligned} Di_{opt} &= 3.9 \times (0.0000)^{0.45} \times (0.001)^{0.13} \\ &= 0.00091165 \text{ in} \end{aligned}$$

Dari App. A.5-1 (Geankoplis, 1993:892) dipilih spesifikasi :

NPS = 1/4 in
OD = 0.540 in = 0 ft = 0.014 m
ID = 0.364 in = 0 ft = 0.009 m



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Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$A = 0.001 \text{ ft}^2 = 1 \text{ in}^2$$
$$Sch = 40$$

Menghitung kecepatan linier :

Kecepatan linier fluida dapat dicari dengan menggunakan persamaan berikut :

$$V = \frac{Q}{A}$$

Sehingga:

$$V = \frac{0.000 \text{ ft}^3/\text{s}}{0.001 \text{ ft}^2}$$
$$= 9\text{E-}05 \text{ ft/s}$$
$$= 3\text{E-}05 \text{ m/s}$$

Menghitung *Reynold Number* (Nre) :

$$Nre = \frac{\rho v D}{\mu}$$

Sehingga :

$$Nre = \frac{63.98 \times 0.000 \times 0.0303}{0.0007823}$$
$$= 0.232 \quad (\text{asumsi aliran turbulen benar})$$

Instalasi pipa

Dari **Fig. 127.** (Brown, 1950:141 1/4 in

- 1 buah <i>gate valve fully open</i> ; Le	= 0.26 ft
ΣLe	= 1 \times 0.26 = 0.26 ft
- 3 buah <i>standard elbow</i> ; Le	= 1.4 ft
ΣLe	= 3 \times 1.4 = 4.2 ft
- 1 buah <i>sudden enlargement</i> ; Le	= 1.4 ft
ΣLe	= 1 \times 1.4 = 1.4 ft
- 1 buah <i>sudden contraction</i> ; Le	= 0.52 ft
ΣLe	= 1 \times 0.52 = 0.52 ft
- 1 buah <i>swing check valve</i> ; Le	= 3 ft
ΣLe	= 1 \times 3 = 3 ft
- Panjang ekuivalen pipa lurus, ΣLe	= 9.38 ft
Panjang pipa lurus	= 2 m
	= 6.562 ft



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Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned} \text{Panjang pipa total} &= 15.94 \text{ ft} \\ &= 4.859 \text{ m} \end{aligned}$$

Menghitung *friction loss*

$$\Sigma F = \frac{4f \cdot v^2 \cdot \Delta L}{2 \cdot ID \cdot gc} \quad \text{Eq. 2.10-6 (Geankoplis, 1993:89)}$$

Dimana :

$$\begin{aligned} \Sigma F &= \text{Friction loss} \quad (\text{ft.lbf/lbm}) \\ f &= \text{Faktor friksi} \\ v &= \text{Kecepatan linier fluida (ft/s)} \\ \Delta L &= \text{Panjang pipa (ft)} \\ ID &= \text{Diameter dalam tangki (ft)} \\ gc &= 32.174 \text{ lbm.ft/lbf.s}^2 \end{aligned}$$

Menghitung *Fanning Friction Factor* (f):

Dari **Fig. 2.10-3.** (Geankoplis, 1993:88) didapat :

$$\begin{aligned} \text{Untuk } commercial \rightarrow \epsilon &= 5E-05 \text{ m} \\ &= 0.0002 \text{ ft} \end{aligned}$$

Sehingga :

$$\frac{\epsilon}{D} = \frac{0.00016}{0.0303} = 0.00519164$$

Dari **Fig. 2.10-3.** (Geankoplis, 1993:88) didapat nilai f = 0.231879193

didapatkan nilai f = 0.0030 sehingga :

$$\begin{aligned} \Sigma F &= \frac{4 \times 0.003 \times (0.000)^2 \times 15.94}{2 \times 0.0303 \times 32.17} \\ &= 0.00000000856 \frac{\text{lb}_f \cdot \text{ft}}{\text{lb}_m} \end{aligned}$$

Menghitung *Static Head* :

$$Z_1 = 1 \text{ m} = 3.28084 \text{ ft}$$

$$Z_2 = 0.2 \text{ m} = 0.656168 \text{ ft}$$

$$\Delta Z = Z_1 - Z_2 = 2.6247 \text{ ft}$$

$$g/gc = 1 \text{ lbf/lbm}$$

$$\Delta Z (g/\xi) = 3 \text{ ft} \times 1 \text{ lbf/lbm} = 3 \text{ ft lbf/lbm}$$

Menghitung *Velocity Head* :

$$V_1 = \text{kecepatan linier fluida dari tangki hidrogen peroksida ke pipa}$$

$$V_2 = \text{kecepatan linier fluida ke heater}$$



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Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

Karena pada 2 titik *reference* dianggap sama, maka $V_1 = V_2 = 0.00009348 \text{ ft/s}$
Sehingga *velocity head* $(\Delta V^2 / 2agc) = 0.000000000136$

Menghitung *Pressure Head* :

$$P_1 = P_2 = 14.7 \text{ Psi}$$

$$\Delta P = 0$$

$$\text{Sehingga, } \Delta P / \rho = 0$$

Menghitung Energi Mekanik Pompa :

$$- W_f = \frac{\Delta V^2}{2 \times \alpha \times gc} + \Delta z \frac{g}{gc} + \frac{\Delta P}{\rho} + \sum F$$

Dimana :

W_f = tenaga yang ditambahkan ke dalam sistem per satuan massa

Sehingga :

$$\begin{aligned} - W_f &= 1E-10 + 2.625 \text{ ft. lbf/lbm} + 0 + 9E-10 \text{ ft. lbf/lbm} \\ &= 2.625 \text{ ft. lbf/lbm} \end{aligned}$$

Menghitung *Broke Horse Power* (BHP) :

$$\text{BHP} = \frac{m \cdot (-W)}{550 \cdot \eta}$$

dari **Fig. 14-37.** (Peters & Timmerhaus, 1991:520), untuk $Q_f =$ ##### gpm diperoleh $\eta_{\text{pompa}} = 0.40$

Sehingga :

$$\begin{aligned} \text{BHP} &= \frac{0.0000 \times 2.6247}{550 \times 0.4000} \\ &= 0.0000000514 \text{ Hp; maka digunakan} = 1 \text{ Hp} \end{aligned}$$

Menghitung Tenaga Motor

Fig. 14.38., (Peters & Timmerhaus, 1991:520), diperoleh $\eta_{\text{motor}} = 0.80$

diperoleh $\eta_{\text{motor}} = 0.80$

Sehingga *power* motor yang diperlukan :

$$\begin{aligned} P_{\text{motor}} &= \frac{\text{BHP}}{\eta} \\ &= \frac{1 \text{ Hp}}{0.80} \\ &= 1.3 \text{ Hp} \end{aligned}$$

Dipilih motor standar dengan 2 Hp (*Standard NEMA*)



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Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Spesifikasi Pompa *Mixer*

Kode : L-123
 Fungsi : Mengalirkan NaOH dari *mixer* menuju *heater*
 Tipe : *Centrifugal pump*
 Bahan Konstru : *Commercial steel*
 Jumlah : 1 buah
 Rate Volumet : 0.0000 ft³/s
 Kecepatan Alir : 0.0001 ft/s
 Ukuran Pipa : NPS = 1/4 in
 : *Sch. Numb* = 40
 : OD = 0.0450 ft = 0.014 m
 : ID = 0.0303 ft = 0.009 m
 : *Flow Area* = 0.0007 ft² = 0.000 m²
 Power Pompa : 1 Hp
 Power Motor : 2 Hp

9 HEATER NaOH (E-124)

Fungsi : Memanaskan natrium hidroksida sebelum masuk *neutraliser* suh 30 °C menjadi 95 °C
 Tipe : *Double pipe heat exchanger*
 Jumlah : 1 buah

KOMPONEN	Massa	FRAKSI (x)	ρ (g/ml)	x/ρ
NaOH	2.81E-05	0.00400000	2.13	0.001878
H2O	0.007003	0.99600000	1.00	1.000000
Total	0.01	1.00	3.13	1.00

$$\begin{aligned} \rho \text{ bahan} &= 0.998 \times 62.43 \frac{\text{lb/ft}^3}{\text{gr/ml}} \\ &= 62.31 \text{ lb/ft}^3 \\ \text{Rate bahan} &= 0.01 \text{ lb/jam} \\ \text{Rate volumetri} &= \frac{\text{Rate massa}}{\rho \text{ bahan}} \\ &= 0.0001128405 \text{ ft}^3/\text{jam} \end{aligned}$$

$$\begin{aligned} T \text{ bahan masuk} &= 30.00 \text{ }^\circ\text{C} = 86 \text{ }^\circ\text{F (t1)} \\ T \text{ bahan keluar} &= 95 \text{ }^\circ\text{C} = 203 \text{ }^\circ\text{F (t2)} \\ T \text{ steam masuk} &= 220.00 \text{ }^\circ\text{C} = 428 \text{ }^\circ\text{F (T1)} \\ T \text{ steam keluar} &= 220.00 \text{ }^\circ\text{C} = 428 \text{ }^\circ\text{F (T2)} \end{aligned}$$



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Pabrik Diocetyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned} Q \text{ Supply} &= 0.24979 \text{ kkal/jam} \\ &= 0.98919 \text{ BTU/jam} \\ \text{Kebutuhan Steam} &= 0.00056 \text{ kg/jam} \\ &= 0.00124 \text{ lb/jam} \end{aligned}$$

Log Mean Temperature Difference (ΔT LMTD)

$$\Delta T \text{ LMTD} = \frac{\Delta t_2 - \Delta t_1}{\ln \frac{\Delta t_2}{\Delta t_1}}$$

$$\begin{aligned} \Delta t_1 &= T_1 - t_2 \\ &= 225 \text{ }^\circ\text{F} \end{aligned}$$

$$\begin{aligned} \Delta t_2 &= T_2 - t_1 \\ &= 342 \text{ }^\circ\text{F} \end{aligned}$$

$$\Delta T \text{ LMTD} = 279.4 \text{ }^\circ\text{F}$$

Tc dan tc

$$\begin{aligned} t_c &= T \text{ bahan} \\ &= 144.5 \text{ }^\circ\text{F} \end{aligned}$$

$$\begin{aligned} T_c &= T \text{ stean} \\ &= 428 \text{ }^\circ\text{F} \end{aligned}$$

Trial :

$$\begin{aligned} UD &= 100 - 500 \\ &= 500 \end{aligned}$$

$$\text{Tipe} = 1-2 \text{ Heat Exchanger}$$

$$\text{Length} = 10 \text{ ft}$$

$$OD = 1.00 \text{ in}$$

$$BWG = 14$$

$$\text{Passes} = 2$$

$$a't = 0.546 \text{ in}^2$$

$$a'' = 0.262 \text{ ft}^2 / \text{ft}$$

$$\begin{aligned} A &= \frac{Q}{\Delta T \text{ LMTD} \times UD} \\ &= 0.000007 \text{ ft}^2 \end{aligned}$$

$$\begin{aligned} Nt &= \frac{A}{L \times a''} \\ &= 1.00000 \end{aligned}$$



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Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Berdasarkan Tabel 9 kern dipilih ukuran yang paling mendekati :

$$\begin{aligned}
 \text{OD tube} &= 1 \text{ in} \\
 \text{Pitch} &= 1 \frac{1}{4} \text{ in} \\
 \text{Passes} &= 2 \quad \text{L/D} = 86.96 \\
 \text{Nt} &= 16 \\
 a'' &= 0.262 \text{ ft}^2 / \text{ft} \\
 \text{Wall thickr} &= 0.083 \text{ in} \\
 \text{ID tube} &= 1.38 \text{ in} ; \quad 0.115 \text{ ft} \\
 \text{ID shell} &= 2.38 \text{ in} \\
 \text{B space} &= 1.428 \text{ in} \\
 \text{Passes} &= 1 \\
 A &= \text{Nt} \times L \times a'' \\
 &= 41.9 \text{ ft}^2 \\
 \text{UD} &= \frac{Q}{\Delta T \text{ LMTD} \times A} \\
 &= 0.000084447
 \end{aligned}$$

Hot Fluid :	Cold fluid
o) Flow area, as $as = \text{ID} \times C' \times B / 144 \text{ Pr}$ $B = \text{ID}$ $\quad 5$ $\quad = 0 \text{ in}$ $C' = \text{Pt} - \text{OD tube}$ $\quad 2$ $\quad = 1/8$ $as = 8E-04 \text{ ft}^2$	o) Flow Area, at $at = \text{Nt} \times a$ $\quad 144 \text{ n}$ $\quad = 0.0303 \text{ ft}^2$
o) Mass Velocity, Gs $Gs = \frac{w}{as}$ $\quad = 2 \text{ lb/jam.ft}^2$	o) Mass Velocity, Gt $Gt = \frac{w}{at}$ $\quad = 0.23 \text{ lb/jam.ft}^2$
o) Bilangan Reynold, Ret $Nre = \frac{Ds \times Gt}{\mu}$ $tc = 428 \text{ }^\circ\text{F}$ $\mu = 0.016 \text{ cP}$ $\quad = 0.039 \text{ lb/jam.ft}$ $Ds = 0.99 \text{ in}$ $\quad = 0.083 \text{ ft}$ $Nre = 3.35490$	o) Bilangan Reynold, Ret $Nre = \frac{D \times Gt}{\mu}$ $\mu = 0.316 \text{ cP}$ $\quad = 0.761 \text{ lb/jam.ft}$ $Nre = 0.03505$
o) jH = 580	o) jH = 1000



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Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

Hot Fluid :	Cold fluid
o) $k = 0.9 k_{\text{water}}$ $= 0.36$ $c = 0.45$ $(c\mu/k)^1 = 0.364$	o) $(c\mu/k)^{1/3}$ $t_c = 144.5$ $c = 0.36$ $k = 0.45$ $(c\mu/k)^1 = 0.8$
o) $h_o = 1500 \text{ BTU/hr.ft}^2 \cdot ^\circ\text{F} \times \phi_t$ $h_o = 1500 \text{ BTU/hr.ft}^2 \cdot ^\circ\text{F}$ ϕ_s	o) $h_i = jH .k/D . (c\mu/k)^{1/3} \times \phi_t$ $\frac{h_i}{\phi_t} = 276.32156$
o) $t_w = t_c + \frac{h_o/\phi_t (T_c-t_c)}{h_i/\phi_t + h_o/\phi_s}$ $= 667.40$	
$\phi_s = \frac{(\mu)^{1/4}}{(\mu w)^{1/4}}$ $\phi_s = 0.938$	$\phi_t = \frac{(\mu)^{1/4}}{(\mu w)^{1/4}}$ $(\mu w)^{1/4} = 0.242$ $\phi_t = 1.331$
o) $h_o = 1599$	o) $h_{io} = 208$

o) $h_o = 1599$

Menghitung UC

$$U_c = \frac{h_{io} \times h_o}{h_{io} + h_o}$$

$$= 183.689209 \text{ BTU/ft}^2 \cdot ^\circ\text{F jam}$$

Menghitung Rd

$$R_d = \frac{U_c - U_D}{U_c \times U_D}$$

$$U_D = 0.0001$$

$$R_d = 11841.7 > 0.002000 \text{ (memenuhi)}$$

Pressure Drop	
o) $N_{re} = 3$	o) $N_{re} = 0.03505$
$f = 0.000001$	$f = 5E-04$
$v = 37.93145 \text{ ft}^3/\text{lb}$	$s = 1$
$\rho = 0.0264 \text{ lb/ft}^3$	
$s = 0.000422$	
o) $N+1 = 12 \text{ L/B}$	o) $\Delta P_t = \dots Gt^2 . L . I$



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Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Pressure Drop	
o) d_e	$= 84.03361$ $= 0.9904 \text{ in}$ $= 0.0825 \text{ ft}$
o) ΔP_s	$= f \cdot G_s^2 \cdot D_s \cdot (\frac{1}{\rho})$ $= 5,22 \cdot 10^{10} \cdot D_s$ $= 9E-12 < 10 \text{ psi}$ (memenuhi)
	$5,22 \cdot 10^{10} \cdot D_s$ $= 9E-14 < 2 \text{ psi}$ (memenuhi)

Spesifikasi Heater NaOH

Type Heater = 2-4 Shell & Tube Exchanger

Jumlah = 1 buah

Bagian shell :

Jenis Bahan = Carbon steel

ID Shell = 2.38 in

Baffle Space = 1.428 in

Passes = 1

Bagian tube :

Jumlah & Panjang Tube = 1 buah, 10 ft

OD, BWG, pitch = 1 in, 14 BWG 1 1/4 Triangular

Passes = 2

10 GUDANG PENYIMPANAN PHTHALIC ANHYDRIDE (F-130)

Temperatur = 30 C

Tekanan = 1 atm

Waktu penyim = 30 hari

Fungsi = Untuk menyimpan bahan baku berupa phthalic anhydride

Tipe = berbentuk balok beratap asbes

Komponen karbon aktif masuk

Komponen	%Berat	Massa (Kg)	ρ (gr/ml)
$C_8H_4O_3$	0.997	1445.34966	1.53
$C_8H_6O_4$	0.002	2.899397513	1.52
H_2O	0.001	1.449698756	0.998
Total	1	1449.698756	

$$\rho_{\text{Campura}} = \frac{1.000}{\text{Fraksi berat}} \times 62$$

$$= \frac{1}{0.653952} \frac{\text{gr}}{\text{ml}} \times 62 \frac{\text{lbm}}{\text{cu}} \frac{\text{gr}}{\text{ml}}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned} &= 95.46575 \text{ lbm/cu} \\ \text{Rate massa} &= 1449.699 \text{ kg/jam} \\ &= 1449.699 \times 2.205 \text{ (konversi ke lb/jam)} \\ &= 3196.586 \text{ lb/jam} \\ \text{Rate volume} &= \frac{\text{Rate Massa}}{\rho \text{ Campuran}} \\ &= \frac{3196.586 \text{ lb/jam}}{95.46575 \text{ lb/cu}} \\ &= 33.48411 \text{ cuft/jam} \end{aligned}$$

DITENTUKAN

$$\begin{aligned} \text{Waktu tinggal} &= 30 \text{ hari} \\ &= 720 \text{ jam} \\ \text{Tinggi} &= 1 \text{ H} \\ \text{Panjang} = \text{Lebar} &= 2 \text{ H} \\ \text{Volume bahan} &= \text{Rate volume} \times \text{Waktu tinggal (jam)} \\ &= 33.484109 \times 720 \\ &= 24109 \text{ cuft} \end{aligned}$$

Volume gudang direncanakan 80% terisi bahan

$$\begin{aligned} \text{Volume gudan} &= \frac{100}{80} \times 24109 \\ &= 30136 \text{ cuft} \\ \text{Volume gudan} &= P \times L \times H \\ 30135.7 \text{ cuf} &= 2H \times 2H \times H \\ 30135.7 \text{ cuf} &= 4 H^3 \\ H^3 &= 7534 \text{ ft} \\ &= 19.6 \\ &= 3.28 \text{ (Konversi ft ke m)} \\ &= 5.977 \text{ m} \end{aligned}$$

Sehingga

$$\begin{aligned} \text{Panjang} &= 2H = 11.95 \text{ m} = 39.21 \text{ ft} \\ \text{Lebar} &= 2H = 11.95 \text{ m} = 39.21 \text{ ft} \\ \text{Tinggi} &= H = 5.977 \text{ m} = 19.6 \text{ ft} \end{aligned}$$

SPEKIFIKASI GUDANG PENYIMPANAN PHTHALIC ANHYDRIDE

$$\begin{aligned} \text{Fungsi} &= \text{Untuk menyimpan bahan baku berupa phthalic anhydride} \\ \text{Kapasitas} &= 24108.56 \text{ cuft} \\ \text{Panjang} &= 39.20761 \text{ ft} \\ \text{Lebar} &= 39.20761 \text{ ft} \\ \text{Tinggi} &= 19.60381 \text{ ft} \\ \text{Bahan Konst} &= \text{Beton} \\ \text{Jumlah} &= 1 \text{ buah} \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

11 BELT CONVEYOR PHTHALIC ANHYDRIDE (J-131)

Tempe = 30 C

Tekana = 1 atm

Fungsi = Mengangkut *phthalic anhydride* dari gudang menuju *bucket elevator*

Tipe = Troughed belt on 45° idlers with rolls of equal length
(Dipilih conveyor jenis belt sesuai dengan bahan)

PERHITUNGAN

Rate Mass: = 1449.699 kg/jam
= 1.4497 ton/jam

Berdasarkan kapasitas bah = 1.45 ton/jam

Dari Perry 7 Ed. T 21-7 dan fig 21-4 dipilih belt conveyor dengan spesifikasi :

Kapasitas mak = 32 ton/jam

hp tiap 10 ft (li = 0.34 hp/ft

ASUMSI

Jarak Belt = 13 ft

Tinggi belt = 10 ft

Slope = α

tg α = $\frac{10}{13}$

α = 0.769230769

Panjang Belt = $\sqrt{(13^2+10^2)}$
= 16.40122 ft

Perhitungan Powe

hp/10 ft, lift = 0.34 hp/ft

hp = $\frac{16.40122}{10} \times 0.3$
= 0.557641 hp

Penambahan power untuk tripper = 2 hp

Power Total = 2.5576 hp = 3 hp

SPEKIFIKASI BELT CONVEYOR

Nama = Belt Conveyor phthalic anhydride

Fungsi = Mengangkut *phthalic anhydride* dari gudang menuju *bucket elevator*

Kapasitas = 1.45 ton/jam

Bel :

Width = 14 in = 0.3556 m

Through wid = 9 in = 0.2286 m

Skirt Seal = 2 in = 0.0508 m



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned} \text{Belt speed} &= 1.45 \times 100 \text{ ft/min} \\ &= 32 \\ &= 4.53 \text{ ft/min} \\ \text{Panjang} &= 16.4 \text{ ft} \\ \text{Power} &= 3 \text{ hp} \\ \text{Jumlah} &= 1 \text{ buah} \end{aligned}$$

12 BUCKET ELEVATOR PHTHALIC ANHYDRIDE (J-132)

Fungsi : Mengangkut *Phthalic Anhydride* dari *belt conveyor* menuju *hopper*

Jumlah : 1 buah

Kondisi Operasi : T = 30 °C = 303.15 K

P = 1 atm = 760 mmHg = 14.7 psi

Komposisi Bahan

Komponen	%Berat	Massa (Kg)	ρ (gr/ml)
C ₈ H ₄ O ₃	0.997	1445.34966	1.2
C ₈ H ₆ O ₄	0.002	2.899397513	1.526
H ₂ O	0.001	1.449698756	0.997
Total	1	1449.698756	

$$\begin{aligned} \rho \text{ Campuran} &= \frac{1.000}{\frac{\sum \text{Fraksi berat}}{\sum \rho \text{ Komponen}}} \times 62.43 \\ &= \frac{1}{\frac{1}{1} + \frac{0}{2} + \frac{0.0}{1}} \times 62.43 \\ &= \frac{1.2}{74.9} \times 62.43 \\ &= 74.9 \text{ lb/cuft} \end{aligned}$$

$$\begin{aligned} \text{Rate massa} &= 1449.6988 \text{ kg/jam} \\ &= 1.4497 \text{ ton/jam} \end{aligned}$$

$$\rho \text{ Campuran} = 74.9 \text{ lbm/cuft}$$

Dari perry 7 Ed. T 21-9 dipilih bucket elevator dengan spesifikasi :

$$\text{Tinggi bucket} = 18.7065 \text{ ft}$$

$$\text{Putaran head shaft} = 28 \text{ rpm}$$

$$\text{kapasitas maksimum} = 35 \text{ ton/jam}$$

$$\text{bucket linear speed} = 150 \text{ ft/mnt}$$

$$\begin{aligned} \text{maka, kecepatan bucket elevator} &= \frac{\text{Rate massa}}{\text{kapasitas maksimum}} \times \text{bucket} \\ &= \frac{1.4497}{35} \times 150 \text{ ft/mnt} \\ &= 6.2130 \text{ ft/mnt} \end{aligned}$$

$$\begin{aligned} \text{Power head shaft} &= 1.8 \text{ hp} \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

power tambahan	=	0.06	hp	tiap ft
	=	0.06	x	18.7065
	=	1.12239	hp	
power total	=	Pwr.head shaft +	Pwr tambahan	
	=	1.8	+	1.12239
	=	2.92239	hp	
ukuran bucket	=	Lebar x Proyeksi x Kedalaman		
	=	8"x5.5"x7.75"		
bucket spacing	=	8	inch	
efisiensi motor	=	0.8		
maka, motor penggerak yang digunakan	=	$\frac{\text{Power total}}{\text{efisiensi motor}}$		
	=	$\frac{2.92239}{0.8}$		
	=	3.6529875	hp	

SPESIFIKASI BUCKET ELEVATOR

Nama	=	Bucket Elevator	$C_8H_4O_3$
Fungsi	=	Mengangkut <i>Phthalic Anhydride</i>	dari belt conveyor menuju hopper
Tipe	=	Continuous bucket elevator	
Kapasitas	=	1.449698756	ton/jam
Bucket = tinggi bucket	=	18.7065	ft
kecepatan	=	6.21299467	ft/mnt
bucket spacing	=	8	in
ukuran bucket	=	8"x5.5"x7.75"	
putaran head shaft	=	28	rpm
power	=	3.6529875	hp = 4 hp
jumlah	=	1	buah

13 HOPPER PHTHALIC ANHYDRIDE (F-133)

Fungsi	:	Menampung sementara <i>phthalic anhydride</i> sebelum masuk <i>melter</i>
Tipe	:	Tangki silinder vertikal dengan tutup atas berupa <i>plate</i> dan tutup bawah berupa <i>conical</i>
Jumlah	:	1 buah
Kondisi Operasi	:	T = 30 °C = 303.15 K
		P = 1 atm = 760 mmHg
		= 14.7 psi
Rate massa	=	1449.6988 kg/jam
	=	3196.5858 lb/cuft
ρ Campuran	=	74.9 lbm/cuft



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned} \text{Rate volumetri} &= \frac{\text{Rate Massa}}{\rho \text{ Campuran}} \\ &= \frac{3196.5858 \text{ lb/jam}}{74.9328} \text{ lb/cuft} \\ &= 42.6594 \text{ cuft/jam} \end{aligned}$$

Direncanakan penyimpanan untuk 1 Jam dengan 1 buah hopper

$$\begin{aligned} \text{Volume :} &= \frac{42.6594 \times 1 \text{ jam}}{43} \\ &= 1.0000 \text{ cuft} \end{aligned}$$

Asumsi volume bahan 80%

$$\begin{aligned} \text{Volume Tangki} &= \frac{1.0000}{80\%} \\ &= 1.2500 \text{ cuft} = 9.350625 \text{ gal} \end{aligned}$$

Menentukan ukuran tangki tangki dan ketebalannya

$$\text{Dimensi ratio} = \frac{H}{D} = 2 \quad [\text{Ulrich; T.4-27:p.248}]$$

Dengan mengabaikan volume dished head.

$$\begin{aligned} \text{Volume tangki} &= \frac{\pi}{4} \times D^2 \times H \\ 1.2500 &= \frac{3.14}{4} \times D^2 \times H \\ 1.2500 &= 1.57 \times D^3 \\ D^3 &= 0.80 \text{ ft} \\ D &= 0.93 \text{ ft}, H = 1.9 \text{ ft} \\ &= 11.12 \text{ in} \quad 22.2 \text{ in} \\ &= 0.28 \text{ m} \quad 0.6 \text{ m} \end{aligned}$$

Tinggi feed dalam tangki :

$$\begin{aligned} \text{Volume feed} &= \frac{\pi}{4} \times D^2 \times H \\ 1.0000 &= \frac{3.14}{4} \times 0.8590 \times H \\ H \text{ feed} &= 1.4829 \text{ ft} \end{aligned}$$

$$P \text{ operasi} = 1 \text{ atr} = 14.7 \text{ psi}$$

P design diambil 10% lebih besar dari P operasi untuk faktor keamanan.

$$\begin{aligned} P \text{ design} &= P \text{ operasi} \times 110\% \quad [\text{Walas; p.623}] \\ &= 14.7 \times 1 \\ &= 16.17 \text{ psi} \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

Penentuan tebal shell :

$$t_{\min} = \frac{P \times r_i}{fE - 0,6P} + C \quad [Brownell\&Young; eq.13-1]$$

dengan

t_{\min} = tebal shell mini ; in

P = tekanan tangki ; psi

r_i = jari-jari tangki ; in ($1/2 D$)

C = faktor korosi ; in (digunakan $1/8$)

E = faktor pengelasan, digunakan $d E = 1$

f = stress allowable,

bahan konstruksi Low-Alloy Steel SA-203 Grade C $2 1/2$ Ni

f = 18750 psi [Brownell&Young; T.13-1]

$$r_i = 1 \times D$$

$$= 1 \times 11.12 \text{ in}$$

$$= 5.561 \text{ in}$$

$$t_{\min} = \frac{P \times r_i}{fE - 0,6P} + C$$

$$= \frac{16.17 \times 5.561}{15000 - 9.702} + 0.125$$

$$= \frac{89.9217}{14990.30} + 0.125$$

$$= 0.006 + 0.125$$

$$= 0.131 \text{ in}$$

digunal $t_{\min} = 4/16 \text{ in}$

Tutup bawah, conis : [Brownell&Young; p.118]

$$t_{\text{tebal conic}} = \frac{P.D}{2 \cos \alpha (fE - 0,6P)} + C$$

Dengan $\alpha = 1/2$ sudut conis = $60^\circ/2 = 30$

$$t_c = \frac{P.D}{2 \cos \alpha (fE - 0,6P)} + C$$

$$t_c = \frac{14.70 \times 11.1220}{1.73 \times 14990.30} + 1/8$$

$$= \frac{163.4941}{25967.94} + 1/8$$

$$= 0.13 \text{ in digunakan } t = 4/16 \text{ in}$$

Tinggi conical :

$$h = \frac{\text{tg } \alpha \times (D - m)}{2} \quad [Hesse; eq.4-17]$$

Keterangan = $\alpha = 1/2$ sudut conis ; 30

D = diameter tangki (ft)

m = flat spot center ; $12 \text{ in } = 1 \text{ ft}$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
 h &= \frac{\text{tg } \alpha \times (D - m)}{2} \\
 &= \frac{0.577 \times 0.0732}{2} \\
 &= 0.02 \text{ ft}
 \end{aligned}$$

SPESIFIKASI HOPPER PHTHALIC ANHYDRIDE

fungsi : Menampung sementara *phthalic anhydride* sebelum masuk melter

type : Silinder tegak dengan tutup alas plat dan bawah conis

volume tangki : 1.25 cuft

diameter tangki : 0.93 ft

tinggi total : 0.02 ft

tebal shell : 4/16 in

tebal tutup bawah : 4/16 in

bahan konstruksi : Low-Alloy Steel SA-203 Grade C 2 1/2 Ni

jumlah : 1 buah

14 MELTER PHTHALIC ANHYDRIDE (X-134)

Fungsi : Tempat meleburkan *Phthalic Anhydride* padat menjadi *Phthalic Anhydride* cair, untuk keperluan proses

Tipe : silinder vertikal dengan alas dan tutup berbentuk *torispherical*

Bahan : *Carbon Steel SA-283 Grade C*

Jumlah : 1 buah

Kondisi Operasi : T = 165 °C = 438.2 K
P = 1 atm = 760 mmHg = 14.7 psi

Menentukan Kapasitas Tangki

Komponen	Massa	%Berat	ρ	%Berat/ ρ
	kg/jam		(gr/m ³)	
C ₈ H ₄ O ₃	1445.35	0.997	1.53	0.651634
H ₂ O	1.44970	0.001	0.997	0.001003
C ₈ H ₆ O ₄	2.899398	0.002	1.522	0.001314
Total	1449.699	1		0.653951

$$\begin{aligned}
 \rho_{\text{campuran}} &= \frac{\sum x_i}{\sum x_i / \rho} \\
 &= \frac{1.000}{0.654} \\
 &= 1.529 \text{ kg/m}^3
 \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$C_p \text{ Campuran} = \frac{1}{1915} = 0.000522$$

$$\begin{aligned} \text{Kebutuhan Phthalic Anhydride} &= 1445.350 \text{ kg/jam} \\ &= 3186.447 \text{ lb/jam} \end{aligned}$$

$$\text{Densitas bahan} = 1.529 \text{ kg/m}^3 = 0.095 \text{ lb/ft}^3$$

$$\begin{aligned} V_{\text{bahan}} &= \frac{m}{\rho} \\ &= \frac{1445.350 \text{ kg}}{1.529166 \text{ kg/m}^3} \\ &= 945.19 \text{ m}^3 \\ &= 26764.7 \text{ ft}^3 \end{aligned}$$

Asumsi 80% volume tangki merupakan volume larutan, sehingga :

$$\begin{aligned} V_t &= \frac{V}{0.8} = \frac{945.1879}{0.8} \text{ m}^3 \\ &= 1181.485 \text{ m}^3 \\ &= 200227.0 \text{ galon} \end{aligned}$$

Diameter Tangki, D

$$\begin{aligned} V_t &= V_s + V_d \\ V_t &= (1/4 \pi D^2 H) + (2 \times 0,000049D^3) \\ &= (1/4 \pi 2 D^3) + (2 \times 0,000049D^3) \\ &= D^3 (2/4 \pi + 0,000098) \\ D^3 &= \frac{V_t}{2/4 \pi + 0,000098} \\ &= \frac{200227.0335}{2/4 \pi + 0,000098} \text{ ft}^3 \\ &= 1.5701 \text{ ft}^3 \\ D &= 1.1623 \text{ ft} \\ &= 13.9473 \text{ in} \\ &= 0.3543 \\ H &= 2 D \\ H &= 2 \times 1.1623 \text{ ft} \\ &= 2.3246 \text{ ft} \\ &= 27.8946 \text{ in} \\ &= 0.7085 \text{ m} \end{aligned}$$

Tinggi Larutan

$$\begin{aligned} \text{Volume larutan dalam dish} &= 5E-05 D^3 \\ &\text{Pers. 5.11 (Brownell \& Young, 1959)} \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
 &= 5E-05 \times 13.9473^3 \\
 &= 0.13 \text{ ft}^3
 \end{aligned}$$

$$\begin{aligned}
 \text{Volume larutan dalam shell} &= V_{\text{larutan}} - V_{\text{dish}} \\
 &= 26764.7 \text{ ft}^3 - 0.13 \text{ ft}^3 \\
 &= 26764.6 \text{ ft}^3
 \end{aligned}$$

$$\begin{aligned}
 \text{Tinggi larutan dalam shell} &= \frac{\text{Volume larutan dalam shell}}{1/4 \pi D^2} \\
 &= \frac{26764.6089 \text{ ft}^3}{1/4 \pi \times 1.16^2 \text{ ft}^2} \\
 &= 25239 \text{ ft} \\
 &= 7692.9 \text{ m}
 \end{aligned}$$

Tekanan *Design*

$$\begin{aligned}
 \text{Tekanan Hidrostatik, } P_H &= \rho \times g/g_c \times H_s/144 \\
 &= 0.0955 \times 1 \times 7692.870 / 144 \\
 &= 5.1000 \text{ psia}
 \end{aligned}$$

$$\begin{aligned}
 \text{Tekanan } Design, P_d &= (P_{op} + P_H) S_f \quad ; \text{ dengan faktor keamanan } 10\% \\
 &= (14,7 + 5.1000) \times 1.1 \\
 &= 21.7800 \text{ psia}
 \end{aligned}$$

Penentuan Tebal *Shell*

$$t_s = \frac{(P_d \times r_i)}{(f \times E) - (0,6 \times P_d)} + C \quad \text{Pers. 13.1 (Brownell \& Young, 1959)}$$

Dimana :

t_s = tebal dinding tangki, in

P_d = tekanan desain

r_i = jari-jari tangki

f = *allowable stress*

E = efisiensi pengelasan

C = faktor korosi

Bahan konstruksi yang dipilih adalah *Carbon Steel SA-283 Grade C* :

f = 12650 psi **Tabel 13.1.** (Brownell & Young, 1959:251)

E = 80% **Tabel 13.2.** (Brownell & Young, 1959:254)

r_i = $D/2$

C = 0.125



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$ts = \frac{21.7800 \times 6.9737}{(12650 \times 80\% - 0.6 \times 21.7800)} + 0$$

$$ts = 0.14003 \text{ in}$$

$$\text{Diambil tebal shell standard} = \frac{1}{5} \text{ in (3/16)}$$

Tabel 5.7. (Brownell & Young, 1959:89)

$$\begin{aligned} \text{OD} &= \text{ID} + (2 \times ts) = 14.3223 \text{ in} \\ \text{OD di standardkan} &= 38 \text{ in} = 3.1667 \text{ ft} \\ \text{ID koreksi} &= \text{OD} - 2ts = 37.6250 \text{ in} \\ &= 0.9557 \text{ m} \\ &= 3.1354 \text{ ft} \end{aligned}$$

$$\begin{aligned} \text{H koreksi} &= 2 \times \text{ID} \\ &= 2 \times 0.95568 \text{ m} \\ &= 1.91135 \text{ m} \\ &= 6.27076 \text{ ft} \end{aligned}$$

Penentuan Tebal *Dish*

$$td = \frac{(0,885 \times Pd \times Rc)}{(f \times E) - (0,1 \times Pd)} + C$$

$$Rc = \text{ID}/2 = 18.8125 \text{ in}$$

$$\begin{aligned} td &= \frac{0.885 \times 21.78 \times 18.8125}{12650 \times 1 - 0.1 \times 21.8} + 0.125 \\ &= 0.1608 \text{ in} \end{aligned}$$

$$\text{Dipakai tebal dish standar 3/16 in} = 0.188 \text{ in}$$

Penentuan Tinggi *Dish*

$$\text{OD standard} = 38 \text{ in}$$

$$r = 36 \text{ in} \text{ **Tabel 5.7.** (Brownell & Young, 1959:89)}$$

$$\text{icr} = 2 \frac{3}{8} \text{ in} \text{ **Tabel 5.7.** (Brownell & Young, 1959:89)}$$

$$\begin{aligned} \text{AB} &= \text{ID}/2 - \text{icr} \\ &= 16 \frac{4}{9} \text{ in} \end{aligned}$$

$$\begin{aligned} \text{BC} &= r - \text{icr} \\ &= 33.625 \text{ in} \end{aligned}$$

$$\begin{aligned} b &= r - (\text{BC}^2 - \text{AB}^2)^{1/2} \\ &= 6.6666 \text{ in} \text{ Pers. 5.8. (Brownell & Young, 1959:87)} \end{aligned}$$

$$\text{Tinggi flange, sf} = 2 \text{ in} \text{ **Tabel 5.8.** (Brownell & Young, 1959:93)}$$

$$\begin{aligned} \text{Tinggi tutup, OA} &= b + sf + t_d \\ &= ##### \text{ in} \\ &= ##### \text{ ft} \end{aligned}$$

Tinggi Tangki Total

$$\begin{aligned} \text{Ht} &= \text{H} + 2\text{Hd} \\ &= 6.2708 \text{ ft} + 2 \times 0.7356 \text{ ft} \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned} &= 7.7420 \text{ ft} \\ &= 92.9039 \text{ in} = 2.36 \text{ m} \end{aligned}$$

Merancang Pengaduk

Komponen	Massa	%Berat	μ (cP)	Massa/ μ (cP)
	kg/jam			
C ₈ H ₄ O ₃	1445.35	0.997	0.82	1.215854
H ₂ O	1.44970	0.001	0.1606	0.006227
C ₈ H ₆ O ₄	2.899398	0.002	3.9398	0.000508
Total	1449.699	1		1.222588

$$\begin{aligned} \mu_{\text{campuran}} &= \frac{\sum x_i}{\sum x_i/\mu} \\ &= \frac{1.0000}{1.2226} \\ &= 0.8179 \text{ cP} \end{aligned}$$

Menghitung Ukuran Pengaduk

Digunakan pengaduk jenis *Propeller three blades pitch*, karena viskositas cairan dalam *melter* $0.818 \text{ cP} = 0.001 \text{ lb/ft.s}$

Dimana :

ID = diameter dalam tangki

d = diameter pengaduk

L = panjang *blade* pengaduk

Z_i = jarak pengaduk dasar tangki

p = jarak antar *blade*

l = tinggi cairan

Data pengaduk diperoleh dari (Brown, 1950:507)

Ukuran Pengaduk :

Diameter pengaduk (d)

$$ID/d = 3$$

$$\begin{aligned} d &= \frac{0.956}{3.000} \\ &= 0.319 \text{ m} \\ &= 12.54 \text{ in} \\ &= 1.045 \text{ ft} \end{aligned}$$

Panjang *blade* pengaduk

$$L = d/4$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}L &= \frac{0.31856}{4.00000} \\ &= 0.07964 \text{ m} \\ &= 3.13541 \text{ in} \\ &= 0.26127 \text{ ft}\end{aligned}$$

Jarak pengaduk dan dasar tangki (Z_i)

$$\begin{aligned}Z_i/d &= 0,75 - 1,3 \quad ; \text{ (dipilih } 0,75 \text{)} \\ Z_i &= 1 \times d \\ &= 0.239 \text{ m} \\ &= 9.406 \text{ in} \\ &= 0.784 \text{ ft}\end{aligned}$$

Lebar *baffle* (W)

$$\begin{aligned}W &= 0,1 \times d \\ &= 0.1 \times 0.3186 \\ &= 0.0319 \text{ m} \\ &= 1.2542 \text{ in} \\ &= 0.1045 \text{ ft}\end{aligned}$$

Jarak antar *blade* (p)

$$\begin{aligned}p &= 2 \times d \\ &= 2 \times 0.3186 \\ &= 0.6371 \text{ m} \\ &= 25.0833 \text{ in} \\ &= 2.0902 \text{ ft}\end{aligned}$$

Kecepatan Putar Pengaduk (N)

$$N = 600/\pi d \sqrt{(WELH/2d)} \quad \text{Eq. 8.8 (Rase \& Holmes, 1977:338)}$$

$$WELH = ZL \times Sg$$

Dimana :

- N = kecepatan putar pengaduk, rpm
- d = diameter pengaduk, m
- ZL = tinggi cairan dalam tangki, m
- Sg = *specific gravity*
- $WELH$ = *Water Equivalent Liquid Height*

$$Sg = \frac{\rho_{\text{cairan}}}{\rho_{\text{air}}}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$= \frac{1.52917 \text{ kg/liter}}{0.99700 \text{ kg/liter}}$$

$$= 1.53377$$

$$Z_L = \text{Tinggi cairan di dalam shell}$$

$$= 7693 \text{ m}$$

$$= 25239 \text{ ft}$$

$$\text{WELH} = 7692.87 \text{ m} \times 1.5338$$

$$= 11799.08 \text{ m}$$

$$= 38710.41 \text{ ft}$$

$$\text{Jumlah pengaduk} = \frac{\text{WELH}}{\text{ID}}$$

$$= \frac{11799}{0.9557} = 12346 \approx 12,346 \text{ buah}$$

$$N = 600/\pi d \sqrt{(\text{WELH}/2d)} = 24880.59 \text{ rpm}$$

$$= 414.68 \text{ rps}$$

$$= 1492836 \text{ rph}$$

Menghitung *Power* Pengaduk (P)

$$N_{re} = \frac{N \cdot d^2 \cdot \rho}{\mu}$$

$$= \frac{414.6766 \times 1.0451^2 \times 73.2100}{0.0005}$$

$$= 60330514.30$$

Karena nilai $N_{re} (>10.000)$ maka digunakan pengaduk jenis *propeller*.

Power pengaduk merupakan fungsi dari $N_p = K_T = 0.87$

Tabel 9.3 (Mc Cabe et al., 1993:254) dan persamaan :

$$P = \frac{N_p \cdot \rho \cdot N^3 \cdot D^5}{g_c} \quad \text{Pers. 9.20 (Mc Cabe et al., 1993:253)}$$

Dimana :

P = daya pengaduk, lb.ft/s

N_p = *power number* $K_T = 0.87$

N = kecepatan putaran pengaduk = 414.6766 rps

ρ = densitas bahan = 0.0955 lb/ft³

d = diameter pengaduk = 1.0451 ft

g_c = gravitasi = 32.1740 lbf/lbm·s²

Sehingga :

$$P = \frac{N_p \cdot \rho \cdot N^3 \cdot d^5}{g_c}$$

$$= \frac{0.87 \times 0.095 \times 414.68^3 \times 1.045^5}{32.1740}$$

$$= \text{#####} \text{ ft.lb/s}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$= 417.3436 \text{ Hp}$$

$$\text{Efisiensi motor penggerak } (\eta) = 80\%$$

Fig . 14.38 (Peters & Timmerhaus, 1991:521)

$$\text{Daya penggerak motor} = P/\eta = \frac{\text{#####}}{80\%} = \text{#####} \text{ Hp}$$

$$\text{Maka dipakai motor dengan daya} = \text{#####} \text{ Hp}$$

$$\text{Power motor yang digunakan} = 1 \text{ Hp}$$

Menghitung Poros Pengaduk

$$\text{Panjang poros} = Z_T + Z_p - Z_i$$

dimana :

$$Z_T = \text{tinggi total tangki} = 2.35976 \text{ m}$$

$$Z_p = \text{tinggi poros di atas tangki, diambil} = 0.30000 \text{ m}$$

$$Z_i = \text{tinggi poros di atas dasar tangki} = 0.23892 \text{ m}$$

$$\text{Panjang poros} = 2.360 + 0 - 0.239 = 2.421 \text{ m}$$

Diameter Poros

Momen puntir (τ)

$$\tau = (\text{HP} \times 75 \times 60)/(2 \pi v)$$

Dimana :

$$\text{HP} = \text{Daya pengaduk} = 521.6796 \text{ Hp}$$

$$V = \text{kecepatan putaran pengaduk} = 24880.59 \text{ rpm}$$

$$\tau = \frac{521.6796 \times 75 \times 60}{2 \times \pi \times 24880.59} = 15.0244 \text{ kg-m}$$

Diambil faktor keamanan 50%

$$\text{Jadi, momen puntir} = 1.5 \times \text{#####}$$

$$= 22.536538 \text{ kg-m}$$

$$f_s = \frac{\tau}{t_p} \quad Z_p = \frac{\pi D_p^3}{16} \quad Z_p = \frac{\tau}{f_s}$$

$$\frac{\tau}{f_s} = \frac{\pi D_p^3}{16}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

dimana :

$$f_s = \text{shear stress} = 550 \text{ kgf/cm}^2$$

D_p = diameter poros

$$D_p^3 = \frac{16 \tau}{\pi f_s}$$

$$D_p^3 = \frac{16 \times 22.5365}{\pi \times 550 \times 10000} = 2.09E-05 \text{ m}^3$$

$$\begin{aligned} D_p &= 0.02754 \text{ m} \\ &= 2.75363 \text{ cm} \\ &= 1.08410 \text{ in} \\ &= 1.08410 \text{ in} \end{aligned}$$

Jadi, diameter poros pengaduk

Pemilihan *Bearing* (bantalan poros)

Tipe : *gear grove ball bearing*

Berat poros : $\frac{1}{4} \pi D_p^2 \times L_p \times \rho$

Dimana :

$$L_p = \text{panjang poros} = 2.421 \text{ m}$$

$$D_p = \text{diameter poros} = 0.028 \text{ m}$$

$$\begin{aligned} \rho &= \text{densitas baja} = 0.283 \text{ lb/in}^3 \\ &= 7924 \text{ kg/m}^3 \end{aligned}$$

$$\begin{aligned} \text{Berat poros (Wp)} &= \frac{1}{4} \times 3, x \quad 0.028^2 \times 2.421 \times 7924 \\ &= 11.418 \text{ kg} \end{aligned}$$

Berat Pengaduk

$$\text{Berat pengaduk} = \text{volume blade} \times \rho \text{ blade}$$

$$\text{Tebal blade} = 1 \text{ in}$$

$$\text{Tinggi blade (h)} = 0,2 d = 0.016 \text{ m} = 0.627 \text{ in}$$

$$\text{Panjang blade (L)} = 0.0796 \text{ m} = 3.135 \text{ in}$$

$$\text{Densitas blade } (\rho) = 0.2790 \text{ lb/in}^3$$

$$\begin{aligned} \text{Volume blade} &= 1 \times 0.6271 \times 3.135 \\ &= 1.966 \text{ in}^3 \end{aligned}$$

$$\begin{aligned} \text{Berat blade} &= \text{Volume blade} \times \text{Densitas blade} \\ &= 1.9662 \times 0.2790 \\ &= 0.5486 \text{ lb} \\ &= 0.2488 \text{ kg} \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
\text{Berat radial} &= \text{Berat poros} + (3 \times \text{berat } blade) \\
&= 11.4180 + 3 \times 0.2488 \\
&= 11.4180 \text{ kg} \\
&= 25.1724 \text{ lb}
\end{aligned}$$

SISTEM PENDINGIN

Perhitungan sistem penjaga suhu : (Kern Hal .719)

Dari neraca panas, suhu yang dijaga : 165.000 C

$$\begin{aligned}
Q &= -484213 \text{ kal/jam} \\
&= -1921.480 \text{ BTU/jam}
\end{aligned}$$

$$\begin{aligned}
\text{Suhu Feed masuk} &= 165 \text{ C} = 329 \text{ F} \\
\text{Suhu Bahan Keluar} &= 165 \text{ C} = 329 \text{ F} \\
\text{Suhu Air pendingin masuk} &= 30 \text{ C} = 86 \text{ F} \\
\text{Suhu air pendingin keluar} &= 45 \text{ C} = 113 \text{ F} \\
\Delta T1 &= 216 \text{ F} \\
\Delta T2 &= 243 \text{ F}
\end{aligned}$$

$$\begin{aligned}
\text{Kebutuhan media} &= 7710.154 \text{ kg/jam} \\
&= 17000.889 \text{ lb/jam} \\
\text{Densitas media} &= 62.430 \text{ lb/cuft} \\
\text{Rate Volumetrik} &= \frac{17000.889}{62.430} \text{ lb/cuft} \\
&= 272.319 \text{ cuft/jam} \\
&= 0.076 \text{ cuft/detik}
\end{aligned}$$

Koefisien perpindahan panas bagian luar jaket :

$$hi = 0,36 (k/Di) \frac{[L^2 N \rho]^{2/3} [C \cdot \mu]^{1/3} [\mu]^{0.14}}{\mu \quad k \quad \mu w}$$

(Kern pg.71 eq 20-1)

Keterangan :

$$\begin{aligned}
L &= \text{Da (diameter impeller)} = 1.162 \text{ ft} \\
N &= \text{Putaran pengaduk} = 415 \text{ rpm} = 24881 \text{ rph} \\
\rho &= \text{Berat jenis air} = 62.43 \text{ lb/cuft} \\
\text{Pada suhu tangki penampungan} &= 30 \text{ C} \text{ maka} \\
\mu &= ##### \text{ cps} = 1.985 \text{ lb/ft.jam} \\
k &= ##### \text{ btu/jam/ft}^2 \\
c &= 1 \text{ btu/lb.F} \\
\text{Re p} &= \frac{[L^2 N \rho]^{2/3}}{\mu} \\
&= \left(\frac{1.35 \times 24881 \times 62.43}{1.985} \right)^{2/3}
\end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$= 10375.85$$

$$\frac{[C \cdot \mu]^{1/3}}{k} = \left(\frac{1.000 \times 1.985}{0.333} \right)^{1/3}$$

$$= 3.776551447$$

$$\frac{[\mu]^{0.14}}{\mu_w} = 1 \quad (\text{Untuk air})$$

$$h_i = 0.36 \times \frac{0.3328}{1.1623} \times 10375.9 \times 3.7766 \times 1$$

$$= 4039.207 \text{ Btu/jam} \cdot \text{ft} \cdot \text{F}$$

Untuk air pendingin yang berada dalam jaket mengacu pada diameter dalam bejana hio = 100 Btu/jamft² F

Menghitung Uc :

$$U_c = \frac{h_i \times h_{io}}{h_i + h_{io}}$$

$$= 97.5841 \text{ Btu /jamft}^2 \text{ F}$$

$$R_d = 0.001 \quad (\text{Kern tabel 12 pg 845})$$

$$h_d = \frac{1}{R_d}$$

$$= 1000$$

Menghitung Ud :

$$U_d = \frac{U_c \times h_d}{U_c + h_d}$$

$$= 88.90806673 \text{ Btu /jamft}^2 \text{ F}$$

$$A = \pi \times D_i \times H \text{ Pengaduk} + \pi/4 \times D_i^2$$

$$= 3.14 \times 1.16 \times 2.32 + 1.06$$

$$= 9.5440 \text{ ft}^2$$

Penghitungan Tinggi Jaket :

$$\text{Tinggi jaket} = \text{Tinggi shell}$$

$$h = 2.3246 \text{ ft}$$

Asumsi :

$$\text{Tebal air pendingin (ts)} = 2 \text{ in}$$

$$\text{Tebal jaket (tj)} = 3/16 \text{ in}$$

$$e = 0.8$$

$$C = 0.13$$

Dipergunakan bahan konstruksi yang terbuat dari carbon steel SA-283 Grade C

$$f \text{ allowable} = 12650$$

$$D_o (\text{Shell}) = D_i + 2ts$$

$$= 13.9473 + 4$$

$$= 17.9473 \text{ in}$$

$$D_i (\text{jaket}) = D_o (\text{Shell}) + 2tj$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
 &= 17.9473 + 0.375 \\
 &= 18.3223 \text{ in} \\
 \text{Do (Jaket)} &= \text{Di (Jaket)} + 2t_j \\
 &= 18.3223 + 0.375 \\
 &= 18.6973 \text{ in} \\
 \text{P desain jaket} &= P_o - P_i + P_h \\
 &= 14.7 - 14.7 + x \text{ g/gc} \times h_{li} \\
 &= 62.43 \frac{\text{lbr}}{\text{cuft}} \times 1 \frac{\text{lbf}}{\text{lbn}} \times \text{##### ft} \\
 &= 1575675 \text{ lbf/ft}^2 \\
 &= 10942.19 \text{ psi}
 \end{aligned}$$

Penentuan tebal jaket :

Tebal jaket berdasarkan ASME Code :

$$t_j = \frac{P \times D_{ij} + C}{2 \cdot f \cdot e - P}$$

$$2 = \frac{10942 \times 18.32}{1.6 f - 10942} + 0$$

$$f = 73667.56669$$

$$f_{\text{allowable}} > f_{\text{desain}}$$

$$12650 > 73667.56669$$

$$\text{Dipilih tebal jaket} = (3/16) \text{ in}$$

SPESIFIKASI MELTER PHTHALIC ANHYDRIDE

- Fungsi : Tempat meleburkan *phthalic anhydride* padat menjadi *phthalic anhydride* cair
- Tipe : silinder vertikal dengan alas dan tutup berbentuk *torispherical*
- Bahan Konstruksi : *Carbon Steel SA-283 Grade C*
- Jumlah : 1 buah
- Kondisi Operasi : Temperatur 165 °C dan Tekanan 1 atm
- Kapasitas : 1181.485 m³
- Dimensi
- Silinder
- Diameter : 0.9557 m
- Tinggi : 1.9114 m
- Tebal : 3/16 in
- Tutup
- Tebal : 3/16 in
- Tinggi : 8.8274 in
- Tinggi Tangki Tot : 2.3598 m
- Power Pengaduk : 1 Hp

Sistem Pendingin :



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

Diameter jaket	=	18.3223	ft
Tinggi Jaket	=	2.3246	ft
Jaket Spacing	=	4	in
Tebal Jaket	=	(3/16)	in

15 POMPA MELTER (L-135)

Fungsi : Mengalirkan *phthalic anhydride* dari *melter* menuju reaktor

Tipe : *Centrifugal pump*

Jumlah : 1 buah

PERHITUNGAN

Bahan Masuk	=	1449.698756	kg/jam
	=	3196.585758	lb/jam
ρ Campuran	=	74.9	lb/cuft
Rate massa	=	1449.698756	lb/jam
Rate volumetrik	=	19.3	cuft/jam
	=	0.322444367	cuft/menit
	=	0.005374073	cuft/detik
	=	2.412206308	gpm

Asumsi aliran turbulen

(Di) optimum untuk aliran turbulen, $NRe > 2100$ digunakan persamaan :

$$\text{Diameter optimum} = 3.9 \times q_f^{0.45} \times \rho^{0.13} \quad (\text{Peters, 4}^{ed}, \text{ pers.15 : 496})$$

Keterangan =

in

q_f = Fluid flow rate; (cuft/detik)

ρ = Fluid Density; (lb/cuft)

$$\begin{aligned} \text{Diameter pipa optimum, I} &= 3.9 \times (0.005)^{0.45} \times (74.93)^{0.13} \\ &= 1.1432 \text{ in} \end{aligned}$$

Dipilih pipa 4 in, sch 10s (Brownell and Young, APP-K)

$$\text{OD} = 4.5 \text{ in}$$

$$= 0.375 \text{ ft}$$

$$\text{ID} = 4.216 \text{ in}$$

$$= 0.351 \text{ ft}$$

$$A = \left(\frac{1}{4} \times \pi \times \text{ID}^2\right)$$

$$= 0.25 \times 3.14 \times 0.351333333^2$$

$$= 0.096896562 \text{ ft}^2$$

$$\text{Kecepatan linear, } v = \frac{q_f}{A}$$

$$= \frac{0.005374073 \text{ cuft/detik}}{0.096896562 \text{ ft}^2}$$

$$= 0.055461955 \text{ ft/dtk}$$

$$\rho \text{ reference} = 62.43 \text{ lb/cuft}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
 \text{sg reference} &= 1 \\
 \mu \text{ reference} &= 0.95 \\
 \text{sg bahan} &= \frac{\rho \text{ bahan}}{\rho \text{ reference}} \times \text{sg reference} \\
 &= \frac{74.9}{62.43} \times 1 \\
 &= 1.20026844 \\
 \mu \text{ berdasarkan sg bahan} \\
 \mu &= \frac{\text{sg bahan}}{\text{sg reference}} \times \mu \text{ reference} \\
 &= \frac{1.20026844}{1} \times 0.95 \\
 &= 1.140255018 \text{ cps} \\
 &= 0.000766216 \text{ lbf/s} \\
 \text{Nre} &= \frac{D \ V \ \rho}{\mu} \\
 &= \frac{0.35133333 \times 0.055461955 \times 74.9}{0.000766216} \\
 &= 1905.614399 > 2100 \text{ (Asumsi aliran turbulen benar)}
 \end{aligned}$$

Dipilih pipa commercial steel,

$$\begin{aligned}
 \varepsilon &= 0.000046 \text{ m} \\
 \varepsilon/D &= 0.00042945 \\
 f &= 0.007
 \end{aligned}$$

Digunakan persamaan Bernoulli :

$$-Wf = \frac{\Delta P}{\rho} + \Delta Z \frac{g}{gc} + \frac{\Delta V^2}{2 \alpha g} + \Sigma F$$

Perhitungan friksi berdasarkan **Geankoplis 3ed T 2.10-1, hal 93**

Taksiran panjang pipa lurus	= 29.4	ft = 8.963414634	m
- 3 elbow 90°	= 3	x 35	x 0.351333
	= 36.89	ft	
- 1 gate valve	= 1	x 9	x 0.351333
	= 3.162	ft	
Panjang total pipa	= 69.452	ft	

Friksi yang terjadi:

1. Friksi karena gesekan bahan dalam pipa

$$\begin{aligned}
 F_1 &= \frac{2f \times v^2 \times Le}{gc \times D} \quad [Geankoplis ed.3; eq 2.10-6] \\
 &= \frac{2 \times 0.007 \times 0.0555^2 \times 69.45}{32.2 \times 0.3513} \\
 &= 0.0003 \text{ ft lbf/lbm}
 \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

2. Friksi karena kontraksi dari tangki ke pipa

$$F_2 = \frac{K_c \times v^2}{2 \times \alpha \times g_c} \quad (\text{Geankoplis 3ed eq. 2.10-16})$$

$$K_c = 0.55$$

untuk aliran turbulen :

$$\alpha = 1$$

$$\frac{A_2}{A_1} = \frac{0.096896562}{406944}$$

$$= 2.38108E-07$$

$$K_c = 1 - \frac{A_2}{A_1}$$

$$= 0.999999762$$

$$\text{Maka : } F_2 = \frac{0.999999762 \times 0.055461955^2}{2 \times 1 \times 32.271}$$

$$= 4.76593E-05 \text{ ft lbf/lbm}$$

3. Friksi karena enlargement (ekspansi) dari pipa ke tangki

$$F_3 = \frac{\Delta v^2}{2 \times \alpha \times g_c}$$

$$= \frac{v_2^2 - v_1^2}{2 \times \alpha \times g_c} \quad ; (A_1 \lll A_2, \text{ maka } V_1 \text{ dianggap} = 0)$$

$$= \frac{0.055^2 - 0}{2 \times 1 \times 32.217}$$

$$= 0.000 \text{ ft lbf/lbm}$$

4. Friksi karena elbow 90°

$$F_4 = n \times K_f \times \frac{v_1^2}{2}$$

n = jumlah elbow

(Geankoplis 3ed, eq 2.10-17)

$$F_4 = 3.000 \times 0.750 \times 0.002$$

$$= 0.003 \text{ ft lbf/lbm}$$

5. Friksi karena Gate Valve

$$F_5 = n \times K_f \times \frac{v_1^2}{2}$$

$$F_5 = 1 \times 0.170 \times 0.002$$

$$= 0.000 \text{ ft lbf/lbm}$$

Sehingga :

$$\Sigma F = F_1 + F_2 + F_3 + F_4 + F_5$$

$$= 0.004 \text{ ft.lbf/lbm}$$

$$1 \text{ atm} = 14.6959 \text{ psi} = 2116.2096 \text{ lbf/ft}^2$$

$$P_1 = 1 \text{ atm} + P \text{ hidrostatik}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
 &= 2116.2096 \text{ lbf/ft}^2 + (74.9 \times 1 \times 14.7) \\
 &= 2116.2096 + 1101.5 \\
 &= 3217.721153 \text{ lbf/ft}^2 \\
 P_2 &= 1 \quad \text{atr} = 2116.2096 \text{ lbf/ft}^2 \\
 \Delta P &= 3217.721153 - 2116.2096 \\
 &= 1101.512 \text{ lbf/ft}^2 \\
 \frac{\Delta P}{\rho} &= \frac{1101.512}{74.933} \frac{\text{lbf/ft}^2}{\text{lb/cuft}} \\
 &= 14.700 \frac{\text{ft.lbf}}{\text{lbm}} \\
 \frac{\Delta v^2}{2 \times \alpha \times gc} &= \frac{0.055462^2}{2 \times 1 \times 32.22} \\
 &= 4.77392\text{E-}05 \text{ ft lbf/lbm}
 \end{aligned}$$

Asumsi :

$$\begin{aligned}
 Z_1 &= 13.400 \text{ ft} \\
 Z_2 &= 19.400 \text{ ft} \\
 \Delta Z &= (Z_2 - Z_1) = 6.000 \text{ ft} \\
 \frac{\Delta Z}{gc} &= \frac{6.000 \text{ ft} \times 1.000}{\text{lbm}} \frac{\text{lbf}}{\text{lbm}} \\
 &= 6.000 \frac{\text{ft.lbf}}{\text{lbm}}
 \end{aligned}$$

Persamaan Bernoulli

$$\begin{aligned}
 -W_f &= \frac{\Delta P}{\rho} + \frac{\Delta Z}{gc} + \frac{\Delta V^2}{2 \alpha gc} + \Sigma F \\
 &= 14.700 + 6.000 + 5\text{E-}05 + 0.004 \\
 &= 20.70413 \frac{\text{ft.lbf}}{\text{lbm}}
 \end{aligned}$$

$$\begin{aligned}
 H_p &= \frac{-W_f \times \text{flowrate(gpm)} \times sg}{3960} \\
 &= \frac{20.70413 \times 2.4122 \times 1.2}{3960} \\
 &= 0.015
 \end{aligned}$$

$$\text{Efisiensi Pompa} = 0.450 \quad (\text{Peters 4}^{\text{ed}}; \text{Figure 14 - 37})$$

$$\begin{aligned}
 Bhp &= \frac{H_p}{\eta \text{ pompa}} \\
 &= \frac{0.015}{0.450} \\
 &= 0.033638925 \text{ Hp}
 \end{aligned}$$

$$\text{Efisiensi motor} = 0.800 \quad (\text{Peters 4}^{\text{ed}}; \text{Figure 14 - 38})$$

$$\text{Power motor} = \frac{Bhp}{\eta \text{ motor}}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
 &= \eta \text{ motor} \\
 &= 0.034 \\
 &= \frac{0.034}{0.800} \\
 &= 0.042048656 \text{ Hp}
 \end{aligned}$$

SPESIFIKASI POMPA MELTER

Fungsi = Mengalirkan *phthalic anhydride* dari *melter* menuju reak
 Type = Centrifugal Pump
 (Sesuai untuk viskositas <10 cP dan bahan liquid)
 Power Motor = 0.0420 Hp
 Rate volumetrik = 2.412206308 gpm
 Total dynamic head = 20.704 ft.lbf/lbm
 Efisiensi pompa = 0.450
 Efisiensi motor = 0.800
 Bahan konstruksi = Commercial Steel
 Jumlah = 1 buah

16 TANGKI PENYIMPANAN 2-ETHYL HEXANOL (F-140)

Fungsi : Menampung dan menyimpan bahan baku *2-ethyl hexanol*
 Tipe : *Cylindrical tank, flat bottom, torispherical dished head*
 Jumlah : 7 buah
 Kondisi Opera : T = 30 °C = 303.15 K
 P = 1 atm
 = 760 mmHg
 = 14.7 psi

Komposisi	%berat	berat (kg/jam)	ρ (gr/cc)
C ₈ H ₁₈ O	0.995	2539.127781	0.833
H ₂ O	0.005	12.75943609	0.998
Total	1	2551.887217	

$$1 \text{ kg} = \text{## lb}$$

$$1 \text{ gr/cc} = 62.4 \text{ lb/cuft}$$

$$\begin{aligned}
 \text{Densitas Campuran} &= \frac{1}{\frac{\text{fraksi berat}}{\rho \text{ komponen}}} \times 62.4 \\
 &= \frac{1}{\frac{1.00}{0.833} + \frac{0.01}{1}} \times 62.43 \\
 &= 0.83 \text{ gr/cc} \times 62.43 \\
 &= 52.05 \text{ lb/cuft}
 \end{aligned}$$

$$\begin{aligned}
 \text{Rate Volumetrik} &= \frac{\text{Rate Massa}}{\text{Densitas}} \\
 &= \frac{5624.36}{\text{Densitas}}
 \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$= 52.0472$$

$$= 108.0626 \text{ cuft/jam}$$

Direncanakan penyimpanan untuk 24 jam dengan 7 buah tangki (mempermudah pengeluaran dan pengisian), sehingga volume liquid adalah

$$V \text{ liquid} = \frac{\text{##### cuft/jam} \times 24 \text{ jam}}{7}$$

$$= 370.50 \text{ cuft}$$

Asumsi volume bahan (liquid) mengisi 80% volume tangki sehingga volume ruang kosong sebesar 20%

$$\text{Volume Tangki} = \frac{370.5005}{80\%}$$

$$= 463.1256 \text{ cuft} = 3464.179 \text{ gal}$$

Disarankan > 10000 gal, menggunakan vertical tank dan pondasi beton

[Walas ed.2; p.xvii]

Menentukan ukuran tangki tangki dan ketebalannya

$$\text{Dimensi ratio} = \frac{H}{D} = 2 \quad [\text{Ulrich; T.4-27:p.248}]$$

$$\text{Volume tangki} = \frac{\pi}{4} \times D^2 \times H$$

$$463.1256 = \frac{3.14}{4} \times D^2 \times 2D$$

$$463.1256 = 2 \times D^3$$

$$D^3 = 294.9845 \text{ cuft}$$

$$D = 6.656813 \text{ ft} = 79.88176 \text{ in} = 1.997044 \text{ m}$$

Karena $H = 2D$, maka H :

$$H = 2 \times D$$

$$H = 2 \times 6.657 \text{ ft}$$

$$H = 13.31363 \text{ ft} = 159.7635 \text{ in} = 3.994088 \text{ m}$$

Tinggi liquid dalam tangki:

$$\text{Volume Liquid} = \frac{\pi}{4} \times D^2 \times H$$

$$370.50 = \frac{3.14}{4} \times 44.313164 \times H$$

$$H \text{ liq} = 10.65 \text{ ft}$$

Penentuan tebal shell :

$$t_{\min} = \frac{P \times r_i}{fE - 0,6P} + C \quad [\text{Brownell\&Young; Eq.13-1}]$$

dengan

t_{\min} = tebal shell minimum; in

P = tekanan tangki ; psi



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$r_i = \text{jari-jari tangki} \quad ; \text{in (1/2 D)}$$

$$C = \text{faktor korosi} \quad ; \text{in (digunakan 1/8 in)}$$

$$E = \text{faktor pengelasan, digunakan double weld} \quad E = 0.8$$

$$f = \text{Stress allowable, baha stainless steel 316}$$

$$f = 35969.92 \text{ psi} \quad [Perry \text{ ed.7; T.28-11}]$$

$$P \text{ operasi} = 1 \text{ atr} = 14.7 \text{ lb/in}^2 \text{ (psi)}$$

$$P \text{ hidrostatik} = \rho \times g/gc \times H_{liq}$$

$$= 52.05 \times 1 \times 10.65$$

$$= 554.3498 \text{ lb/ft}^2$$

$$= 3.8497 \text{ lb/in}^2$$

P Design 10% lebih besar untuk faktor keamanan

$$P \text{ design} = (P \text{ operasi} + P \text{ hidrostatik}) \times 110\%$$

$$= 14.7 + 3.8497 \times 110\%$$

$$= 20 \text{ psi}$$

$$r_i = 1 \times D$$

$$= 1 \times 79.8818$$

$$= 39.94088 \text{ in}$$

$$t_{\min} = \frac{P \times r_i}{fE - 0.6P} + C$$

$$= \frac{20 \times 39.94088}{28775.9 - 12.2428} + 0.125$$

$$= \frac{814.9783}{28763.69} + 0.125$$

$$= 0.0283 + 0.125$$

$$= 0.1533 \text{ in}$$

digunakan $t_{\min} = 4/16 \text{ in}$

Dimensi Tutup Atas :

Tutup atas berbentuk standart dished head

$$OD = D + 2 \text{ ts} = 80 \text{ in}$$

$$rc = 130 \text{ in} \quad [Brownell\&Young; T.5-7]$$

$$= 10.8 \text{ ft}$$

Tebal standart toripherical dished (atas) :

$$t_h = 0,885 \times P \times rc + C$$

$$f_e - 0,1P$$

dengan

$$t_h = \text{tebal shell minimur;in}$$

$$P = \text{tekanan tangki} \quad ; \text{psi}$$

$$rc = \text{crown radius} \quad ; \text{in}$$

$$C = \text{faktor korosi} \quad ; \text{in (digunakan 1/8)}$$

$$E = \text{faktor pengelasan, digunakan double weld} \quad E = 0.8$$

$$f = \text{tekanan tangki} \quad \text{stainless steel 316}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
 f &= 35969.92 \text{ psi} \quad [Perry \text{ ed.7; T.28-11}] \\
 P \text{ design} &= 20.40 \text{ psi} \\
 t_h &= \frac{0,885 \times P \times r_c}{f.E - 0,1.P} + C \\
 &= \frac{0.885 \times 20.40 \times 130}{28775.94 - 2.04} + 1/8 \\
 &= 0.207 \text{ in digunakan } t = 4/16 \text{ in} \\
 h &= r_c - \sqrt{r_c^2 - \frac{D^2}{4}} \\
 &= 0.52 \text{ ft}
 \end{aligned}$$

untuk tebal tutup bawah datar karena tutup menumpang diatas beton, maka tebal tutup = 1/3 in [Brownell&Young; p.58]

Tangki Penyimpanan 2-Ethyl Hexanol

Dimensi Shell :

Diameter tangki	: 6.66 ft
Tinggi tang	: 13.31 ft
Tebal Shell	: 4/16 in
Tebal tutup atas	: 4/16 in
Tebal tutup bawah	: 1/3 in
Bahan konstruksi	: Stainless steel 316
Jumlah tangki	: 7 buah

17 POMPA 2-ETHYL HEXANOL (L-141)

Fungsi : Mengalirkan 2-ethyl hexanol dari tangki penyimpanan 2-ethyl hexanol menuju heater

Tipe : Centrifugal pump

Jumlah : 1 buah

Perhitungan :

Bahan masuk = 2551.887 kg/jam = 5626.91 lb/jam

ρ Campuran = 52.05 lb/cuft

$$\begin{aligned}
 \text{Rate volumetri} &= \frac{\text{Rate massa}}{\rho \text{ campuran}} \\
 &= \frac{5626.91 \text{ lb/jam}}{52.04722 \text{ lb/cuft}} \\
 &= 108.1117 \text{ cuft / jam} \\
 &= 1.8019 \text{ cuft / menit} \\
 &= 13.4797 \text{ gpm} \\
 &= 0.0300 \text{ cuft/detik}
 \end{aligned}$$

Asumsi aliran turbulen

(Di) optimum untuk aliran turbulen, $NRe > 2100$ digunakan persamaan :



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\text{Diameter optimum} = 3.9 \times q_f^{0.45} \times \rho^{0.13} \text{ in} \quad [Peters \text{ ed.4; eq.15-496}]$$

Dengan :

$$q_f = \text{Fluid flow rate; (cuft/detik)}$$

$$\rho = \text{Fluid Density; (lb/cuft)}$$

$$\begin{aligned} \text{Diameter pipa optimum, I} &= 3.9 \times 0.03^{0.45} \times 52.05^{0.13} \\ &= 1.3462 \text{ in} \end{aligned}$$

Dipilih pipa 2.5 in, sch 80 [Kern; Tab.11]

$$\text{OD} = 2.880 = 0.24 \text{ ft}$$

$$\text{ID} = 2.323 = 0.1936 \text{ ft}$$

$$\begin{aligned} A &= \left(\frac{1}{4} \times \pi \times \text{ID}^2\right) \\ &= \frac{1}{4} \times 3.14 \times 0.1936^2 \\ &= 0.0294 \text{ ft}^2 \end{aligned}$$

$$\begin{aligned} \text{Kecepatan linier, } v &= \frac{q_f}{A} \\ &= \frac{0.0300 \text{ cuft/detik}}{0.0294 \text{ ft}^2} \\ &= 1.0209 \text{ ft/ detik} \end{aligned}$$

$$\rho \text{ reference} = 62.43 \text{ lb/cuft}$$

$$\text{sg reference} = 1$$

$$\mu \text{ reference} = 0.89$$

$$\text{sg bahan} = \frac{\rho \text{ bahan}}{\rho \text{ reference}} \times \text{sg reference}$$

$$\begin{aligned} &= \frac{52.0472}{62.430} \times 1 \\ &= 0.8337 \end{aligned}$$

μ berdasarkan sg bahan :

$$\begin{aligned} \mu \text{ bahan} &= \frac{\text{Sg bahan}}{\text{Sg reference}} \times \mu \text{ reference} \\ &= \frac{0.8337}{1} \times 0.89 \\ &= 0.7420 \text{ cps} = 0.000499 \text{ lb/ft s} \end{aligned}$$

$$\begin{aligned} \text{NRe} &= \frac{D v \rho}{\mu} \\ &= \frac{0.194 \times 1.0209 \times 52.0472}{0.000499} \\ &= 20629.40 > 2100 \quad (\text{Asumsi turbulen benar}) \end{aligned}$$

Dipilih pipa: Stainless Steel type 302 (Cocok [Perry ed.7; tab.28-2])

$$\varepsilon = 0.000007 \text{ ft} \quad \text{untuk [Peters ed.4; tab.12-4]}$$

$$\varepsilon/D = 0.00004$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$f \text{ (Fanning factor)} = 0.0130 \quad [\text{Moody Chart fanning factor}]$$

Digunakan persamaan Bernoulli : (Mechanical energy balance eq.)

$$-W_f = \frac{\Delta P}{\rho} + \Delta Z \frac{g}{gc} + \frac{\Delta V^2}{2 \alpha gc} + \Sigma F$$

[Geankoplis ed.3; eq.2.10-20]

Perhitungan friksi berdasarkan [Geankoplis ed.3; T2.10-1, p.93]

$$\text{Taksiran panjang pipa lurus} = 35 \text{ ft} = 10.6707 \text{ m}$$

$$\begin{aligned} - 3 \text{ elbow } 90^\circ &= 3 \times 35 \times 0.1936 \\ &= 20.3263 \text{ ft} \end{aligned}$$

$$\begin{aligned} - 1 \text{ gate valve} &= 1 \times 9 \times 0.1936 \\ &= 1.7423 \text{ ft} \end{aligned}$$

$$\text{Panjang total pipa} = 57.0685 \text{ ft}$$

Friksi yang terjadi:

1. Friksi karena gesekan bahan dalam pipa

$$F_1 = \frac{2f \times v^2 \times L_e}{gc \times D} \quad [\text{Geankoplis ed.3; eq 2.10-6}]$$

$$\begin{aligned} &= \frac{2 \times 0.013 \times 1.0209^2 \times 57.07}{32.2170 \times 0.1936} \\ &= 0.2479 \text{ ft lbf/lbm} \end{aligned}$$

2. Friksi karena kontraksi dari tangki ke pipa

$$F_2 = \frac{K_c \times v^2}{2 \times \alpha \times gc} \quad [\text{Geankoplis ed.3 eq. 2.10-16}]$$

untuk aliran turbulen :

$$D1 = 6.657 \text{ ft} \quad (\text{Diameter pre-pompa})$$

$$\alpha = 1$$

$$\frac{A2}{A1} = \frac{0.0294}{34.7858} = 8E-04$$

$$K_c = 0.55 \times \left(1 - \frac{A2}{A1} \right)$$

$$= 0.55 \times 1 - 0.0008 = 0.550$$

Maka :

$$\begin{aligned} F_2 &= \frac{0.55 \times 1.02086^2}{2 \times 1 \times 0.013} \\ &= 22.02677 \text{ ft lbf/lbm} \end{aligned}$$

3. Friksi karena enlargement (ekspansi) dari pipa ke tangki

$$D3 = 0.041 \text{ ft} \quad (\text{Diameter pasca-pompa})$$

$$\frac{A2}{A3} = \frac{0.0294}{0.0013} = 22.18$$

$$K_{ex} = 1 \times \left(1 - \frac{A2}{A3} \right)^2$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$= 1 \times 1 - 22.185 = 448.8$$

Maka:

$$F_3 = \frac{Kex \cdot \Delta v^2}{2 \times \alpha \times gc} \quad [Geankoplis \text{ ed.3 eq. 2.10-15}]$$

$$= \frac{449}{2 \times 1} \times \frac{1.021^2}{32.2170} - 0$$

$$= 7.2586 \text{ ft lbf/lbm}$$

4. Friksi karena elbow 90°

$$F_4 = n \times K_f \times \frac{v_1^2}{2} \quad n = \text{jumlah elbow} \quad [Geankoplis \text{ ed.3; eq 2.10-17}]$$

$$F_4 = 3 \times 0.75 \times 0.5211$$

$$= 1.1724 \text{ ft lbf/lbm}$$

5 Friksi karena Gate Valve

$$F_5 = n \times K_f \times \frac{v_1^2}{2}$$

$$F_5 = 1 \times 0.17 \times 0.5211$$

$$= 0.0886 \text{ ft lbf/lbm}$$

Sehingga :

$$\Sigma F = F_1 + F_2 + F_3 + F_4 + F_5$$

$$= 30.794 \text{ ft.lbf/lbm}$$

$$1 \text{ at} = 14.7 \text{ psi} = 2117 \text{ lbf/ft}^2$$

$$P_1 = 1 \text{ atm} + P \text{ hidrostatik} \quad (H \text{ liquid pre-pump})$$

$$= 2116.8 \text{ lbf/ft}^2 + (52.05 \times 1 \times 10.65)$$

$$= 2116.8 + 554.35$$

$$= 2671.1 \text{ lbf/ft}^2$$

$$P_2 = 1 \text{ atm} = 2117 \text{ lbf/ft}^2$$

$$\Delta P = 2671 - 2116.80$$

$$= 554.3 \text{ lbf/ft}^2$$

$$\frac{\Delta P}{\rho} = \frac{554.3 \text{ lbf/ft}^2}{52.05 \text{ lb/cuft}} = 10.651 \frac{\text{ft.lbf}}{\text{lbm}}$$

$$\frac{\Delta v^2}{2 \times \alpha \times gc} = \frac{1.021^2}{2 \times 1 \times 32.217}$$

$$= 0.016 \text{ ft lbf/lbm}$$

Asumsi	Z1	=	3.281	ft	Tinggi tutup bawah tangki
	Z2	=	0.7	ft	Tinggi total



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\Delta Z = Z_1 - Z_2 = 2.62 \text{ ft} \quad \text{reaktor + asumsi}$$

$$\Delta Z \frac{\text{g}}{\text{gc}} = 2.62 \text{ ft} \times 1 \frac{\text{lbf}}{\text{lbm}}$$

$$= 2.625 \frac{\text{ft.lbf}}{\text{lbm}}$$

Persamaan Bernoulli

$$-W_f = \frac{\Delta P}{\rho} + \Delta Z \frac{\text{g}}{\text{gc}} + \frac{\Delta V^2}{\alpha \text{gc}} + \Sigma F$$

$$= 10.65 + 2.625 + 0.0162 + 30.79$$

$$= 44.09 \frac{\text{ft.lbf}}{\text{lbm}}$$

$$\text{hp} = \frac{-W_f \times \text{flowrate}(\text{gpm}) \times \text{sg}}{3960}$$

$$= \frac{44.09 \times 13.4797 \times 0.834}{3960}$$

$$= 0.1251 \text{ hp}$$

$$\text{Efisiensi Pompa} = 30\% \quad [Peters \text{ ed.4; Fig.14-36}]$$

$$\text{Bhp} = \frac{\text{hp}}{\eta \text{ pompa}}$$

$$= \frac{0.1251}{30\%}$$

$$= 0.4170 \text{ hp}$$

$$\text{Efisiensi motor} = 82\% \quad [Peters \text{ ed.4; Fig.14-38}]$$

$$\text{Power motor} = \frac{\text{Bhp}}{\eta \text{ motor}}$$

$$= \frac{0.4170}{82\%}$$

$$= 0.5086 \text{ hp}$$

$$= 1 \text{ hp}$$

SPESIFIKASI POMPA 2-ETHYL HEXANOL

Fungsi : Mengalirkan 2-ethyl hexanol dari tangki penyimpanan 2-ethyl hexanol menuju heater

Type : *Centrifugal pump*

Power Motor : 1 hp

Rate volumetrik : 13.4797 gpm

Total Dynamic Head : 44.0860 ft.lbf/lbm

Efisiensi Pompa : 30%

Efisiensi Motor : 82%

Bahan konstruksi pipa : Stainless Steel type 302

Bahan konstruksi pump & val : Stainless Steel type 303



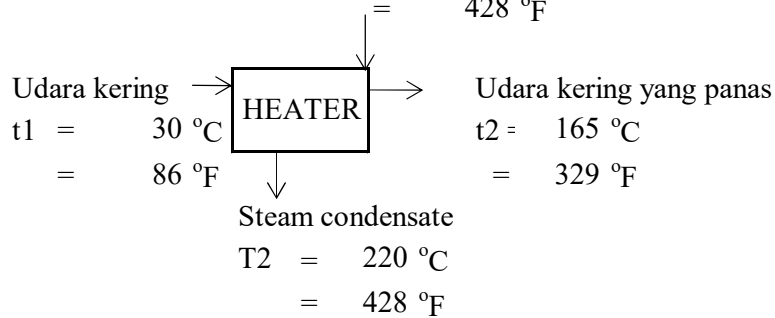
PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Jumlah : 1 Buah

18 HEATER 2-ETHYL HEXANOL (E-142)

Diagram suhu : Steam $T_1 = 220\text{ }^{\circ}\text{C}$
 $= 428\text{ }^{\circ}\text{F}$



Fungsi : Memanaskan *2-ethyl hexanol* cair sebelum masuk reaktor dari su
30 °C menjadi 165 °C

Tipe : Shell and Tube

Jumlah : 1 buah

Heat Balance

Data dari Appendix A dan Appendix B diperoleh :

Rate massa, $W_{\text{massa}} = 2,552\text{ kg/jam}$
 $= 5,627\text{ lb/jam}$

$Q_{\text{supply}} = 57,245\text{ Kkal/jam}$
 $= 227,262\text{ Btu/jam}$

$W_{\text{steam}} = 129\text{ kg/jam}$
 $= 285\text{ lb/jam}$

Log Mean Temperature Difference

Suhu bahan masuk = 30 °C = 86 °F (t_1)

Suhu bahan keluar = 165 °C = 329 °F (t_2)

Suhu steam masuk = 220 °C = 428 °F (T_1)

Suhu steam condensat = 220 °C = 428 °F (T_2)

$$\begin{aligned}\Delta t_1 &= T_2 - t_1 \\ &= 428 - 86 \\ &= 342\text{ }^{\circ}\text{F}\end{aligned}$$

$$\begin{aligned}\Delta t_2 &= T_1 - t_2 \\ &= 428 - 329 \\ &= 99\text{ }^{\circ}\text{F}\end{aligned}$$

(Kern : Hal. 90)

$$R = \frac{T_1 - T_2}{t_2 - t_1} = \frac{0}{243} = 0.000000000000$$

$$S = \frac{t_2 - t_1}{T_1 - t_1} = \frac{243}{342} = 0.711$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned} \Delta t \text{ LMTD} &= \frac{\Delta t_2 - \Delta t_1}{\ln \frac{\Delta t_2}{\Delta t_1}} && (\text{Kern : Pers.5.14, Hal. 149}) \\ &= \frac{99 - 342}{\ln \frac{99}{342}} \\ &= 196 \text{ } ^\circ\text{F} \end{aligned}$$

(Kern : Pers.5.14, Hal. 89)

$$F_T = 1 \quad (\text{Kern : Fig. 18, Hal. 828})$$

$$\begin{aligned} \Delta t &= \Delta t \text{ LMTD} \times F_T \\ &= 196.0166 \times 1 \\ &= 196 \text{ } ^\circ\text{F} \end{aligned} \quad (\text{Kern : Pers.7.42, Hal. 149})$$

Temperatur Rata-rata

$$\begin{aligned} t_c &= t_{av} \text{ Udara} && T_c = T_{av} \text{ Steam} \\ &= \frac{86 + 329}{2} && = \frac{428 + 428}{2} \\ &= 207.5 \text{ } ^\circ\text{F} && = 428 \text{ } ^\circ\text{F} \end{aligned}$$

Ukuran Heater

Asumsi:

Shell Side

$$\text{ID} = 35 \text{ in}$$

$$\text{Baffle space} = 2 \text{ in}$$

$$\text{Passes} = 4$$

$$\text{Berarti : } a't = 0.639 \text{ in}^2$$

$$a'' = 0.262 \text{ ft}^2/\text{ft panjang}$$

Heater Steam-Water, $U_D = 100\text{-}500$

$$U_D = 500 \text{ Btu/jam.ft}^2.\text{ } ^\circ\text{F}$$

Tube Side

$$\text{Number and Length} = 1 \text{ } 16 \text{ } 0''$$

$$\text{OD, BWG, pitch} = 1 \text{ in, 16 BWG, } 1 \text{ } 1/4\text{-in square}$$

$$\text{Passes} = 2$$

(Kern : T.10, Hal. 843)

$$\begin{aligned} A &= \frac{Q}{\Delta t \text{ LMTD} \times U_D} && N_t = \frac{A}{L \times a''} \\ &= \frac{227,262}{196.0166 \times 500} && = \frac{2.3188}{16 \times 0.2618} \\ &= 2.3188 \text{ ft}^2 && = 0.5536 \\ &= 27.826 \text{ in}^2 && = 1 \end{aligned}$$

(Kern : Pers.7.6, Hal. 140)

Sehingga dipilih ukuran yang paling mendekati, yaitu:



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
 \text{OD tube} &= 1 \text{ in} \\
 \text{ID tube} &= 0.902 \text{ in} \\
 \text{Pitch} &= 1 \frac{1}{4} \text{ in square} \\
 \text{ID Shell} &= 35 \text{ in} \\
 \text{Passes tube} &= 1 \\
 \text{Jumlah tube} &= 1
 \end{aligned}$$

Koreksi Nilai U_D (Overall Design Coefficient)

$$\begin{aligned}
 A &= N_t \times L \times a'' \\
 &= 1 \times 16 \times 0.2618 \\
 &= 2.32 \text{ ft}^2
 \end{aligned}$$

$$\begin{aligned}
 U_D &= \frac{Q}{\Delta t \text{ LMTD} \times A} \\
 &= \frac{227,262}{196.0166 \times 2.32} \\
 &= 500 \text{ Btu/jam.ft}^2 \text{ } ^\circ\text{F}
 \end{aligned}$$

Fluida Panas (Tube Side) Steam	Fluida Dingin (Shell Side) Udara
(4) Flow area (a_t) $a_t = \frac{N_t \times a'_t}{144 \times n}$ $= \frac{1 \times 0.639}{144 \times 1}$	(4') Flow area (a_s) $C' = P_t - OD$ $= 1.25 - 1$ $= 1/4 \text{ in}$ $a_s = ID \times C' \times B$
Fluida Panas (Tube Side) Steam $= 0.0025 \text{ ft}^2$	Fluida Dingin (Shell Side) Udara $= \frac{144 \times P_t}{35 \times 1/4 \times 2}$ $= \frac{144 \times 1.25}{35 \times 1.25}$ $= 0.0972 \text{ ft}^2$
(5) Kecepatan massa (Gt) $G_t = \frac{W}{a_t}$ $= \frac{284.526}{0.0025}$ $= 115,826.930 \text{ lb/jam ft}^2$	(5') Kecepatan massa (Gs) $G_s = \frac{W}{a_s}$ $= \frac{5,626.9113}{0.0972}$ $= 57,876.8021 \text{ lb/jam ft}^2$
(6) Pada $T_c = 428 \text{ } ^\circ\text{F}$ $\mu = 0.014 \text{ cps}$	(6') Pada $t_c = 208 \text{ } ^\circ\text{F}$ $\mu = 0.02 \text{ cps}$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Fluida Panas (Tube Side) Steam	Fluida Dingin (Shell Side) Udara
$= 0.034 \text{ lb/jam.ft}$	$= 0.048 \text{ lb/jam.ft}$
$D_i = \frac{0.902}{12}$ $= 0.075 \text{ ft}$	$D_e = \frac{4 \times a_s}{\text{wetted perimeter}}$ $= \frac{4 \times 0.0972}{1 \times 3.14 \times \frac{1}{12}}$ $= 2.6847 \text{ ft}$
$Re_t = \frac{D_i \times G_t}{\mu}$ $= \frac{0.075 \times 115,826.930}{0.03388}$ $= 256975.3$	$Re_s = \frac{D_e \times G_s}{\mu}$ $= \frac{2.6847 \times 57,876.8021}{0.0484}$ $= 3210423.635$
<p>(7) $jH = 250$ (Kern: Fig.24)</p>	<p>(7) $jH = 28$ (Kern: Fig.28)</p>
<p>(9) Condensation of steam :</p> $h_{io} = 1000 \text{ Btu/hr.ft}^2 \cdot ^\circ\text{F}$ <p>(Kern:164)</p>	<p>(8') Pada $t_c = 208 \text{ } ^\circ\text{F}$</p> $k = (\text{Kern: T.5, Hal.801})$ $= 0.017 \text{ Btu/hr.ft}^2 \cdot (^\circ\text{F/ft})$ $C_p = 0.25 \text{ Btu / lb.}^\circ\text{F}$ $\left[\frac{C_p \cdot \mu}{k} \right] = \left[\frac{0.250 \times 0.048}{0.017225} \right]^{1/3}$ $= 0.8889$ <p>(9')</p> $h_o = jH \times \frac{k}{D_e} \times \left[\frac{C_p \cdot \mu}{k} \right]^{1/3} \phi_s$ $\frac{h_o}{\phi_s} = 28 \times \frac{0.0172}{2.6847} \times 0.889$ $= 0.159694$
<p>(10) t_w^*</p> $t_w = t_c + \frac{h_{io}}{h_{io} + h_o} (T_c - t_c)$ $= 207.5 + \frac{1000}{1000.1597} \times 220.5$ $= 427.96479 \text{ } ^\circ\text{F}$	<p>(11') Pada $t_w = 427.96479 \text{ } ^\circ\text{F}$</p> $\mu_w = (\text{Kern: Fig.15})$ $= 0.0145 \text{ cps}$ $= 0.03509 \text{ lb / jam.ft}$ $\phi_s = (\mu/\mu_w)^{0.14}$ $= 13.01946$



PERANCANGAN PABRIK

Pabrik Diocetyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Fluida Panas (Tube Side) Steam	Fluida Dingin (Shell Side) Udara
	(12') Corrected Coefficient $h_o = \frac{h_o \cdot \phi_s}{\phi_s}$ $= 2.079134 \text{ Btu/hr.ft}^2 \cdot ^\circ\text{F}$
(13) Clean Overall coefficient, U_c : $U_c = \frac{h_{io} \times h_o}{h_{io} + h_o}$ $= \frac{1000 \times 2.0791342}{1000 + 2.0791342}$ $= 2.074820357 \text{ Btu/hr.ft}^2 \cdot ^\circ\text{F}$	
(14) Dirt factor, R_d : $R_d = \frac{U_c - U_D}{U_c \times U_D}$ $= \frac{2.0748204 - 500}{2.0748204 \times 500}$ $= 0.479969437 \text{ Btu/hr.ft}^2 \cdot ^\circ\text{F}$	
0.002 (Kern: T-12, Hal. 845)	

Pressure Drop

Fluida Panas (Tube Side) Steam	Fluida Dingin (Shell Side) Larutan NaOH
(1) Specific vol of steam from table 7 : $v = 6.3000 \text{ ft}^3/\text{lb}$ $s = \frac{0.1587}{62.5} = 0.0025$ $Re_t = 256975.3$ $f = 0.00011 \text{ ft}^2/\text{in}^2$ (Kern: Fig.26, Hal. 836) $\Delta P_t = \frac{1}{2} \times \frac{f \cdot G_t^2 \cdot L \cdot n}{5.22 \times 10^{10} \times D \times s \times \phi_t}$ $= 1.1847 \text{ Psi}$ $\frac{V^2}{2g'} = 0.001$ (Kern: Fig.27, Hal. 837)	(1') Flow area (a_s) $N+1 = 12L/B$ $= 96$ $De' = \frac{4 \times \text{flow area}}{\text{frictional wetted perimeter}}$ $= \frac{4 \times 0.0972}{1 \times 3.14 \times \frac{1}{12}}$ $= 3.14 \times \frac{35}{12}$ $= 0.0418 \text{ ft}$ $Re's = \frac{De' \times G_s}{\mu}$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Fluida Panas (Tube Side) Steam	Fluida Dingin (Shell Side) Larutan NaOH
$\Delta P_r = \frac{4 n}{sg} \times \frac{V^2}{2g'}$ $= \frac{4 \cdot 1}{0.16} \times 0.001$ $= 0.0252$	$= \frac{0.0418 \times 57,876.8021}{0.0484}$ $= 49986.5381$
$\Delta P_T = \Delta P_t + \Delta P_r$ $= 1.1847 + 0.0252$ $= 0.0012 \text{ psi}$	$f = 0.0015 \text{ ft}^2/\text{in}^2$ <p>(Kern: Fig.29, Hal. 839)</p>
$\Delta P_T < 2 \text{ psi}$ <p>(memenuhi untuk steam)</p>	$\Delta P_s = \frac{f \cdot G_s^2 \cdot D_s \cdot (N+1)}{5.22 \times 10^{10} \times D_e \times s \times \phi_s}$ $= \frac{1295014778.1730}{52102286941.7491}$ $= 0.0249 \text{ Psi}$
	$\Delta P_s < 10 \text{ psi}$ <p>(memenuhi untuk liquid)</p>

SPESIFIKASI HEATER 2-ETHYL HEXANOL

Bagian Tube

OD	=	1	in
ID	=	0.902	in
BWG	=	16	
Panjang	=	16	ft
Pitch	=	1 1/4	in square
Jumlah tube, Nt	=	1	
Passes	=	1	

Bagian Shell

ID	=	35	in
Passes	=	4	
A	=	2.32	ft ²
U _D	=	500	Btu/jam.ft ² °F
Jumlah heater	=	1	buah

19 TANGKI PENYIMPANAN TETRABUTYL TITANATE (F-150)

Fungsi	:	Tempat penyimpanan katalis <i>tetrabutyl titanate</i>
Tipe	:	Tangki silinder vertikal dengan <i>flat bottom</i> dan <i>conical roof</i>
Jumlah	:	1 buah
Kondisi Operasi	:	T = 30 °C = 303.15 K
	:	P = 1 atm



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$= 760 \text{ mmHg}$$

$$= 14.7 \text{ psi}$$

KOMPOSISI

Komponen	% Berat	berat (kg/jam)	ρ (gr/cc)
$C_{16}H_{36}O_4T$	0.996	0.059767162	0.998
H_2O	0.004	0.000240029	0.997
Total	1	0.06000719	

$$1 \text{ kg} = 2.2 \text{ lb}$$

$$1 \text{ gr/cc} = 62.4 \text{ lb/cuft}$$

$$\begin{aligned} \text{Densitas Campuran} &= \frac{1}{\frac{\text{fraksi berat}}{\rho \text{ komponen}}} \times 62.4 \\ &= \frac{1}{\frac{0.996}{0.998} + \frac{0.004}{1}} \times 62.43 \\ &= 0.9980 \text{ gr/cc} \times 62.43 \\ &= 62.30 \text{ lb/cuft} \end{aligned}$$

$$\begin{aligned} \text{Rate Volumetrik} &= \frac{\text{Rate Massa}}{\text{Densitas}} \\ &= \frac{0.13}{62.3049} \\ &= 0.0021 \text{ cuft/jam} \end{aligned}$$

Direncanakan penyimpanan untuk 30 hari dengan 1 buah tangki

$$\begin{aligned} V \text{ liquid} &= \frac{0.00 \text{ cuft/jar}}{24 \text{ jam}} \times 30 \text{ hari} \\ &= 1.53 \text{ cuft} \end{aligned}$$

Asumsi volume bahan (liquid) mengisi 80% volume tangki sehingga volume ruang kosong sebesar 20%

$$\begin{aligned} \text{Volume Tangki} &= \frac{1.5284}{80\%} \\ &= 1.9104 \text{ cuft} = 14.29015 \text{ gal} \end{aligned}$$

Disarankan > 10000 gal, menggunakan vertical tank dan pondasi beton

[Walas ed.2; p.xvii]

Menentukan ukuran tangki dan ketebalannya

$$\text{Dimensi ratio} = \frac{H}{D} = 2 \quad [\text{Ulrich; T.4-27:p.248}]$$

Dengan mengabaikan volume dished head.

$$\text{Volume tangki} = \frac{\pi}{4} \times D^2 \times H$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$1.9104 = \frac{3.14}{4} \times D^2 \times 2D$$

$$1.9104 = 2 \times D^3$$

$$D^3 = 1.217 \text{ cuft}$$

$$D = 1.068 \text{ ft} = 12.811297 \text{ in} = 0.320282 \text{ m}$$

Karena $H = 2D$, maka H :

$$H = 2 D$$

$$H = 2 \times 1.0676081 \text{ ft}$$

$$H = 2.135216 \text{ ft} = 25.62259 \text{ in} = 0.640565 \text{ m}$$

Tinggi liquid dalam tangki:

$$\text{Volume Liquid} = \frac{\pi}{4} \times D^2 \times H$$

$$1.53 = \frac{3.14}{4} \times 1.1397871 \times H$$

$$H_{\text{liq}} = 1.708173 \text{ ft}$$

Penentuan tebal shell :

$$t_{\min} = \frac{P \times r_i}{fE - 0,6P} + C \quad [Brownell\&Young; Eq.13-1]$$

dengan

t_{\min} = tebal shell minimum; in

P = tekanan tangki ; psi

r_i = jari-jari tangki ; in ($1/2 D$)

C = faktor korosi ; in (digunakan $1/8$ in)

E = faktor pengelasan, digunakan double weld $E = 1$

f = Stress allowable, baha stainless steel 316

$$f = 35970 \text{ psi} \quad [Perry ed.7; T.28-11]$$

$$P_{\text{operasi}} = 1 \text{ atr} = 14.7 \text{ lb/in}^2 \text{ (psi)}$$

$$P_{\text{hidrostatik}} = \rho \times g/gc \times H_{\text{liq}}$$

$$= 62.30 \times 1 \times 1.708$$

$$= 106.4275 \text{ lb/ft}^2$$

$$= 0.7391 \text{ lb/in}^2$$

P Design 10% lebih besar untuk faktor keamanan

$$P_{\text{design}} = (P_{\text{operasi}} + P_{\text{hidrostatik}}) \times 110\%$$

$$= 14.7 + 0.7391 \times 110\%$$

$$= 17 \text{ psi}$$

$$r_i = 0.5 \times D$$

$$= 0.5 \times 12.81$$

$$= 6.405649 \text{ in}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
t_{\min} &= \frac{P \times r_i}{fE - 0,6P} + C \\
&= \frac{17}{28775.94} \times \frac{6.4056487}{10.1898} + 0.125 \\
&= \frac{108.7871}{28765.75} + 0.125 \\
&= 0.0038 + 0.125 \\
&= 0.1288 \text{ in}
\end{aligned}$$

digunakan $t_{\min} = 4/16$ in

Dimensi Tutup Atas :

Tutup atas berbentuk standart dished head

$$\begin{aligned}
OD = D + 2 \text{ ts} &= 13 \text{ in} \\
rc &= 130 \text{ in} \quad [Brownell\&Young; T.5-7] \\
&= 10.8 \text{ ft}
\end{aligned}$$

Tebal standart toripherical dished (atas) :

$$t_h = \frac{0,885 \times P \times rc}{fe - 0,1P} + C$$

lengan

$$\begin{aligned}
t_h &= \text{tebal shell minimum; in} \\
P &= \text{tekanan tangki ; psi} \\
rc &= \text{crown radius ; in} \\
C &= \text{faktor korosi ; in (digunakan } 1/8 \text{)} \\
E &= \text{faktor pengelasan, digunakan double weld } E = 1 \\
f &= \text{faktor korosi stainless steel 316} \\
f &= 35970 \text{ psi} \quad [Perry \text{ ed. 7; T.28-11}]
\end{aligned}$$

$P_{\text{design}} = 14.70$ psi

$$\begin{aligned}
t_h &= \frac{0,885 \times P \times rc}{fE - 0,1.P} + C \\
&= \frac{0.885 \times 14.7 \times 130}{28775.94 - 1.47} + 1/8 \\
&= 0.184 \text{ in} \text{ digunakan } t = 4/16 \text{ in}
\end{aligned}$$

$$\begin{aligned}
h &= rc - \sqrt{rc^2 - \frac{D^2}{4}} \\
&= 0.01 \text{ ft}
\end{aligned}$$

untuk tebal tutup bawah datar karena tutup menumpang diatas beton, maka tebal tutup = $1/3$ in *[Brownell&Young; p.58]*

SPEKIFIKASI TANGKI PENYIMPANAN TETRABUTYL TITANATE

Diameter tangki : 1.07 ft
Tinggi tang : 2.14 ft



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Tebal Shell : 4/16 in
Tebal tutup atas : 4/16 in
Tebal tutup bawah : 1/3 in
Bahan konstruksi : Stainless steel 316
Jumlah tangki : 1 buah

20 POMPA TETRABUTYL TITANATE (L-151)

Fungsi : Mengalirkan *tetrabutyl titanate* dari tangki penyimpanan *tetribut titanate* menuju *heater*
Tipe : *Centrifugal pump*
Jumlah : 1 buah
Tujuan : 1. Menentukan tipe pompa
2. Menentukan bahan konstruksi pompa
3. Menghitung tenaga pompa
4. Menghitung tenaga motor

3) Menghitung Tenaga Pompa

Rate masuk = 0.0600 kg/jam = 0.0000367 lb/s
Densitas = 0.9980 kg/m³ = 0.0623 lb/ft³
 μ = 1.2369 cP = 0.0008 lb/ft.s

$$\begin{aligned} \text{Rate volumetrik (Q)} &= \frac{\text{massa}}{r} \\ &= \frac{0.000037 \text{ lb/s}}{0.0623 \text{ lb/ft}^3} = 0.000590 \text{ ft}^3/\text{s} \\ &= 0.2647 \text{ gpm} \end{aligned}$$

Diperkirakan aliran fluida laminar ($N_{re} < 2100$), sehingga digunakan persamaan $Di < 1$ in yaitu :

$$Di_{opt} = 3.6 Q^{0.40} \mu^{0.20} \quad \text{Eq. 48. (Peters \& Timmerhaus, 1991:365)}$$

Dimana:

Di_{opt} = diameter dalam optimum, in
 Q = kecepatan volumetrik, ft³/s
 μ = viskositas fluida, lb/ft.s

Sehingga:

$$\begin{aligned} Di_{opt} &= 3.6 \times (0.0005898)^{0.40} \times (0.000831)^{0.20} \\ &= 0.0445 \text{ in} \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Dari App. A.5-1 (Geankoplis, 1993:892) dipilih spesifikasi :

$$\begin{aligned}
 \text{NPS} &= 1/8 \text{ in} \\
 \text{OD} &= 0.4050 \text{ in} = 0.03375 \text{ ft} = 0.010287 \text{ m} \\
 \text{ID} &= 0.2690 \text{ in} = 0.0224167 \text{ ft} = 0.006833 \text{ m} \\
 \text{A} &= 0.0004 \text{ ft}^2 = 0.3716 \text{ in}^2 \\
 \text{Sch} &= 40
 \end{aligned}$$

Menghitung kecepatan linier :

Kecepatan linier fluida dapat dicari dengan menggunakan persamaan berikut :

$$V = \frac{Q}{A}$$

Sehingga :

$$V = \frac{6.E-04 \text{ ft}^3/\text{s}}{0.0004 \text{ ft}^2} = 1.4745 \text{ ft/s} = 0.4494 \text{ m/s}$$

Menghitung *Reynold Number* (Nre) :

$$\text{Nre} = \frac{\rho v D}{\mu}$$

Sehingga :

$$\begin{aligned}
 \text{Nre} &= \frac{0.0623 \times 1.4745 \times 0.0224}{0.0008} \\
 &= 2.4776 \quad (\text{asumsi aliran laminar benar})
 \end{aligned}$$

Instalasi pipa

Dari **Fig. 127** (Brown, 1950:141) dengan NPS = 1/8 in

- 1 buah <i>gate valve fully open</i> ; Le	= 0.2 ft
$\sum \text{Le} = 1 \times 0.2$	= 0.2 ft
- 4 buah <i>standard elbow</i> ; Le	= 1 ft
$\sum \text{Le} = 4 \times 1$	= 4 ft
- 1 buah <i>sudden enlargement</i> ; Le	= 4.5 ft
$\sum \text{Le} = 1 \times 4.5$	= 4.5 ft
- 1 buah <i>sudden contraction</i> ; Le	= 2.5 ft
$\sum \text{Le} = 1 \times 2.5$	= 2.5 ft
- 1 buah <i>swing check valve</i> ; Le	= 2.5 ft
$\sum \text{Le} = 1 \times 2.5$	= 2.5 ft
- Panjang ekivalen pipa lurus, $\sum \text{Le}$	= 13.7 ft
Panjang pipa lurus	= 2 m
	= 6.5616 ft
Panjang pipa total	= 20.2616 ft
	= 6.1757 m



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Menghitung *friction loss*

$$\Sigma F = \frac{4f \cdot v^2 \cdot \Delta L}{2 \cdot ID \cdot gc} \quad \text{Eq. 2.10-6 (Geankoplis, 1993:89)}$$

Dimana :

$$\Sigma F = \text{Friction loss} \quad (\text{ft.lbf/lbm})$$

f = Faktor friksi

v = Kecepatan linier fluida (ft/s)

ΔL = Panjang pipa (ft)

ID = Diameter dalam tangki (ft)

gc = 32.1740 lbm.ft/lbf.s²

Menghitung *Fanning Friction Factor* (f) :

Dari **Fig. 2.10-3.** (Geankoplis, 1993:88) didapat :

$$\begin{aligned} \text{Untuk commercial steel} \quad \rightarrow \quad \varepsilon &= 5.E-05 \quad \text{m} \\ &= 0.0002 \quad \text{ft} \end{aligned}$$

Sehingga :

$$\frac{\varepsilon}{D} = \frac{0.00016}{0.0224} = 0.0070$$

Untuk aliran laminar dari Eq. 2.10-7. (Geankoplis, 1993:92) diperoleh :

$$f = \frac{16}{N_{Re}} = \frac{16}{2.48E+00} = 6.4578$$

$$\begin{aligned} \Sigma F &= \frac{4 \times 6.4578 \times (1.4745)^2 \times 20.2616}{2 \times 0.0224 \times 32.1740} \\ &= 788.8724 \quad \frac{\text{lb}_f \cdot \text{ft}}{\text{lb}_m} \end{aligned}$$

Menghitung *Static Head* :

$$Z_1 = 1 \quad \text{m} = 3.2808 \quad \text{ft}$$

$$Z_2 = 0.2 \quad \text{m} = 0.6562 \quad \text{ft}$$

$$\Delta Z = Z_1 - Z_2 = 2.6247 \quad \text{ft}$$

$$g/gc = 1 \quad \text{lbf/lbm}$$

$$\Delta Z (g/gc) = 2.624672 \quad \text{ft} \times 1 \quad \text{lbf/lbm} = 2.624672 \quad \text{ft lbf/lbm}$$

Menghitung *Velocity Head* :

$$V_1 = \text{kecepatan linier fluida dari tangki hidrogen peroksida ke pipa}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

V_2 = kecepatan linier fluida ke *heater*

Karena pada 2 titik *reference* dianggap sama, maka $V_1 = V_2 = 1.4745$ ft/s

Sehingga *velocity head* ($\Delta V^2 / 2gc$) = 0.03379

Menghitung *Pressure Head* :

$$P_1 = P_2 = 14.7 \text{ Psi}$$

$$\Delta P = 0$$

Sehingga, $\Delta P/\rho = 0$

Menghitung Energi Mekanik Pompa :

$$-W_f = \frac{\Delta V^2}{2 \times \alpha \times gc} + \Delta z \frac{g}{gc} + \frac{\Delta P}{\rho} + \Sigma F$$

Dimana :

W_f = tenaga yang ditambahkan ke dalam sistem per satuan massa

Sehingga :

$$\begin{aligned} -W_f &= 0.034 + 2.625 \frac{\text{ft.}}{\text{lbf/lbm}} + 0 + 788.8724 \frac{\text{ft.}}{\text{lbf/lb}} \\ &= 791.5309 \text{ ft. lbf/lbm} \end{aligned}$$

Menghitung *Broke Horse Power* (BHP) :

$$\text{BHP} = \frac{m. (-W_f)}{550 \cdot \eta}$$

dari **Fig. 14-37.** (Peters & Timmerhaus, 1991:520), untuk $Q = 0.2647$ gpm diperoleh η pompa = 0.40

Sehingga :

$$\begin{aligned} \text{BHP} &= \frac{0.000037 \times 791.5309}{550 \times 0.40} \\ &= 1.E-04 \text{ Hp}; \text{ maka digunakan } power = 1 \text{ Hp} \end{aligned}$$

4) Menghitung Tenaga Motor

Fig 14.38. (Peters & Timmerhaus, 1991:521) untuk $E = 1$ Hp

diperoleh η motor = 0.80

Sehingga *power* motor yang diperlukan :

$$\begin{aligned} P_{\text{motor}} &= \frac{\text{BHP}}{\eta} \\ &= \frac{1}{0.8} \text{ Hp} \\ &= 1.3 \text{ Hp} \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Dipilih motor standar dengan $pow = 2 \text{ Hp}$ (Standard NEMA)

Spesifikasi Pompa $C_{16}H_{36}O_4Ti$

Nama Alat	: Pompa $C_{16}H_{36}O_4Ti$
Kode	: L-151
Fungsi	: Mengalirkan <i>tetrabutyl titanate</i> dari tangki penyimpanan <i>tetrabutyl titanate</i> menuju <i>heater</i>
Tipe	: <i>Centrifugal pump</i>
Bahan Konstruksi	: <i>Commercial steel</i>
Jumlah	: 1 buah
Rate Volumetrik	: 0.00059 ft ³ /s
Kecepatan Aliran	: 1.47451 ft/s
Ukuran Pipa	: NPS = 1/8 in
	: <i>Sch. Number</i> = 40
	: OD = 0.0338 ft = 0.0103 m
	: ID = 0.0224 ft = 0.0068 m
	: <i>Flow Area</i> = 0.0004 ft ² = 3.7E-05 m ²
<i>Power</i> Pompa	: 1.0 Hp
<i>Power</i> Motor	: 2.0 Hp

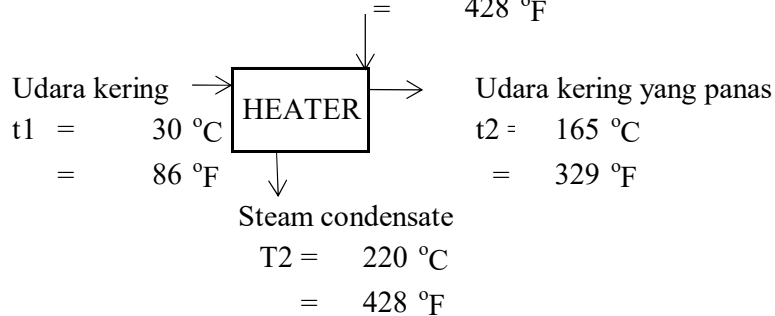
21 HEATER TETRABUTYL TITANATE (E-152)

Fungsi : Memanaskan katalis cair *tetrabutyl titanate* sebelum masuk reaktor dari suhu = 30 °C menjadi = 165 °C

Tipe : Shell and Tube

Jumlah : 1 buah

Diagram suhu : Steam T1 = 220 °C
= 428 °F



Heat Balance

Data dari Appendix A dan Appendix B diperoleh :

Rate massa, W massa	= 0.0600 kg/jam
	= 0.1323 lb/jam
Q supply	= 4.2768 Kkal/jam



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
 &= 16.9790 \text{ Btu/jam} \\
 W \text{ steam} &= 0.0096 \text{ kg/jam} \\
 &= 0.0213 \text{ lb/jam}
 \end{aligned}$$

Log Mean Temperature Difference

$$\begin{aligned}
 \text{Suhu bahan masuk} &= 30 \text{ }^\circ\text{C} = 86 \text{ }^\circ\text{F} (t_1) \\
 \text{Suhu bahan keluar} &= 165 \text{ }^\circ\text{C} = 329 \text{ }^\circ\text{F} (t_2) \\
 \text{Suhu steam masuk} &= 220 \text{ }^\circ\text{C} = 428 \text{ }^\circ\text{F} (T_1) \\
 \text{Suhu steam condensat} &= 220 \text{ }^\circ\text{C} = 428 \text{ }^\circ\text{F} (T_2)
 \end{aligned}$$

$$\begin{aligned}
 \Delta T \text{ LMTD} &= \frac{342 - 99}{\ln 3.4545} \\
 &= 196.0166 \text{ }^\circ\text{F}
 \end{aligned}$$

$$R = \frac{T_1 - T_2}{t_2 - t_1} = \frac{0}{243} = 0.0000$$

$$S = \frac{t_2 - t_1}{T_1 - t_1} = \frac{243}{342} = 0.711$$

Untuk type heater dipilih jenis 2-4 shell & tube exchanger

$$\text{Untuk 2-4 shell \& tube } F_T = 0.97 \quad [Kern; App fig. 18]$$

$$\begin{aligned}
 \Delta T &= F_T \times \Delta T \text{ LMTD} \\
 &= 0.97 \times 196.02 \\
 &= 190.1361 \text{ }^\circ\text{F}
 \end{aligned}$$

Tc dan tc

$$T_c = T_{av} \text{ steam}$$

$$T_c = \frac{T_1 + T_2}{2} = \frac{428 + 428}{2} = 428 \text{ }^\circ\text{F}$$

$$t_c = t_{av} \text{ bahan}$$

$$t_c = \frac{t_1 + t_2}{2} = \frac{86 + 329}{2} = 208 \text{ }^\circ\text{F}$$

$$\begin{aligned}
 Sg \text{ bahan} &= \frac{\rho \text{ bahan}}{\rho \text{ reference}} \times Sg \text{ reference} \\
 &= \frac{0.0727 \text{ lb/c}}{6.524 \text{ lb/c}} \times 1 \\
 &= 0.0111
 \end{aligned}$$

μ berdasarkan Sg bahan :

$$[Kern; tab.6 p. 808] \text{ didapat } Sg \text{ reference} = 1$$

$$[Kern; fig.14, p. 823] \text{ didapat } \mu \text{ reference} = 0.899 \text{ Cp}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned} \mu \text{ bahan} &= \frac{\text{Sg bahan}}{\text{Sg reference}} \times \mu \text{ reference} \\ &= \frac{0.0111}{1} \times 0.899 \\ &= 0.0100 \quad \text{Cp} \end{aligned}$$

Pemilihan UD Heater:

Hot Fluid : Steam

Cold Fluid : Gas

Sehingga di dapat nilai U_D dengan rang 5 - 50 Btu/jam.ft²°F

[Kern; tab.8 , p.840]

Dipilih nilai $U_D = 31$ Btu/jam.ft²°F

Digunakan 1 buah heater, sehingga luas perpindahan panas heater:

$$A = \frac{Q}{UD \times \Delta T}$$

Dimana,

A = Luas Perpindahan Panas

Q = Q supply dari steam

UD = Overall Design Coefficients

ΔT = Perubahan Suhu

$$\begin{aligned} A &= \frac{Q}{UD \times \Delta T} \\ &= \frac{16.9790}{31 \times 190.1361} \\ &= 0.0029 \quad \text{ft}^2 \\ &= 0.0003 \quad \text{m}^2 \end{aligned}$$

Dikarenakan luas perpindahan panas > 200 ft², maka design heater menggunakan shell and tube [Walas; p.xvi]

$$\begin{aligned} \text{Jumlah tube (Nt)} &= \frac{A}{L \times a''} \\ &= \frac{0.0029}{16 \times 0.3271} \\ &= 0.0006 \quad \text{buah} \end{aligned}$$

Dalam perencanaan ini digunakan HEATER dengan spesifikasi :

Digunakan TUBE dengan ukuran : [Kern; tab. 10, p. 843]

OD, BWG = 1 1/4 in , 16 BWG

Pitch = 1 3/4 in triangular pitch [Kern; tab.9]

Panjang tube,L = 16 ft

Passes = 4

ID Tube = 1.120 in 0.0933 ft



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
 a't &= 0.985 \text{ in}^2 \\
 a'' &= 0.3271 \text{ ft}^2/\text{lin ft} \\
 \text{Digunakan Nt} &= 155 \text{ buah}
 \end{aligned}$$

Dari tube passes dan jumlah tube, didapat SHELL:

Untuk pipa 1 1/4 OD, 1 9/16 in triangular

$$\text{ID shell} = 25 \text{ in} \quad [Kern; \text{tab.9}]$$

$$\text{Baffle} = 20 \text{ in}$$

$$\text{Passes (n)} = 2$$

*Koreksi U_D

$$\begin{aligned}
 A &= Nt \times L \times a'' \\
 &= 811.2080 \text{ ft}^2
 \end{aligned}$$

$$\begin{aligned}
 U_D \text{ koreksi} &= \frac{Q}{A \times \Delta T} \\
 &= \frac{16.979}{811.2 \times 190.1} \\
 &= 0.00011 \text{ Btu/jam.ft}^2 \text{ } ^\circ\text{F}
 \end{aligned}$$

Nilai U_D asumsi sama dengan U_D koreksi (memenuhi)

Fluida Panas (Tube Side) Steam	Fluida Dingin (Shell Side) Gas
(4) Flow area (a_t) $ \begin{aligned} a_t &= \frac{Nt \times a'_t}{n} \\ &= \frac{155 \times 0.99}{144 \times 4} \\ &= \text{#####} \text{ ft}^2 \end{aligned} $	(4') Flow area (a_s) $ \begin{aligned} C' &= \text{Pitch} - \text{OD} = 0.50 \text{ in} \\ a_s &= \frac{\text{IDshell} \times C' \times E}{144 \times P_T} \\ &= \frac{25 \times 1/2 \times 20}{144 \times 1 \ 3/4} \\ &= 0.9921 \text{ ft}^2 \end{aligned} $
(5) Kecepatan massa (Gt) $ \begin{aligned} G_t &= \frac{W \text{ mass}}{a_t} \\ &= \frac{0.0213}{0.2651} \\ &= 0.0802 \text{ lb/jam ft}^2 \end{aligned} $	(5') Kecepatan massa (Gs) $ \begin{aligned} G_s &= \frac{W \text{ mass}}{a_s} \\ &= \frac{0.1323}{0.9921} \\ &= 0.1334 \text{ lb/jam ft}^2 \end{aligned} $
(6) pada $t_c = 208 \text{ } ^\circ\text{F}$ $ \begin{aligned} \mu \text{ steam} &= \text{#####} \text{ cps} \\ &= 0.036 \text{ lb/jam.ft} \\ &[Kern; \text{Fig.15 p. 825}] \end{aligned} $	(6') $T_c = 0 \text{ } ^\circ\text{F}$ $ \begin{aligned} \mu \text{ bahan} &= 0.023 \text{ cps} \\ &= 0.056 \text{ lb/jam.ft} \\ De &= 0.0758 \text{ ft} \\ &[Kern; \text{Fig.28}] \end{aligned} $



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Fluida Panas (Tube Side) Steam	Fluida Dingin (Shell Side) Gas
$Di = 1.12 \text{ in} = 0.0933 \text{ ft}$ $Re_t = \frac{Di \times G_t}{\mu}$ $= 0.206$	$Re_s = \frac{D_e \times G_s}{\mu}$ $= 0.1817$ <p>(7') $jH = 150$ [Kern; Fig.28 shell side data]</p> <p>(8') pada $T_c = 428 \text{ }^\circ\text{F}$ $k = 0.0198 \text{ Btu/hr.ft}^2.(^\circ\text{F/ft})$ [Kern; Tab.5] [Kern; fig.3] $C_p = 0.25 \text{ Btu / lb.}^\circ\text{F}$ $\frac{C_p \cdot \mu}{k} \Big ^{1/3} = \frac{0.25 \times 0.056}{0.0198}$ $= 0.8891$</p> <p>(9) $h_o = J_H \times (k/De) \times (c \times \mu / k)^{1/3} \times \phi_s$ $\frac{h_o}{\phi_s} = 35 \text{ Btu/hr.ft}^2.\text{F}$</p> <p>(10) $\phi_s = 1$ $h_o = 35 \text{ Btu/hr.ft}^2.\text{F}$</p>
<p>(13) Clean Overall coefficient, U_c :</p> $U_c = \frac{h_{io} \times h_o}{h_{io} + h_c}$ $= 34.0310 \text{ Btu/hr.ft}^2.\text{F}$ <p>(14) Dirt Factor, R_d :</p> $R_d \text{ hitung} = \frac{U_c - U_D}{U_c \times U_D} \quad [Kern; T-12 p.845]$ <p style="text-align: center;">(Air)</p> $= 0.0029 \text{ Btu/hr.ft}^2.\text{F}$ <p>$R_d \text{ perhitungan} > R_d \text{ data} \quad 0.0029 > 0.0020$</p>	
Fluida Dingin (Tube Side) Steam	Fluida Panas (Shell Side) Gas
<p>(1) $Re_t = 0.206$ $f = 0.00025 \text{ ft}^2/\text{in}^2$ [Kern; Fig.26] $s = \frac{6.52}{62.43} = 0$ [Kern; Tab 7]</p>	<p>(1') $Re's = 0.1817$ $f = 0.002 \text{ ft}^2/\text{in}^2$ [Kern; Fig.29] $s = \frac{0.0727}{62.43} = 0.0012$</p> <p>(2') $N+1 = 12L/B$</p>



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Fluida Dingin (Tube Side)	Fluida Panas (Shell Side)
Steam	Gas
$\Delta P_t = \frac{(1/2) \times f \cdot G_t^2 \cdot L \cdot n}{5.22 \times 10^{10} \times D \times s \times \phi^5}$ $= \frac{1}{2} \times \frac{0.0001}{509129073}$ $= 0.000000 \text{ Psi}$	$= 9.6$ $D_s = 2.083 \text{ ft}$ $\Delta P_s = \frac{f \cdot G_s^2 \cdot D_s \cdot (N+1)}{5.22 \times 10^{10} \times D_e \times s \times \phi^5}$ $= \frac{0.0006}{4609690.0529}$ $= 1.31205E-10 \text{ Psi}$
$\Delta P_T < 2 \text{ psi}$ (memenuhi)	$\Delta P_s < 10 \text{ psi}$ (memenuhi)

SPEKIFIKASI HEATER TETRABUTYL TITANATE

Type Heater = 2-4 Shell & Tube Exchanger

Jumlah = 1 buah

Bagian shell :

Jenis Bahan = Carbon steel

ID Shell = 25 in

Baffle Space = 20 in

Passes = 2

Bagian tube :

Jumlah & Panjang Tul = 155 buah, 16 ft

OD, BWG, pitch = 1 1/4 in, 16 BWG 1 3/4 Triangular

Passes = 4

22 REAKTOR (R-210)

Temperatur = 165 C

Tekanan = 1 atm

Waktu Operas = 1,5 jam

Fungsi = Mereaksikan *phthalic anhydride* dengan *2-ethyl hexanol* menggunakan katalis *tetrabutyl titanate* untuk membentuk *dioctyl phthalate*

Tipe = Silinder tegak, tutup atas dan bawah berbentuk elliptical dishead

PERHITUNGAN

Bahan Masuk

Komponen	%Berat	Massa (Kg)	ρ (gr/ml)
$C_8H_4O_3$	0.361189	1445.35	1.2
$C_8H_6O_4$	0.000725	2.899398	1.522



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Komponen	%Berat	Massa (Kg)	ρ (gr/ml)
$C_8H_{18}O$	0.634521	2539.128	0.8207
$C_{16}H_{36}O_4$	1E-05	0.059767	0.998
H_2O	0.003551	14.20937	1
Total	1	4001.646	5.5407

$$\rho \text{ Campuran} = \frac{1.000}{\sum \text{Fraksi berat}} \times 62$$

$$= \frac{1.000 \text{ gr}}{1.078 \text{ ml}} \times 62 \frac{\text{lbm/cuft}}{\text{gr/ml}}$$

$$= 57.90322 \text{ lbm/cuft}$$

$$\text{Rate massa} = 4001.646 \text{ kg/jam}$$

$$= 4001.646 \times 2.205 \text{ (konversi ke lb/jam)}$$

$$= 8823.629 \text{ lb/jam}$$

$$\text{Rate volumetrik} = \frac{\text{Rate Massa}}{\rho \text{ Campuran}}$$

$$= \frac{8823.629 \text{ lb/jam}}{57.90322 \text{ lb/cuft}}$$

$$= 152.3858 \text{ cuft/jam}$$

DITENTUKAN

$$\text{Waktu tinggal} = 0.3 \text{ jam}$$

$$\text{Tinggi (Hs)} = 1.5 \text{ D}$$

$$\text{Volume bahan} = \text{Rate volumetrik} \times \text{Waktu tinggal (jam)}$$

$$= 152.3858264 \times 0.3$$

$$= 45.71574793 \text{ cuft}$$

Volume tangki direncanakan 80% terisi bahan

$$\text{Volume tangki} = \frac{100}{80} \times 46 \text{ cuft}$$

$$= 57.144685 \text{ cuft}$$

$$\text{Volume silinder} = (\pi/4) \times D^2 \times Hs$$

$$= 3.14 \times D^2 \times 2 \times D$$

$$= 1.1775 \text{ D}^3$$

$$\text{Volume tutup atas} = 0.000049 \text{ D}^3 \text{ (Brownell, hal 95)}$$

$$\text{Volume tutup bawah} = 0.000049 \text{ D}^3 \text{ (Brownell, hal 95)}$$

$$\text{Volume tangki} = V \text{ silinder} + V \text{ tutup atas} + V \text{ tutup bawah}$$

$$57.14468 \text{ cuf} = 1.1775 \text{ D}^3 + 0.00005 \text{ D}^3 + 5E-05 \text{ D}^3$$

$$57.14468 \text{ cuf} = 1.177598 \text{ D}^3$$

$$D = 3.64748 \text{ ft}$$

$$= 1.112037 \text{ m}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
 H_s &= 1.5D = 43.76976 \text{ in} \\
 &= 5.47122 \text{ ft} \\
 &= 65.65464 \text{ in}
 \end{aligned}$$

MENENTUKAN TEBAL SHELL

1 MENENTUKAN TINGGI LIQUIDA DALAM SHELL

DITENTUKAN

$$\begin{aligned}
 \text{Tinggi liquida} &= 1.4023 && \text{tinggi shell} \\
 H_l &= 1.4023 && H_s \\
 H_l &= 1.4023 && \times 5.471220106 \text{ ft} \\
 &= 7.672291955 \text{ ft} \\
 &= 2.339113401 \text{ m}
 \end{aligned}$$

2 MENENTUKAN TEKANAN DESIGN

Apabila didalam bejana terdapat liquida, maka

$$\begin{aligned}
 P_{\text{design}} &= P_o\text{-}p_i + P_{\text{hidrostatik}} \\
 P_{\text{design}} &= 14.7 - 14.7 + P_{\text{Hidrostatik}} \\
 P_{\text{design}} &= P_{\text{hidrostatik}} \\
 P_{\text{design}} &= \rho \times (\text{g/gc}) \times h_{\text{liquid}} \\
 &= 57.90322 \frac{\text{lb}_m}{\text{cuft}} \times 1 \frac{\text{lb}_f}{\text{lb}_m} \times 7.672 \text{ ft} \\
 &= 444.2504 \frac{\text{lb}_f}{\text{ft}^2} \\
 &= 3.085072 \text{ psi}
 \end{aligned}$$

P_{design} diambil 20% lebih besar untuk faktor keamanan

$$\begin{aligned}
 P_{\text{design}} &= 1.2 \times (P_{\text{operasi}} + P_{\text{hidrostatik}}) \\
 &= 21.34208647 \text{ psi}
 \end{aligned}$$

Dipergunakan bahan konstruksi dari carbon steel dengan spesifik SA 283 Grade C (Brownell T. 13-1)

$$\begin{aligned}
 f_{\text{allowable}} &= 12650 && \text{carbon stell SA-283 grade C} \\
 c &= 0 && \text{in}
 \end{aligned}$$

Sambungan las dengan tipe double welded butt joint

$$\begin{aligned}
 \text{efisiensi las (e)} &= 0.8 \\
 r &= 0.5 \times 43.77 \text{ in} \\
 &= 21.88 \text{ in}
 \end{aligned}$$

Rumus tebal shell yang digunakan

$$\begin{aligned}
 t_{\text{min}} &= \frac{P \times D}{2(f \times e) - (0.6 \times P)} \\
 &= \frac{21.34 \times 43.77}{10120 - 1.851} \\
 &= 0.092 \text{ in}
 \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

digunakan ts = 3/16 in

3 MENENTUKAN TEBAL TUTUP ATAS ELIPTICAL

Tutup alas berbentuk standart dished head

$$\begin{aligned} \text{OD} &= \text{ID} + 2ts \\ &= 43.76976085 + 0.375 \\ &= 44.14476085 \text{ in} \\ \text{rc} &= 22.07238043 \text{ in} \\ &= 1.839365035 \text{ ft} \end{aligned}$$

$$\begin{aligned} \text{tinggi tutup (h)} &= \text{rc} - (\text{rc}^2 - (\text{D}^2/4)^{0.5} \\ &= 1.600124536 \text{ ft} \end{aligned}$$

$$\begin{aligned} \text{Vol dished} &= 1,1 \times h^2 \times (3R-h) \\ &= 1.1 \times 2.5603985 \times (5.52 - 1.60) \\ &= 11.0347227 \text{ cuft} \end{aligned}$$

Bentuk = Flanged and standart dished head

Tebal standart eliptical dished (atas)

$$t = \frac{P \times \text{Di}}{2 \cdot f \cdot e} + C$$

$$t = \frac{21.34 \times 43.77}{20240 - 4.268} + 0.1$$

$$= 0.171 \text{ in} \quad (\text{Brownell, pers 13.10 hal 25})$$

digunakan ts tebal head = 3/16 in

Asumsi = Tebal tutup atas = Tebal tutup bawah = 3/1 in

Sistem Pengaduk

Jumlah baffle = 4 buah

Jumlah impeller (pengaduk) antara 4-16 tetapi umumnya 6 atau 8
(Mc Cabe 5 ed pg 243)

Dipilih pengaduk type flat blade turbine dengan jumlah blade 6

1 PENENTUAN DIMENSI PENGADUK

$$\begin{aligned} \text{Tinggi bahan total, HL} &= 7.6723 \text{ ft} \\ &= 92.0675 \text{ in} \end{aligned}$$

$$\text{Diameter dalam tangki} = 3.6475 \text{ ft}$$

$$\text{Dt} = 43.7698 \text{ in}$$

Ukuran pengaduk diambil dari Mc Cabe Ed 5th hal 243

$$\frac{\text{Da}}{\text{Dt}} = \frac{1}{3} = \frac{\text{E}}{\text{Da}} = 1$$

$$\frac{\text{L}}{\text{Da}} = \frac{1}{4} \quad \frac{\text{J}}{\text{Dt}} = \frac{1}{12}$$

$$\frac{\text{W}}{\text{Da}} = \frac{1}{5}$$

Keterangan



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

Da = Diameter Impeller (pengaduk)
 Dt = Diameter Tangki
 L = Panjang Blade
 W = Lebar Blade
 E = Jarak Impeller dari dasar tangki
 J = Lebar Blade

$$\begin{aligned} \text{Diameter impeler (Da)} &= 1/3 Dt = 0.333333333 \times 4 \\ &= 1.21582669 \text{ ft} \\ \text{Lebar blade (W)} &= 1/5 Da = 0.2 \times 1 \\ &= 0.243165338 \text{ ft} \\ \text{Panjang blade (L)} &= 1/4 Da = 0.25 \times 1 \\ &= 0.303956673 \text{ ft} \\ \text{jarak impeller dari dasar} &= 1/3 Dt = 0.333333333 \times 4 \\ &= 1.21582669 \text{ ft} \\ \text{Lebar baffle (J)} &= 1/12 Dt = 0.083333333 \times 4 \\ &= 0.303956673 \text{ ft} \\ \text{Tebal pengaduk} &= \frac{1}{10} \times 0.303956673 \\ &= 0.030395667 \text{ ft} \end{aligned}$$

2 PENENTUAN JUMLAH PENGADUK

$$\begin{aligned} \text{Tinggi bahan total, HL} &= 7.672292 \text{ ft} \\ \text{Diameter dalam tangki, D} &= 3.6474801 \text{ ft} \\ \text{sg} &= \frac{\rho \text{ Bahan}}{\rho \text{ reference}} \\ &= \frac{57.903216 \text{ lbm/cuft}}{62.43 \text{ lbm/cuft}} \\ &= 0.9274902 \\ \text{Jumlah impeler} &= \frac{\text{tinggi larutan}}{\text{diameter bejana}} \times \text{sg} \\ &= \frac{7.672292}{3.6474801} \times 1 \\ &= 1.9509293 \\ \text{Jadi jumlah impeller sebanyak} &= 2 \text{ buah} \end{aligned}$$

3 PENENTUAN POWER MOTOR

$$\begin{aligned} \text{sg reference} &= 1 && \text{(Kern T.6 hal.808)} \\ \mu \text{ reference} &= 0.95 \text{ cps} && \text{(Kern Fig.14 hal 822-823)} \\ \mu \text{ bahan} &= \frac{\text{sg bahan}}{\text{sg reference}} \times \mu \text{ reference} \\ &= \frac{0.92749}{1} \times 0.95 \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
 &= 0.881115729 \text{ cps} \\
 &= 0.000592082 \text{ lb/ft.s} \\
 \rho \text{ Campuran} &= 57.90321576 \text{ lb/cuft} \\
 \text{Dari Joshi hal 415 didapat, kecepatan putaran pengadukan jenis turbin} \\
 &\text{antara 200-250 m/min} \\
 \text{Ditetapkan kecepatan pengaduk (N)} &= 190 \text{ rpm} \\
 &= 3.167 \text{ rps} \\
 \text{Putaran pengaduk (V)} &= \pi \times N \times Da \\
 &= 3.14 \times 190 \times 0.37 \\
 &= 221.1470132 \text{ m/min (memenuhi)} \\
 \text{Bilangan Reynold} &= \frac{\rho \times Da^2 \times N}{\mu} \\
 &= \frac{57.9 \times 1.478 \times 3.17}{0.000592082} \\
 &= ##### \text{ (Aliran Turbulen)} \\
 \text{Perhitungan power pengaduk yang dibutuhkan :} \\
 \text{Diperoleh nilai Nre} > 10000, \text{ sehingga } N_p = KT \\
 KT = N_p &= 5.75 \quad (\text{Ludwig vol-1 T.5-1,hal301}) \\
 P &= \frac{KT \times N^3 \times Da^5 \times \rho}{gc} \quad (\text{McCabe \%ed,tabel 9.2,hal254}) \\
 &\quad \quad \quad \quad \quad \quad \quad \quad \quad (\text{McCabe \%ed,tabel 9.24,hal253}) \\
 P &= \frac{5.75 \times 31.75^3 \times 2.6568 \times 57.9}{32.2} \\
 &= 872.3 \text{ ft.lbf/s} \\
 &= \frac{872.3}{550} \\
 &= 1.586 \text{ hp} \\
 \text{Power Losess pada} &= 0.1 \times 1.586051861 \\
 \text{gland 10\% hp} &= 0.158605186 \text{ hp} \\
 \text{Power input dengan} &= 1.586051861 + 0.158605186 \\
 \text{gland losess} &= 1.744657047 \text{ hp} \\
 \text{Transmission sistem} &= 0.2 \times 1.744657047 \\
 \text{losses 20\%} &= 0.348931409 \text{ hp} \\
 \text{Power total} &= 1.744657047 + 0.348931409 \\
 &= 2.093588456 \text{ hp} \\
 \text{Efisiensi motor} &= 0.8 \\
 \text{Sehingga power moto} &= \frac{2.093588456}{0.8} \\
 &= 2.61698557 \text{ hp}
 \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

4 SISTEM PENDINGIN

Perhitungan sistem penjaga suhu : (Kern Hal .719)

Dari neraca panas, suhu yang dijaga : 165.000 C

$$Q = 232012.558 \text{ kal/jam}$$

$$= 920.685 \text{ BTU/jam}$$

Suhu Feed masuk = 165 C = 329 F

Suhu Bahan Keluar = 165 C = 329 F

Suhu Air pendingin masuk = 30 C = 86 F

Suhu air pendingin keluar = 45 C = 113 F

$\Delta T_1 = 216 \text{ F}$

$\Delta T_2 = 243 \text{ F}$

Kebutuhan media = 15467.50 kg/jam

= 34105.83 lb/jam

Densitas media = 62.430 lb/cuft

Rate Volumetrik = $\frac{34105.83}{62.430}$ lb/jam

= 546.305 cuft/jam

= 0.152 cuft/detik

Koefisien perpindahan panas bagian luar jaket :

$$h_i = 0,36 (k/Di) \frac{[L^2 N \rho]^{2/3}}{\mu} \frac{[C \cdot \mu]^{1/3}}{k} \frac{[\mu]^{0.14}}{\mu_w}$$

(Kern pg.71 eq 20-1)

Keterangan :

L = Da (diameter impeller) = 1.216 ft

N = Putaran pengaduk = 190 rpm = 11400 rph

ρ = Berat jenis air = 62.43 lb/cuft

Pada suhu tangki penampungan = 30 C maka

μ = 0.821 cps = 1.985 lb/ft.jam

k = 0.333 btu/jam/ft²

c = 1 btu/lb.F

$$Re_p = \frac{[L^2 N \rho]^{2/3}}{\mu}$$

$$= \left(\frac{1.478 \times 11400 \times 62.43}{1.985} \right)^{2/3}$$

$$= 6548.426$$

$$\frac{[C \cdot \mu]^{1/3}}{k} = \frac{(1.000 \times 1.985)^{1/3}}{0.333}$$

$$= 3.776551447$$

$$\frac{[\mu]^{0.14}}{\mu_w} = 1 \text{ (Untuk air)}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
 h_i &= 0.36 \times \frac{0.333}{3.647} \times 6548.4 \times 3.7766 \times 1 \\
 &= 812.3164 \text{ Btu/jam} \cdot \text{ft} \cdot \text{F}
 \end{aligned}$$

Untuk air pendingin yang berada dalam jaket mengacu pada diameter dalam bejana hio = 100 Btu/jamft² F

Menghitung Uc :

$$\begin{aligned}
 U_c &= \frac{h_i \times h_{io}}{h_i + h_{io}} \\
 &= 89.0389 \text{ Btu /jamft}^2 \text{ F} \\
 R_d &= 0.001 \quad (\text{Kern tabel 12 pg 845}) \\
 h_d &= \frac{1}{R_d} \\
 &= 1000
 \end{aligned}$$

Menghitung Ud :

$$\begin{aligned}
 U_d &= \frac{U_c \times h_d}{U_c + h_d} \\
 &= 81.75914643 \text{ Btu /jamft}^2 \text{ F} \\
 A &= \pi \times D_i \times H \text{ Pengaduk} + \pi/4 \times D_i^2 \\
 &= 3.14 \times 3.65 \times 5.47 + 10.44 \\
 &= 73.1061 \text{ ft}^2
 \end{aligned}$$

Penghitungan Tinggi Jaket :

$$\begin{aligned}
 \text{Tinggi jaket} &= \text{Tinggi shell} \\
 h &= 5.4712 \text{ ft}
 \end{aligned}$$

Asumsi :

$$\begin{aligned}
 \text{Tebal air pendingin (ts)} &= 2 \text{ in} \\
 \text{Tebal jaket (tj)} &= 3/16 \text{ in} \\
 e &= 0.8 \\
 C &= 0.13
 \end{aligned}$$

Dipergunakan bahan konstruksi yang terbuat dari carbon steel SA-283 Grade C

$$\begin{aligned}
 f \text{ allowable} &= 12650 \\
 D_o \text{ (Shell)} &= D_i + 2ts \\
 &= 43.7698 + 4 \\
 &= 47.7698 \text{ in} \\
 D_i \text{ (jaket)} &= D_o \text{ (Shell)} + 2tj \\
 &= 47.7698 + 0.375 \\
 &= 48.1448 \text{ in} \\
 D_o \text{ (Jaket)} &= D_i \text{ (Jaket)} + 2tj \\
 &= 48.1448 + 0.375 \\
 &= 48.5198 \text{ in} \\
 P \text{ desain jaket} &= P_o - P_i + P_h
 \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Diocetyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
&= 14.7 - 14.7 + x \text{ g/gc} \times \text{hli} \\
&= 62.43 \frac{\text{lbm}}{\text{cuft}} \times 1 \frac{\text{lbm}}{\text{lbm}} \times 8 \text{ ft} \\
&= 478.9812 \text{ lbf/ft}^2 \\
&= 3.326258 \text{ psi}
\end{aligned}$$

Penentuan tebal jaket :

Tebal jaket berdasarkan ASME Code :

$$t_j = \frac{P \times D_{ij} + C}{2 \cdot f \cdot e - P}$$

$$2 = \frac{3.326 \times 48.14}{1.6 f - 3.326} + 0.1$$

$$f = 55.45954726$$

$$f_{\text{allowable}} > f_{\text{desain}}$$

$$12650 > 55.45954726$$

Dipilih tebal jaket = 3/16 in

SPEKIFIKASI REAKTOR

Reaktor

Dimensi Shell :

Diameter shell inside	=	3.647480071	ft
Tinggi Shell	=	5.471220106	ft
Tebal Shell	=	0.1875	in
Tebal tutup atas (Eliptical dished)	=	3/16	in
Tebal tutup bawah (eliptical dished)	=	3/16	in

Sistem Pengaduk :

Dipakai impeller jenis turbin dengan 6 buah flat blade dengan 2 impeller .

Diameter impeller	=	1.21582669	ft
Panjang Blade	=	0.303956673	ft
Lebar blade	=	0.243165338	ft
Power Motor	=	3.61698557	hp
Bahan Konstruksi	=	Carbon Steel SA-283	Grade C
Jumlah Tangki	=	1	buah

Sistem Pendingin :

Diameter jaket	=	4.012063404	ft
Tinggi Jaket	=	5.4712	ft
Jaket Spacing	=	4	in
Tebal Jaket	=	3/16	in

23 POMPA REAKTOR (L-211)

Fungsi = Mengalirkan *diocetyl phthalate* dari reaktor menuju *cooler*

Tipe = Centrifugal Pump

(Sesuai untuk viskositas <10 cP dan bahan liquid)



PERANCANGAN PABRIK

Pabrik Diocetyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

PERHITUNGAN

$$\begin{aligned}
\text{Bahan Masuk} &= 4001.646 \text{ kg/jam} \\
&= 8823.629 \text{ lb/jam} \\
\rho \text{ Campuran} &= 53.2404 \text{ lb/cuft} \\
\text{Rate massa} &= 8823.629 \text{ lb/jam} \\
\text{Rate volumetrik} &= 152.3858 \text{ cuft/jam} \\
&= 2.539764 \text{ cuft/menit} \\
&= 0.042329 \text{ cuft/detik} \\
&= 18.99997 \text{ gpm}
\end{aligned}$$

Asumsi aliran turbulen

(Di) optimum untuk aliran turbulen, $NRe < 2100$ digunakan persamaan :

$$\text{Diameter optir } 3.9 \times q_f^{0.45} \times \rho^{0.13} \quad (\text{Peters, 4}^{\text{ed}}, \text{ pers.15 : 496})$$

Keterangan =

in

q_f = Fluid flow rate; (cuft/detik)

ρ = Fluid Density; (lb/cuft)

Diameter pipa optimum, Di

$$\begin{aligned}
&= 3.9 \times 0.042329 \times 53.2404 \\
&= 1.576 \text{ in}
\end{aligned}$$

Dari App. A.5-1 (Geankoplis, 1993:892) dipilih spesifikasi :

$$\text{OD} = 0.675 \text{ in}$$

$$= 0.056 \text{ ft}$$

$$\text{ID} = 0.493 \text{ in}$$

$$= 0.041 \text{ ft}$$

$$A = \left(\frac{1}{4} \times \pi \times \text{ID}^2\right)$$

$$= 0.25 \times 3.14 \times 0.041^2$$

$$= 0.001325 \text{ ft}^2$$

$$\text{Kecepatan linear, } v = \frac{q_f}{A}$$

$$= \frac{0.0423294 \text{ cuft/detik}}{0.001325 \text{ ft}^2}$$

$$= 31.947808 \text{ ft/dtk}$$

$$\rho \text{ reference} = 62.43 \text{ lb/cuft}$$

$$\text{sg reference} = 1$$

$$\mu \text{ reference} = 0.95$$

$$\text{sg bahan} = \frac{\rho \text{ bahan}}{\rho \text{ reference}} \times \text{sg reference}$$

$$= \frac{53.2404}{62.43} \times 1$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$= 0.8528015$$

μ berdasarkan sg bahan

$$\mu = \frac{\text{sg bahan}}{\text{sg reference}} \times \mu \text{ reference}$$

$$= \frac{0.8528015}{1} \times 1$$

$$= 0.8101615 \text{ cps}$$

$$= 0.0005444 \text{ lbf/s}$$

$$\text{Nre} = \frac{D \ v \ \rho}{\mu}$$

$$= \frac{0.0410833 \times 31.95 \times 53.24}{0.000544403}$$

$$= 128359.27 > 2100 \text{ (Asumsi aliran turbulen benar)}$$

Dipilih pipa commercial steel,

$$\varepsilon = 0.000048 \text{ m}$$

$$\varepsilon/D = 0.003832211$$

$$f = 0.004$$

Digunakan persamaan Bernoulli :

$$-Wf = \frac{\Delta P}{\rho} + \Delta Z \frac{g}{gc} + \frac{\Delta V^2}{2 \alpha} + \Sigma F$$

Perhitungan friksi berdasarkan **Geankoplis 3ed T 2.10-1, hal 93**

$$\text{Taksiran panjang pipa lurus} = 6.6 \text{ ft} = 2 \text{ m}$$

$$\begin{aligned} - 3 \text{ elbow } 90^\circ &= 3 \times 35 \times 0.041 \\ &= 4.3 \text{ ft} \end{aligned}$$

$$\begin{aligned} - 1 \text{ gate valve} &= 1 \times 9 \times 0.041 \\ &= 0.4 \text{ ft} \end{aligned}$$

$$\text{Panjang total pipa (Le)} = 11 \text{ ft}$$

Friksi yang terjadi:

1. Friksi karena gesekan bahan dalam pipa

$$\Sigma F = \frac{2f \times v^2 \times Le}{gc \times D} \quad (\text{Geankoplis 3ed, eq 2.10-6})$$

$$= \frac{4 \times 0.004 \times 1020.7 \times 11.28}{32.27 \times 0.0411}$$

$$= 138.9849774 \text{ lbf/lbm}$$

$$1 \text{ atm} = 14.6959 \text{ psi} = 2116.2096 \text{ lbf/ft}^2$$

$$P_1 = 1 \text{ atm} + P \text{ hidrostatik}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
 &= 2116 \text{ lbf/ft}^2 \cdot 53.24 \times 1 \times 7.672 \\
 &= 2116 + 408.5 \\
 &= 2525 \text{ lbf/ft}^2
 \end{aligned}$$

$$P_2 = 1 \text{ atr} = 2116.2 \text{ lbf/ft}^2$$

$$\begin{aligned}
 \Delta P &= 2525 - 2116 \\
 &= 408.476 \text{ lbf/ft}^2
 \end{aligned}$$

$$\begin{aligned}
 \frac{\Delta P}{\rho} &= \frac{408.476 \text{ lbf/ft}^2}{53.240 \text{ lb/cuft}} \\
 &= \frac{7.672 \text{ ft.lbf}}{\text{lbm}}
 \end{aligned}$$

$$\begin{aligned}
 \frac{\Delta v^2}{2 \times \alpha \times gc} &= \frac{31.95^2}{2 \times 1 \times 32.217} \\
 &= 15.84 \text{ ft lbf/lbm}
 \end{aligned}$$

Asumsi :

$$Z_1 = 3.281 \text{ ft}$$

$$Z_2 = 0.656 \text{ ft}$$

$$\Delta Z = (Z_2 - Z_1) = 2.625 \text{ ft}$$

$$\begin{aligned}
 \frac{\Delta Z}{gc} &= 2.625 \text{ ft} \times 1 \frac{\text{lbf}}{\text{lbm}} \\
 &= 2.625 \frac{\text{ft.lbf}}{\text{lbm}}
 \end{aligned}$$

Persamaan Bernoulli

$$\begin{aligned}
 -W_f &= \frac{\Delta P}{\rho} + \frac{\Delta Z}{gc} + \frac{\Delta v^2}{2 \alpha gc} + \Sigma F \\
 &= 7.672 + 2.6247 + 15.84 + 138.985 \\
 &= 165.1224 \frac{\text{ft.lbf}}{\text{lbm}}
 \end{aligned}$$

$$\begin{aligned}
 H_p &= \frac{-W_f \times \text{flowrate(gpm)} \times sg}{3960} \\
 &= \frac{165.1224 \times 19 \times 0.853}{3960} \\
 &= 0.675634
 \end{aligned}$$

$$\text{Efisiensi Pompa} = 0.400 \quad (\text{Peters 4}^{\text{ed}}; \text{Figure 14 - 37})$$

$$\begin{aligned}
 Bhp &= \frac{H_p}{\eta \text{ pompa}} \\
 &= \frac{0.676}{0.400} \\
 &= 1.689 \text{ Hp}
 \end{aligned}$$

$$\text{Efisiensi motor} = 0.800 \quad (\text{Peters 4}^{\text{ed}}; \text{Figure 14 - 38})$$

$$\text{Power motor} = \frac{Bhp}{\eta \text{ motor}}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$= \frac{1.689}{0.800}$$

$$= 2.111 \text{ Hp}$$

SPEKIFIKASI POMPA REAKTOR

Fungsi	= Mengalirkan <i>dioctyl phthalate</i> dari reaktor menuju <i>cooler</i> .
Type	= Centrifugal Pump (Sesuai untuk viskositas <10 cP dan bahan liquid)
Power Motor	= 2.111357179 Hp
Rate volumetrik	= 18.99997279 gpm
Total dynamic head	= 165.122 ft.lbf/lbm
Efisiensi pompa	= 0.400
Efisiensi motor	= 0.800
Bahan konstruksi	= Commercial Steel
Jumlah	= 1 buah

24 COOLER REAKTOR (E-212)

Fungsi : Mendinginkan produk reaktor sebelum masuk *neutraliser*

Komponen	kg/jam	Fraksi Berat	ρ (gr/cm ³)	fraksi/ ρ
C ₈ H ₄ O ₃	2.891	0.0007	1.200	0.0006
C ₈ H ₆ O ₄	2.899	0.0007	1.522	0.0005
C ₈ H ₁₈ O	5.078	0.0013	0.821	0.0015
C ₁₆ H ₃₆ O ₄ T	0.060	0.0000	0.998	0.0000
C ₂₄ H ₃₈ O ₄	3801.074	0.9499	0.984	0.9653
H ₂ O	189.644	0.047391392	1.000	0.0474
Total	4001.646	1.000	6.525	1.015

$$\rho \text{ bahan} = 0.985 \times 62.43 \frac{\text{lb/ft}^3}{\text{gr/ml}}$$

$$= 61.49 \text{ lb/ft}^3$$

$$\text{Rate bahan} = 4001.65 \text{ lb/jam}$$

$$\text{Rate volumetrik} = \frac{\text{Rate massa}}{\rho \text{ bahan}}$$

$$= 65.08 \text{ ft}^3/\text{jam}$$

$$T \text{ bahan masuk} = 165.00 \text{ }^\circ\text{C} ; 329 \text{ }^\circ\text{F (t1)}$$

$$T \text{ bahan keluar} = 95 \text{ }^\circ\text{C} ; 203 \text{ }^\circ\text{F (t2)}$$

$$T \text{ air pendingin masuk} = 30.00 \text{ }^\circ\text{C} ; 86 \text{ }^\circ\text{F (T1)}$$

$$T \text{ air pendingin keluar} = 45.00 \text{ }^\circ\text{C} ; 113 \text{ }^\circ\text{F (T2)}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned} Q_{\text{loss}} &= 232012.5580 \text{ kkal/jam} \\ &= 918769.730 \text{ BTU/jam} \\ n_{\text{CW}} &= 15467.50 \text{ kg/jam} \\ &= 34028.49 \text{ lb/jam} \end{aligned}$$

Log Mean Temperature Difference (ΔT_{LMTD})

$$\Delta T_{\text{LMTD}} = \frac{\Delta t_2 - \Delta t_1}{\ln \frac{\Delta t_2}{\Delta t_1}}$$

$$\begin{aligned} \Delta t_1 &= T_1 - t_2 \\ &= 117 \text{ }^\circ\text{F} \end{aligned}$$

$$\begin{aligned} \Delta t_2 &= T_2 - t_1 \\ &= 216 \text{ }^\circ\text{F} \end{aligned}$$

$$\Delta T_{\text{LMTD}} = 161.5 \text{ }^\circ\text{F}$$

Tc dan tc

$$\begin{aligned} t_c &= T_{\text{bahan}} \\ &= 266 \text{ }^\circ\text{F} \end{aligned}$$

$$\begin{aligned} T_c &= T_{\text{stean}} \\ &= 99.5 \text{ }^\circ\text{F} \end{aligned}$$

Trial :

$$\begin{aligned} \text{UD} &= 75 - 150 \\ &= 75 \end{aligned}$$

Type = Double Pipe Heat Exchanger (DPHE)

$$\text{Length} = 20 \text{ ft}$$

$$\text{OD} = 1.00 \text{ in}$$

$$\text{BWG} = 14$$

$$\text{Passes} = 2$$

$$a't = 0.546 \text{ in}^2$$

$$a'' = 0.262 \text{ ft}^2 / \text{ft}$$

$$\begin{aligned} A &= \frac{Q}{\Delta T_{\text{LMTD}} \times \text{UD}} \\ &= 75.87 \text{ ft}^2 \end{aligned}$$

$$\begin{aligned} N_t &= \frac{A}{L \times a''} \\ &= 15.00 \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

Berdasarkan Tabel 9 kern dipilih ukuran yang paling mendekati :

$$\begin{aligned}
 \text{OD tube} &= 1 \text{ in} \\
 \text{Pitch} &= 1 \frac{1}{4} \text{ in} \\
 \text{Passes} &= 2 \quad \text{L/D} = 315.8 \\
 \text{Nt} &= 220 \\
 a'' &= 0.262 \text{ ft}^2 / \text{ft} \\
 \text{Wall thickr} &= 0.083 \text{ in} \\
 \text{ID tube} &= 0.76 \text{ in} ; \quad 0.0633 \text{ ft} \\
 \text{ID shell} &= 8 \text{ in} \\
 \text{B space} &= 4.8 \text{ in} \\
 \text{Passes} &= 1 \\
 A &= \text{Nt} \times L \times a'' \\
 &= 1152.8 \text{ ft}^2 \\
 \text{UD} &= \frac{Q}{\Delta T \text{ LMTD} \times A} \\
 &= 4.935737
 \end{aligned}$$

Cold Fluid	Hot Fluid
o) Flow area, as $as = \text{ID} \times C' \times B / 144 \text{ Pr}$ $B = \text{ID}$ 5 $= 2 \text{ in}$ $C' = \frac{\text{Pt} - \text{OD tube}}{2}$ $= 1/8$ $as = 0 \text{ ft}^2$	o) Flow Area, at $at = \frac{\text{Nt} \times at}{144 n}$ $= 0.4171 \text{ ft}^2$
o) Mass Velocity, Gs $Gs = \frac{w}{as}$ $= 3,828,206 \text{ lb/jam.ft}^2$	o) Mass Velocity, Gt $Gt = \frac{w}{at}$ $= 9,594 \text{ lb/jam.ft}^2$
o) Bilangan Reynold, Ret $\text{Nre} = \frac{Ds \times Gt}{\mu}$ $t_c = 99.5 \text{ }^\circ\text{F}$ $\mu = 0.013 \text{ cP}$ $= 0.031 \text{ lb/jam.ft}$ $Ds = 0.99 \text{ in}$ $= 0.083 \text{ ft}$ $\text{Nre} = 10,039,001$	o) Bilangan Reynold, Ret $\text{Nre} = \frac{D \times Gt}{\mu}$ $\mu = 0.09 \text{ cP}$ $= 0.217 \text{ lb/jam.ft}$ $\text{Nre} = 2,801$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Cold Fluid	Hot Fluid
o) $jH = 1000$ o) $k = k_{water}$ $= 0.3600$ $c = 0.45$ $(c\mu/k)^1 = 0.34$ o) $ho = 1500 \text{ BTU/hr.ft}^2 \times \phi_s$ $ho = 1500 \text{ BTU/hr.ft}^2 \cdot \phi_s$ ϕ_s o) $tw = tc + \frac{ho/\phi_t (Tc-tc)}{ho/\phi_t + ho/\phi_s}$ $= #####$ $\phi_s = \frac{(\mu)^{1/4}}{(\mu_w)^{1/4}}$ $\phi_s = 0.891$ o) $ho = 1684$ o) $ho = 1684.202$	o) $jH = 600$ o) $(c\mu/k)^{1/3}$ $tc = 266$ $c = 0.36$ $k = 0.45$ $(c\mu/k)^1 = 0.6$ o) $hi = jH .k/D . (c\mu/k)^{1/3} \times \phi_t$ $\frac{hi}{\phi_t} = 198.152711$ $\phi_t = \frac{(\mu)^{1/4}}{(\mu_w)^{1/4}}$ $(\mu_w)^{1/4} = 0.242$ $\phi_t = 0.973$ o) $hio = 204$

Menghitung UC

$$Uc = \frac{hio \times ho}{hio + ho}$$

$$= 181.68 \text{ BTU/ft}^2 \text{ } ^\circ\text{F jam}$$

Menghitung Rd

$$Rd = \frac{Uc \times U_D}{Uc + U_D}$$

$$U_D = 4.936$$

$$Rd = 0.20 > 0.002 \text{ (memenuhi)}$$

Pressure Drop	
o) $Nre = 10,039,001$ $f = 1E-07$ $v = 37.93145 \text{ ft}^3/\text{lb}$ $\rho = 0.0264 \text{ lb/ft}^3$	o) $Nre = 2,801$ $f = 0.0005$ $s = 1$



PERANCANGAN PABRIK

Pabrik Diocetyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$s = 0.000422$ o) $N+1 = 12 \text{ L/B}$ $= 50$ o) $d_e = 0.9904 \text{ in}$ $= 0.0825 \text{ ft}$ o) $\Delta P_s = f \cdot G_s^2 \cdot D_s \cdot (N+1)$ $5,22 \cdot 10^{10} \cdot D_e \cdot S \cdot \Phi_s$ $= 3.32640 < 10 \text{ psi}$ (memenuhi)	o) $\Delta P_t = f \cdot G_t^2 \cdot L \cdot n$ $5,22 \cdot 10^{10} \cdot D_t \cdot s \cdot \phi_t$ $= 0.00 < 2 \text{ psi}$ (memenuhi)
--	--

SPESIFIKASI COOLER REAKTOR

Nama Alat	: Cooler Reaktor
Fungsi	: Mendinginkan produk reaktor sebelum masuk <i>neutraliser</i> dari suhu 165 °C menjadi 95 °C
Tipe	: <i>Double Pipe Heat Exchanger</i>
Inside Diameter	= 0.76 in
Outside Diameter	= 1 in
Panjang	= 20 ft
Pitch	= 1 in
Jumlah Tube	= 220 buah
Shell	
Inside Diameter	= 8 in
Baffle space	= 4 4/5 in
Jumlah cooler	= 1

25 NEUTRALISER (M-220)

Fungsi	: Menetralkan katalis <i>tetrabutyl titanate</i> menggunakan natrium hidroksida
Tipe	: Tangki berbentuk silinder tegak, tutup atas dan bawah berbentuk <i>torispherical</i>
Jumlah	: 1 buah
Kondisi Operasi	: $T = 95 \text{ °C} = 368.15 \text{ K}$ $P = 1 \text{ atm} = 760 \text{ mmHg} = 14.7 \text{ psi}$

1) Menentukan Kapasitas Tangki



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Komponen	Massa (kg/jam)	%Berat
C ₁₆ H ₃₆ O ₄ Ti	0.0598	0.000015
C ₁₆ H ₃₅ O ₄ TiNa	0.0636	0.000016
C ₂₄ H ₃₈ O ₄	3801.0743	0.949860
H ₂ O	189.6467	0.047391
NaOH	0.0070	0.000002
C ₈ H ₁₈ O	5.0783	0.001269
C ₈ H ₆ O ₄	2.8994	0.000725
C ₈ H ₄ O ₃	2.8907	0.000722
Total	4001.7198	1.0000

Komponen	%Berat	ρ (kg/m ³)	ρ _{campuran} (kg/m ³)	μ (cP)	μ _{campuran} (cP)
C ₁₆ H ₃₆ O ₄ Ti	0.00001	809.2010	1.85E-08	1.4400	1.04E-05
C ₁₆ H ₃₅ O ₄ TiNa	0.00002	834.6840	1.91E-08	0.7720	2.06E-05
C ₂₄ H ₃₈ O ₄	0.94986	919.0110	0.001034	3.3040	0.287488
H ₂ O	0.04739	960.6370	4.93E-05	0.2940	0.161195
NaOH	0.00000	1880.9150	9.34E-10	210.3030	8.36E-09
C ₈ H ₁₈ O	0.00127	774.0010	1.64E-06	0.9910	0.001281
C ₈ H ₆ O ₄	0.00072	1243.5200	5.83E-07	155.9670	4.65E-06
C ₈ H ₄ O ₃	0.00072	1241.1410	5.82E-07	3.9880	0.000181
Total	1.0000	8663.11	0.0011	377.0590	4.50E-01

$$\rho_{\text{campuran}} = \frac{\sum x_i}{\sum x_i / \rho} = \frac{1.0000}{0.0011} = 921.0 \text{ g/m}^3$$

$$= 57.50 \text{ lb/ft}^3$$

$$\mu_{\text{campuran}} = \frac{\sum x_i}{\sum x_i / \mu} = \frac{1.0000}{0.4502} = 2.221 \text{ cP}$$

$$= 0.001 \text{ lb/ft.s}$$

Aliran keluar *neutraliser* dijaga selalu berada pada 4001.7 kg/jam. Sehingga tidak terjadi defisit aliran, karena beroperasi kontinyu (tidak ada waktu tinggal) dianggap pengadukan sempurna 100%. Maka :

$$V_{\text{cairan}} = \frac{m}{\rho} = \frac{4001.7198 \text{ kg/jam}}{921.0277 \text{ kg/m}^3} = 4.3448 \text{ m}^3$$



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$$= 153.4367 \text{ ft}^3$$

Untuk perancangan diasumsikan tangki berisi 80% sehingga volume desain tangki sebagai berikut :

$$\begin{aligned} \text{Volume tangki} &= 4.345 \text{ m}^3 \div 80\% = 5.4311 \text{ m}^3 \\ &= 1434.732 \text{ galon} \end{aligned}$$

2) Menentukan Diameter dan Tinggi *Shell*

Tangki dirancang dengan tutup atas berupa *torispherical*.

Untuk ukuran optimum tangki silinder, harga $H_s/ID = 2$

$$\begin{aligned} \text{Asumsi : } H_s/ID &= 2 \\ H_s &= 2 \text{ ID} \end{aligned}$$

Volume tangki = V silinder + 2 × V tutup

$$\begin{aligned} \text{Volume tangki} &= \left(\frac{1}{4} \times \pi \times ID^2\right) \times H_s + 2 \times (0,000049 ID^3) \\ 5.4311 &= \left(\frac{1}{4} \times (3,14 \times ID^2)\right) \times + 2 \times (0,000049 ID^3) \\ 5.4311 &= 1,1775 ID^3 + 2 \times (0,000049 ID^3) \\ ID^3 &= 3.459 \\ ID &= 1.512 \text{ m} = 4.962 \text{ ft} = 59.41 \text{ in} \\ H_s &= 2 \times ID \\ &= 2 \times 1.5124 \text{ m} \\ &= 3.025 \text{ m} = 9.924 \text{ ft} \end{aligned}$$

3) Menentukan Tinggi Cairan pada Tangki

$$\begin{aligned} H_L &= \frac{V_L}{\frac{1}{4} \times \pi \times D^2} = \frac{153.4367}{\frac{1}{4} \times \pi \times (4.962)^2} = 7.939 \text{ ft} \\ &= 2.420 \text{ m} \\ &= 95.05 \text{ in} \end{aligned}$$

4) Menentukan Tekanan Desain

$$P_{\text{operasi}} = 1 \text{ atm} = 14.7 \text{ psi}$$

$$\begin{aligned} P_{\text{hidrostatik}} &= \rho_{\text{Campuran cai}} \times \frac{g}{gc} \times H_s \\ &= 57.4998 \text{ lb/f} \times 1 \times 9.924 \text{ ft} \\ &= 570.6023 \text{ lb/f} = 3.963 \text{ psi} \end{aligned}$$

$$P_{\text{desain}} = P_{\text{operasi}} + P_{\text{hidrostatik}}$$



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$$\begin{aligned} &= 14.6960 + 3.9627 \\ &= 18.6587 \text{ psi} \end{aligned}$$

5) Menentukan Tebal *Shell*

Dengan faktor keamanan 20%, maka :

$$\begin{aligned} P_d &= \text{tekanan desain} = 1.2 \times 18.66 \text{ psi} = 22.39 \text{ Psi} \\ r &= \text{jari-jari shell} = 59.41 : 2 = 29.70 \text{ in} \\ f &= \text{allowable stress} = 12650 \text{ psi} \\ &\quad \text{Carbon Steel SA-283 Grade C} \quad (\text{Brownell \& Young, 1959:25}) \\ E &= \text{efisiensi pengelasan double-welded butt joint} = 80\% \\ c &= \text{faktor korosi} = 0.1250 \end{aligned}$$

$$\begin{aligned} \text{Sehingga : } t_s &= \frac{P_d \cdot r}{f \cdot E - 0.6 P_d} + c \\ &\quad (\text{Pers 13.1 Brownell \& Young, 1959:254}) \\ &= \frac{22.3904 \times 29.7026}{12650 \times 80\% - 1 \times 22.39} + 0.125 \\ &= 0.191 \text{ in} \\ &\approx 1/4 \text{ in} \quad \text{Tabel 5.7. (Brownell \& Young, 1959:89)} \end{aligned}$$

6) Menentukan Diameter dan Tinggi yang Distandarkan

Diameter Standar (D_s)

$$\begin{aligned} \text{OD} &= \text{ID} + 2 t_s \\ &= 59.4051 \text{ in} + 2 \times 1/4 \text{ in} \\ &= 59.9051 \text{ in} \end{aligned}$$

Sehingga : $\text{OD}_{\text{standar}} = 60 \text{ in}$ Tabel 5.7. (Brownell & Young, 1959:91)

$$\begin{aligned} \text{ID} &= \text{OD}_{\text{standar}} - 2 t_s \\ &= 60 \text{ in} - 2 \times 1/4 \text{ in} \\ &= 60 \text{ in} = 1.511 \text{ m} = 4.958 \text{ ft} \end{aligned}$$

Diperoleh OD standar 60 in dengan :

$$\begin{aligned} \text{icr} &= 3 \frac{5}{8} \text{ in} \\ r &= 60 \text{ in} \end{aligned}$$

Tinggi Standar (H_s)

$$\begin{aligned} H_s &= 2 \times 60 \text{ in} \\ &= 120 \text{ in} \\ &= 3.048 \text{ m} \\ &= 393.7 \text{ ft} \end{aligned}$$



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7) Menentukan Dimensi pada Tutup

Digunakan *head* jenis standard *dished/torispherical* dengan bahan konstruk yang sama dengan *shell mixer*

$$\begin{aligned}
 th &= \frac{0.885 P_{d,r}}{f \cdot E - 0.1 P_d} + c \\
 &= \frac{0.8850 \times 22.3904 \times 29.7026}{12650 \times 80\% - 0.1 \times 22.39} + 0 \\
 &= 0.183 \text{ in}
 \end{aligned}$$

Digunakan tebal stand = 3/16 in **Tabel 5.7.** (Brownell & Young, 1959:91)

Tabel 5.6. (Brownell & Young, 1959:88) untuk tebal tutup 3/16 in didapat t sf 1½ -2 in. Diambil sf: 1 1/2 in

Fig. 5.8. (Brownell & Young, 1959:87), dihitung ukuran-ukuran sebagai berikut :

$$\begin{aligned}
 a &= \frac{1}{2} ID = \frac{1}{2} \times 60 = 29.75 \text{ in} \\
 AB &= a - icr = 29.75 - 3 \frac{5}{8} = 26.13 \text{ in} \\
 BC &= r - icr = 60 - 3 \frac{5}{8} = 56.38 \text{ in} \\
 AC &= \sqrt{(BC)^2 - (AB)^2} = 49.96 \text{ in} \\
 b &= r - AC = 60 - 49.96 = 10.04 \text{ in} \\
 OA &= th + b + sf \\
 &= \frac{3}{16} + 10.04 + 1 \frac{1}{2} = 11.73 \text{ in} \\
 &= 0.298 \text{ m}
 \end{aligned}$$

$$\begin{aligned}
 \text{Tinggi tangki total} &= H_s + 2 OA \\
 &= 3.025 + 2 \times 0.298 \\
 &= 3.621 \text{ m} = 142.2 \text{ in}
 \end{aligned}$$

8) Menentukan Jumlah Pengaduk

Dipilih jenis pengaduk *propeler three-blade* dengan alasan :

Karena campuran *liquid* mempunyai viskositas yang rendah dan hasil campuran diharapkan homogen karena pengaduk jenis ini cocok untuk viskositas rendah hingga sedang persamaan berikut :

$$\text{SpGr} = \frac{\rho_{\text{campuran}}}{\rho_{\text{air}}} = \frac{57.50 \text{ lb/f}}{59.97 \text{ lb/ft}^3} = 0.959$$

$$\begin{aligned}
 \text{Jumlah pengaduk} &= \frac{H_L \times \text{SpGr}}{ID} = \frac{2.420 \text{ m} \times 0.959}{1.5113 \text{ m}} \\
 &= 1.535 \\
 &\approx 2 \text{ buah}
 \end{aligned}$$



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9) Menentukan Diameter dan Tinggi Pengaduk

Diameter Pengaduk

$$ID/d = 3 \quad (\text{Brown, 1950:507})$$

$$\begin{aligned} Da &= \frac{ID}{ID/d} = \frac{1.511}{3} = 0.5038 \text{ m} \\ &= 19.83 \text{ in} \\ &= 1.65 \text{ ft} \end{aligned}$$

Jarak Pngaduk dari Dasar Tangki

$$Zi/c = 0,75 - 1,3 \quad ; c(\text{Brown, 1950:507})$$

$$\begin{aligned} Zi &= 0,75 \times Da = 0.75 \times 0.504 = 0.378 \text{ m} \\ &= 14.87 \text{ in} \\ &= 1.240 \text{ ft} \end{aligned}$$

10) Menentukan Dimensi Pengaduk

Dari Tabel 3.4-1 (Geankoplis, 1993:144)

$$W/Da = 1/5$$

$$Dd/Da = 2/3$$

$$H/Dt = 1$$

$$L/Da = 1/4$$

$$J/Dt = 1/12$$

$$\begin{aligned} H &= Dt = 60 \text{ in} \\ &= 1.524 \text{ m} \\ &= 5.000 \text{ ft} \end{aligned}$$

$$\begin{aligned} Dd &= 2/3 \times Da = 2/3 \times 19.83 \text{ in} \\ &= 13.22 \text{ in} \\ &= 0.336 \text{ m} \\ &= 1.102 \text{ ft} \end{aligned}$$

$$\begin{aligned} L &= 1/4 \times Da = 1/4 \times 19.83 \text{ in} \\ &= 4.958 \text{ in} \\ &= 0.126 \text{ m} \\ &= 0.413 \text{ ft} \end{aligned}$$

$$\begin{aligned} W &= 1/5 \times Da = 1/5 \times 19.83 \text{ in} \\ &= 3.967 \text{ in} \\ &= 0.101 \text{ m} \\ &= 0.331 \text{ ft} \end{aligned}$$



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$$\begin{aligned}
 J &= 1/12 \times OD = 1/12 \times 60 \text{ in} \\
 &= 5.000 \text{ in} \\
 &= 0.127 \text{ m} \\
 &= 0.417 \text{ ft}
 \end{aligned}$$

11) Menentukan *Power* Motor Pengaduk Kecepatan Putar Pengaduk

$$\begin{aligned}
 WELH &= \frac{\rho_{\text{Campurar}}}{\rho_{\text{air}}} \times H_L = \frac{57.50 \text{ lb/f} \times 7.939 \text{ ft}}{59.97 \text{ lb/ft}^3} \\
 &= 7.612 \text{ ft}
 \end{aligned}$$

$$N = \frac{600}{\pi d} \sqrt{\frac{WELH}{2d}} \quad \text{Eq. 8.8. (Rase \& Holmes, 1977:338)}$$

Keterangan :

Da = diameter pengaduk (ft)

N = kecepatan putar pengaduk (rpm)

WELH = *water equivalent liquid height* (ft)

$$\begin{aligned}
 N &= \frac{600}{\pi d} \sqrt{\frac{WELH}{2d}} = \frac{600}{3,14 \times 1.653} \sqrt{\frac{7.6120}{2 \times 1.653}} \\
 &= 133.1166
 \end{aligned}$$

Diambil kecepatan pengaduk standar 155 rpm, maka (Walas, 1990:288) :

$$\begin{aligned}
 N &= 155 \text{ rpm} \\
 &= \text{#####} \text{ rps} \\
 &= 9300 \text{ rph}
 \end{aligned}$$

Bilangan Reynold

$$\begin{aligned}
 N_{Re} &= \frac{\rho \times Da^2 \times N}{\mu} \quad \text{Pers. 3.4-1. (Geankoplis, 1993:158)} \\
 &= \frac{57.4998 \text{ lb/ft}^3 \times 1.6528^2 \text{ ft} \times 2.5833 \text{ rps}}{0.0015 \text{ lb/ft.s}} \\
 &= 272230.9580
 \end{aligned}$$

Karena $N_{Re} > 10.000$, maka sesuai untuk digunakan *baffle* (Perry, 1997:9-8)

Power Pengaduk

Karena $N_{Re} > 10.000$ maka *power* tidak tergantung pada bilangan Reynold c bukan merupakan fungsi viskositas. Maka untuk pengaduk jenis *propeller* didapat $N_p = KT = 0,87$ **Tabel 9.3.** (Mc Cabe et al., 1993:254). Sehingga :



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$$P = \frac{N_p \cdot \rho \cdot N^3 \cdot D_a^5}{g_c} \quad \text{Pers. 9.20. (Mc Cabe et al., 1993:253)}$$

Keterangan :

P = daya pengaduk (lb.ft/s)

N_p = *power number*

ρ = densitas campuran (lb/ft³)

N = kecepatan putar pengaduk (rps)

D_a = diameter pengaduk (ft)

g_c = percepatan gravitasi = 32.1740 lb.ft/lbf.s²

$$\begin{aligned} P &= \frac{0.87 \times 57.4998 \times 2.5833^3 \times 1.6528^5}{32.1740} \\ &= 330.5899 \text{ lb.ft/s} \\ &= 0.6011 \text{ hp} \end{aligned}$$

Selama proses pengadukan, pengaduk mengalami :

- *Gain Losses* diperhitungkan 10% dari daya masuk (dengan mempertimbangkan adanya kebocoran daya pada proses dan *bearing*) = 10% × P_o

$$\text{Gain losses} = 0.0601 \text{ hp}$$

- *Transmission Losses* diperhitungkan 20% dari daya masuk (dengan meninjau kemungkinan terjadinya kebocoran *belt* atau *gear*) = 20% × P_o

$$\text{Transmission losses} = 0.1202 \text{ hp}$$

$$\begin{aligned} \text{Power input} &= 0.6011 + 0.0601 + 0.1202 \\ &= 0.7814 \text{ hp} \end{aligned}$$

Fig. 14.38. (Peters & Timmerhaus, 1991:521) didapat efisiensi motor pengaduk 80%, maka *power* motor pengaduk sebesar :

$$\text{Power motor} = \frac{P}{\eta} = \frac{0.7814}{80\%} = 0.9767 \text{ hp}$$

Maka digunakan *power* motor pengaduk standar sebesar 1 hp

12) Menentukan Pendingin *Neutraliser*

$$Q \text{ Pendingin} = 255662.4 \text{ kJ/jam}$$

$$= 242324.5 \text{ Btu/jam}$$

$$m \text{ pendingin} = 4070.928 \text{ kg/jam}$$

$$= 3858.548 \text{ lb/jam}$$

$$\text{Suhu bahan masuk } T_1 = 95 \text{ }^\circ\text{C} = 203 \text{ }^\circ\text{F}$$

$$\text{Suhu bahan keluar } T_2 = 95 \text{ }^\circ\text{C} = 203 \text{ }^\circ\text{F}$$

$$\text{Suhu pendingin masuk } t_1 = 30 \text{ }^\circ\text{C} = 86 \text{ }^\circ\text{F}$$

$$\text{Suhu pendingin keluar } t_2 = 45 \text{ }^\circ\text{C} = 113 \text{ }^\circ\text{F}$$



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Fluida panas ($^{\circ}\text{F}$)		Fluida dingin ($^{\circ}\text{F}$)		Selisih ($^{\circ}\text{F}$)	
T_1	203	t_2	113	90	Δt_1
T_2	203	t_1	86	117	Δt_2

$$\begin{aligned} \text{LMTD} &= \frac{\Delta t_2 - \Delta t_1}{\ln\left(\frac{\Delta t_2}{\Delta t_1}\right)} \\ &= \frac{117 - 90}{\ln(117 : 90)} \\ &= 103 \text{ }^{\circ}\text{F} \end{aligned}$$

Suhu rata-rata pemanas :

$$\begin{aligned} t_c &= \frac{(t_1 + t_2)}{2} \\ &= \frac{86 + 113}{2} \\ &= 100 \text{ }^{\circ}\text{F} \\ &= 38 \text{ }^{\circ}\text{C} \end{aligned}$$

$$\begin{aligned} \rho \text{ air} &= 960.6 \text{ kg/m}^3 \\ &= 61.70 \text{ lb/ft}^3 \end{aligned}$$

$$\begin{aligned} \text{Debit a} &= \frac{\text{massa air}}{\rho \text{ air}} \\ &= \frac{3858.55}{61.70} \\ &= 62.54 \text{ ft}^3/\text{jam} \\ &= 1.771 \text{ m}^3/\text{jam} \\ &= 0.000 \text{ m}^3/\text{s} \\ &= 0.017 \text{ ft}^3/\text{s} \end{aligned}$$

Lama waktu tinggal dalam *coil* = 60 menit

$$\begin{aligned} \text{Volum} &= \frac{1.771 \times 60}{60} \\ &= 1.771 \text{ m}^3 \end{aligned}$$

Menghitung Luas Penampang Aliran (A)

Harga kecepatan untuk cairan dalam pipa 1,5 - 2,5 m/s (Coulson, 1989:534)

Dipilih harga kecepatan = 2.5 m/s

$$A = \frac{F_{vp}}{v} = \frac{\pi}{4} (ID)^2$$



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$$\begin{aligned}ID &= \sqrt{\frac{4 \cdot F_{vp}}{\pi \cdot v}} = \\&= 0.0158 \text{ m} \\&= 0.0519 \text{ ft} \\&= 0.6233 \text{ in}\end{aligned}$$

Dari **Tabel 11.** (Kern, 1965:844) dipilih :

$$\begin{aligned}\text{IPS} &= 1 \text{ in} \\&= 0.083 \text{ ft} \\&= 0.025 \text{ m} \\OD &= 1.32 \text{ in} \\&= 0.110 \text{ ft} \\&= 0.034 \text{ m} \\ID &= 1.05 \text{ in} \\&= 0.087 \text{ ft} \\&= 0.027 \text{ m}\end{aligned}$$

$$A^I \left(\frac{\text{flowarea}}{\text{pipe}} \right) = 0.864 \text{ in}^2 = 0.0600 \text{ ft}^2$$

$$A^{II} \left(\frac{\text{surface}}{\text{lin.ft}} \right) = 0.344 \text{ ft}^2/\text{ft}$$

Menghitung *Mass Velocity* (v)

$$\begin{aligned}G_t &= \frac{W_t}{A^I} = \frac{3858.548 \text{ lb/jam}}{0.0600 \text{ ft}^2} \\&= 64309.55 \text{ lb/ft}^2\text{jam}\end{aligned}$$

$$\begin{aligned}v &= \frac{G_t}{\rho_{air}} = \frac{64309.55 \text{ lb/ft}^2\text{jam}}{61.7017 \text{ lbm/ft}^3} \\&= 1042.265 \text{ ft/jam} \\&= 0.2895 \text{ ft/s}\end{aligned}$$

Menghitung h dan hio

$$\begin{aligned}\mu_{air} &= 0.294 \text{ cp} \\&= 0.711 \text{ lb/ft.jam}\end{aligned}$$

Re dalam pipa :

$$\begin{aligned}Re_t &= \frac{ID \cdot G_t}{\mu} = \frac{0.087 \times 64309.55}{0.7112} \\&= 7904.393 \text{ Btu/jam.ft}^2 \text{ } ^\circ\text{F}\end{aligned}$$

$$\begin{aligned}\text{Pada T} &= 100 \text{ } ^\circ\text{F} \\v &= 0.290 \text{ ft/s,}\end{aligned}$$



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maka diperoleh $h_i = 500 \text{ Btu/jam.ft}^2 \text{ } ^\circ\text{F}$ **Fig. 25.** (Kern, 1965:835)

$$h_{io} = h_i \cdot \frac{ID}{OD} = \frac{500 \times 0.0874}{0.1100} = 397.3485$$

Menghitung Koefisien Transfer Panas

$$h_j = \frac{0.74 \cdot k}{Dt} \cdot \left[\frac{L^2 \cdot N \cdot \rho}{\mu} \right]^{\frac{2}{3}} \cdot \left[\frac{cp \cdot \mu}{k} \right]^{\frac{1}{3}} \cdot \left[\frac{\mu}{\mu_w} \right]^{0.14}$$

Dimana :

h_c = Koefisien transfer panas *coil*, Btu/jam.ft².°F

D_t = Diameter dalam tangki = 4.958 ft

k = Konduktivitas panas = 3.486 Btu/jam.ft.°F

L = Diameter putaran pengaduk = 1.653 ft

N = Kecepatan putaran pengaduk = 9300.000 rph

μ = Viskositas air = 2.221 cp

= 5.376 lb/jam.ft

ρ = Densitas campuran = 57.4998 lb/ft³

C_p = Kapasitas panas = #REF! Btu/lb.ft

$$Re_j = \frac{N \cdot D_t^2 \cdot \rho}{\mu} = 2445633.87$$

$$\left(\frac{\mu}{\mu_w} \right)^{0.14} = 1$$

$$h_c = h_o = \text{\#REF!} \text{ Btu/jam.ft}^2 \text{ } ^\circ\text{F}$$

Menghitung U_c dan U_d

Eq. 6. 38. (Kern, 1965:231) :

$$U_c = \frac{h_o \cdot h_{io}}{h_o + h_{io}} = \frac{\text{\#REF!} \times 397.3485}{\text{\#REF!} + 397.3485} = \text{\#REF!} \text{ Btu/jam.ft}^2 \text{ } ^\circ\text{F}$$

Dimana: U_c = *clean overall coefficient*

U_d = *dirty overall coefficient*

Dirt Factors (R_d) = #####

$$\frac{1}{U_d} = \frac{1}{U_c} + R_d = \frac{1}{\text{\#REF!}} + 0.0020 = \text{\#REF!} \text{ Btu/ft}^2 \text{ } \cdot \text{hr.} \text{ } ^\circ\text{F}$$



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Menghitung Luas Permukaan Panas

$$\begin{aligned} A &= \frac{Q}{U_D \times \text{LMTD}} \\ &= \frac{242324.5375}{\#REF! \times 102.9} \\ &= \#REF! \text{ ft}^2 \\ &= \#REF! \text{ m}^2 \end{aligned}$$

Menghitung Luas Selubung *Neutraliser*

$$\begin{aligned} A &= \pi D L \\ A &= 3.14 \times 1.511 \text{ m} \times 3.048 \text{ m} \\ &= 14.4642 \text{ m}^2 \end{aligned}$$

Luas transfer panas *neutraliser* lebih kecil dibandingkan dengan selubung *neutraliser* sehingga digunakan *jacket* pendingin.

$$(A_{\text{panas}} = \#REF! \text{ m}^2 < A_{\text{selubung}} = 14.46 \text{ m}^2)$$

A yang diperoleh dari persamaan diatas tidak boleh lebih besar dari luas *bottom* + luas *shell* reaktor yang tercelup cairan

Maka :

$$\begin{aligned} A_{\text{Bott}} &= (3,14/4) \times OD^2 \\ A_{\text{Shell}} &= OD \times 3,14 \times H \end{aligned}$$

dimana :

$$\begin{aligned} OD &= \text{Diameter luar reaktor} = 60 \text{ in} \\ &= 4.9998 \text{ ft} \\ L &= \text{Tinggi cairan di reaktc} = 95.0541 \text{ in} \\ &= 7.9212 \text{ ft} \\ H &= \text{Tinggi total tangki} = 142.2191 \text{ in} \\ &= 11.8516 \text{ ft} \end{aligned}$$

$$\begin{aligned} A_{\text{bottc}} &= 19.6234 \text{ ft}^2 \\ A_{\text{Shell}} &= 186.0625 \text{ ft}^2 \\ A_{\text{total}} &= A_{\text{bottom}} + A_{\text{Shell}} \\ &= 19.6234 + 186.0625 \\ &= 205.6860 \text{ ft}^2 \\ &= 19.1089 \text{ m}^2 \end{aligned}$$

Mencari nilai Ud

Dimana :

$$\begin{aligned} l &= 0.1259 \text{ m} = 0.4069 \text{ ft} \\ N &= 155 \text{ rpm} = 9300 \text{ rpj} \end{aligned}$$



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$$\rho = 921.0277 \text{ kg/m}^3 = 57.50 \text{ lb/ft}^3 \text{ (Yaws, 1999)}$$

$$\mu = 2.2213 \text{ cP} = 5.3312 \text{ lb/ft hr (Yaws, 1999)}$$

Sehingga :

$$\begin{aligned} \text{Re}_j &= \frac{L^2 \times N \times \rho}{\mu} \\ &= 16606.04 \end{aligned}$$

Diketahui :

$$\text{Re}_j = 16606.04$$

$$j = \text{#####} \quad \text{Fig. 24. (Kern, 1965:834)}$$

$$k = \text{#####} \text{ Btu/ft.jam. Tabel 4. (Kern, 1965:800)}$$

$$\text{IDs} = 1.51 \text{ m}$$

$$= 4.88 \text{ ft}$$

$$\begin{aligned} h_i &= j \times (k/\text{ID}) \times (\text{Cp} \times \mu/k)^{1/3} \\ &= 100 \times \left(\frac{0.33280}{4.88271} \right) \times \left(\frac{5.33120}{0.33280} \right)^{1/3} \\ &= 36.3951 \end{aligned}$$

$$h_0 = jH \times (k/\text{De}) \times (\text{Cp} \times \mu/k)^{1/3}$$

Dimana :

Jarak dari *jacket* (*jacket spacing*) berkisar 1 - 4 in, di ambil 1 in

$$\begin{aligned} \text{Jarak dari jacket} &= 1 \text{ in} \\ &= 0.083 \text{ ft} \end{aligned}$$

$$\begin{aligned} D_1 &= \text{Diameter dalam} + (2 \times \text{tebal dinding}) \\ &= 4.883 + (2 \times 1/4) \\ &= 4.924 \text{ ft} \end{aligned}$$

$$\begin{aligned} D_2 &= D_1 + (2 \times \text{jarak jacket}) \\ &= 4.924 + (2 \times 0) \\ &= 5.091 \text{ ft} \end{aligned}$$

$$\begin{aligned} \text{De} &= \frac{D_2^2 - D_1^2}{D_1} \\ &= \frac{5.091^2 - 4.924^2}{4.924} \\ &= 0.339 \text{ ft} \end{aligned}$$

$$h_0 = j \times (k/\text{De}) \times (\text{Cp} \times \mu/k)^{1/3}$$



PERANCANGAN PABRIK

Pabrik Diethyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraethyl Titanate Melalui Proses Esterifikasi

$$= 100 \times \left(\frac{0.333}{0.339} \right) \times \left(\frac{5.3312}{0.3328} \right)^{1/3}$$
$$= 247.5036$$

$$\text{hoi} = \text{ho} \times \frac{\text{ID}}{\text{OD}}$$
$$= 247.5 \times \frac{4.9}{5.1}$$
$$= 239.4013$$

Menghitung Uc

$$\text{Uc} = \frac{\text{hoi} \times \text{hi}}{\text{hoi} + \text{hi}}$$
$$= \frac{239.4 \times 36.40}{239.4 + 36.40}$$
$$= 31.59$$

Menghitung Ud

$$\frac{1}{\text{Ud}} = \frac{1}{\text{Uc}} + \text{Rd}$$
$$\frac{1}{\text{Ud}} = \frac{1}{31.59} + 0.001$$
$$\frac{1}{\text{Ud}} = 0.03$$
$$\text{Ud} = 30.62$$

Menghitung A

$$\text{A} = \frac{\text{Q}}{\text{Ud} \times \Delta T}$$
$$= \frac{242324.54}{30.62 \times \text{\#REF!}}$$
$$= \text{\#REF! ft}^2$$
$$= \text{\#REF! m}^2$$

Massa air pendingin = 4070.9284 kg/jam

Kecepatan volumetris air

$$\text{Fv} = \frac{\text{W}}{\rho \text{ air}}$$
$$\text{Fv} = \frac{4,070.93 \text{ kg/jam}}{960.6370 \text{ kg/m}^3}$$
$$= 4.2377 \text{ m}^3/\text{jam}$$



PERANCANGAN PABRIK

Pabrik Diocetyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
 \text{Diameter dalam } jacket \text{ (Di)} &= h_{ell} + (2 \times \text{jarak } jac) \\
 &= 95 + (2 \times 1) \\
 &= 97.05 \text{ in} \\
 &= ##### \text{ ft} \\
 &= ##### \text{ m}
 \end{aligned}$$

$$\begin{aligned}
 \text{Tinggi Jacket} &= \text{Tinggi total tangki} - \text{Tinggi tutup atas} \\
 &= ##### - 0.2980 \\
 &= 3.3227 \text{ m} \\
 &= 10.9010 \text{ ft} \\
 &= 130.8124 \text{ in}
 \end{aligned}$$

Kecepatan *supercritical cooling* (v_i)

$$\begin{aligned}
 v_i &= \frac{Fv}{A} \\
 &= \frac{4.238 \text{ m}^3/\text{jam}}{##### \text{ m}^2} \\
 &= ##### \text{ m/jam}
 \end{aligned}$$

Bahan *Carbon Steel SA-283 Grade C*

Tabel 13.1. (Brownell & Young, 1959:251)

$$\begin{aligned}
 f &= 12650 \text{ psia} \\
 E &= 0.8 \\
 H_{jacket} &= 130.8124 \text{ in} \\
 &= 10.9006 \text{ ft} \\
 \rho_{\text{air}} &= 61.7017 \text{ lb/ft}^3
 \end{aligned}$$

$$\begin{aligned}
 Ph &= \frac{(H - 1) \times \rho}{144} \\
 &= \frac{[10.9006 - 1] \text{ ft} \times 61.70 \text{ lb/ft}^3}{144} \\
 &= 4.2422 \text{ Psia}
 \end{aligned}$$

$$\begin{aligned}
 P_{\text{desig}} &= P + Ph \\
 &= 14.70 + ##### \\
 &= 18.94 \text{ psia}
 \end{aligned}$$

Maka :

$$\begin{aligned}
 t &= \frac{P \times D}{(f \times E) - (0.6 \times P)} \\
 &= \frac{18.9382 \text{ psia} \times 97.054128 \text{ in}}{2 \times [12650 \times 1] - [1 \times 18.94]}
 \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$= 0.0909 \text{ in}$$

Maka dipilih tebal *jacket* standar $3/16 \text{ in} = ##### \text{ m}$

Tabel 5.7. (Brownell & Young, 1959:89)

$$\begin{aligned} \text{Diameter luar } jacket &= \text{ID } jacket + (2 \times \text{tebal } jacket) \\ &= 97.05 \text{ in} + [2 \times ##### \text{ in}] \\ &= 97.43 \text{ in} \\ &= 8.1191 \text{ ft} \end{aligned}$$

SPEKIFIKASI NEUTRALISER

Nama Alat	: <i>Neutraliser</i>
Kode	: M-220
Fungsi	: Menetralkan katalis <i>tetrabutyl titanate</i> menggunakan natrium hidroksida
Tipe	: Tangki berbentuk silinder tegak, tutup atas dan bawah berbentuk <i>torispherical</i>
Bahan Konstruksi	: <i>Carbon Steel SA-283 Grade C</i>
Jumlah	: 1 buah
Kondisi Operasi	: Temperatur 95 °C dan Tekanan 1 atm
Kapasitas	: 4.345 m ³
Dimensi	
Silinder	
Diameter	: 1.512 m
Tinggi	: 3.025 m
Tebal	: 0.191 in
Tutup	
Tebal	: 0.183 in
Tinggi	: 0.298 in
Tinggi Tangki Total	: 3.621 m
Power Pengaduk	: 1 Hp
<i>Jacket Pendingin</i>	
Tinggi	: 3.323 m
Diameter	: 2.475 m

26 POMPA NEUTRALISER (L-221)

Fungsi	: Mengalirkan <i>dioctyl phthalate</i> dari <i>neutraliser</i> menuju <i>rotary drum vacuum filter</i>
Tipe	: <i>Centrifugal pump</i>
Jumlah	: 1 buah

1. Menentukan Bahan Konstruksi Pompa

Bahan konstruksi yang dipilih adalah *Commercial Steel* karena :



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

- Tahan terhadap korosi
- Memiliki batas tekanan yang diijinkan besar (s.d 22500 psi)
- Memiliki batas suhu yang diizinkan besar (-65°F s.d 650°F)

2. Menghitung Tenaga Pompa

$$\begin{aligned} \text{Rate masuk} &= 4001.653 \text{ kg/jam} = 2.4506 \text{ lb/s} \\ \text{Densitas} &= 789.4158 \text{ kg/m}^3 = 49.2832 \text{ lb/ft}^3 \\ \mu &= 1.018 \text{ cP} = 0.001 \text{ lb/ft.s} \end{aligned}$$

$$\begin{aligned} \text{Rate volumetr} &= \frac{\text{massa}}{r} \\ &= \frac{2.4506 \text{ lb/s}}{49.2832 \text{ lb/ft}^3} \\ &= 0.0497 \text{ ft}^3/\text{s} \\ &= 22.3178 \text{ gpm} \end{aligned}$$

Diperkirakan aliran fluida turbulen ($N_{re} > 2100$), sehingga digunakan persamaan $Di \geq 1$ in yaitu :

$$Di_{opt} = 4 Q^{0.4} \mu^{0.13} \quad \text{Eq. 45. (Peters \& Timmerhaus, 1991:365)}$$

Dimana :

$$\begin{aligned} Di_{opt} &= \text{diameter dalam optimum, in} \\ Q &= \text{kecepatan volumetrik, ft}^3/\text{s} \\ \mu &= \text{viskositas fluida, lb/ft.s} \end{aligned}$$

Sehingga :

$$\begin{aligned} Di_{opt} &= 4 \times (0.050)^{0.45} \times (0.001)^{0.13} \\ &= 0.39179 \text{ in} \end{aligned}$$

Dari App. A.5-1 (Geankoplis, 1993:892) dipilih spesifikasi :

$$\begin{aligned} NP &= 1/2 \text{ in} \\ OD &= 0.840 \text{ in} = 0.070 \text{ ft} = 0.021 \text{ m} \\ ID &= 0.622 \text{ in} = 0.052 \text{ ft} = 0.016 \text{ m} \\ A &= 0.002 \text{ ft}^2 = 1.960 \text{ in}^2 \\ Sch &= 40 \end{aligned}$$

Menghitung kecepatan linier :

Kecepatan linier fluida dapat dicari dengan menggunakan persamaan berikut

$$V = \frac{Q}{A}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

Sehingga :

$$\begin{aligned}
 V &= \frac{0.0497 \text{ ft}^3/\text{s}}{0.0021 \text{ ft}^2} \\
 &= 23.5660 \text{ ft/s} \\
 &= 7.1829 \text{ m/s}
 \end{aligned}$$

Menghitung *Reynold Number* (Nre) :

$$Nre = \frac{\rho v D}{\mu}$$

Sehingga :

$$\begin{aligned}
 Nre &= \frac{49.28 \times 23.57 \times 0.0518}{0.0007} \\
 &= 88042 \text{ (asumsi aliran turbulen benar)}
 \end{aligned}$$

Instalasi pipa

Dari **Fig. 127.** (Brown, 1950:141) dengan NPS = 1/2 in

- 1 buah <i>gate valve fully open</i> ; Le	= 0.26	ft
$\Sigma L = 1 \times 0$	= 0.26	ft
- 3 buah <i>standard elbow</i> ; Le	= 1.4	ft
$\Sigma L = 3 \times 1$	= 4.2	ft
- 1 buah <i>sudden enlargement</i> ; Le	= 1.4	ft
$\Sigma L = 1 \times 1$	= 1.4	ft
- 1 buah <i>sudden contraction</i> ; Le	= 0.52	ft
$\Sigma L = 1 \times 1$	= 0.52	ft
- 1 buah <i>swing check valve</i> ; Le	= 3	ft
$\Sigma L = 1 \times 3$	= 3	ft
- Panjang ekivalen pipa lurus, ΣLe	= 9.38	ft
Panjang pipa lurus	= 2	m
	= 6.562	ft
Panjang pipa total	= 15.94	ft
	= 4.859	m

Menghitung *friction loss*

$$\Sigma F = \frac{4f \cdot v^2 \cdot \Delta L}{2 \cdot ID \cdot gc} \quad \text{Eq. 2.10-6 (Geankoplis, 1993:89)}$$

Dimana :

$$\begin{aligned}
 \Sigma F &= \text{Friction loss} \quad (\text{ft.lbf/lbm}) \\
 f &= \text{Faktor friksi} \\
 v &= \text{Kecepatan linier fluida (ft/s)} \\
 \Delta L &= \text{Panjang pipa (ft)} \\
 ID &= \text{Diameter dalam tangki (ft)} \\
 gc &= 32.174 \text{ lbf.ft/lbf.s}^2
 \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

Menghitung *Fanning Friction Factor* (f) :

Dari **Fig. 2.10-3.** (Geankoplis, 1993:88) didapat :

$$\begin{aligned}\text{Untuk } commercial \rightarrow \varepsilon &= 5.E-05 \text{ m} \\ &= 0.0002 \text{ ft}\end{aligned}$$

Sehingga :

$$\begin{aligned}\frac{\varepsilon}{D} &= \frac{0.000}{0.052} \\ &= 0.003\end{aligned}$$

Dari **Fig. 2.10-3.** (Geankoplis, 1993:88) dengan nilai $N_r = 88042$ didapatkan = 0.005 sehingga :

$$\begin{aligned}\Sigma F &= \frac{4 \times 0.005 \times (23.57)^2 \times 15.94}{2 \times 0.051833 \times 32.17400000} \\ &= 53.09 \frac{\text{lb}_f \cdot \text{ft}}{\text{lb}_m}\end{aligned}$$

Menghitung *Static Head* :

$$\begin{aligned}Z_1 &= 1 \text{ m} \\ &= 3 \text{ ft}\end{aligned}$$

$$\begin{aligned}Z_2 &= 0 \text{ m} \\ &= 1 \text{ ft}\end{aligned}$$

$$\begin{aligned}\Delta Z &= Z_1 - Z_2 \\ &= 2.625 \text{ ft}\end{aligned}$$

$$g/gc = 1 \text{ lbf/lbm}$$

$$\begin{aligned}\Delta Z (g/gc) &= 3 \text{ ft} \times 1 \text{ lbf/lbm} \\ &= 2.625 \text{ ft lbf/lbm}\end{aligned}$$

Menghitung *Velocity Head* :

$$V_1 = \text{kecepatan linier fluida dari tangki hidrogen peroksida ke pipa}$$

$$V_2 = \text{kecepatan linier fluida ke heater}$$

Karena pada 2 titik *reference* dianggap sama, maka $V_1 = V_2 = 23.57 \text{ ft/s}$

$$\text{Sehingga velocity head } (\Delta V^2 / 2agc) = 8.630$$

Menghitung *Pressure Head* :

$$P_1 = P_2 = 14.7 \text{ Psi}$$

$$\Delta P = 0$$

$$\text{Sehingga, } \Delta P/\rho = 0$$

Menghitung Energi Mekanik Pompa :



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$- W_f = \frac{\Delta V^2}{2 \times \alpha \times g_c} + \Delta z \frac{g}{g_c} + \frac{\Delta P}{\rho} + \Sigma F$$

Dimana :

W_f = tenaga yang ditambahkan ke dalam sistem per satuan massa

Sehingga :

$$\begin{aligned} - W_f &= 9 + 3 \text{ ft. lbf/ll} + 0 + 53.09 \text{ ft. lbf/lbm} \\ &= 64.34 \text{ ft. lbf/lbm} \end{aligned}$$

Menghitung *Broke Horse Power* (BHP) :

$$BH = \frac{m. (-W_f)}{550 \cdot \eta}$$

dari **Fig. 14-37.** (Peters & Timmerhaus, 1991:520), untu = 22.3178 gpm
diperoleh η pompa = 0.45

Sehingga :

$$\begin{aligned} &= \frac{2.451 \times 64.34}{550 \times 0.45} \\ &= 0.637 \text{ Hp}; \text{ maka digunakan } power = 1 \text{ Hp} \end{aligned}$$

3. Menghitung Tenaga Motor

Fig. 14.38. (Peters & Timmerhaus, 1991:521) untuk BH = 1.0 Hp
diperoleh η motor = 0.80

Sehingga *power* motor yang diperlukan :

$$\begin{aligned} P \text{ moto} &= \frac{BHP}{\eta} \\ &= \frac{1}{0.80} \text{ Hp} \\ &= 1.25 \text{ Hp} \end{aligned}$$

Dipilih motor standar dengan *power* = 2 Hp (*Standard NEMA*)

SPEKIFIKASI POMPA NEUTRALISER

Nama Alat : Pompa *Neutraliser*

Kode : L-221

Fungsi : Mengalirkan *dioctyl phthalate* dari *neutraliser* menuju *rotary drum vacuum filter*

Tipe : *Centrifugal pump*

Bahan Kor : *Commercial steel*

Jumlah : 1 buah

Rate Volur : 0.050 ft³/s



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Kecepatan : 23.5660 ft/s
 Ukuran Pip : NPS = 1/2 in
 : Sch. Number = 40
 : OD = 0.0700 ft = 0.021 m
 : ID = 0.0518 ft = 0.016 m
 : Flow Area = 0.0021 ft² = 0.000 m²
 Power Pompa : 1 Hp
 Power Motor : 2 Hp

27 ROTARY DRUM VACUM FILTER (H-310)

Fungsi : Memisahkan C₁₆H₃₅O₄TiNa dari C₈H₄O₃
 C₈H₆O₄, C₂₄H₃₈O₄, C₈H₁₈O, dan H₂O
 Tipe : Rotary Drum Vacuum Filter
 Jumlah : 1 buah
 Kondisi Op : T = 95 °C = 368.15 K
 P = 3 in F = 212.19 psf

Komposisi bahan masuk

Komponen	Berat (kg/jam)	Fraksi berat	ρ bahan (gr/ml)	ρ bahan x fraksi berat
C ₂₄ H ₃₈ O ₄	3801.074	0.9499	0.984	0.935
C ₈ H ₁₈ O	5.0783	0.0013	0.833	0.001
C ₈ H ₄ O ₃	2.8907	0.0007	1.2	0.001
C ₈ H ₆ O ₄	2.8994	0.0007	1.526	0.001
6H ₃ SO ₄ Ti	0.0636	0.0000	0.834684	0.000
H ₂ O	189.6467	0.0474	0.997	0.047
Total	4001.653	1		0.985

rate massa = 4001.653 kg/jam
 = 8822.124 lb/jam

ρ campuran = $\sum \rho_i \times X_i$
 = 0.985 gr/ml
 = 61.4897 lb/cuft

rate volumetri = $\frac{\text{rate massa}}{\rho \text{ campuran}}$
 = $\frac{8822.124 \text{ lb/jam}}{61.4897 \text{ lb/cuft}}$
 = 143.4731 cuft/jam
 = 17.8875 gal/min



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
 \text{sg} &= \frac{\rho \text{ bahan}}{\rho \text{ reference}} \\
 &= \frac{61.4897 \text{ lb/cuft}}{62.43 \text{ lb/cuft}} \\
 &= 0.9849 \\
 \text{bahan} &= \frac{\text{sg bahan}}{\text{sg referenc}} \times \mu \text{ reference} \\
 &= \frac{0.9849}{0.996} \times 0.0009 \\
 &= 0.0008 \text{ lb/ft s} \quad (\text{berdasarkan sg bahan}) \\
 &= \text{##### cP} \quad (1 \text{ lb/fts} = 1488 \text{ cP})
 \end{aligned}$$

Perhitungan

$$\begin{aligned}
 \text{Asumsi tebal cake} &= 3.81 \text{ cm} \\
 \text{Minimum Tebal Cake} &= 6 \text{ cm} \quad (\text{Perry } 7^{\text{ed}}; T. 18-8) \\
 \text{Berat Cake kering (tiap cycle)} &= 10 \text{ kg/ m}^2 \quad (\text{Perry } 7^{\text{ed}}; \text{fig. 18-98}) \\
 \Theta_d/W \text{ untuk 25\% moist} &= 0.02 \quad (\text{Perry } 7^{\text{ed}}; \text{fig. 18-102}) \\
 \text{Waktu Pengeringan } \Theta_d &= \frac{\Theta_d}{W} \times W \\
 &= 0.02 \times 10 \\
 &= 0.2 \text{ Menit} \\
 \text{Asumsi Waktu Pencucian} &= 0.2 \text{ Menit} \\
 \% \text{ cycle dalam kondisi vaccun} &= 80\% \quad (\text{Perry Ed 7 T18-9}) \\
 \% \text{ cycle washing} &= 29\% \\
 \% \text{ cycle discharge} &= 20\%
 \end{aligned}$$

$$\begin{aligned}
 \% \text{ cycle untuk washing dan final dry} &= \% \text{ Cycle Total} - \% \text{ Cycle Washing} - \% \text{ Cycle Discharge} \\
 &= 80\% - 29\% - 20\% \\
 &= 31\%
 \end{aligned}$$

$$\begin{aligned}
 \text{Cycle time} &= \frac{\Theta \text{ Dry} + \Theta \text{ Wash}}{\% \text{ Cycle}} \\
 &= \frac{0 + 0}{31\%} \\
 &= 2.581
 \end{aligned}$$

$$\begin{aligned}
 \text{Asumsi Overall Scale-Up faktor} &= 1 \quad \text{Perry 18-87 ed 7} \\
 \text{Rate Filtrate} &= \frac{W}{\text{cycle time}} \times (60 \times \text{scale up}) \\
 &= \frac{10}{2.580645} \times (60 \times 0.9) \\
 &= 209.3 \text{ kg/jam m}^2
 \end{aligned}$$

$$\text{Dry cake pada produk} = 0.064 \text{ kg/jam}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
 \text{Total filtrate area} &= \frac{\text{Dry Cake}}{\text{Rate Filtrate}} \\
 &= \frac{0.0636344}{0.0000159} \\
 &= 4001.653 \text{ m}^2
 \end{aligned}$$

Dimensi Filter

Perbandingan L/D komersial untuk single unit (L/D = 2

$$\begin{aligned}
 \text{Total Filtrat Area (A)} &= 4001.653 \text{ m}^2 \\
 A &= \pi D L \\
 4001.65 &= 3.14 \times D \times 2 D \\
 D^2 &= \frac{4001.653}{5.024} \\
 D^2 &= 796.5074 \\
 D &= 28.2225 \text{ m} \\
 L &= 45.1559 \text{ m}
 \end{aligned}$$

Kondisi Vaccum

$$\begin{aligned}
 \text{Waktu awal pengeringan} &= 2.5806 \times 0 \\
 &= 0.2065 \text{ Min} \quad \text{Perry ed7 fig18-105} \\
 \text{Udara mengalir pd awal} &= 0.2065 \times 1 \\
 &= 0.2684 \text{ m}^3/\text{m}^2.\text{min}
 \end{aligned}$$

Akhir pengeringan

$$\begin{aligned}
 \text{Udara mengalir pd akhir} &= 0.2 \times 2.75 \\
 &= 0.55 \text{ m}^3/\text{m}^2.\text{min} \\
 \text{Total Filtrate x Keamanan} &= 0.8184 \text{ m}^3/\text{m}^2.\text{min} \times 1 \\
 &= 0.9002 \text{ m}^3/\text{m}^2.\text{min}
 \end{aligned}$$

$$\begin{aligned}
 \text{Total Udara (Log P)} &= \frac{0.9002}{2.5806} \\
 &= 0.3488 \text{ m}^3/\text{m}^2.\text{min} \\
 &= 1.4174 \times 10 \\
 &= 14.174 \text{ in Hg}
 \end{aligned}$$

Menghitung power rotary drum vacuum filter

$$\begin{aligned}
 \text{Power total} &= A^{0.75} \quad (\text{Ulrich T. 4 - 23, h : 222}) \\
 \text{Power total} &= 4001.653^{0.75} \\
 &= 503.1293 \text{ kWh} \\
 &= 674.4360 \text{ Hp} = 675 \text{ Hp}
 \end{aligned}$$

SPEKIFIKASI ROTARY DRUM VACUM FILTER

Fungsi : Memisahkan $C_{16}H_{35}O_4TiNa$ dari $C_8H_4O_3$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$C_8H_6O_4$, $C_{24}H_{38}O_4$, $C_8H_{18}O$, dan H_2O
 Type : Continuous Rotary Drum Vacuum Filter
 Bahan : Carbon Steel
 Diameter : 28.22 m area = 28234.13
 Panjang : 45.16 m 92630.52 ft
 Putaran : 6 rpm (Perry ed7 18-96)
 Vacuum : 14.17 inHg
 Power Total : 675 Hp
 Jumlah : 1 Buah

28 POMPA ROTARY DRUM VACUM FILTER (L-311)

Fungsi : Mengalirkan *dioctyl phthalate* dari *rotary drum vacum filter* menuju evaporator

Tipe : *Centrifugal pump*

Jumlah : 1 buah

Komponen	Massa (kg/jam)	Fraksi berat	ρ bahan (gr/ml)	ρ bahan x fraksi berat
$C_{24}H_{38}O_4$	3801.074	0.7036	0.99	0.6966
$C_8H_{18}O$	5.0783	0.0009	0.8207	0.0008
$C_8H_4O_3$	2.8907	0.0005	1.2	0.0006
$C_8H_6O_4$	2.8994	0.0005	1.522	0.0008
6H35O4Ti	0.0636	0.0000	0.8346	0.0000
H_2O	1590.225	0.2944	1	0.2944
Total	5402.232	1		0.9932

rate massa = 5402.23 kg/jam
 = 11909.87 lb/jam

$$\rho \text{ campuran} = \sum \rho_i \times X_i = 0.9932 \text{ g/ml}$$

$$= 62.002 \text{ lb/cuft}$$

$$\begin{aligned}
 \text{rate volumetrik} &= \frac{\text{rate massa}}{\rho \text{ campuran}} = \frac{11909.87}{62.0023} \\
 &= 192.0877 \text{ cuft/jam} \\
 &= 0.0534 \text{ cuft/det} \\
 &= 23.95 \text{ gpm}
 \end{aligned}$$

Asumsi aliran turbulen ($N_{re} > 2100$)

$$\begin{aligned}
 \text{Diameter optimum} &= 3.9 \times Q^{0.45} \times \mu^{0.13} \\
 &= 3.9 \times 0.2674 \times 1.71
 \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$= 1.784 \text{ inch}$$

Dari *App*. A.5-1 (Geankoplis, 1993:892) dipilih spesifikasi :

$$NP = 3/8 \text{ in}$$

$$OD = 0.675 \text{ in} = 0.056 \text{ ft} = 0.017 \text{ m}$$

$$ID = 0.493 \text{ in} = 0.041 \text{ ft} = 0.013 \text{ m}$$

$$A = 0.001 \text{ ft}^2 = 1.236 \text{ in}^2$$

$$Sch = 40$$

Menghitung kecepatan linier :

Kecepatan linier fluida dapat dicari dengan menggunakan persamaan berikut :

$$V = \frac{Q}{A}$$

Sehingga :

$$\begin{aligned} V &= \frac{0.0534 \text{ ft}^3/\text{s}}{0.0013 \text{ ft}^2} \\ &= 40.1186 \text{ ft/s} \\ &= 12.2281 \text{ m/s} \end{aligned}$$

Menghitung *Reynold Number* (Nre) :

$$Nre = \frac{\rho v D}{\mu}$$

Sehingga :

$$\begin{aligned} Nre &= \frac{62.00 \times 40.12 \times 0.0411}{0.0002} \\ &= 510961.8 \quad (\text{asumsi aliran turbulen benar}) \end{aligned}$$

Instalasi pipa

Dari **Fig. 127.** (Brown, 1950:141) = 3/8 in

- 1 buah *gate valve fully open* ; Le = 0.26 ft
 $\sum Le = 1 \times 0.26 = 0.26 \text{ ft}$
 - 3 buah *standard elbow* ; Le = 1.4 ft
 $\sum Le = 3 \times 1.4 = 4.2 \text{ ft}$
 - 1 buah *sudden enlargement* ; Le = 1.4 ft
 $\sum Le = 1 \times 1.4 = 1.4 \text{ ft}$
 - 1 buah *sudden contraction* ; Le = 0.52 ft
 $\sum Le = 1 \times 0.52 = 0.52 \text{ ft}$
 - 1 buah *swing check valve* ; Le = 3 ft
 $\sum Le = 1 \times 3 = 3 \text{ ft}$
 - Panjang ekivalen pipa lurus, $\sum Le = 9.38 \text{ ft}$
Panjang pipa lurus = 2 m
= 6.562 ft
-



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}\text{Panjang pipa total} &= 15.94 \text{ ft} \\ &= 4.859 \text{ m}\end{aligned}$$

Menghitung *friction loss*

$$\Sigma F = \frac{4f \cdot v^2 \cdot \Delta L}{2 \cdot ID \cdot gc} \quad \text{Eq. 2.10-6 (Geankoplis, 1993:89)}$$

Dimana :

$$\begin{aligned}\Sigma F &= \text{Friction lo. (ft.lbf/lbm)} \\ f &= \text{Faktor friksi} \\ v &= \text{Kecepatan linier fluida (ft/s)} \\ \Delta L &= \text{Panjang pipa (ft)} \\ ID &= \text{Diameter dalam tangki (ft)} \\ gc &= 32.17 \text{ lbm.ft/lbf.s}^2\end{aligned}$$

Menghitung *Fanning Friction Factor* (f) :

Dari **Fig. 2.10-3.** (Geankoplis, 1993:88) didapat :

$$\begin{aligned}\text{Untuk } commercial \rightarrow \varepsilon &= 5.E-05 \text{ m} \\ &= 0.0002 \text{ ft}\end{aligned}$$

Sehingga :

$$\frac{\varepsilon}{D} = \frac{0.000}{0.041} = 0.004$$

Dari **Fig. 2.10-3.** (Geankoplis, 1993:88) dengan nilai $N_{re} = 510962$

didapatkan nilai $f = 0.007$ sehingga :

$$\begin{aligned}\Sigma F &= \frac{4 \times 0.007 \times (40.1)^2 \times 15.9}{2 \times 0.0411 \times 32.1740} \\ &= 271.7565 \frac{\text{lb}_f \cdot \text{ft}}{\text{lb}_m}\end{aligned}$$

Menghitung *Static Head* :

$$Z_1 = 1 \text{ m} = 3.281 \text{ ft}$$

$$Z_2 = 0.2 \text{ m} = 0.656 \text{ ft}$$

$$\Delta Z = Z_1 - Z_2 = 2.625 \text{ ft}$$

$$g/g = 1 \text{ lbf/lbm}$$

$$\begin{aligned}\Delta Z (g/g) &= 2.625 \text{ ft} \times 1 \text{ lbf/lbm} \\ &= 2.625 \text{ ft lbf/lbm}\end{aligned}$$

Menghitung *Velocity Head* :



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Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

V_1 = kecepatan linier fluida dari tangki hidrogen peroksida ke pipa

V_2 = kecepatan linier fluida ke heater

Karena pada 2 titik *reference* dianggap sama, maka $V_1 = V_2 = 40.1186$ ft/s

Sehingga *velocity head* ($\Delta V^2 / 2agc$) = 25.0124

Menghitung *Pressure Head* :

$$P_1 = P_2 = 14.7 \text{ Psi}$$

$$\Delta P = 0$$

$$\text{Sehingga, } \Delta P/\rho = 0$$

Menghitung Energi Mekanik Pompa :

$$-W_f = \frac{\Delta V^2}{2 \times \alpha \times gc} + \Delta z \frac{g}{gc} + \frac{\Delta P}{\rho} + \sum F$$

Dimana :

W_f = tenaga yang ditambahkan ke dalam sistem per satuan massa

Sehingga :

$$\begin{aligned} -W_f &= 25 + 2.625 \text{ ft. lbf/lbm} + 0 + 271.8 \text{ ft. lbf/lbm} \\ &= 299.4 \text{ ft. lbf/lbm} \end{aligned}$$

Menghitung *Broke Horse Power* (BHP) :

$$\text{BHP} = \frac{m. (-W_f)}{550 \cdot \eta}$$

dari **Fig. 14-37.** (Peters & Timmerhaus, 1991:520)

$$Q_f = 23.9 \text{ gpm} \text{ diperoleh } \eta \text{ pompa} = 0.45$$

Sehingga :

$$\begin{aligned} \text{BHP} &= \frac{3.308 \times 299.4}{550 \times 0.45} \\ &= 4.002 \text{ Hp; maka digunakan power} = 4 \text{ Hp} \end{aligned}$$

4) Menghitung Tenaga Motor

Fig. 14.38. (Peters & Timmerhaus, 1991:521) untuk BHP = 4 Hp

diperoleh η motor = 1

Sehingga power motor yang diperlukan :

$$\begin{aligned} P_{\text{moto}} &= \frac{3\text{BHP}}{\eta} \\ &= \frac{4}{1} \text{ Hp} \\ &= 5 \text{ Hp} \end{aligned}$$



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Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

Dipilih motor standar dengan power = 5 Hp (Standard NEMA)

SPEKIFIKASI POMPA ROTARY DRUM VACUM FILTER

Nama Alat : Pompa *Rotary Drum Vacum Filter*
 Kode : L-311
 Fungsi : Mengalirkan *dioctyl phthalate* dari *rotary drum vacum filter* menuju evaporator
 Tipe : *Centrifugal pump*
 Bahan Konstru : *Commercial steel*
 Jumlah : 1 buah
 Rate Volumet : 0.053 ft³/s
 Kecepatan Alir : 40.12 ft/s
 Ukuran Pipa : NPS = 3/8 in
 : *Sch. Number* = 40
 : OD = 0.056 ft = 0.017 m
 : ID = 0.041 ft = 0.013 m
 : *Flow Area* = 0.001 ft² = 1.236 in²
 Power Pompa : 4 Hp
 Power Motor : 5 Hp

29 EVAPORATOR (V-320)

Fungsi : untuk memisahkan kandungan C₃H₈O dan H₂O dari C₈H₁₈O dan C₂₄H₃₈O₄
 Tipe : Evaporator berbentuk *short vertical tube* dengan tutup atas dan bawah berbentuk *torispherical*
 Jumlah : 1 buah
 Kondisi Operasi : T_{in} = 95 °C = 368.2 K
 T_{out} = 102.14 °C = 375.3 K
 P = 1 atm = 760 mmHg = 14.7 psi
 1 BTU/lb = 1 kkal/kg
 1 kkal/kg = 2 btu/lb

KOMPOSISI

Komponen	Berat (kg)	Fraksi berat	ρ (gr/cc)	C _p (btu/lb)
C ₈ H ₁₈ O	5.0783	0.0009	0.833	0.215773
C ₂₄ H ₃₈ O ₄	3801.0743	0.7073	0.984	0.268309
H ₂ O	1562.2124	0.2907	0.997	0.516967
C ₈ H ₄ O ₃	2.8907	0.0005	1.2	1.000481
C ₈ H ₆ O ₄	2.8994	0.0005	1.526	1.000481
Total	5374.1551	1		



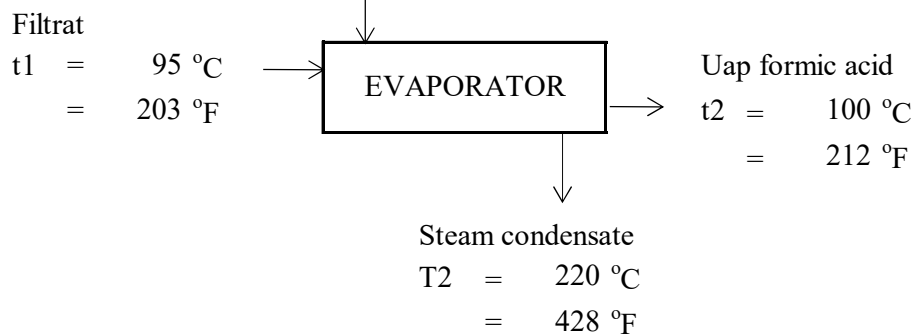
PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned} \text{Densitas Campuran} &= \frac{1}{\frac{\text{fraksi berat}}{\rho \text{ komponen}}} \times 62.43 \\ &= \frac{1}{\frac{0}{1} + \frac{1}{1} + \frac{0}{1} + \frac{0}{1} + \frac{0}{2}} \times 62 \\ &= 0.99 \text{ gr/cc} \times 62 \\ &= 61.67 \text{ lb/cuft} \end{aligned}$$

$$\begin{aligned} \text{Rate Volumetrik} &= \frac{\text{Rate Massa}}{\text{Densitas}} \\ &= \frac{5374.16}{61.6721} \\ &= 87.1408 \text{ cuft/jam} \end{aligned}$$

Diagram suhu : Steam $T_1 = 220 \text{ }^\circ\text{C}$
 $= 428 \text{ }^\circ\text{F}$



1. Heat Balance

Data dari Appendix A dan Appendix B diperoleh :

$$\begin{aligned} \text{Rate massa, W massa} &= 5,374 \text{ kg/jam} \\ &= 11,850 \text{ lb/jam} \\ \text{Q supply} &= 782,602 \text{ kkal/jam} \\ &= 3,106,930 \text{ Btu/jam} \\ \text{W steam} &= 1,764 \text{ kg/jam} \\ &= 3,890 \text{ lb/jam} \end{aligned}$$

2. Log Mean Temperature Difference

$$\begin{aligned} \text{Suhu bahan masuk} &= 95 \text{ }^\circ\text{C} = 203 \text{ }^\circ\text{F} (t_1) \\ \text{Suhu bahan keluar} &= 100 \text{ }^\circ\text{C} = 212 \text{ }^\circ\text{F} (t_2) \\ \text{Suhu steam masuk} &= 220 \text{ }^\circ\text{C} = 428 \text{ }^\circ\text{F} (T_1) \\ \text{Suhu steam condensat} &= 220 \text{ }^\circ\text{C} = 428 \text{ }^\circ\text{F} (T_2) \\ \Delta T \text{ LMTD} &= \frac{225 - 216}{\ln 1.0417} \\ &= 220.4694 \text{ }^\circ\text{F} \end{aligned}$$



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Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$R = \frac{T_1 - T_2}{t_2 - t_1} = \frac{0}{9} = 0.00$$

$$S = \frac{t_2 - t_1}{T_1 - t_1} = \frac{9}{225} = 0.04$$

Untuk type evaporator standart tube vertical dipilih jenis 2-4 shell & tube exchanger.

$$\text{Untuk 2-4 shell \& tube } F_T = 0.97 \quad [Kern; App fig. 18]$$

$$1/2 \cdot 0.85$$

$$\begin{aligned} \Delta T &= F_T \times \Delta T \text{ LMTD} \\ &= 0.97 \times 220.47 \\ &= 213.8553 \text{ } ^\circ\text{F} \end{aligned}$$

3. Tc dan tc

$$T_c = T_{av} \text{ steam}$$

$$T_c = \frac{T_1 + T_2}{2} = \frac{428 + 428}{2} = 428 \text{ } ^\circ\text{F}$$

$$t_c = t_{av} \text{ bahan}$$

$$t_c = \frac{t_1 + t_2}{2} = \frac{203 + 212}{2} = 208 \text{ } ^\circ\text{F}$$

$$\begin{aligned} Sg \text{ bahan} &= \frac{\rho \text{ bahan}}{\rho \text{ reference}} \times Sg \text{ reference} \\ &= \frac{61.6721 \text{ lb/c}}{62.43 \text{ lb/c}} \times 1 \\ &= 0.9879 \end{aligned}$$

μ berdasarkan Sg bahan :

$$[Kern; tab.6 p. 808] \text{ didapat } Sg \text{ reference} = 1$$

$$[Kern; fig.14, p. 823] \text{ didapat } \mu \text{ reference} = 0.899 \text{ Cp}$$

$$\begin{aligned} \mu \text{ bahan} &= \frac{Sg \text{ bahan}}{Sg \text{ reference}} \times \mu \text{ reference} \\ &= \frac{0.9879}{1} \times 0.899 \\ &= 0.8881 \text{ Cp} \end{aligned}$$

Evaporator termasuk sistem heater karena :

Hot Fluid : Steam

Cold Fluid : Aqueous Solution dimana μ bahan kurang dari 2 Cp

Sehingga di dapat nilai U_D dengan rang 200 - 700 Btu/jam.ft²°F

[Kern; tab.8, p.840]

$$\text{Dipilih nilai } U_D = 220 \text{ Btu/jam.ft}^2 \text{ } ^\circ\text{F}$$



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Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

Digunakan 1 buah evaporator, sehingga luas perpindahan panas evaporator :

$$A = \frac{Q}{UD \times \Delta T}$$

Dimana,

A = Luas Perpindahan Panas

Q = Q supply dari steam

UD = Overall Design Coefficients

ΔT = Perubahan Suhu

$$\begin{aligned} A &= \frac{Q}{UD \times \Delta T} \\ &= \frac{3,106,930.0393}{220 \times 213.8553} \\ &= 66.0372 \text{ ft}^2 \\ &= 7.3375 \text{ m}^2 \end{aligned}$$

Dikarenakan luas perpindahan panas > 200 ft², maka design heater menggunakan shell and tube *[Walas; p.xvi]*

$$\begin{aligned} \text{Jumlah tube (Nt)} &= \frac{A}{L \times a''} \\ &= \frac{66.0372}{16 \times 0.3271} \\ &= 13 \text{ buah} \end{aligned}$$

Dalam perencanaan ini digunakan HEATER dengan spesifikasi :

Digunakan TUBE dengan ukuran : *[Kern; tab. 10, p. 843]*

OD, BWG = 1 1/4 in , 12 BWG

Pitch = 1 4/7 in triangular *[Kern; tab.9]*

Panjang tube, L = 16 ft *(L tube: 7 - 13 ft)*

Passes = 6

ID Tube = 1.030 in 0.1 ft *(min ID: 1 in)*

a't = 0.836 in²

a'' = 0.3271 ft²/lin ft

Digunakan Nt = 91 buah

Dari tube passes dan jumlah tube didapat SHELL:

Untuk pipa 1 1/4 OD tubes on 1 9/16 in square pitch

ID shell = 21 1/4 in *[Kern; tab.9]*

Baffle = 5 in

Passes (n) = 2

*Koreksi U_D

$$A = Nt \times L \times a''$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
 &= 476.2576 \text{ ft}^2 \\
 U_D \text{ koreksi} &= \frac{Q}{A \times \Delta T} \\
 &= \frac{3,106,930.039}{476.3 \times 213.9} \\
 &= 31 \text{ Btu/jam.ft}^2 \text{ } ^\circ\text{F}
 \end{aligned}$$

Nilai U_D asumsi sama dengan U_D koreksi (memenuhi)

Fluida Panas (Tube Side) Steam	Fluida Dingin (Shell Side) Larutan NaOH
<p>(4) Flow area (a_t)</p> $ \begin{aligned} a_t &= \frac{N_t \times a'_t}{144 \times n} \\ &= \frac{91 \times 0.84}{144 \times 6} \\ &= 0.0881 \text{ ft}^2 \end{aligned} $	<p>(4') Flow area (a_s)</p> $ \begin{aligned} C' &= P_t - O.D. = 0.31 \text{ in} \\ a_s &= \frac{I.D.s \times C' \times B}{144 \times P_T} \\ &= \frac{21.25 \times 1/3 \times 5}{144 \times 1 \ 4/7} \\ &= 0.1476 \text{ ft}^2 \end{aligned} $
<p>(5) Kecepatan massa (Gt)</p> $ \begin{aligned} G_t &= \frac{W}{a_t} \\ &= \frac{3,889.797}{0.0881} \\ &= 44176.66883 \text{ lb/jam ft}^2 \end{aligned} $	<p>(5') Kecepatan massa (Gs)</p> $ \begin{aligned} G_s &= \frac{W}{a_s} \\ &= \frac{11,850.0119}{0.1476} \\ &= 80,301.2572 \text{ lb/jam ft}^2 \end{aligned} $
<p>(6) pada $t_c = 208 \text{ } ^\circ\text{F}$ $\mu_{\text{stean}} = 0.0140 \text{ cps}$ $= 0.034 \text{ lb / jam.ft}$ (Kern, Fig 15 ; Page 825) $D_i = 1.03 \text{ in} = 0.0858 \text{ ft}$ $Re_t = \frac{D_i \times G_t}{\mu}$ $= 111919.4434$</p>	<p>(6') pada $T_c = 428 \text{ } ^\circ\text{F}$ $\mu_{\text{bahan}} = 0.888 \text{ cps}$ $= 2.1492 \text{ lb/jam.ft}$ $D_e = 0.0198 \text{ ft}$ $Re_s = \frac{D_e \times G_s}{\mu}$ $= 740.9834$</p>
<p>(7) $jH = 180 \text{ [Fig.24]}$</p>	<p>(7') $jH = 110$ <i>[Fig.28 shell side data]</i></p>
	<p>(8') pada $T_c = 428 \text{ } ^\circ\text{F}$ $k = 0.9 k_{\text{water}}$ $= 0.015 \text{ Btu/hr.ft}^2 \text{ } (^{\circ}\text{F/ft})$</p>



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Pabrik Diocetyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Fluida Panas (Tube Side) Steam	Fluida Dingin (Shell Side) Larutan NaOH
	$C_p = 0.4500 \text{ Btu / lb}$ $\left[\frac{C_p \cdot \mu}{k} \right]^{1/3} = \frac{0.45 \times 0.034}{0.015}$ $= 1.0068$
(9) Condensation of steam : $h_{io} = 1500 \text{ Btu/hr.ft}^2 \cdot ^\circ\text{F}$ <p style="text-align: right;"><i>[Kern:164]</i></p>	(9') $h_o = J_H \times (k/De) \times (c \times \mu / k)^{1/3} \times \phi_s$ $h_o = 83.430 \text{ Btu/hr.ft}^2 \cdot ^\circ\text{F}$
(10) tw^* $tw = tc + \frac{h_{io}}{h_{io} + h_o} (Tc - tc)$ $= 208 + \frac{1500}{1583.430} \times 221$ $= 416.382 \text{ } ^\circ\text{F}$	(10') Pada $tw = 416.382 \text{ } ^\circ\text{F}$ $\mu_w = \mu_{\text{water}} \text{ [Fig.14]}$ $= 1.776 \text{ cps}$ $= 4.298 \text{ lb / jam.ft}$ $\phi_s = (\mu/\mu_w)^{0.14}$ $= 0.9075$
	(11') Corrected coefficient $h_o = \frac{h_o \cdot \phi_s}{\phi_s}$ $= 75.714 \text{ Btu/hr.ft}^2 \cdot ^\circ\text{F}$
(12) Clean Overall coefficient, U_c : $U_c = \frac{h_{io} \times h_o}{h_{io} + h_o}$ $= 72.0761 \text{ Btu/hr.ft}^2 \cdot ^\circ\text{F}$	
(13) Dirt Factor, R_d : $R_d \text{ hitung} = \frac{U_c - U_D}{U_c \times U_D}$ $= 0.0189 \text{ Btu/hr.ft}^2 \cdot ^\circ\text{F}$	
[Kern, T-12 Page 845]	
$R_d \text{ perhitungan} > R_d \text{ data} \quad 0.0189 > 0.0030$	
Fluida Panas (Tube Side) Steam	Fluida Dingin (Shell Side) Larutan Caustic Soda
(1) Sg bahan: $s = \frac{0.9879}{62.5} = 0.0158$ $Re_t = 111,919.443$ $f = 0.0002 \text{ ft}^2/\text{in}^2 \text{ [Fig.26]}$	(1') Flow area (a_s) $N+1 = 12L/B$ $= 38.4$ $D_s = 1.7708 \text{ ft}$ $Re's = \frac{D_s' \cdot G_s}{\mu}$



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Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

Fluida Panas (Tube Side) Steam	Fluida Dingin (Shell Side) Larutan Caustic Soda
$\Delta P_t = \frac{1}{2} \times \frac{f \cdot G^2 \cdot L \cdot n}{5.22 \times 10^{10} \times D \times s \times \phi t}$ $= 0.2646 \text{ Psi}$	$= 66,165.1882$ $f = 0.0012 \text{ ft}^2/\text{in}^2$ <p style="text-align: right;"><i>[Fig.29]</i></p>
$\Delta P_T < 2 \text{ psi}$ (memenuhi)	$(2')$ $\Delta P_s = \frac{f \cdot G_s^2 \cdot D_s \cdot (N+1)}{5.22 \times 10^{10} \times D_e \times s \times \phi s}$ $= 0.0280 \text{ Psi}$ $\Delta P_s < 10 \text{ psi}$ (memenuhi)

Nilai U_D asumsi sama dengan U_D koreksi (memenuhi)

Menentukan dimensi evaporator

Panjang Tube (L) = 16 ft

Tinggi evaporator 1.5 - 2) dari panjang tube *[Hugot ed.3; p.508]*

Tinggi evaporator = 2 x 16
= 24 ft = 7.2 m

Asumsi:

$$\frac{H}{D} = 2$$

$$D = \frac{24}{2}$$

$$= 12 \text{ ft} = 3.6 \text{ m}$$

Diameter Vessel = 4 m (max = 4 m) *[Ulrich; p.94]*

Diameter Center well: ## m (1/4 Diameter Vessel) *[Hugot ed.3; p.509]*

Tinggi Evaporator = ## m (3.6 - 4 m) *[Hugot ed.3; p.508]*

Menentukan volume shell

Volume cairan yang ditampung di evapora = 87.1408 cuft/jam

Waktu tinggal cairan : 2.8 menit untuk 3.03 m³ liquid

Menggunakan waktu tingg = 15 menit *[Coulson ed.5 vol.2; p.901]*

$$\text{Volume bahan} = 87.1408 \frac{\text{cuf}}{\text{jam}} \times \frac{1 \text{ jam}}{60 \text{ menit}} \times 15 \text{ menit}$$

$$= 21.2944 \text{ cuf} = 0.60 \text{ m}^3$$

Volume tangki = 26.6180 cuft

Tinggi cairan dalam evaporator :

$$V = \frac{1}{4} \pi D^2 Z_L$$



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Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$Z_L = \frac{4}{3.14} \times \frac{26.6180}{144}$$

$$= 0.2355 \text{ ft} = 0.0754 \text{ m}$$

Menentukan Tekanan Design

Jika didalam bejana terdapat liquid , maka :

$$P \text{ operasi} = 1 \text{ atr} = 14.7 \text{ psi}$$

$$P \text{ hidrostatik} = \rho \times g/gc \times h$$

$$= 61.67 \frac{\text{lbm}}{\text{cuft}} \times 1 \frac{\text{lbf}}{\text{lbm}} \times 0.2355 \text{ ft}$$

$$= 14.52 \frac{\text{lbf}}{\text{ft}^2}$$

$$= 0.10 \text{ psi}$$

Asumsi P design 10% lebih besar untuk faktor keamanan

$$P \text{ total} = P \text{ hidrostatik} + P \text{ Operasi}$$

$$= 0.10 + 14.7$$

$$= 14.80 \text{ psi}$$

$$P \text{ desig} = 110\% \times 14.80$$

$$= 16.28 \text{ psi}$$

Menentukan tebal minimum shell

Tebal shell berdasarkan ASME code untuk cylindrical tank :

$$t_{\min} = \frac{P \times r_i}{f_e - 0,6P} + C \quad [Brownell\&Young; \text{ eq.13-1, p.254}]$$

dengan :

$$t_{\min} = \text{tebal shell minimum}; \text{in}$$

$$P = \text{tekanan tangki} ; \text{psi}$$

$$r_i = \text{jari-jari tangki} ; \text{in (1/2 D)}$$

$$C = \text{faktor korosi} ; \text{in (digunakan } 1/8 \text{ in)}$$

$$E = \text{faktor pengelasan, digunakan double weld} \#\#$$

$$f = \text{stress allowable, bahan konstruksi Stainless steel 316 grade D, maka } f = 35970$$

[Perry ed.7; T.28-11]

$$r_i = 1 \times \frac{144}{72} \text{ in}$$

$$t_{\min} = \frac{P \times r_i}{fE - 0,6P} + C$$

$$= \frac{16.2809 \times 72}{28775.9 - 9.7686} + 0.125$$

$$= \frac{1172.23}{28766.17} + 0.125$$

$$= 0.0408 + 0.125$$

$$= 0.1658 \text{ in}$$



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Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

digunakan $t_{min} = 3/16$ in

Dimensi tutup atas (Elliptical dished)

$$\begin{aligned} OD &= ID + 2 t_s \\ &= 144 + 0.4 \\ &= 144.4 \text{ in} \end{aligned}$$

Berdasarkan [Brownell & Young; Tab. 5-7]

$$\begin{aligned} OD &= 144 \text{ in} \\ t_{shell} &= 3/16 \text{ in} \\ icr &= 7 \frac{5}{8} \text{ in} \\ rc &= 120 \text{ in} \end{aligned}$$

karena icr lebih besar dari 6% maka digunakan persamaan berikut

[Brownell & Young; eq.13.12 p.258]

Tebal tutup atas Elliptical dished :

$$\begin{aligned} t_h &= \frac{P \times r_c \times W}{2 f \cdot e - 0.2 P} + C \\ W &= \frac{1}{4} \times \left(3 + \sqrt{\frac{rc}{icr}} \right) \\ &= \frac{1}{4} \times (3 + 6) \\ &= 2.344 \end{aligned}$$

dengan:

t_h = tebal tutup (head) shell minimu ; in

r_c = *radius of curfative* sama deng; in

W = faktor stress intensif

P = tekanan tangki ; psia

E = faktor pengelasan, digunakan jenis *double welded* = 1

C = faktor korosi = 1/8 in

f = allowable stress, bahan konstruksi Stainless stell 316
maka $f = 35970$ psi [Perry ed.7; T.28-11]

$$\begin{aligned} t_h &= \frac{P \times r_c \times W}{2 f \cdot E - 0.2 P} + C \\ &= \frac{16.2809 \times 120 \times #####}{57551.87 - 3.2562} + 1/8 \\ &= \frac{#####}{57548.62} + 1/8 \\ &= 0.2046 \end{aligned}$$

digunakan = 4/16 in

$$\begin{aligned} h &= rc - \sqrt{rc^2 - \frac{D^2}{4}} \\ &= 24.1409 \text{ in} = 2.012 \text{ ft} \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

Dimensi tutup bawah (Conical)

$$\text{Tebal conical} = \frac{\text{P.D}}{2 \cos \alpha (\text{fe} - 0,6\text{P})} + C \quad [\text{Brownell\&Young};$$

Dengan $\alpha = 1/2$ sudut conis = $60^\circ/2 = 30$

$$tc = \frac{\text{P.D}}{2 \cos \alpha (\text{fe} - 0,6\text{P})} + C$$

$$tc = \frac{16.28}{1.73} \times \frac{1.0}{28766.17} + 1/8$$

$$= \frac{16.28}{49824.46} + 1/8$$

$$= 0.1253 \text{ in digunakan } t = 3/16 \text{ in}$$

Tinggi conical :

$$h = \frac{\text{tg } \alpha \times (\text{D} - m)}{2} \quad [\text{Hesse}; \text{eq.4-17}]$$

eterang: = $\alpha = 1/2$ sudut conis ; 30

D = diameter tangk ; ft

m = flat spot center ; 12 in = 1 ft

$$h = \frac{\text{tg } \alpha \times (\text{D} - m)}{2}$$

$$= \frac{0.577 \times 11}{2} = 3.18 \text{ ft}$$

SPESIFIKASI EVAPORATOR

Type Heater : 2-4 Shell & Tube Exchanger

Jumlah = 1 buah

Bagian Shell :

Jenis Bahan = Stainless steel 316

Diameter Vessel = 3.6 m

Diameter Centerwall = 0.9 m

Tinggi Vessel = 7.2 m

Tebal Shell = 3/16 in

Tebal Tutup Atas = 1/4 in

Tinggi Tutup Atas = ##### ft

Tebal Tutup Bawah = 3/16 in

Tinggi Tutup Bawah = ##### ft

Tinggi total vessel = 26.7872 ft

ID Shell = 21 1/4 in

Baffle Space = 5 in

Passes = 2

Bagian Tube :

Jumlah & Panjang Tul = 91 buah, 16 ft

OD, BWG, pitch = 1 in, 12 BWG 1 4/7 Triangular



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Passes = 6

30 POMPA EVAPORATOR (L-321)

Fungsi : Mengalirkan *dioctyl phthalate* dari evaporator menuju tangki dekolorisasi

Tipe : *Centrifugal pump*

Jumlah : 1 buah

Tujuan : 1. Menentukan tipe pompa
2. Menentukan bahan konstruksi pompa
3. Menghitung tenaga pompa
4. Menghitung tenaga motor

3) Menghitung Tenaga Pompa

Rate masuk	=	3812.0989	kg/jam	=	2.3345	lb/s
Densitas	=	0.9879	kg/m ³	=	0.0617	lb/ft ³
μ	=	2.8134	cP	=	0.0019	lb/ft.s

$$\begin{aligned} \text{Rate volumetrik (Q)} &= \frac{\text{massa}}{r} \\ &= \frac{2.3345 \text{ lb/s}}{0.0617 \text{ lb/ft}^3} = 37.8532 \text{ ft}^3/\text{s} \\ &= 16989.7011 \text{ gpm} \end{aligned}$$

Diperkirakan aliran fluida turbulen ($N_{re} > 2100$), sehingga digunakan persamaan $Di \geq 1$ in yaitu :

$$Di_{opt} = 3.9 Q^{0.45} \mu^{0.13} \quad \text{Eq. 45. (Peters \& Timmerhaus, 1991:365)}$$

Dimana :

Di_{opt} = diameter dalam optimum, in

Q = kecepatan volumetrik, ft³/s

μ = viskositas fluida, lb/ft.s

Sehingga :

$$\begin{aligned} Di_{opt} &= 3.9 \times (37.8532)^{0.45} \times (0.0019)^{0.13} \\ &= 8.8546 \text{ in} \end{aligned}$$

Dari App. A.5-1 (Geankoplis, 1993:892) dipilih spesifikasi :

NPS = 1/2 in

OD = 0.8400 in = 0.07 ft = 0.021336 m

ID = 0.6220 in = 0.0518333 ft = 0.015799 m

A = 0.0021 ft² = 1.9603 in²

Sch = 40



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Menghitung kecepatan linier :

Kecepatan linier fluida dapat dicari dengan menggunakan persamaan berikut :

$$V = \frac{Q}{A}$$

Sehingga :

$$V = \frac{37.8532 \text{ ft}^3/\text{s}}{0.0021 \text{ ft}^2} = 17939.9093 \text{ ft/s} = 5468.0844 \text{ m/s}$$

Menghitung *Reynold Number* (Nre) :

$$Nre = \frac{\rho v D}{\mu}$$

Sehingga :

$$Nre = \frac{0.0617 \times 17939.9093 \times 0.0518}{0.0019} = 30333.14104 \text{ (asumsi aliran turbulen benar)}$$

Instalasi pipa

Dari **Fig. 127.** (Brown, 1950:141) dengan NPS = 1/2 in

- 1 buah <i>gate valve fully open</i> ; Le	= 0.35 ft	
$\sum Le = 1 \times 0.35$	= 0.35 ft	
- 4 buah <i>standard elbow</i> ; Le	= 1.7 ft	
$\sum Le = 4 \times 1.7$	= 6.8 ft	
- 1 buah <i>sudden enlargement</i> ; Le	= 1 ft	
$\sum Le = 1 \times 1$	= 1 ft	
- 1 buah <i>sudden contraction</i> ; Le	= 0.58 ft	
$\sum Le = 1 \times 0.58$	= 0.58 ft	
- 1 buah <i>swing check valve</i> ; Le	= 3.5 ft	
$\sum Le = 1 \times 3.5$	= 3.5 ft	
- Panjang ekivalen pipa lurus, $\sum Le$	= 12.23 ft	
Panjang pipa lurus	= 2 m	
	= 6.5616 ft	
Panjang pipa total	= 18.7916 ft	
	= 5.7277 m	

Menghitung *friction loss*

$$\Sigma F = \frac{4f \cdot v^2 \cdot \Delta L}{2 \cdot ID \cdot gc} \quad \text{Eq. 2.10-6 (Geankoplis, 1993:89)}$$

Dimana :

$$\Sigma F = \text{Friction loss} \quad (\text{ft.lbf/lbm})$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

- f = Faktor friksi
 v = Kecepatan linier fluida (ft/s)
 ΔL = Panjang pipa (ft)
 ID = Diameter dalam tangki (ft)
 gc = 32.1740 lbm.ft/lbf.s²

Menghitung *Fanning Friction Factor* (f):

Dari **Fig. 2.10-3.** (Geankoplis, 1993:88) didapat :

Untuk *commercial steel* $\rightarrow \epsilon = 5.E-05$ m
 $= 0.0002$ ft

Sehingga :

$$\frac{\epsilon}{D} = \frac{0.0002}{0.0518} = 0.0030$$

Dari **Fig. 2.10-3.** (Geankoplis, 1993:88) dengan nilai $Nre = 30333.1410$ didapatkan nilai $f = 0.008$ sehingga :

$$\Sigma F = \frac{4 \times 0.0080 \times (17939.9093)^2 \times 18.7916}{2 \times 0.0518 \times 32.1740}$$

$$= 58024316.55 \frac{\text{lb}_f \cdot \text{ft}}{\text{lb}_m}$$

Menghitung *Static Head* :

$$Z_1 = 1 \text{ m} = 3.28084 \text{ ft}$$

$$Z_2 = 0.2 \text{ m} = 0.656168 \text{ ft}$$

$$\Delta Z = Z_1 - Z_2 = 2.6247 \text{ ft}$$

$$g/gc = 1 \text{ lbf/lbm}$$

$$\Delta Z (g/gc) = 2.624672 \text{ ft} \times 1 \text{ lbf/lbm} = 2.624672 \text{ ft lbf/lbm}$$

Menghitung *Velocity Head* :

V_1 = kecepatan linier fluida dari tangki hidrogen peroksida ke pipa

V_2 = kecepatan linier fluida ke *heater*

Karena pada 2 titik *reference* dianggap sama, maka $V_1 = V_2 = 17939.91$ ft/s

$$\text{Sehingga velocity head } (\Delta V^2 / 2agc) = 5001559.442$$

Menghitung *Pressure Head* :

$$P_1 = P_2 = 14.7 \text{ Psi}$$

$$\Delta P = 0$$

$$\text{Sehingga, } \Delta P/\rho = 0$$



PERANCANGAN PABRIK

Pabrik Diocetyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Menghitung Energi Mekanik Pompa :

$$- W_f = \frac{\Delta V^2}{2 \times \alpha \times g_c} + \Delta z \frac{g}{g_c} + \frac{\Delta P}{\rho} + \sum F$$

Dimana :

W_f = tenaga yang ditambahkan ke dalam sistem per satuan massa

Sehingga :

$$\begin{aligned} - W_f &= ##### + 3 \text{ ft. lbf/lbm} + 0 + 58024317 \text{ ft. lbf/lbm} \\ &= 63025878.62 \text{ ft. lbf/lbm} \end{aligned}$$

Menghitung *Broke Horse Power* (BHP) :

$$\text{BHP} = \frac{m \cdot (-W_f)}{550 \cdot \eta}$$

dari **Fig. 14-37.** (Peters & Timmerhaus, 1991:520), untuk $Q_f = 16989.7011 \text{ gpm}$ diperoleh η pompa = 0.40

Sehingga :

$$\begin{aligned} \text{BHP} &= \frac{2.3345 \times 63025878.62}{550 \times 0.4000} \\ &= 668786.8 \text{ Hp}; \text{ maka digunakan } power = 1 \text{ Hp} \end{aligned}$$

4) Menghitung Tenaga Motor

Fig. 14.38. (Peters & Timmerhaus, 1991:521) untuk $BH = 1.0 \text{ Hp}$ diperoleh η motor = 0.80

Sehingga *power* motor yang diperlukan :

$$\begin{aligned} P_{\text{motor}} &= \frac{\text{BHP}}{\eta} \\ &= \frac{1 \text{ Hp}}{0.80} \\ &= 1.3 \text{ Hp} \end{aligned}$$

Dipilih motor standar dengan $power = 2 \text{ Hp}$ (*Standard NEMA*)

Spesifikasi Pompa Evaporator

Nama Alat : Pompa Evaporator
Kode : L-321
Fungsi : Mengalirkan *diocetyl phthalate* dari evaporator menuju tangki dekolorisasi
Tipe : *Centrifugal pump*
Bahan Konstruksi : *Commercial steel*
Jumlah : 1 buah



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

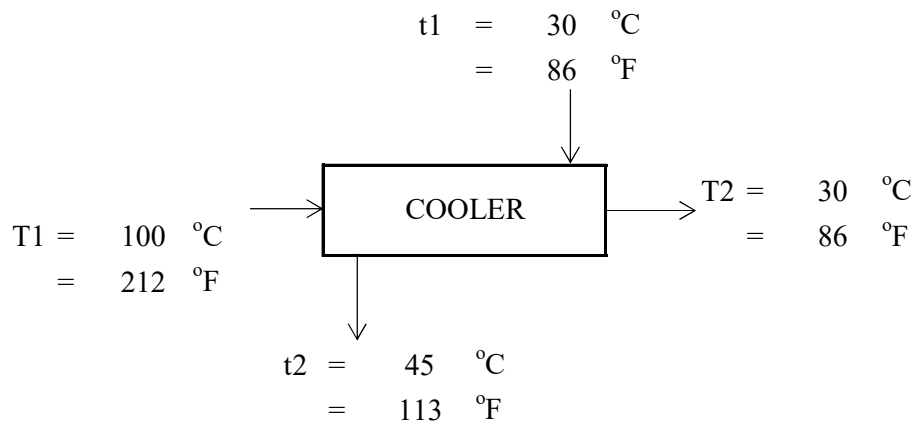
Rate Volumetrik	:	37.8532	ft ³ /s
Kecepatan Aliran	:	17939.9093	ft/s
Ukuran Pipa	:	NPS	= 1/2 in
	:	Sch. Number	= 40
	:	OD	= 0.0700 ft = 0.021336 m
	:	ID	= 0.0518 ft = 0.015799 m
	:	Flow Area	= 0.0021 ft ² = 0.0002 m ²
Power Pompa	:	1	Hp
Power Motor	:	2	Hp

31 COOLER EVAPORATOR (E-322)

Fungsi : Mendinginkan produk keluaran bawah evaporator sebelum masuk tangki dekolorisasi dari suhu 102,1353 °C menjadi 30 °C

Tipe : *Shell and Tube Heat Exchanger*

Jumlah : 1 buah



Perhitungan :

Komponen	Berat (kg/Jam)
C ₈ H ₁₈ O	5.07826
C ₂₄ H ₃₈ O ₄	3793.47214
H ₂ O	7.75837
C ₈ H ₄ O ₃	2.89070
C ₈ H ₆ O ₄	2.89940
Total	3812.09886



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned} \text{Berat Bahan Masuk} &= 3812.0989 \text{ kg/jam} \\ &= 8405.6780 \text{ lb/cuft} \end{aligned}$$

$$\begin{aligned} \text{Panas yang diserap (Q)} &= 117681.56 \text{ Kkal/Jam} \\ &= 467195.80 \text{ BTU/Jam} \end{aligned}$$

$$\begin{aligned} \text{Kebutuhan Cooling water} &= 3514.5312 \text{ Kg/Jam} \\ &= 7749.5414 \text{ Lb/Jam} \end{aligned}$$

a. Perhitungan ΔT LMTD (pers 5.14, Kern, 2010)

$$\Delta T \text{ LMTD} = \frac{\Delta t_2 - \Delta t_1}{\ln \frac{\Delta t_2}{\Delta t_1}} = \frac{99 - 0}{\ln \frac{99}{0}} = 21.5446 \text{ } ^\circ\text{F}$$

$$\Delta t_1 = 0 \text{ } ^\circ\text{F}$$

$$\Delta t_2 = 99 \text{ } ^\circ\text{F}$$

$$\begin{aligned} R &= \frac{T_1 - T_2}{t_2 - t_1} \\ &= 4.6667 \end{aligned}$$

$$\begin{aligned} S &= \frac{t_2 - t_1}{T_1 - t_1} \\ &= 0.214286 \end{aligned}$$

$$F_t = 0.86$$

(Kern, Fig. 18, Page. 828)

$$\begin{aligned} \Delta T \text{ LMTD} &= \Delta T \text{ LMTD} \times F_t \\ &= 18.5284 \text{ } ^\circ\text{F} \end{aligned}$$

$$\begin{aligned} \text{b. } T_c &= T_{\text{average bahan}} \\ &= 149 \text{ } ^\circ\text{F} \end{aligned}$$

$$\begin{aligned} t_c &= t_{\text{average air}} \\ &= 99.5 \text{ } ^\circ\text{F} \end{aligned}$$

Trial :

a. Asumsi U_D (Kern, 2010 Table.8 hal 840)

$$U_D = 75 - 150 \text{ Btu/jam.ft}^2.\text{ } ^\circ\text{F}$$

$$\text{dengan ketentuan : } P_{\text{allow}} = 5-10 \text{ psi}$$

$$\text{Dirt factor (} R_d \text{)} = 0.004$$

Sehingga, Trial U_D

$$U_D = 100 \text{ Btu/jam.ft}^2.\text{ } ^\circ\text{F}$$

$$\begin{aligned} A &= \frac{Q}{\Delta t \times U_D} \\ &= \frac{467195.8043 \text{ Btu/jam}}{99.00 \text{ } ^\circ\text{F} \times 100 \text{ Btu/jam.ft}^2.\text{ } ^\circ\text{F}} \\ &= 47.1915 \text{ ft}^2 \\ &= 566.2979 \text{ in}^2 \\ a'' &= 0.1963 \text{ ft}^2/\text{lin ft} \end{aligned} \quad \text{(Kern, 2010 Table.10)}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
 \text{Number of tubes } N_t &= \frac{A}{L \times a''} \\
 &= \frac{566.2979}{12 \times 0.1963} \\
 &= 240.4050 \\
 &= 240
 \end{aligned}$$

Berdasarkan Tabel 9 & 10 Kern Maka Dipilih

Tube	
OD	= 3/4 in
BWG	= 16
a't	= 0.302 in ²
a''	= 0.1963 surface/ lin ft, ft ²
Pitch	= 1 in square pitch
Nt	= 240
L	= 12
Passes	= 6
ID	= 0.62 in
Wall Thickness	= 0.065 in

Shell	
Shell ID	= 35 in
Passes	= 6
B space	= 7 in

$$\begin{aligned}
 A \text{ baru} &= N_t \times L \times a'' \\
 &= 566 \text{ ft}^2
 \end{aligned}$$

$$\begin{aligned}
 \text{Koreksi } U_D &= \frac{Q}{A \text{ baru} \times \Delta T_{LMTD}} \\
 &= 100
 \end{aligned}$$

Karena hasil U_D baru sama dengan U_D asumsi, maka asumsi dapat digunakan

antisipasi korosi : bahan di tube, dan air di shell

HOT FLUID: Tube (Minyak)	COLD FLUID : Shell (Air Pendingin)
(4) Flow area (a_t) =	(4') Flow area (a_s)
$ \begin{aligned} a_t &= \frac{N_t \times a't}{144 \times n} \\ &= \frac{240 \times 0.302}{144 \times 6} \\ &= 0.0840 \text{ ft}^2 \end{aligned} $	$ \begin{aligned} C'' &= P_t - OD_{\text{tube}} \\ &= 1 - 3/4 \\ &= 1/4 \text{ in} \\ B &= \frac{ID}{5} = \frac{35}{5} \\ &= 7.0 \text{ in} \\ a_s &= ID \times C'' \times B / 144 \times Pr \end{aligned} $



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

HOT FLUID: Tube (Minyak)	COLD FLUID : Shell (Air Pendingin)
	$= \frac{35 \times 0 \times 7.0}{144 \times 1}$ $= 0.4253 \text{ ft}^2$
(5) Mass Velocity, G_t $G_t = \frac{W}{at}$ $= \frac{8405.678}{0.0840}$ $= 100031.3451 \text{ lb/jam ft}^2$ Velocity = $\frac{G_t}{3600 \times \rho}$ $= \frac{100031.3451}{3600 \times \text{\#REF!}}$ $= \text{\#REF!} \text{ ft/s}$	(5') Mass Velocity, G_s $G_s = \frac{W}{as}$ $= \frac{7749.541}{0.42535}$ $= 18219.33 \text{ lb/jam ft}^2$
(6) Pada $T_c = 149 \text{ }^\circ\text{F}$ $\mu = 7.5 \text{ cps (fig 14)}$ $= 18.15 \text{ lbm/jam.ft}$ $D_i = \frac{0.62}{12} \text{ in}$ $= 0.0517 \text{ ft}$ $R_{et} = \frac{D \times G_t}{\mu}$ $= \frac{0.0517 \times 100031.35}{18.15}$ $= 284.7540586$	(6') Pada $t_c = 100 \text{ }^\circ\text{F}$ $\mu = 0.7 \text{ cps (fig 14)}$ $= 1.694 \text{ lbm/jam.ft}$ $D_e = \frac{0.95}{12} \text{ (fig.28)}$ $= 0.079167 \text{ ft}$ $R_{es} = \frac{D_e \times G_s}{\mu}$ $= \frac{0.079 \times 18219.33}{1.694}$ $= 851.4543$
(7) $j_H = 3 \text{ (fig.24)}$	(7') $j_H = 40 \text{ (fig.28)}$
	(8') Pada $t_c = 100 \text{ }^\circ\text{F}$ $k = 0.381 \text{ Btu/hr.ft}^2(\text{ }^\circ\text{F/ft})$ <i>(Kern, T.4, hh:800)</i> $C_p = 1 \text{ Btu/lb.}^\circ\text{F}$ <i>(Kern, F.2, hh:804)</i> $\left(\frac{C_p \cdot \mu}{k} \right)^{1/2} = 2.1086$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

HOT FLUID: Tube (Minyak)	COLD FLUID : Shell (Air Pendingin)
<p>(9) h_i :</p> $h_i = 460 \text{ Btu/hr.ft}^2.\text{°F}$ <p>(Kern; fig. 25)</p>	<p>(9') $\frac{h_o}{\phi_s} = j_H \frac{k}{De} \left(\frac{c\mu}{k} \right)^{1/2}$</p> $= 405.9166$
<p>(10) t_w^*</p> $t_w = t_c + \frac{h_i}{h_i + h_o} (T_c - t_c)$ $= 149 + \frac{460}{865.9166} 50$ $= 175.296 \text{ °F}$	<p>(11') Pada $t_w = 175.296 \text{ °F}$</p> $\mu_w = 0.45 \text{ cps}$ $= 1.089 \text{ lbm/jam.ft}$ $\phi_s = (\mu/\mu_w)^{0.14}$ $= 1.06381 \text{ Btu/hr.ft}^2.\text{°F}$
<p>(11) h_{io} (eq. 6.5)</p> $h_{io} = h_i \times ID/OD$ $= 380.3 \text{ Btu/hr.ft}^2.\text{°F}$	<p>(12') Corrected coefficient</p> $h_o = \frac{h_o \times \phi_s}{\phi_s}$ $= 431.8181$
<p>13. Clean overall coefficient, U_c :</p> $U_c = \frac{h_{io} \times h_o}{h_{io} + h_o}$ $= \frac{460 \times 431.8181}{460 + 431.8181}$ $= 222.7319 \text{ Btu/hr ft}^2 \text{ °F}$	
<p>14 Dirt factor, R_d :</p> $R_d = \frac{U_c - U_D}{U_c \times U_D}$ $= \frac{222.732 - 100}{222.732 \times 100}$ $= 0.0055$ <p>Rd perhitungan > Rd data : 0.0055 > 0.004 (memenuhi) (Kern, T.12, hh:845)</p>	



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

HOT FLUID: Tube	COLD FLUID : Shell
(Bahan)	(Air Pendingin)
Pressure Drop	
a. Untuk R_{ct} : 284.7540586 $f = 0.001 \text{ ft}^2/\text{in}^2$ <i>(Kern, Fig 26)</i> $s = 0.55$ <i>(Kern, Fig 6)</i>	a'. Untuk R_{cs} : 851.4543 $f = 0.0025 \text{ ft}^2/\text{in}^2$ <i>(Kern, Fig 29)</i> $s = 1$ <i>(Kern, Table 6)</i>
b. $\Delta P_t = \frac{1}{2} \times \frac{f \times G_t^2 \times L \times n}{5,22 \times 10^{10} D \cdot s \cdot \phi s}$ $= 0.2283$	b'. No of crosses $N + 1 = \frac{12L}{B}$ $= 21$
c. $G_t = 100031.345 \text{ lb/jam ft}^2$ $\frac{V^2}{2g'} = 0.001$ <i>(Kern, fig, 27)</i> $\Delta P_r = \frac{4n}{s} \times \frac{V^2}{2g'}$ $= \frac{24}{0.55} \times 0.001$ $= 0.043636$	c'. $\Delta P_s = \frac{f \times G_s^2 \times De \times (N + 1)}{5,22 \times 10^{10} \times De \times s \times \phi s}$ $= 0.000307 \text{ psi}$
d. $\Delta P_T = \Delta P_t + \Delta P_r$ $= 0.2719 \text{ psi}$ $\Delta P_T < 10 \text{ psi}$ (memenuhi)	$\Delta P_S < \Delta P \text{ allowable}$ (memenuhi)

SPEKIFIKASI COOLER EVAPORATOR

Fungsi : Mendinginkan produk keluaran bawah evaporator sebelum masuk tangki dekolrisasi dari suhu 102,1353 °C menjadi 30 °C

Tipe : 1 - 2 Shell and Tube Heat Exchanger (Fixed Tube)

Inside Diameter = 0.62 in
 Outside Diameter = 3/4 in
 Panjang = 12 ft
 Pitch = 1 in square pitch
 Jumlah Tube = 240 buah

Shell

Inside Diameter = 35 in
 Baffle space = 7 in
 Jumlah cooler = 1



PERANCANGAN PABRIK

Pabrik Diocetyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

32 CONDENSOR (E-323)

Fungsi = Mengembunkan uap formic acid

Type = Vertikal kondenser shell&tube

Komposisi vapor:

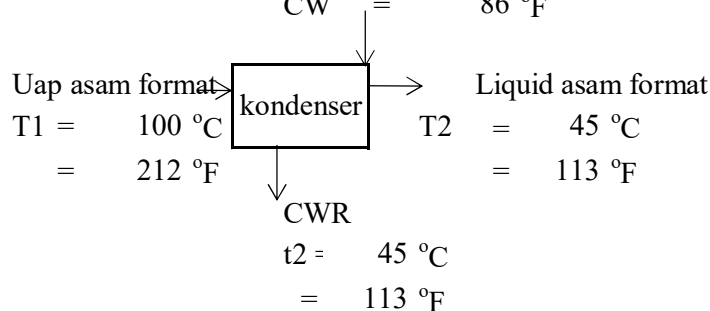
Komponen	Berat (kg)	Fraksi berat	ρ (gr/cc)
H ₂ O	1562.0562	1.0000	1
	1562.0562	1.0000	

$$\begin{aligned} \text{Densitas Campuran} &= \frac{1}{\frac{\text{fraksi berat}}{\rho \text{ komponen}}} \times 62.43 \\ &= \frac{1}{\frac{1.0}{1}} \times 62 \\ &= 1.00 \text{ gr/cc} \times 62 \\ &= 62.43 \text{ lb/cuft} \end{aligned}$$

$$\begin{aligned} \text{Rate Volumetrik} &= \frac{\text{Rate Massa}}{\text{Densitas}} \\ &= \frac{3444.33}{62.4300} \\ &= 55.1711 \text{ cuft/jam} \end{aligned}$$

Diagram suhu : $t_1 = 30 \text{ }^\circ\text{C}$

CW = $86 \text{ }^\circ\text{F}$



1. Heat Balance

Data dari Appendix A dan Appendix B diperoleh :

$$\begin{aligned} \text{Rate massa, W massa} &= 1,562 \text{ kg/jam} \\ &= 3,444 \text{ lb/jam} \\ \text{Q air pendingin} &= \text{btu/jam} \\ \text{M air pendingin} &= 3,515 \text{ kg/jam} \\ &= 7,750 \text{ lb/jam} \\ \text{Cp air pendingin} &= 1 \text{ kkal/kg.C} \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

2. Log Mean Temperature Difference

$$\text{Suhu CW masuk} = 30 \text{ }^{\circ}\text{C} = 86 \text{ }^{\circ}\text{F} (t_1)$$

$$\text{Suhu CWR} = 45 \text{ }^{\circ}\text{C} = 113 \text{ }^{\circ}\text{F} (t_2)$$

$$\text{Suhu bahan masuk} = 100 \text{ }^{\circ}\text{C} = 212 \text{ }^{\circ}\text{F} (T_1)$$

$$\text{Suhu bahan keluar} = 45 \text{ }^{\circ}\text{C} = 113 \text{ }^{\circ}\text{F} (T_2)$$

$$\begin{aligned} \Delta T \text{ LMTD} &= \frac{99 - 27}{\ln 4} \\ &= 55.4152 \text{ }^{\circ}\text{F} \end{aligned}$$

$$R = \frac{T_1 - T_2}{t_2 - t_1} = \frac{99}{27} = #####$$

$$S = \frac{t_2 - t_1}{T_1 - t_1} = \frac{27}{126} = 0.214$$

Dengan nilai R dan S tersebut, didapatkan Design Shell & Tube:

Untuk 2-4 Shell & Tube $F = 0.95$ [Kern; fig.19] 1 1'2

$$\begin{aligned} \Delta T &= F_T \times \Delta T \text{ LMTD} \\ &= 0.95 \times 55.4152 \\ &= 52.6444 \text{ }^{\circ}\text{F} \end{aligned}$$

3. Tc dan tc

$T_c = T_{av}$ vapor

$$T_c = \frac{T_1 + T_2}{2} = \frac{212 + 113}{2} = 163 \text{ }^{\circ}\text{F}$$

$t_c = t_{av}$ water

$$t_c = \frac{t_1 + t_2}{2} = \frac{86 + 113}{2} = 100 \text{ }^{\circ}\text{F}$$

$$\begin{aligned} \text{Sg bahan} &= \frac{\rho \text{ bahan}}{\rho \text{ reference}} \times \text{Sg reference} \\ &= \frac{62.4300 \text{ lb/c}}{62.43 \text{ lb/c}} \times 1 \\ &= 1.0000 \end{aligned}$$

μ berdasarkan Sg bahan :

[Kern; tab.6 p. 808] didapat Sg reference = 1

[Kern; fig.14 , p. 823] didapat μ reference = 0.89 Cps

$$\begin{aligned} \mu \text{ bahan} &= \frac{\text{Sg bahan}}{\text{Sg reference}} \times \mu \text{ reference} \\ &= \frac{1.0000}{1.0000} \times 0.89 \\ &= 0.8900 \text{ Cps} \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Komposisi kondenser

Hot Fluid : Uap formic ($> 1 \text{ cps, heavy organic}$)

Cold Fluid : Air pendingin

Sehingga di dapat nilai $U_D = 75 - 150 \text{ Btu/jam.ft}^2\text{ }^\circ\text{F}$
dengan range *[Kern; tab.8, p.840]*

Dipilih nilai $U_D = 150 \text{ Btu/jam.ft}^2\text{ }^\circ\text{F}$

$$A = \frac{Q}{UD \times \Delta T}$$

Dimana,

A = Luas Perpindahan Panas

Q = Q supply dari steam

UD = Overall Design Coefficients

ΔT = Perubahan Suhu

$$\begin{aligned} A &= \frac{Q}{UD \times \Delta T} \\ &= \frac{0.0000}{150 \times 52.6444} \\ &= 0.0000 \text{ ft}^2 \\ &= 0.0000 \text{ m}^2 \end{aligned}$$

Dikarenakan luas perpindahan panas $> 200 \text{ ft}^2$, maka design kondenser menggunakan shell an *[Walas; p.xvi]*

$$\begin{aligned} \text{jumlah tube } (n) &= \frac{A}{L \times a''} \\ &= \frac{0.0000}{19 \times 0.327} \\ &= 0 \text{ buah} \end{aligned}$$

Dalam perencanaan ini digunakan Condenser dengan spesifikasi :

Digunakan TUBE dengan *[Kern; tab. 10, p. 843]*

OD, BWG = 1 1/4 in , 16 BWG

Pitch = 1 3/4 in triangular *[Kern; tab.9]*

Panjang tube, L = 19 ft

Passes = 1

ID Tube = 1.120 in 0.0933 ft

$a't$ = 0.985 in²

a'' = 0.3271 ft²/lin ft

Digunakan Nt = 235 buah



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

Dari tube passes dan jumlah tube didapat SHELL:

Untuk pipa 1 1/4 in, OD tubes on 1 3/4 in triangular pitch

$$\text{ID shell} = 29 \text{ in} \quad [Kern; \text{tab.9}]$$

$$\text{Baffle} = 20 \text{ in}$$

$$\text{Passes (n)} = 2$$

*Koreksi U_D

$$A = N_t \times L \times a''$$

$$= 1460.502 \text{ ft}^2$$

$$U_D \text{ koreksi} = \frac{Q}{A \times \Delta T}$$

$$= \frac{0.000}{1461 \times 52.6}$$

$$= 0 \text{ Btu/jam.ft}^2 \text{ } ^\circ\text{F}$$

Nilai U_D asumsi sama dengan U_D koreksi (memenuhi)

Fluida Panas (Shell Side) Water	Fluida Dingin : (Tube side) Vapor
(4') Flow area (a_s) $C' = \text{Pitch} - \text{OD} = \text{## in}$ $a_s = \frac{\text{IDshell} \times C' \times E}{144 \times P_T}$ $= \frac{29 \times 1/2 \times 20}{144 \times 1 \ 3/4}$ $= 1.1508 \text{ ft}^2$	(1) Flow area (a'_t) $a'_t = \frac{N_t \times a'_t}{144 \times n}$ $= \frac{235 \times 0.99}{144 \times 1}$ $= 1.6075 \text{ ft}^2$
(5') Kecepatan massa (Gs) $G_s = \frac{W}{a_s}$ $= \frac{7,749.54 \text{ lb/jam}}{1.1508}$ $= 6,734 \text{ lb/jam ft}^2$ $De = \text{#####} \quad [Kern; \text{fig 28}]$ Pada $T_c = 163 \text{ } ^\circ\text{F}$, didapat μ $\mu = \text{##} \times 2.42 = 1 \text{ lb/ft.hr}$ $NR_e = \frac{D_e \times G_s}{\mu}$ $= \frac{469}{1}$	(2) Kecepatan massa (Gt) $G_t = \frac{W}{a_t}$ $= \frac{3,444.334}{1.6075}$ $= 2,142.71 \text{ lb/jam ft}^2$ Pada $t_c = 100 \text{ } ^\circ\text{F}$, didapat μ $\mu = \text{##} \times 2.42 = 2 \text{ lb/ft.hr}$ $NR = \frac{G_t \times \text{ID}}{\mu}$ $= \frac{2,143 \times 0.09}{2.15}$ $= 93$
(3) Asumsi $h_o = 400$ $\text{Btu/jam.ft}^2 \text{ } ^\circ\text{F}$	(3') Menghitung h_i



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Fluida Panas (Shell Side) Water	Fluida Dingin : (Tube side) Vapor
$t_w = t_c + \frac{h_o}{h_o + h_{ic}}(T_c - t_c)$ $= 135.3 \text{ } ^\circ\text{F}$ $t_f = \frac{T_c + t_w}{2}$ $= \frac{163 + 135.3}{2}$ $= 148.9 \text{ } ^\circ\text{F}$ <p>Dari nilai t_f tersebut, diperoleh:</p> $k_f = 0.381 \text{ Btu/jam.ft}^2 \text{ } ^\circ\text{F/ft}$ $s_f = 1 \quad [\text{Kern; Tab. 6}]$ $\mu_f = 0.6 \text{ cp} \quad [\text{Kern; Fig.14}]$ <p>Dipilih vertical condenser</p> $G' = \frac{W}{3.14 \times N_t \times D}$ $= \frac{7,749.5414}{3.14 \times 235 \times 0.09}$ $= 112.5231 \text{ lb/jam lin ft}$ <p>Dari [Kern; Fig 12-9, p.267] di dapat :</p> $h_o = 410 \text{ Btu/jam.ft}^2 \text{ } ^\circ\text{F}$ <p>Karena nilai h_o tersebut berbeda dengan h_o trial, jika berbeda tidak boleh lebih dari 20%. Jika lebih dari 20% maka ulangi trial h_o.</p>	$v = \frac{G_t}{3600 \rho}$ $= \frac{2,143}{3600 \times 62.43}$ $= 0.010 \text{ ft/s}$ <p>Dari [Kern; Fig. 25] Dari nilai t_c dan v, diperoleh h_i:</p> $h_i = 340 \text{ Btu/jam.ft}^2 \text{ } ^\circ\text{F}$ $h_{io} = h_i \times \frac{ID}{OD}$ $= 340 \times \frac{1.12}{1 \frac{1}{4}}$ $= 305 \text{ Btu/jam.ft}^2 \text{ } ^\circ\text{F}$
<p>(12) Clean Overall coefficient, U_c :</p> $U_c = \frac{h_{io} \times h_o}{h_{io} + h_c}$ $= 174.7767 \text{ Btu/hr.ft}^2 \text{ } ^\circ\text{F}$ <p>(13) Dirt Factor, R_d :</p> $R_d \text{ hitu} = \frac{U_c - U_D}{U_c \times U_D}$ $= \#DIV/0! \text{ Btu/hr.ft}^2 \text{ } ^\circ\text{F}$ <p>R_d perhitungan > R_d data $\#DIV/0! > 0.001$ [Kern; T-12, p.845] (Condenser) R_d Hitung > R_d Ketentuan , Alat dapat digunakan</p>	



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Pressure Drop	
Bagian shell (water)	Bagian tube (vapor)
$NR = 469$ $f = 0.0028 \text{ ft}^2/\text{in}^2$ <i>[Kern; fig.29 p.839]</i> $N+ = 12L/B$ $= 11.4 \text{ in}$ $s = 1.000$ $Ds = 2 \text{ ft}$ <i>[Kern; Eq. 12-47 p.273]</i> $\Delta Ps = \frac{1}{2} \times \frac{f \cdot Gs^2 \cdot Ds \cdot (N+1)}{5.22 \times 10^{10} \times De \times s}$ $= 1 \times 0.000884$ $= 0.000 \text{ Psi}$	$NR = 93$ $f = 4E-04 \text{ ft}^2/\text{in}^2$ <i>[Kern; fig.26 p.839]</i> <i>[Kern; eq.7-45 p.148]</i> $\Delta Pt = \frac{f \cdot Gt^2 \cdot L \cdot n}{5.22 \times 10^{10} \times D \times s \times \phi t}$ $= 0.00001 \text{ Psi}$ $\Delta Pt = \frac{4 n}{s} \times \frac{V^2}{2 G'}$ Dari <i>[Kern; Fig.27 p.837]</i> $Gt = 2,142.711 \text{ lb/jam.ft}^2$ Didapat harga $\frac{V^2}{2 G} = 0.002$ $s = 1.000$ $\Delta Pr = \frac{4 n}{s} \times \frac{V^2}{2 G'}$ $\Delta Pr = \frac{4}{1.000} \times 0.002$ $= 0.00960 \text{ Psi}$ $\Delta P_T \text{ Total} = \Delta Pt + \Delta Pr$ $= 0.000 + 0.010$ $= 0.0096 \text{ psi}$

Jadi $P_i = \Delta Ps + \Delta Pt \text{ total}$
 $= 0.0004 + 0.0096$
 $= 0.0101 \text{ Psi}$

a. Menghitung diameter kondenser

Untuk counter-current Condenser dengan rate 1562.056 kg uap/jam
 Diperoleh kondenser dengan $H = 3 \text{ m}$ *[Hugot ed.3; tab 40.2]*

Asumsi: $H = 2 D$
 $D = 1.5 \text{ m}$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Spesifikasi :

Type Condens 2-4 Shell & Tube Exchanger

Jumlah = 2 buah (+ Cooler)

Bagian shell :

Jenis Bahan = Stainless Steel 302

Diameter kond = 1.5 m

Tinggi kondens = 3 m

ID Shell = 29 in

Baffle Space = 20 in

Passes = 2

Bagian tube :

Jumlah & Panj = 235 buah 19 ft

OD, BWG, pit = 1.25 in, 16 BWG 1 3/4 triangular

Passes = 1

33 TANGKI DEKOLORISASI (D-330)

Fungsi : Tempat karbon aktif melakukan proses penyerapan warna *Dioctyl Phthalate* menjadi bening

Tipe : Tangki silinder vertikal dengan alas dan tutup berbentuk *torispherical* serta pengaduk jenis *propeller*

Jumlah : 1 buah

Kondisi O_T : $T = 30\text{ }^\circ\text{C} = 303.15\text{ K}$

$P = 1\text{ atm} = 760\text{ mmHg} = 14.7\text{ psi}$

Menentukan Dimensi Tangki

Komponen	Laju Massa	%Berat	ρ (g/ml)	ρ_{Campuran}	μ (cP)	μ_{campuran}
	(kg/jam)			(kg/m^3)		(cP)
$\text{C}_{24}\text{H}_{38}\text{O}_4$	3793.4721	0.99462	0.9840	1.0108	52.6447	0.0189
$\text{C}_8\text{H}_{18}\text{O}$	5.0783	0.00133	0.8330	2E-03	5.0819	0.0003
$\text{C}_8\text{H}_4\text{O}_3$	2.8906993	0.00076	1.2000	6E-04	3.9880	0.0002
$\text{C}_8\text{H}_6\text{O}_4$	2.8993975	0.00076	1.5260	5E-04	155.97	0.0000
H_2O	7.7584	0.00203	0.9970	2E-03	0.8150	0.0025
Karbon	1.9060	0.00050	460.00	1E-06	0.0000	0.00
Total	3814.0049	1.0000	465.54	1.0156	218.497	0.02

$$\rho_{\text{campuran}} = \frac{\sum x_i}{\sum x_i / \rho} = \frac{1.000}{1.016} = 0.985\text{ kg/m}^3$$

$$= 0.061\text{ lb/ft}^3$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}\mu_{\text{campuran}} &= \frac{\sum x_i}{\sum x_i/\mu} = \frac{1.0000}{218.4966} = 0.005 \text{ cP} \\ &= 0.000003 \text{ lb/ft.s}\end{aligned}$$

Menentukan Volume Cairan dalam *mixer* (V_L) dan volume *mixer* (V_s)

Digunakan persamaan :

$$V_L = m / \rho_{\text{mix}}$$

dimana :

$$\begin{aligned}m &= 3814.00 \text{ kg/jam} \\ &= 8408.43 \text{ lb/jam}\end{aligned}$$

maka :

$$\begin{aligned}V_L &= \frac{8408.432 \text{ lb/jam} \times 1 \text{ jam}}{0.061 \text{ lb/ft}^3} \\ &= 136781.3 \text{ ft}^3\end{aligned}$$

$$\begin{aligned}V_s &= (1 + 0) \times V_L \\ &= 1.2 \times 136781.3 \text{ ft}^3 \\ &= 164137.5 \text{ ft}^3 \\ &= 1227913 \text{ galon}\end{aligned}$$

Menentukan Diameter dan Tinggi *Shell*

Dipilih tangki silinder dengan rasio $H_s/ID = 1.5$
Eq 3.10. (Brownell & Young, 1959:43)

$$H_s/ID = 2$$

$$H_s = 2 ID$$

$$\begin{aligned}\text{Volume head bentuk} &= 0,000049 ID^3 \\ &\text{(Brownell \& Young, 1959:88)}\end{aligned}$$

Maka,

$$V_s = \text{Volume silinder} + 2 \text{ volume head}$$

$$V_s = (1/4 \times \pi \times ID^2 + (2 \times 0,000049 ID^3))$$

$$164138 = (1/4 \times (3,14 \times ID^2) + (2 \times 0,000049 ID^3))$$

$$164138 = 1.178 ID + 2 \times 5E-05 ID^3$$

$$ID^3 = 139383.3$$

$$ID = 51.8486 \text{ ft}$$

$$ID = 15.8035 \text{ m}$$

$$ID = 622.1818 \text{ in}$$

Sehingga diameter *shell* (D_s) adalah sebesar = 15.8035 m

Maka tinggi *shell* :

$$H_s = 1.5 \times ID$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned} &= 1.5 \times 51.8486 \text{ ft} \\ &= 77.7729 \text{ ft} \\ &= 23.7055 \text{ m} \end{aligned}$$

Jadi diameter dan tinggi *shell* sebesar = 15.8035 m dan 23.7055 m

Menentukan Tinggi Larutan dalam Tangki

$$\begin{aligned} V_{\text{larutan}} &= \left(\frac{\pi}{4} \times Di^2 \right) \times H_{\text{liquid}} \\ H_{\text{liquid}} &= \frac{V_{\text{larutan}}}{\left(\frac{\pi}{4} \times Di^2 \right)} \\ &= \frac{136781.2825 \text{ ft}^3}{\frac{3.14}{4} \times (51.85)^2 \text{ ft}^2} \\ &= 64.8161 \text{ ft} \\ &= 19.7560 \text{ m} \end{aligned}$$

Menentukan Tebal *Shell*

$$t_s = \frac{P_d \cdot r}{f \cdot E - 0,6 P} + C \quad \text{Eq. 13.1. (Brownell \& Young, 1959:254)}$$

Dimana :

P_d = Tekanan desain

dengan :

$$P_d = P_{\text{operasi}} + P_{\text{hidrostatik}}$$

$$P_{\text{op}} = 1 \text{ atm} = 14.7 \text{ psi}$$

$$P_{\text{hid}} = \rho \times \frac{g}{gc} \times h$$

Dimana :

$$h = \text{tinggi cairan dalam} = 64.8161 \text{ ft}$$

$$\rho = \text{densitas cairan} = 0.0615 \text{ lb/ft}^3$$

maka,

$$\begin{aligned} P_{\text{hid}} &= 0.061 \text{ lbm/ft}^3 \times 1 \text{ lbm/lbf} \times 64.82 \text{ ft} \\ &= 3.984 \text{ lbf/ft}^2 \\ &= 0.028 \text{ psi} \end{aligned}$$

$$\begin{aligned} P_d &= P_{\text{op}} + P_{\text{hid}} \\ &= 14.7 \text{ psi} + 0.0277 \text{ psi} \\ &= 14.7277 \text{ psi} \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

dengan faktor keamanan 20%, maka :

$$P_d = 1.2 \times 14.7277 \text{ psi}$$

$$= 17.6732 \text{ psi}$$

$$r = \text{jari-jari shell } (D_s/2) = 311.0909 \text{ in}$$

$$f = \text{Allowable stress untuk Carbon Steel SA-283 Grade C} \\ \text{sebesar} = 12650 \text{ psi} \quad (\text{Brownell \& Young, 1959:251})$$

$$E = \text{Efisiensi pengelasan, diambil double-welded butt joint, sebesar 80\%}$$

$$c = \text{Faktor korosi sebesar 0,125}$$

Maka,

$$t_s = \frac{17.6732 \times 311.0909}{(12650) \times (0,8) - (0,6) \times 17.6732} + 0.125$$

$$t_s = 0.6688 \text{ in}$$

Digunakan tebal *shell* standar = 3/16 in

Tabel 5.7. (Brownell & Young, 1959:89)

Menentukan Diameter dan Tinggi yang Distandarkan

Diameter (D_s)

$$OD = ID + 2 t_s$$

Dimana :

ID = diameter *shell* (D_s)

Sehingga :

$$OD = 622.2 \text{ in} + 2 \times 0.1875 \text{ in}$$

$$= 622.6 \text{ in}$$

OD standar dapat dicari dari **Tabel 5.7.** (Brownell & Young, 1959:90), sehingga,

$$OD = 60 \text{ in} = 1.524 \text{ m}$$

$$ID = OD_{\text{standar}} - 2 t_s$$

$$= 60 \text{ in} - 2 \times 0.1875 \text{ in}$$

$$= 59.63 \text{ in}$$

$$= 1.51 \text{ m}$$

$$\text{Diameter dalam shell standar} = 59.6250 \text{ in}$$

$$= 1.5145 \text{ m}$$

$$= 4.9687 \text{ ft}$$

Tinggi (H_s)

$$H_s = 1.5 \times 59.6250$$

$$= 89.4375 \text{ in}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$= 2.2717 \text{ m}$$

Tinggi *shell* standar ialah = 2.2717 m

Menentukan Tebal dan Tinggi *Head*

Tebal *Head*

Tebal tutup berupa *torispherical* mengikuti persamaan 7.57.

(Brownell & Young, 1959:133) :

$$th = \frac{Pd \times r_i \times w}{2(f \cdot E - 0,1 P_d)} + c$$

Dimana :

$$w = 1/4 \left[3 + \frac{r}{icr} \right] + 0.5$$

$$r = 60 \text{ Tabel 5.7. (Brownell \& Young, 1959:90)}$$

$$icr = 3 \frac{5}{8} \text{ Tabel 5.7. (Brownell \& Young, 1959:90)}$$

$$w = 1/4 \left[3 + \frac{60}{3 \frac{5}{8}} \right] + 0.5$$
$$= 1.1054$$

$$Pd = 17.6732 \text{ psi}$$

$$r = \text{jari-jari shell } (D_s/2) = 311.0909 \text{ in}$$

$$f = \text{Allowable stress untuk Carbon Steel SA-283 grade C}$$
$$\text{sebesar} = 12650 \text{ psi (Brownell \& Young, 1959:251)}$$

$$E = \text{Efisiensi pengelasan, diambil double-welded butt joint, sebesar 80\%}$$

$$c = \text{Faktor korosi sebesar 0,125}$$

Maka :

$$th = \frac{17.67 \text{ psi} \times 311.1 \text{ in} \times 1.1054}{2[12650 \times 1 - 0.1 \times 17.67 \text{ psi}]} + 0.125$$
$$= 0.4253 \text{ in}$$

Digunakan tebal *head* = 3/16 in

Tabel 5.7. (Brownell & Young, 1959:89)

Tabel 5.7. (Brownell & Young, 1959:90)

$$OD = 60 \text{ in}$$

$$r = 60 \text{ in}$$

$$icr = 3 \frac{5}{8} \text{ in}$$

Sehingga untuk dimensi tutup atas dan bawah dapat dihitung sebagai berikut

$$a = 0.5 \times ID = 29.81 \text{ in}$$

$$AB = a - icr$$

$$= 29.81 \text{ in} - 3 \frac{5}{8} \text{ in}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}
 &= 26.19 \text{ in} \\
 BC &= r - icr \\
 &= 60 - 3 \frac{5}{8} \\
 &= 56.38 \text{ in} \\
 AC &= (BC^2 - AB^2)^{1/2} \\
 &= 49.92 \text{ in} \\
 b &= r - (BC^2 - AB^2)^{1/2} \\
 &= 10.08 \text{ in}
 \end{aligned}$$

Tinggi *straight flange* (sf)

Untuk tebal *head* = 3/16 in **Tabel 5.7.** (Brownell & Young, 1959:89), nilai sf 1 1/2-2 diambil :

$$sf = 1 \frac{1}{2} \text{ in}$$

Dari data di atas dapat dihitung tinggi tutup tangki (OA) yaitu sebesar :

$$\begin{aligned}
 OA &= b + sf + t \\
 &= 10.08 + 1 \frac{1}{2} + 0.1875 \\
 &= 11.76 \text{ in} \\
 &= 0.299 \text{ m}
 \end{aligned}$$

Karena tutup atas dan bawah sama maka tutup bawah juga mempunyai tinggi sebesar = 11.76 in atau 0.2995 m

Sehingga didapat tinggi total *shell* (H_V)

$$\begin{aligned}
 H_{V(\text{total})} &= H_S + 2 \times \text{tinggi head} \\
 &= 2.272 + 0.5990 \\
 &= 2.871 \text{ m}
 \end{aligned}$$

Menentukan Dimensi Pengaduk

Menentukan Jenis Pengaduk dan Jumlah *Impeller*

Dipilih jenis pengaduk *three blade propeller* dengan alasan

Untuk menentukan jumlah pengaduk ditentukan berdasarkan data berikut (Walas, 1965:288) :

<i>Viscosity</i>	<i>Maximum level</i>	<i>Number of</i>
[cP (Pa sec)]	h/Dt	<i>Impellers</i>
<25.000 (<25)	1.4	1
<25.000 (<25)	2.1	2
>25.000 (>25)	0.8	1
>25.000 (>25)	1.6	2

$$\begin{aligned}
 \text{Maksimu} &= \frac{H_L}{\text{m level}} \\
 &= \frac{64.82}{4.969} \text{ ft} \\
 &= 13.04
 \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

Jadi jumlah pengaduk (*impeller*) untuk h/Dt sebesar = 2 buah

Menentukan Dimensi Pengaduk

Tabel 3.4-1. (Geankoplis, 1993:144) didapat data berikut :

$$Da/Dt = 0,3 - 0,5 \text{ (diambil } 0,3)$$

$$W/Da = 1/5$$

$$L/Da = 1/4$$

$$C/Dt = 1/3$$

$$J/Dt = 1/12$$

Dimana :

Da = diameter *impeller*

C = jarak pengaduk dari dasar tangki

Dt = diameter luar tangki = OD

L = panjang *propeller*

W = tinggi *propeller*

Sehingga dimensi pengaduk dapat dihitung sebagai berikut :

Diameter *impeller* (Da)

$$Da = 0.3 \times OD$$

$$Da = 0.3 \times 66$$

$$Da = 19.8 \text{ in}$$

$$= 1.65 \text{ ft}$$

$$= 0.50 \text{ m}$$

Jarak pengaduk dari dasar tangki (C)

$$C = 1/3 \times OD$$

$$C = 1/3 \times 66$$

$$C = 22 \text{ in}$$

$$= 1.83 \text{ ft}$$

$$= 0.56 \text{ m}$$

Panjang *flat* (L)

$$L = 1/4 \times Da$$

$$L = 1/4 \times 19.8$$

$$L = 4.95 \text{ in}$$

$$= 0.41 \text{ ft}$$

$$= 0.13 \text{ m}$$

Tinggi *flat* (W)

$$W = 1/5 \times Da$$

$$W = 1/5 \times 19.8$$

$$W = 3.96 \text{ in}$$

$$= 0.33 \text{ ft}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$= 0.10 \text{ m}$$

Menentukan *Power Motor*

a. Menentukan Kecepatan Putar Pengaduk

$$N = \frac{v}{p \times Da} \quad (\text{Joshi, hal 389})$$

Dimana :

Da = diameter *impeller*

N = putaran putar pengaduk (rpm)

v = kecepatan linier untuk = 300 mpm

Sehingga,

$$N = \text{##### rpm} = 3.1662 \text{ rps}$$

Diambil kecepatan pengaduk standar (komersial) 190 rpm (Walas, 1965:288)

$$N = 190 \text{ rpm} = 3.1667 \text{ rps}$$

Menentukan Bilangan Reynolds (N_{Re})

Dengan menggunakan persamaan :

$$\begin{aligned} N_{Re} &= \frac{r \times Da^2 \times N}{m} \\ &= \frac{0.061 \times (1.650)^2 \times 3.167}{0.0000030709860} \end{aligned}$$

$$N_{Re} = 172576.1$$

Menentukan *Power Pengaduk*

Karena tangki yang digunakan tanpa *baffle* maka perhitungan *power* sebagai berikut :

$$\begin{aligned} N_{FR} &= \frac{n^2 \times Da}{g} \\ &= \frac{3.167^2 \times 1.650}{32.1740} \\ &= 0.5143 \end{aligned}$$

Dari **Tabel 9.1.** (Mc Cabe et al., 1993:252) didapatkan nilai konstanta a dan

$$a = 1$$

$$b = 40$$

maka :

$$\begin{aligned} m &= \frac{1 - \log_{10} N_{Re}}{40} \\ &= -0.1059 \quad \text{Eq. 9.19 (Mc Cabe et al., 1993:251)} \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

N_p = *power number*, dari Grafik 9.12 didapatkan sebesar 1

maka :

$$\begin{aligned} N_{p \text{ correct}} &= N_p \times NFR^m \\ &= 1 \times [0.514]^{-0.106} \\ &= 1.073 \end{aligned}$$

$$\begin{aligned} P &= \frac{p \times \rho \times N^3 \times D}{gc} \\ &= \frac{1.073 \times 0.061 \times 3.1667^3 \times 1.650^5}{32.1740} \\ &= 0.796 \text{ ft/lbf.s} \\ &= 0.001 \text{ hp} \end{aligned}$$

Fig. 14.38. (Peters & Timmerhaus, 1991:521) didapat efisiensi motor penggerak sebesar 80% maka daya penggerak motornya sebesar :

$$\begin{aligned} \text{power mot} &= P/\eta \\ &= \frac{0.001}{80\%} \\ &= 0.002 \text{ hp} \end{aligned}$$

Maka digunakan *power* motor = 0 hp

SPEKIFIKASI TANGKI DEKOLORISASI

Nama Alat : Tangki Dekolorisasi

Kode : M-330

Fungsi : Tempat karbon aktif melakukan proses penyerapan warna *dioctyl phthalate* menjadi bening

Tipe : Tangki berupa silinder vertikal dengan alas dan tutup berbentuk *torispherical* serta pengaduk jenis *propeller*

Bahan Kor : *Carbon Steel SA-283 Grade C*

Jumlah : 1 buah

Kondisi Operasi

Tekanan : 1 atm

Suhu : 30 °C

Kapasitas : 164137.5 ft³

Tekanan D : 14.73 psi

Dimensi

Silinder

Diameter Dalam : 622.2 in = 15.8034 m

Diameter Luar : 60 in = 1.5240 m



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Tebal : 3/16 in = 0.0048 m

Tinggi : 89.44 in = 2.2717 m

Tutup

Jenis : *Torispherical Dished Head*

Tebal Tutup Atas : 3/16 in = 0.005 m

Tebal Tutup Bawah : 3/16 in = 0.005 m

Tinggi : 11.76 in = 0.299 m

Tinggi Tangki Total : 113.0 in = 2.871 m

Pengaduk

Jenis : *Three Blade Propeller*

Jumlah : 1 buah

Diameter : 19.8 in = 0.5029 m

Kecepatan : 190 rpm = 3.1667 rps

Power : 0 hp

34 POMPA TANGKI DEKOLORISASI (L-331)

Fungsi : Mengalirkan *dioctyl phthalate* dari tangki dekolorisasi menuju *filter press*

Tipe : *Centrifugal pump*

Jumlah : 1 buah

- Tujuan :
1. Menentukan tipe pompa
 2. Menentukan bahan konstruksi pompa
 3. Menghitung tenaga pompa
 4. Menghitung tenaga motor

3) Menghitung Tenaga Pompa

Rate masuk = 3814.0049 kg/jam = 2.3357 lb/s

Densitas = 0.9847 kg/m³ = 0.0615 lb/ft³

μ = 2.8134 cP = 0.0019 lb/ft.s

$$\begin{aligned} \text{Rate volumetrik (Q)} &= \frac{\text{massa}}{r} \\ &= \frac{2.3357 \text{ lb/s}}{0.0615 \text{ lb/ft}^3} = 37.9945 \text{ ft}^3/\text{s} \\ &= 17053.0973 \text{ gpm} \end{aligned}$$

Diperkirakan aliran fluida turbulen ($N_{re} > 2100$), sehingga digunakan persamaan $Di \geq 1$ in yaitu :

$$Di_{opt} = 3.9 Q^{0.45} \mu^{0.13} \quad \text{Eq. 45. (Peters \& Timmerhaus, 1991:365)}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

Dimana :

Di opt = diameter dalam optimum, in

Q = kecepatan volumetrik, ft³/s

μ = viskositas fluida, lb/ft.s

Sehingga :

$$\begin{aligned} \text{Di opt} &= 3.9 \times (37.9945)^{0.45} \times (0.0019)^{0.13} \\ &= 8.8695 \text{ in} \end{aligned}$$

Dari App. A.5-1 (Geankoplis, 1993:892) dipilih spesifikasi :

NPS = 1/2 in

OD = 0.8400 in = 0.07 ft = 0.021336 m

ID = 0.6220 in = 0.0518333 ft = 0.015799 m

A = 0.0021 ft² = 1.9603 in²

Sch = 40

Menghitung kecepatan linier :

Kecepatan linier fluida dapat dicari dengan menggunakan persamaan berikut :

$$V = \frac{Q}{A}$$

Sehingga :

$$V = \frac{37.9945 \text{ ft}^3/\text{s}}{0.0021 \text{ ft}^2} = 18006.8512 \text{ ft/s} = 5488.4882 \text{ m/s}$$

Menghitung *Reynold Number* (Nre) :

$$\text{Nre} = \frac{\rho v D}{\mu}$$

Sehingga :

$$\begin{aligned} \text{Nre} &= \frac{0.0615 \times 18006.8512 \times 0.0518}{0.0019} \\ &= 30348.30761 \text{ (asumsi aliran turbulen benar)} \end{aligned}$$

Instalasi pipa

Dari **Fig. 127**. (Brown, 1950:141) dengan NPS = 1/2 in

- 1 buah *gate valve fully open*; Le = 0.35 ft

ΣLe = 1 × 0.35 = 0.35 ft

- 4 buah *standard elbow*; Le = 1.5 ft

ΣLe = 4 × 1.5 = 6 ft

- 1 buah *sudden enlargement*; Le = 1 ft



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

ΣLe	$= 1 \times 1$	$= 1$	ft
- 1 buah <i>sudden contraction</i> ; Le		$= 0.52$	ft
ΣLe	$= 1 \times 0.52$	$= 0.52$	ft
- 1 buah <i>swing check valve</i> ; Le		$= 3.5$	ft
ΣLe	$= 1 \times 3.5$	$= 3.5$	ft
- Panjang ekivalen pipa lurus, ΣLe		$= 11.37$	ft
Panjang pipa lurus		$= 2$	m
		$= 6.5616$	ft
Panjang pipa total		$= 17.9316$	ft
		$= 5.4656$	m

Menghitung *friction loss*

$$\Sigma F = \frac{4f \cdot v^2 \cdot \Delta L}{2 \cdot ID \cdot gc} \quad \text{Eq. 2.10-6 (Geankoplis, 1993:89)}$$

Dimana :

$$\begin{aligned} \Sigma F &= \text{Friction loss} \quad (\text{ft} \cdot \text{lb}_f / \text{lb}_m) \\ f &= \text{Faktor friksi} \\ v &= \text{Kecepatan linier fluida (ft/s)} \\ \Delta L &= \text{Panjang pipa (ft)} \\ ID &= \text{Diameter dalam tangki (ft)} \\ gc &= 32.1740 \quad \text{lb}_m \cdot \text{ft} / \text{lb}_f \cdot \text{s}^2 \end{aligned}$$

Menghitung *Fanning Friction Factor* (f) :

Dari **Fig. 2.10-3.** (Geankoplis, 1993:88) didapat :

$$\begin{aligned} \text{Untuk commercial steel} \quad \rightarrow \quad \varepsilon &= 5.E-05 \quad \text{m} \\ &= 0.0002 \quad \text{ft} \end{aligned}$$

Sehingga :

$$\frac{\varepsilon}{D} = \frac{0.0002}{0.0518} = 0.0030$$

Dari **Fig. 2.10-3.** (Geankoplis, 1993:88) dengan nilai $N_{re} = 30348.3076$ didapatkan nilai $f = 0.0055$ sehingga :

$$\begin{aligned} \Sigma F &= \frac{4 \times 0.0055 \times (18006.8512)^2 \times 17.9316}{2 \times 0.0518 \times 32.1740} \\ &= 38350681.2230 \frac{\text{lb}_f \cdot \text{ft}}{\text{lb}_m} \end{aligned}$$

Menghitung *Static Head* :



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$Z_1 = 1 \text{ m} = 3.28084 \text{ ft}$$

$$Z_2 = 0.2 \text{ m} = 0.656168 \text{ ft}$$

$$\Delta Z = Z_1 - Z_2 = 2.6247 \text{ ft}$$

$$g/gc = 1 \text{ lbf/lbm}$$

$$\Delta Z (g/gc) = 2.624672 \text{ ft} \times 1 \text{ lbf/lbm} = 2.624672 \text{ ft lbf/lbm}$$

Menghitung *Velocity Head* :

V_1 = kecepatan linier fluida dari tangki hidrogen peroksida ke pipa

V_2 = kecepatan linier fluida ke *heater*

Karena pada 2 titik *reference* dianggap sama, maka $V_1 = V_2 = 18006.8512 \text{ ft/s}$

Sehingga *velocity head* ($\Delta V^2 / 2gc$) = 5038955.2101

Menghitung *Pressure Head* :

$$P_1 = P_2 = 14.7 \text{ Psi}$$

$$\Delta P = 0$$

Sehingga, $\Delta P/\rho = 0$

Menghitung Energi Mekanik Pompa :

$$-Wf = \frac{\Delta V^2}{2 \times \alpha \times gc} + \Delta z \frac{g}{gc} + \frac{\Delta P}{\rho} + \sum F$$

Dimana :

Wf = tenaga yang ditambahkan ke dalam sistem per satuan massa

Sehingga :

$$\begin{aligned} -Wf &= 5038955 + 2.625 \text{ ft. lbf/lbm} + 0 + 38350681 \text{ ft. lbf/lbm} \\ &= 43389639.06 \text{ ft. lbf/lbm} \end{aligned}$$

Menghitung *Broke Horse Power* (BHP) :

$$BHP = \frac{m \cdot (-Wf)}{550 \cdot \eta}$$

dari **Fig. 14-37**. (Peters & Timmerhaus, 1991:520), untuk $Q_f = 17053.0973 \text{ gpm}$ diperoleh η pompa = 0.40

Sehingga :

$$\begin{aligned} BHP &= \frac{2.3357 \times 43389639.06}{550 \times 0.4000} \\ &= 460650.9 \text{ Hp}; \text{ maka digunakan } power = 1 \text{ Hp} \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

4) Menghitung Tenaga Motor

Fig. 14.38. (Peters & Timmerhaus, 1991:521) untuk BH = 1.0 Hp diperoleh η motor = 0.80

Sehingga *power* motor yang diperlukan :

$$\begin{aligned} P \text{ motor} &= \frac{\text{BHP}}{\eta} \\ &= \frac{1 \text{ Hp}}{0.80} \\ &= 1.3 \text{ Hp} \end{aligned}$$

Dipilih motor standar dengan *power* = 2 Hp (*Standard NEMA*)

Spesifikasi Pompa Tangki Dekolorisasi

Nama Alat	: Pompa Tangki Dekolorisasi
Kode	: L-331
Fungsi	: Mengalirkan <i>dioctyl phthalate</i> dari tangki dekolorisasi menuju <i>filter press</i>
Tipe	: <i>Centrifugal pump</i>
Bahan Konstruksi	: <i>Commercial steel</i>
Jumlah	: 1 buah
Rate Volumetrik	: 37.9945 ft ³ /s
Kecepatan Aliran	: 18006.8512 ft/s
Ukuran Pipa	: NPS = 1/2 in
	: <i>Sch. Number</i> = 40
	: OD = 0.0700 ft = 0.021336 m
	: ID = 0.0518 ft = 0.015799 m
	: <i>Flow Area</i> = 0.0021 ft ² = 0.0002 m ²
<i>Power</i> Pompa	: 1 Hp
<i>Power</i> Motor	: 2 Hp

35) FILTER PRESS (H-340)

Fungsi	: Memisahkan <i>dioctyl phthalate</i> dari karbon aktif
Tipe	: <i>Plate and Frame Filter Press</i>
Jumlah	: 1 buah
Kondisi Operasi	: T = 30 °C = 303.15 K
	: P = 1 atm = 760 mmHg = 14.7 psi



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetraabutyl Titanate Melalui Proses Esterifikasi

Komposisi filtrat

Komponen	Massa (kg/jam)	Fraksi berat	ρ bahan (g/ml)	ρ bahan x
				fraksi berat (x)
$C_{24}H_{38}O_4$	3766.9558	0.9951	0.984	0.9792
$C_8H_{18}O$	5.0428	0.0013	0.833	0.0011
$C_8H_4O_3$	2.8704933	0.0008	1.2	0.0009
$C_8H_6O_4$	2.8791307	0.0008	1.526	0.0012
H_2O	7.7041388	0.0020	0.997	0.0020
Total	3785.4522910	1.0000	5.5400	0.9844

Komposisi cake

Komponen	Massa (kg/jam)	Fraksi berat	ρ bahan (g/ml)	ρ bahan x
				fraksi berat (x)
$C_{24}H_{38}O_4$	18.9294	0.6630	0.984	0.6524
$C_8H_{18}O$	0.0253	0.0009	0.833	0.0007
$C_8H_4O_3$	0.0144246	0.0005	1.2	0.0006
$C_8H_6O_4$	0.0144246	0.0005	1.526	0.0008
H_2O	0.0387143	0.0014	0.997	0.0014
Karbon Aktif	9.5302472	0.3338	0.4600	0.1535
Total	28.5525771	1.0000	6.0000	0.8094

$$\begin{aligned}
 \text{rate massa filtrat} &= 3785.4523 \quad \text{kg/jam} \\
 &= 8345.4838 \quad \text{lb/jam} \\
 \rho \text{ campuran} &= \sum \rho_i \times X_i = 0.9844 \quad \text{gr/ml} \\
 &= 61.45418032 \quad \text{lb/cuft}
 \end{aligned}$$

$$\begin{aligned}
 \text{rate volumetrik filtrat} &= \frac{\text{rate massa filtrat}}{\rho \text{ campuran}} = \frac{8345.4838}{61.4542} \\
 &= 135.8001 \quad \text{cuft/jam}
 \end{aligned}$$

$$\begin{aligned}
 \text{waktu pembersihan} &= 0.5 \\
 \text{waktu siklus operasi} &= 1 \\
 \text{Volume filtrat} &= \text{rate volumetrik filtrat} \times \text{waktu operasi} \\
 &= 135.8001 \quad \text{x} \quad 1 \\
 &= 135.8001 \quad \text{cuft} = 1015.853 \quad \text{gal}
 \end{aligned}$$

$$\begin{aligned}
 \text{rate massa cake} &= 28.5526 \quad \text{kg/jam} \\
 &= 62.9470 \quad \text{lb/jam}
 \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned}\rho \text{ campuran} &= 0.8094 \text{ gr/ml} \\ &= 50.5271 \text{ lb/cuft} \\ \text{rate volumetrik cake} &= \frac{\text{rate massa cake}}{\rho \text{ campuran}} = \frac{62.9470}{50.5271} \\ &= 1.2458 \text{ cuft/jam}\end{aligned}$$

$$\begin{aligned}\text{Volume cake} &= \text{rate volumetrik cake} \times \text{waktu operasi} \\ &= 1.2458 \times 1 \\ &= 1.2458 \text{ cuft} \\ &= 9.3193 \text{ gal}\end{aligned}$$

Dari literatur perry 5th ed. Tabel 19-18 didapatkan

$$\text{rate filtrasi} = 5 - 20 \text{ gal/ft}^2 \cdot \text{Jam} \quad (\text{diambil } 20 \text{ gal/ft}^2 \cdot \text{jam})$$

$$\begin{aligned}\text{Luas frame total} &= \frac{\text{Volume filtrat}}{\text{rate filtrasi} \times \text{waktu operasi}} \\ &= \frac{1015.8526}{20 \times 1} \\ &= 50.7926 \text{ ft}^2 \\ &= 4.7188 \text{ m}^2\end{aligned}$$

Digunakan ukuran plate and frame = 18 x 18 inch

Dari literatur perry 5th ed. Didapatkan

$$\begin{aligned}\text{Luas efektif} &= 3.9 \text{ ft}^2 \\ \text{Total Kapasitas} &= 0.16 \text{ ft}^2 / \text{in tebal}\end{aligned}$$

Jumlah frame

$$\begin{aligned}N &= \frac{\text{Luas total}}{\text{Luas efektif}} \\ &= \frac{50.7926}{3.9} \\ &= 13.0238 \text{ buah} \\ &= 128 \text{ buah}\end{aligned}$$

$$\begin{aligned}\text{Plate \& Frame yang dipakai} &= (2 \times N) - 1 \\ &= (2 \times 128) - 1 \\ &= 255 \text{ buah}\end{aligned}$$

$$\begin{aligned}\text{Volume cake tiap frame} &= \frac{\text{rate volumetrik cake}}{\text{jumlah frame}} \\ &= \frac{1.2458}{128} \\ &= 0.0097 \text{ cuft} \\ &= 0.0003 \text{ m}^3\end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned} \text{Tebal frame} &= \frac{\text{Volume cake tiap frame}}{\text{luas efektif}} \\ &= \frac{0.0097}{3.9} \\ &= 0.0025 \quad \text{ft} = 0.0299 \quad \text{inch} = 1 \quad \text{inch} \end{aligned}$$

Diambil tebal frame

$$\begin{aligned} \text{Panjang total filter press} &= \text{jumlah plate \& frame} \times \text{tebal frame} \\ &= 255 \times 1 \\ &= 255 \quad \text{inch} \\ &= 6.48 \quad \text{m} \end{aligned}$$

Pressure drop

Dari literatur Christopher, halamann 469 didapatkan Filtrasi dilakukan pada pressure drop(DP) yang konstan sebesar = 30 psi

SPEKIFIKASI FILTER PRESS

Fungsi	=	Memisahkan <i>dioctyl phthalate</i> dari karbon aktif
Type	=	Plate & Frame filter press
Jumlah plate	=	127
Jumlah frame	=	128
Tebal frame	=	1 inch
Panjang alat	=	6.48 m
Pressure drop	=	30 psi
Bahan konstruksi	=	Stainless steel
Jumlah	=	1 buah

36) POMPA FILTER PRESS (L-341)

Fungsi	:	Mengalirkan <i>dioctyl phthalate</i> dari <i>filter press</i> menuju tangki penyimpanan <i>dioctyl phthalate</i>
Tipe	:	<i>Centrifugal pump</i>
Jumlah	:	1 buah
Tujuan	:	1. Menentukan tipe pompa 2. Menentukan bahan konstruksi pompa 3. Menghitung tenaga pompa 4. Menghitung tenaga motor

3) Menghitung Tenaga Pompa

Rate masuk	=	3785.4523	kg/jam	=	2.3182	lb/s
Densitas	=	0.9842	kg/m ³	=	0.0614	lb/ft ³
μ	=	52.8232	cP	=	0.0355	lb/ft.s



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

$$\begin{aligned} \text{Rate volumetrik (Q)} &= \frac{\text{massa}}{r} \\ &= \frac{2.3182 \text{ lb/s}}{0.0614 \text{ lb/ft}^3} = 37.7288 \text{ ft}^3/\text{s} \\ &= 16933.8780 \text{ gpm} \end{aligned}$$

Diperkirakan aliran fluida laminer ($N_{re} < 2100$), sehingga digunakan persamaan $Di < 1$ in yaitu :

$$Di_{opt} = 3.6 Q^{0.40} \mu^{0.20} \quad \text{Eq. 48. (Peters \& Timmerhaus, 1991:36)}$$

Dimana :

Di_{opt} = diameter dalam optimum, in

Q = kecepatan volumetrik, ft^3/s

μ = viskositas fluida, $\text{lb}/\text{ft}\cdot\text{s}$

Sehingga :

$$\begin{aligned} Di_{opt} &= 3.6 \times (37.7288)^{0.40} \times (0.0355)^{0.20} \\ &= 7.8888 \text{ in} \end{aligned}$$

Dari *App. A.5-1* (Geankoplis, 1993:892) dipilih spesifikasi :

$NPS = 1/2 \text{ in}$

$OD = 0.8400 \text{ in} = 0.07 \text{ ft} = 0.021336 \text{ m}$

$ID = 0.6220 \text{ in} = 0.0518333 \text{ ft} = 0.015799 \text{ m}$

$A = 0.0021 \text{ ft}^2 = 1.9603 \text{ in}^2$

$Sch = 40$

Menghitung kecepatan linier :

Kecepatan linier fluida dapat dicari dengan menggunakan persamaan berikut :

$$V = \frac{Q}{A}$$

Sehingga :

$$V = \frac{37.7288 \text{ ft}^3/\text{s}}{0.0021 \text{ ft}^2} = 17880.9642 \text{ ft/s} = 5450.118 \text{ m/s}$$

Menghitung *Reynold Number* (N_{re}) :

$$N_{re} = \frac{\rho v D}{\mu}$$



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Sehingga :

$$\begin{aligned} N_{re} &= \frac{0.0614 \times 17880.9642 \times 0.0518}{0.0355} \\ &= 1604.2712 \quad (\text{asumsi aliran laminar benar}) \end{aligned}$$

Instalasi pipa

Dari **Fig. 127.** (Brown, 1950:141) dengan NPS = 1/2 in

- 1 buah <i>gate valve fully open</i> ; Le	= 0.35 ft
$\sum Le = 1 \times 0.35$	= 0.35 ft
- 4 buah <i>standard elbow</i> ; Le	= 1.5 ft
$\sum Le = 4 \times 1.5$	= 6 ft
- 1 buah <i>sudden enlargement</i> ; Le	= 1 ft
$\sum Le = 1 \times 1$	= 1 ft
- 1 buah <i>sudden contraction</i> ; Le	= 0.57 ft
$\sum Le = 1 \times 0.57$	= 0.57 ft
- 1 buah <i>swing check valve</i> ; Le	= 3.5 ft
$\sum Le = 1 \times 3.5$	= 3.5 ft
- Panjang ekivalen pipa lurus, $\sum Le$	= 11.42 ft
Panjang pipa lurus	= 2 m
	= 6.5616 ft
Panjang pipa total	= 17.9816 ft
	= 5.4808 m

Menghitung *friction loss*

$$\Sigma F = \frac{4f \cdot v^2 \cdot \Delta L}{2 \cdot ID \cdot gc} \quad \text{Eq. 2.10-6 (Geankoplis, 1993:89)}$$

Dimana :

ΣF	=	<i>Friction loss</i>	(ft.lbf/lbm)
f	=	Faktor friksi	
v	=	Kecepatan linier fluida (ft/s)	
ΔL	=	Panjang pipa (ft)	
ID	=	Diameter dalam tangki (ft)	
gc	=	32.1740	lbm.ft/lbf.s ²

Menghitung *Fanning Friction Factor* (f) :

Dari **Fig. 2.10-3.** (Geankoplis, 1993:88) didapat :

Untuk <i>commercial steel</i>	→	ϵ	= 5.E-05	m
			= 0.0002	ft

Sehingga :

$$\frac{\epsilon}{D} = \frac{0.0002}{0.0518} = 0.0030$$



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Dari **Fig. 2.10-3.** (Geankoplis, 1993:88) dengan nilai 1604.2712
didapatkan nilai $f = 0.04$ sehingga :

$$\Sigma F = \frac{4 \times 0.0400 \times (17880.9642)^2 \times 17.9816}{2 \times 0.0518 \times 32.1740}$$

$$= 275794746.68 \frac{\text{lb}_f \cdot \text{ft}}{\text{lb}_m}$$

Menghitung *Static Head* :

$$Z_1 = 1 \text{ m} = 3.28084 \text{ ft}$$

$$Z_2 = 0.2 \text{ m} = 0.656168 \text{ ft}$$

$$\Delta Z = Z_1 - Z_2 = 2.6247 \text{ ft}$$

$$g/gc = 1 \text{ lbf/lbm}$$

$$\Delta Z (g/gc) = 2.624672 \text{ ft} \times 1 \text{ lbf/lbm} = 2.624672 \text{ ft lbf/lbm}$$

Menghitung *Velocity Head* :

V_1 = kecepatan linier fluida dari tangki hidrogen peroksida ke pipa

V_2 = kecepatan linier fluida ke *heater*

Karena pada 2 titik *reference* dianggap sama, maka $V_1 = 17880.9642 \text{ ft/s}$

Sehingga *velocity head* ($\Delta V^2 / 2gc$) = #####

Menghitung *Pressure Head* :

$$P_1 = P_2 = 14.7 \text{ Psi}$$

$$\Delta P = 0$$

Sehingga, $\Delta P/\rho = 0$

Menghitung Energi Mekanik Pompa :

$$- Wf = \frac{\Delta V^2}{2 \times \alpha \times gc} + \Delta Z \frac{g}{gc} + \frac{\Delta P}{\rho} + \Sigma F$$

Dimana :

Wf = tenaga yang ditambahkan ke dalam sistem per satuan massa

Sehingga :

$$- Wf = 4968746 + 2.625 \text{ ft. lbf/lbm} + 0$$

$$+ 275794746.7 \text{ ft. lbf/lbm}$$

$$= 280763495.494 \text{ ft. lbf/lbm}$$

Menghitung *Broke Horse Power* (BHP) :



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$$\text{BHP} = \frac{m \cdot (-Wf)}{550 \cdot \eta}$$

dari **Fig. 14-37**. (Peters & Timmerhaus, 1991:520), untuk $Q_f = 16933.8780$ gpm diperoleh η pompa = 0.40

Sehingga :

$$\begin{aligned} \text{BHP} &= \frac{2.3182 \times 280763495.5}{550 \times 0.4000} \\ &= \text{##### Hp}; \text{ maka digunakan } power = 2.0 \text{ Hp} \end{aligned}$$

4) Menghitung Tenaga Motor

Fig. 14.38. (Peters & Timmerhaus, 1991:521) untuk = 2.0 Hp diperoleh η motor = 0.83

Sehingga *power* motor yang diperlukan :

$$\begin{aligned} P \text{ motor} &= \frac{\text{BHP}}{\eta} \\ &= \frac{2 \text{ Hp}}{0.83} \\ &= 2.4 \text{ Hp} \end{aligned}$$

Dipilih motor standar dengan $power = 3 \text{ Hp}$ (Standard NEMA)

Spesifikasi Pompa *Filter Press*

Nama Alat	: Pompa <i>Filter Press</i>
Kode	: L-341
Fungsi	: Mengalirkan <i>dioctyl phthalate</i> dari <i>filter press</i> menuju tangki penyimpanan <i>dioctyl phthalate</i>
Tipe	: <i>Centrifugal pump</i>
Bahan Konstruksi	: <i>Commercial steel</i>
Jumlah	: 1 buah
Rate Volumetrik	: 37.7288 ft ³ /s
Kecepatan Aliran	: 17880.9642 ft/s
Ukuran Pipa	: NPS = 1/2 in
	: <i>Sch. Number</i> = 40
	: OD = 0.0700 ft = 0.021336 m
	: ID = 0.0518 ft = 0.015799 m
	: <i>Flow Area</i> = 0.0021 ft ² = 0.0002 m ²
Power Pompa	: 2.0 Hp
Power Motor	: 3 Hp

37 TANGKI PENYIMPANAN DIOCTYL PHTHALATE (F-410)

Fungsi	: Tempat menyimpan produk <i>Dioctyl Phthalate</i>
Tipe	: <i>Cylindrical Tank, Flat Bottom, Conical Roof</i>



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Jumlah : 1 buah
 Kondisi Operasi : T = 30 °C = 303.15 K
 P = 1 atm = 760 mmHg
 = 14.7 psi

Komposisi

Komponen	Massa (Kg/Jam)	Fraksi berat	ρ bahan (g/ml)	ρ bahan x
				fraksi berat (x)
C ₂₄ H ₃₈ O ₄	3766.9558	0.9951	0.984	0.9792
C ₈ H ₁₈ O	5.0428	0.0013	0.833	0.0011
C ₈ H ₄ O ₃	2.8705	0.0008	1.2	0.0009
C ₈ H ₆ O ₄	2.8791	0.0008	1.526	0.0012
H ₂ O	7.7041	0.0020	0.997	0.0020
Total	3785.4523	1.0000	5.5400	0.9844

$$\begin{aligned} \text{Densitas Campuran} &= \frac{1}{\frac{\text{fraksi berat}}{\rho \text{ komponen}}} \times 62.43 \\ &= \frac{1}{0.18} \times 62.43 \\ &= 345.86 \times 62.43 \\ &= 21592.18 \text{ lb/cuft} \end{aligned}$$

$$\begin{aligned} \text{Rate Volumetrik} &= \frac{\text{Rate Massa}}{\text{Densitas}} \\ &= \frac{8346.92}{21592.1771} \\ &= 0.3866 \text{ cuft/jam} \end{aligned}$$

Direncanakan penyimpanan untuk 14 har dengan jumlah tangki 1 buah sehingga volume liquid adalah

$$\begin{aligned} V_{\text{liquid}} &= \frac{0.39 \times 24 \times 14}{1} \\ &= 129.89 \text{ cuft} \end{aligned}$$

Asumsi volume bahan (liquid) mengisi 80% volume tangki sehingga volume ruang kosong sebesar 20%

$$\begin{aligned} \text{Volume Tangki} &= \frac{129.8881}{80\%} \\ &= 162.3601 \text{ cuft} = 1214.453 \text{ gal} \end{aligned}$$

Disarankan > 10000 gal, menggunakan vertical tank dan pondasi beton

[Walas ed.2; p.xvii]

Menentukan ukuran tangki tangki dan ketebalannya



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$$\text{Dimensi ratio} = \frac{H}{D} = 2 \quad [Ulrich; T.4-27:p.248]$$

$$\text{Volume tangki} = \frac{\pi}{4} \times D^2 \times H$$

$$162.3601 = \frac{3.14}{4} \times D^2 \times 2D$$

$$162.3601 = 2 \times D^3$$

$$D^3 = 103.4141 \text{ cuft}$$

$$D = 4.6938 \text{ ft} = 56.32585 \text{ in} = 1.408146 \text{ m}$$

Karena $H = 2D$, maka H :

$$H = 2 D$$

$$H = 2 \times 4.693821 \text{ ft}$$

$$H = 9.387642 \text{ ft} = 112.6517 \text{ in} = 2.816293 \text{ m}$$

Tinggi liquid dalam tangki:

$$\text{Volume Liquid} = \frac{\pi}{4} \times D^2 \times H$$

$$129.89 = \frac{3.14}{4} \times 22.0320 \times H$$

$$H_{\text{liq}} = 7.51 \text{ ft} = 2.253034 \text{ m}$$

P design

$$P_{\text{operasi}} = 1 \text{ atm} = 14.7 \text{ lb/in}^2 \text{ (psi)}$$

$$\begin{aligned} P_{\text{hidrostatik}} &= \rho \times g/gc \times H_{\text{liq}} \\ &= 21592.18 \times 1 \times 7.5101 \\ &= 162159.7 \text{ lb/ft}^2 \\ &= 1126.1 \text{ lb/in}^2 \end{aligned}$$

P Design 10% lebih besar untuk faktor keamanan

$$\begin{aligned} P_{\text{design}} &= (P_{\text{operasi}} + P_{\text{hidrostatik}}) \times 110\% \\ &= 14.7 + 1126.1091 \times 110\% \\ &= 1255 \text{ psi} \end{aligned}$$

Penentuan tebal shell :

$$t_{\min} = \frac{P \times r_i}{fE - 0.6P} + C \quad [Brownell\&Young; Eq.13-1]$$

dengan :

$$t_{\min} = \text{tebal shell minimum} \quad ; \text{ in}$$

$$P = \text{tekanan tangki} \quad ; \text{ psi}$$

$$r_i = \text{jari-jari tangki} \quad ; \text{ in (1/2 D)}$$

$$C = \text{faktor korosi} \quad ; \text{ in (digunakan 1/8 in)}$$

$$E = \text{faktor pengelasan, digunakan double welded} \quad E = 0.8$$

$$f = \text{Stress allowable, bahan konstruksi stainless steel 316}$$

$$f = 35969.92 \text{ psi} \quad [Perry ed.7; T.28-11]$$



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$$\begin{aligned}
 r_i &= 1 \times D \\
 &= 1 \times 56.33 \\
 &= 28.1629 \text{ in}
 \end{aligned}$$

$$\begin{aligned}
 t_{\min} &= \frac{P \times r_i}{fE - 0,6P} + C \\
 &= \frac{1255 \times 28.16}{28776 - 752.93} + 0.125 \\
 &= \frac{35341.37}{28023.00} + 0.125 \\
 &= 1.261 + 0.125 \\
 &= 1.386 \text{ in}
 \end{aligned}$$

digunakan $t_{\min} = 4/16 \text{ in}$

Dimensi Tutup Atas :

Tutup atas berbentuk standart dished head

$$\begin{aligned}
 OD &= D + 2 \ t_s = 57 \text{ in} \\
 r_c &= 57 \text{ in} \quad 47 \quad [Brownell\&Young; T.5-7] \\
 &= 4.7 \text{ ft} \quad \text{Rumus iman}
 \end{aligned}$$

Tebal standart toripherical dished (atas) :

$$t_h = \frac{0,885 \times P \times r_c}{f_e - 0,1P} + C$$

dengan

$$\begin{aligned}
 t_h &= \text{tebal shell minimum} \quad ;\text{in} \\
 P &= \text{tekanan tangki} \quad ;\text{psi} \\
 r_c &= \text{crown radius} \quad ;\text{in} \quad [Brownell \text{ and } Young, T-5.7] \\
 C &= \text{faktor korosi} \quad ;\text{in} \quad (\text{digunakan } 1/8) \\
 E &= \text{faktor pengelasan, digunakan dobel welded} \quad E = 1 \\
 f &= \text{Stress allowable, bahan konstruksi stainless steel 316} \\
 f &= 35970 \text{ psi} \quad [Perry \text{ ed. } 7; T.28-11]
 \end{aligned}$$

P design = 1254.89 psi

$$\begin{aligned}
 t_h &= \frac{0,885 \times P \times r_c}{f.E - 0,1.P} + C \\
 &= \frac{0.885 \times 1254.89 \times 57}{28775.94 - 125.49} + 1/8 \\
 &= 2.328 \text{ in} \quad \text{digunakan } t = 5/16 \text{ in}
 \end{aligned}$$

$$\begin{aligned}
 h &= r_c - \sqrt{r_c^2 - \frac{D^2}{4}} \\
 &= 0.62 \text{ ft}
 \end{aligned}$$



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Tutup bawah

untuk tebal tutup bawah datar karena tutup menumpang diatas beton,
maka tebal tutup = $1/3$ in [Brownell&Young; p.58]

SPESIFIKASI TANGKI PENYIMPANAN DIOCTYL PHTHALATE

Dimensi Shell :

Diameter tangki	:	4.69	ft
Tinggi tangki	:	9.39	ft
Tebal Shell	:	4/16	in
Tebal tutup atas	:	5/16	in
Tinggi tutup atas	:	10/16	ft
Tebal tutup bawah	:	1/3	in
Bahan konstruksi	:	Stainless steel 316	
Jumlah tangki	:	1 buah	



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APPENDIX D PERHITUNGAN ANALISA EKONOMI

Kapasitas Produksi	=	30,000 ton/tahun	
	=	3,787.8788 kg/jam	
Waktu operasi	=	330 hari	
Asumsi kurs dollar	=	Rp 15,279 ,-/US \$	(BI, Februari 2020)
Dengan bahan baku :			
$C_8H_4O_3$	=	\$ 1,000	= Rp 15,279,000 /Ton
$C_8H_{18}O$	=	\$ 850	= Rp 12,987,150 /Ton
$C_{16}H_{36}O_4Ti$	=	\$ 0.80	= Rp 12,223 /Kg
NaOH 48%	=	\$ 0.70	= Rp 10,695 /Kg
Karbon Aktif	=	\$ 600	= Rp 9,167,400 /Ton
Produk yang dihasilkan :			
$C_{24}H_{38}O_4$	=	\$ 2.20	= Rp 33,614 /Kg
$C_{16}H_{35}O_4TiNa$	=	\$ 0.30	= Rp 4,584 /Liter

Analisa ekonomi di dalam suatu perencanaan pabrik adalah sangat penting, karena perhitungan ekonomi ini dapat diketahui apakah pabrik yang direncanakan ini layak untuk didirikan atau tidak dalam artian Faktor-faktor yang perlu untuk ditinjau antara lain:

1. Laju pengembalian modal (*Rate of Return, ROR*)
2. Laju investasi yang sehat (*Rate of Investment, ROI*)
3. Lama pengembalian modal (*Pay-Out Periode*)
4. Perhitungan Resiko Hutang (*Rate on Equity*)
5. Titik impas (*Break Even Point, BEP*)

Untuk meninjau faktor-faktor diatas, perlu adanya penaksiran terhadap beberapa faktor, yaitu:

1. Penaksiran modal industri (*Total Capital Investment*) yang terdiri atas:
 - a. Modal tetap (*Fixed Capital Investment*)
 - b. Modal kerja (*Working Capital Investment*)
2. Penentuan biaya produksi total (*Production Cost*) yang terdiri atas:
 - a. Biaya pembuatan (*Manufacturing Cost*)
 - b. Biaya pengeluaran umum (*General Expenses*)
3. Total pendapatan

1. Harga Peralatan

Harga peralatan berubah menurut waktu resmi sesuai dengan kondisi ekonomi dunia. Untuk memperkirakan harga peralatan saat ini, digunakan indeks seperti pada persamaan sebagai berikut :



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$$C_p = \frac{I_p}{I_o} \times C_o$$

Dimana :

- C_p = Harga alat pada tahun 2024
- C_o = Harga alat pada tahun data 2014
- I_p = Cost Index pada tahun 2019
- I_o = Cost Index pada tahun data 2014

Perhitungan peralatan didasarkan pada cost equipment www.matche.com. Sedangkan Cost indeks didasarkan pada 'Peters and Timmerhauss 5ed Plant De -sign and Economic for Chemical Engineering'

Tabel D.1 Indeks harga Peralatan

Tahun	Indeks
2011	585.7
2012	584.6
2013	567.3
2014	576.1
2015	556.8
2016	541.7
2017	562.1
2018	603.1
2019	607.5
2020	596.2
2021	708

sumber: CEPCI tahun 2022 annual index

Dengan metode least square dan data-data pada tabel di atas dilakukan pendekatan atau penafsiran indeks harga peralatan pada awal tahun dimana data-data tersebut dibentuk dalam persamaan :

$$Y = a + bX$$

keterangan :

Y = indeks harga peralatan pada tahun ke-n

X = tahun ke-n

n	X	Y	X ²	Y ²	XY
1	2011	585.7	4,044,121	343,044	1,177,843
2	2012	584.6	4,048,144	341,757	1,176,215
3	2013	567.3	4,052,169	321,829	1,141,975
4	2014	576.1	4,056,196	331,891	1,160,265
5	2015	556.8	4,060,225	310,026	1,121,952
6	2016	541.7	4,064,256	293,439	1,092,067
7	2017	562.1	4,068,289	315,956	1,133,756
8	2018	603.1	4,072,324	363,730	1,217,056



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n	X	Y	X ²	Y ²	XY
9	2019	607.5	4,076,361	369,056	1,226,543
10	2020	596.2	4,080,400	355,454	1,204,324
11	2021	708	4,084,441	501,264	1,430,868
Total	22176	6489	44,706,926	3,847,448	13,082,863

Jumlah data = n = 11

Dengan menggunakan metode Least Square Pers 17-21, Peters&Timmerhauss, diperoleh :

$$\sum (\bar{x} - x)^2 = \sum x^2 - \frac{(\sum x)^2}{n} = 110$$

$$\sum (\bar{y} - y)^2 = \sum y^2 - \frac{(\sum y)^2}{n} = 19,409.9$$

Pers 17-20, Peters & Timmerhauss

$$\sum (\bar{x} - x)(\bar{y} - y) = \sum xy - \frac{\sum x \sum y}{n} = 837.8$$

$$b = \frac{\sum (\bar{x} - x)(\bar{y} - y)}{\sum (\bar{x} - x)^2} = 7.62$$

$$\text{Rata-rata } y = S_y / 9 = a = 590$$

$$\text{Rata-rata } x = S_x / 9 = c = 2016$$

$$\begin{aligned} y &= a + b (x-c) \\ &= 589.918 + 7.62 (x - 2016) \\ &= 589.918 + 7.62 x - 15354.6 \\ &= -14764.7 + 7.62 x \end{aligned}$$

Dari persamaan di atas diperoleh indeks harga pada tahun 2025 sebesar

$$\begin{aligned} y &= -14764.7 + 7.62 x (2025) \\ &= 658.4655 \end{aligned}$$

Kurs Dollar pada tahun 2025

$$(\text{US } \$) 1 = \text{Rp } 15,449 \quad \text{http://www.kursdollar.net}$$

Contoh perhitungan harga peralatan

1. Screw conveyor - 1

Panjang : 20 ft

Diameter : 4 in

Indeks harga tahun 2014 = 708 (US \$)



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Indeks harga tahun 2025 = 658.47 (US \$)

Harga alat pada tahun 2014 = 3500 (US \$)

<http://matche.com/equipcost>

Harga alat pada tahun 2025 = $\frac{\text{Index Tahun 2025}}{\text{Index tahun 2021}} \times \frac{\text{Harga Alat tahun 2021}}{2021}$

Harga alat pada tahun 2025 = $\frac{658.47}{708} \times 3500$

= 3255.13 (US \$)

= \$ 3,255.13 x Rp 15,449

= Rp 50,288,439.02

Tabel I-2. Hasil Perhitungan Peralatan Proses

No	Nama Alat	Harga Unit (US \$)		Jml	Harga (US \$)
		2014	2025		
1	Gudang karbon aktif (F-110)	391	447	1	447
2	Screw Conveyor (J-111)	2,100	2,400	1	2,400
3	Bucket Elevator-1	10,800	12,344	1	12,344
4	Hopper-1 (J-112)	600	686	1	686
5	Tangki NaOH (F-120)	5,200	5,943	1	5,943
6	Pompa-1 (L-121)	400	457	1	457
7	Mixer NaOH (M-22)	55,800	63,778	1	63,778
8	Pompa-2 (L-123)	400	457	1	457
9	Heater-1 (E-124)	700	800	1	800
10	Gudang Phthalic Anhidride (F-130)	5,500	6,286	1	6,286
11	Belt Conveyor (J-131)	9,400	10,744	1	10,744
12	Bucket Elevator-2 (J-132)	10,800	12,344	1	12,344
13	Hopper-2 (F-133)	400	457	1	457
14	Melter (X-134)	42,400	48,462	1	48,462
15	Pompa-3 (L-135)	400	457	1	457
16	Tangki 2-Ethylhexanol (F-140)	87,400	99,896	7	699,270
17	Pompa-4 (L-141)	400	457	1	457
18	Heater-2 (E-142)	18,800	21,488	1	21,488
19	Tangki Tetrabutyl Titanate (F-150)	5,600	6,401	1	6,401
20	Pompa-5 (L-151)	400	457	1	457
21	Heater-3 (E-152)	650	743	1	743
22	Reaktor (R-210)	22,000	25,145	1	25,145
23	Pompa-6 (L-211)	400	457	1	457
24	Cooler-1 (E-212)	20,300	23,202	1	23,202
25	Neutraliser (M-220)	44,000	50,291	1	50,291
26	Pompa-7 (L-221)	400	457	1	457



PERANCANGAN PABRIK

Pabrik Diocetyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

No	Nama Alat	Harga Unit (US \$)		Jml	Harga (US \$)
		2014	2025		
27	Rotary Drum Vacum Filter (H-310)	117,500	134,299	1	134299.064
28	Pompa-8 (L-311)	400	457	1	457
29	Evaporator (V-320)	25,300	28,917	1	28,917
30	Pompa-9 (L-321)	400	457	1	457
31	Cooler-2 (E-322)	93,600	106,982	1	106,982
32	Tangki Dekolorisasi (D-330)	39,600	45,262	1	45,262
33	Pompa-10 (L-331)	400	457	1	457
34	Filter Press (H-340)	68,900	78,751	1	78,751
35	Pompa-11 (L-341)	400	457	1	457
36	Tangki Diocetyl Phthalate (F-410)	174,800	199,791	1	199,791
TOTAL					1,590,262

Tabel I-3. Hasil Perhitungan Peralatan Utilitas

No.	Nama Alat	Harga unit (US \$)		Jumlah Unit	Harga Total (US \$)
		2021	2025		
1	Boiler	375,700	429,414	1	429414.11
2	Cooling tower	197,600	225,851	1	225851.022
3	Pompa air sungai	9,700	11,087	1	11086.8164
4	Bak penampung air sungai	5,500	6,286	1	6286.3392
5	Pompa tangki koagulasi	9,700	11,087	1	11086.8164
6	Tangki koagulasi	14,600	16,687	1	16687.3731
7	Tangki Flokulasi	20,100	22,974	1	22973.7123
8	Pompa clarifier	7,000	8,001	1	8000.7953
9	Clarifier	20,300	23,202	1	23202.3064
11	Bak penampung flok	1,500	1,714	1	1714.4561
12	Bak penampung air bersih-1	1,500	1,714	1	1714.4561
13	Pompa sand filter	9,700	11,087	1	11086.8164
14	Sand filter	8,500	9,715	2	19430.5029
15	Bak penampung air bersih-2	1,500	1,714	1	1714.4561
16	Pompa tangki kation exchange	6,300	7,201	1	7200.7158
17	Kation exchanger	12,500	14,287	1	14287.1345
18	Pompa tangki anion exchanger	6,300	7,201	1	7200.7158
19	Anion exchanger	12,000	13,716	1	13715.6491
20	Pompa air umpan boiler	4,900	5,601	2	11201.1134
21	Bak penampung air lunak	2,500	2,857	1	2857.4269
22	Pompa bak air sanitasi	3,200	3,658	1	3657.5064
23	Bak penampung air sanitasi	1,800	2,057	1	2057.3474
24	Pompa Bak air pendingin	8,600	9,830	1	9829.5485



PERANCANGAN PABRIK

Pabrik Diocetyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

No.	Nama Alat	Harga unit (US \$)		Jumlah Unit	Harga Total (US \$)
		2021	2025		
25	Pompa air pendingin dari CT	8,600	9,830	1	9829.5485
26	Pompa air pendingin	8,600	9,830	1	9829.5485
27	Pompa recycle air pendingin	8,600	9,830	1	9829.5485
28	Bak penampung air pendingin	2,500	2,857	1	2857.4269
29	Generator set	30,000	34,289	2	68578.2456
30	Tangki penyimpan bahan baka	85,000	97,153	1	97152.515
Jumlah					1,060,334

Total harga peralatan = Harga peralatan proses + Harga peralatan utilitas
= \$ 1,590,262 + \$ 1,060,334
= \$ 2,650,596
= **Rp 40,949,058,424**

II. Harga Bahan Baku

1. Phthalic Anhydride

Harga = \$ 1,000.00 per Ton
Kebutuhan per jam = 1,445.35 Kg/jam
Biaya per tahun (330 hari) = \$ 11,447,169 per tahun
= Rp 174,901,299,852 per tahun

2 2-Ethyl Hexanol

Harga = \$ 850.00 per Ton
Kebutuhan per jam = 2,539.13 Kg/jam
Biaya per tahun (330 hari) = \$ 17,093,408 per tahun
= Rp 261,170,184,239 per tahun

3 Tetrabutyl Titanate

Harga = Rp 12,223.20 per Kg
Kebutuhan per jam = 0.05976716 Kg/jam
Biaya per tahun (330 hari) = Rp 5,785,924.08 per tahun

4 Natrium Hidroksida

Harga = Rp 10,695.30 per Kg
Kebutuhan per jam = 0.0070314 Kg/jam
Biaya per tahun (330 hari) = Rp 595,609.83 per tahun

5 Karbon aktif

Harga = \$ 600.00 per Ton
Kebutuhan per jam = 1.91 Kg/jam
Biaya per tahun (330 hari) = \$ 9,058 per tahun



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

= Rp 138,390,259 per tahun

Total biaya bahan baku per tahun Rp 436,216,255,884

III. Harga Jual Produk

Produk Utama : Dioctyl Phthalate

Produk yang dihasilkan = 3,766.96 Ton/jam

Harga produk yang dihasilkan = \$ 2.200 /Ton

Harga produk per tahun = \$ 65,635,437

= Rp 1,014,001,871,305

Total harga jual produk per tah = Rp 1,014,001,871,305

IV. Biaya Pengemasan

Produk Utama : Disodium Phosphat hephydrat

Produk yang dihasilkan = 29,834,290 Kg/tahun

Produk dikemas dalam bag = 100 Kg

Kebutuhan bag per tahun = 298,343 buah

Harga 1 bag = Rp 2,000

Biaya pengemasan per tahu = Rp 596,685,794

Total biaya pengemasan per tahun = Rp 596,685,794

Biaya pendukung (15% pengemasan) = Rp 89,502,869 +

Total biaya pengemasan produk = Rp 686,188,663

Total biaya penjualan = Total harga jual produk + Biaya pengemasan

= **Rp1,014,688,059,968**

V. Gaji Karyawan

No	Jabatan	Gaji per Bulan	Orang	Jumlah Gaji per Bulan
1	Direktur Utama	Rp. 40,000,000	1	Rp. 40,000,000
2	Direktur Komersil	Rp. 25,000,000	1	Rp. 25,000,000
3	Direktur Teknik dan Produksi	Rp. 25,000,000	1	Rp. 25,000,000
4	Direktur SDM dan Umum	Rp. 25,000,000	2	Rp. 50,000,000
5	Staff Ahli	Rp. 25,000,000	4	Rp. 100,000,000
6	Litbang	Rp. 15,000,000	1	Rp. 15,000,000
7	Kepala Bagian Produksi	Rp. 9,500,000	1	Rp. 9,500,000
8	Kepala Bagian Teknik	Rp. 9,500,000	1	Rp. 9,500,000
9	Kepala Bagian SDM	Rp. 9,500,000	1	Rp. 9,500,000
10	Kepala Bagian Umum	Rp. 9,500,000	1	Rp. 9,500,000
11	Kepala Bagian Keuangan	Rp. 9,500,000	1	Rp. 9,500,000



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

No	Jabatan	Gaji per Bulan	Orang	Jumlah Gaji per Bulan
12	Kepala Bagian Pemasaran	Rp. 9,500,000	1	Rp. 9,500,000
13	Kepala Bagian Proses	Rp. 7,000,000	1	Rp. 7,000,000
14	Kepala Seksi Pengendalian	Rp. 7,000,000	1	Rp. 7,000,000
15	Kepala Seksi Laboratorium	Rp. 7,000,000	1	Rp. 7,000,000
16	Kepala Seksi K3 dan Lingkungan	Rp. 7,000,000	1	Rp. 7,000,000
17	Kepala Seksi Pemeliharaan	Rp. 7,000,000	1	Rp. 7,000,000
18	Kepala Seksi Utilitas	Rp. 7,000,000	1	Rp. 7,000,000
19	Kepala Seksi Administrasi	Rp. 7,000,000	1	Rp. 7,000,000
20	Kepala Seksi Akuntansi	Rp. 7,000,000	1	Rp. 7,000,000
21	Kepala Seksi Pembelian	Rp. 7,000,000	1	Rp. 7,000,000
22	Kepala Seksi Penjualan	Rp. 7,000,000	1	Rp. 7,000,000
23	Kepala Seksi Personalia	Rp. 7,000,000	1	Rp. 7,000,000
24	Kepala Seksi Diklat	Rp. 7,000,000	1	Rp. 7,000,000
25	Kepala Seksi Humas	Rp. 7,000,000	1	Rp. 7,000,000
26	Kepala Seksi Pelayanan Umum	Rp. 7,000,000	1	Rp. 7,000,000
27	Sekretaris	Rp. 6,000,000	4	Rp. 24,000,000
28	Karyawan Bagian Proses	Rp. 4,500,000	32	Rp. 144,000,000
29	Karyawan Bagian Pengendalian	Rp. 4,500,000	10	Rp. 45,000,000
30	Karyawan Bagian Laboratorium	Rp. 4,500,000	10	Rp. 45,000,000
31	Karyawan K3 dan Lingkungan	Rp. 4,500,000	7	Rp. 31,500,000
32	Karyawan Bagian Pemeliharaan	Rp. 4,500,000	7	Rp. 31,500,000
33	Karyawan Bagian Utilitas	Rp. 4,500,000	12	Rp. 54,000,000
34	Karyawan Bagian Administrasi	Rp. 4,500,000	4	Rp. 18,000,000
35	Karyawan Bagian Akuntansi	Rp. 4,500,000	4	Rp. 18,000,000
36	Karyawan Bagian Pembelian	Rp. 4,500,000	4	Rp. 18,000,000
37	Karyawan Bagian Penjualan	Rp. 4,500,000	4	Rp. 18,000,000
38	Karyawan Bagian Personalia	Rp. 4,500,000	4	Rp. 18,000,000
39	Karyawan Bagian Diklat	Rp. 4,500,000	4	Rp. 18,000,000
40	Karyawan Bagian Humas	Rp. 4,500,000	4	Rp. 18,000,000
41	Karyawan Pelayanan Umum	Rp. 4,500,000	4	Rp. 18,000,000
42	Dokter	Rp. 7,500,000	2	Rp. 15,000,000
43	Perawat	Rp. 5,000,000	4	Rp. 20,000,000
44	Sopir dan Pesuruh	Rp. 4,200,000	4	Rp. 16,800,000
45	Kebersihan dan Keamanan	Rp. 4,200,000	10	Rp. 42,000,000
Jumlah			Rp.	1,022,800,000

$$\begin{aligned} \text{Gaji karyawan per tahun} &= \text{Rp } 1,022,800,000 \\ &= \text{Rp } 1,022,800,000 \times 13 \text{ bulan} \\ &= \text{Rp } \mathbf{13,296,400,000} \end{aligned}$$



PERANCANGAN PABRIK

Pabrik Dioctyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

VI. Biaya Utilitas

a. Kebutuhan Air

1. Air Sanitasi

Kebutuhan per hari	=	41.904 m ³ /hari
Harga air mengolah sendiri	= Rp	150 /m ³
Biaya pengolahan per tahun	= Rp	2,074,248

2. Air Umpan Boiler

Kebutuhan air umpan boiler	=	65.4416 m ³ /hari
H. air boiler mengolah sendiri	= Rp	500 /m ³
Biaya pengolahan per tahun	= Rp	10,797,859

3. Air Pendingin

Kebutuhan air pendingin	=	1,159.5432 m ³ /hari
Harga air pendingin	= Rp	300 /m ³
Biaya pengolahan per tahun	= Rp	114,794,781

b. Kebutuhan Penunjang Pengolahan Air

Kebutuhan Al ₂ (SO ₄) ₃	=	9030 kg/tahun
Harga Al ₂ (SO ₄) ₃	= Rp	4,000 /kg
Biaya Al ₂ (SO ₄) ₃ per tahun	= Rp	36,121,534

Kebutuhan PAC	=	1355 kg/tahun
Harga PAC	= Rp	15,500 /kg
		(tokopedia.com)
Biaya PAC per tahun	= Rp	20,997,321

Kebutuhan resin kation	=	14,348.0637 L/tahun
Harga resin dowex	= Rp	1,281,000 /25 L (tokopedia.com)
Biaya Dowex per tahun	= Rp	735,194,784

Kebutuhan resin anion	=	12,913.2573 L/tahun
Harga resin dowex	= Rp	2,671,100 /25 L (tokopedia.com)
Biaya APS per tahun	= Rp	1,379,704,065

Kebutuhan HCl 33%	=	714.1423 liter/tahun
Harga HCl	= Rp	10,000 /liter
Biaya HCl per tahun	= Rp	7,141,423

Kebutuhan NaOH	=	571 kg/tahun
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PERANCANGAN PABRIK

Pabrik Diocetyl Phthalate Dari Phthalic Anhydride Dan 2-Ethyl Hexanol Dengan Katalis Tetrabutyl Titanate Melalui Proses Esterifikasi

Harga NaOH = Rp 19,000 /kg
Biaya NaOH per tahun = Rp 10,848,598

c. Kebutuhan Bahan Bakar (*fuel oil*)

Kebutuhan bahan bakar = 91.6386 liter/jam
= 725,777.9089 liter/tahun
Harga bahan bakar = Rp 18,700 /liter
(solarindustri.com)
Biaya bahan bakar per tahun = Rp 13,572,046,897

d. Kebutuhan Listrik

Kebutuhan listrik = 59.0760 kWh/jam
= 467,881.9200 kWh/tahun
Harga listrik = Rp 1,444 /kWh
(PLN per Oktober 2022)
Biaya listrik per tahun = Rp 675,621,492

Total biaya utilitas per tahun = Rp 16,565,343,001

VII. Harga Tanah dan Bangunan

Luas tanah = 20,465 m²
Harga tanah per m² = Rp 1,200,000
Harga tanah total = Rp 24,557,520,000

Luas bangunan pabrik = 15,431 m²
Harga bangunan pabrik per m² = Rp 2,000,000
Harga bangunan pabrik total = Rp 30,862,000,000

Luas bangunan gedung = 1,750 m²
Harga bangunan gedung per m² = Rp 1,500,000
(urbanindo.com)
Harga bangunan gedung total = Rp 2,625,000,000

Harga bangunan total = Rp 33,487,000,000
Total harga tanah dan bangunan = Rp 58,044,520,000